

Introduction to Internal Combustion Engine

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ABSTRACT: *The combustion of a fuel takes place with the help of an oxidizer often air in a combustion chamber that is a crucial component of the working fluid flow circuit in an internal combustion engine ICE or IC engine. In an internal combustion engine, a component is subjected to direct force as a result of the expansion of the high-temperature and high-pressure gases produced during combustion. Typically, the force is applied to a rotor Winkle engine, a piston engine, turbine blades gas turbine, or a nozzle jet engine.*

KEYWORDS: *Combustion Engine, Exhaust Valve, Four Stroke, Heat Engine, Internal Combustion.*

INTRODUCTION

Internal combustion engines were developed with the help of numerous scientists and engineers. John Barber created the gas turbine in 1791. Thomas Mead obtained a gas engine patent in 1794. Also in 1794, Robert Street created an engine and obtained a patent for an internal combustion device that used liquid fuel for the first time. The first internal combustion engine in America was created by John Stevens in 1798. Napoleon Bonaparte granted a patent for the Pyrophore, a prototype internal combustion engine that was developed in 1807 by Nicéphore and Claude Niece of France who later invented photography. On the French Saone River, a boat was propelled by this engine. The same year, François Isaac de Rivas, a Swiss engineer, created a hydrogen-based internal combustion engine that was ignited by an electric spark. The world's first internal combustion powered automobile was the crude operating vehicle that De Rivas attached his invention to in 1808 [1], [2].

The first industrially used internal combustion engine was patented by Samuel Brown in 1823. Eugenio Bar Santi and Felicia Matteucci, two Italian innovators, were granted a patent for Obtaining Motive Power by the Explosion of Gases in the UK in 1854. Their invention of an Improved Apparatus for Obtaining Motive Power from Gases was granted patent No. 1655 by the Great Seal Patent Office in 1857. Between 1857 and 1859, Bar Santi and Matteucci received further patents for the same idea in Piedmont, France, and Belgium. An internal combustion engine powered by petrol was created in 1860 by Belgian engineer Jean Joseph Etienne Lenoir. Nicolaus Otto obtained a patent for the first atmospheric gas engine in 1864. George Brayton, an American, created the first industrial internal combustion engine in 1872. Working with Wilhelm May Bach and Gottlieb Daimler, Nicolaus Otto

invented the compressed charge, four-cycle engine in 1876. Karl Benz obtained a patent for a dependable two-stroke gasoline engine in 1879. Later, in 1886, Benz started the first industrial production of internal combustion-powered cars using a three-wheeled, four-cycle engine that was integrated into the chassis [3], [4].

Rudolf Diesel created the initial compressed charge, compression ignition engine in 1892. Robert Goddard fired off the first liquid-fueled rocket in 1926. The Henkel He 178 became the first jet aero plane in history in 1939. The combustion of a fuel takes place with the help of an oxidizer often air in a combustion chamber that is a crucial component of the working fluid flow circuit in an internal combustion engine ICE or IC engine. In an internal combustion engine, a component is subjected to direct force as a result of the expansion of the high-temperature and high-pressure gases produced during combustion. Typically, the force is applied to a rotor Winkle engine, a piston engine, turbine blades gas turbine, or a nozzle jet engine. The component is propelled across a distance by this force, which converts chemical energy into kinetic energy that is then utilised to move or power whatever the engine is connected to. Étienne Lenoir developed the first internal combustion engine that was a commercial success around 1860, and Nicolaus Otto developed the Otto engine, the first modern internal combustion engine, in 1876. Typically, when you hear the word internal combustion engine, you're thinking of a machine with intermittent combustion, like the more well-known two- and four-stroke piston engines, as well as variations like the six-stroke piston engine and the Winkle rotary engine [5], [6].

Continuous combustion is used by a second class of internal combustion engines, including gas turbines, jet engines, and the majority of rocket engines, all of which operate on the same fundamental design as the one just mentioned. Although they are of a sort

so specialized that they are frequently considered as a different category, weapons like mortars and anti-aircraft cannons are also a type of internal combustion engine. In contrast, energy is given to a working fluid that does not contain, is mixed with, or is contaminated by combustion products in external combustion engines like steam or Sterling engines. External combustion engines can run on air, hot water, pressurized water, or even sodium-based liquid that has been heated in a boiler. Despite having several stationary uses, ICEs are mostly used in mobile applications and serve as the main power source for cars, boats, and other moving objects. Hydrocarbon-based fuels like ethanol, natural gas, petrol, or diesel fuel are frequently used to power ICEs. Compression ignition CI engines use biodiesel as a fuel, while spark ignition SI engines use bioethanol or ETBE ethyl tert-butyl ether, which is made from bioethanol. Rudolf Diesel, the creator of the diesel engine, was using peanut oil to power his machines as early as 1900. Frequently, fossil fuels and renewable energy are combined. Rarely used hydrogen can be produced using fossil fuels or clean energy sources [7], [8].

Application of IC Engine

For land and water vehicles, such as cars, motorbikes, ships, and to a lesser extent locomotive most use diesel engines, some are electrical, but most are powered by reciprocating piston engines, reciprocating piston engines are by far the most prevalent form of propulsion. Some cars, planes, and motorbikes employ rotary engines of the Winkle design. Internal-combustion-engine vehicles ICEV refer to all of these. Internal combustion engines often take the form of combustion turbines or Winkle engines where high power-to-weight ratios are required. A conventional ICE, which could be a reciprocating engine, is used by powered aircraft. Instead of using jet engines, which are forms of turbines, aero planes and helicopters can use turbo shafts [9], [10].

Airliners may use an additional ICE as an auxiliary power source in addition to propulsion. Many unmanned aerial vehicles have Winkle engines installed. Large electric generators that power electricity grids are propelled by ICEs. They are typically found in the form of combustion turbines, which have an average electrical output of about 100 MW. In combined cycle power plants, water steam is boiled and superheated using the high temperature exhaust to drive a steam turbine. Because more energy is recovered from the fuel than could be by the combustion engine alone, the efficiency is higher

as a result. Energy efficiency for combined cycle power plants is between 50 and 60%.

DISCUSSION

Conversion of Energy

The prevalent use of mechanical power is the defining characteristic of our civilization today, setting it apart from all others. Once upon a time, the main Man's muscles were the main source of power for either peace or war. Animals were later trained to assist, and even the wind and a flowing stream were harnessed. However, the greatest advancement in this regard came when man discovered how to convert energy from one form to another. An engine is the technical term for the device that converts energy.

The Meaning of Engine

An engine is a tool that converts one type of energy into another. The effectiveness of the conversion, however, is a crucial factor when changing energy from one form to another. The term heat engines refer to the majority of engines that typically transform thermal energy into mechanical work.

Heat Engine

A heat engine is a machine that converts chemical energy from fuel into thermal energy and then uses this thermal energy to carry out meaningful work. As a result, a heat engine transforms thermal energy into mechanical energy. Two general categories can be used to classify heat engines: External and internal combustion engines EC and IC engines, respectively. A heat engine is a device used in thermodynamics and engineering to transform heat into useable energy, notably mechanical energy, which can subsequently be used to do mechanical labor. Although the heat engine concept was first developed in relation to mechanical energy, it has been used to refer to several other forms of energy, particularly electrical, from at least the late 19th century. The heat engine accomplishes this by lowering the temperature of a working substance from one condition to another. Thermal energy produced by a heat source raises the temperature of the working substance. The working substance transfers heat to the cooler sink until it achieves a lower temperature state while producing work in the engine's working body.

By taking use of the qualities of the working substance, some of the thermal energy is transformed into work throughout this process. The working material can be any system with a heat capacity greater than zero, however it is often a gas or liquid. Some heat is typically lost to the

environment during this process and is not put to use. A certain amount of energy is also lost due to drag and friction. Any device that transforms energy into mechanical work is referred to as an engine. Because Carnot's theorem essentially limits heat engines' efficiency, they stand apart from other forms of engines. Heat engines have the advantage that most forms of energy can be easily converted into heat by processes like exothermic reactions such as combustion, nuclear fission, absorption of light or energetic particles, friction, dissipation, and resistance, even though this efficiency limitation can be a disadvantage. Heat engines have a wide range of uses since the heat source that provides thermal energy to the engine can thus be fueled by almost any type of energy. With the cycles they try to implement, heat engines are frequently misunderstood. The words engine and cycle are typically used to refer to mechanical devices and models, respectively.

Heat Engine Classification and Some Fundamental Information

Both internal and external combustion engines fall into one of two categories. Reciprocating engines, rotary engines, and more In Figure 1, a thorough classification of heat engines is shown. The internal combustion engine with reciprocating motion, the gas turbine, and the steam turbine are the three most popular forms of heat engines. Today, the steam engine is being gradually phased out. Because the working fluid does not pass through heat exchangers boilers and condensers in a steam turbine plant, the reciprocating internal combustion engine has several benefits over a steam turbine. As a result, the internal combustion engine's power plant is far more efficient and has a simpler mechanical design. The reciprocating internal combustion engine has an additional benefit over the other two types in that all of its parts operate at an average temperature that is far lower than the highest temperature of the working fluid in the cycle. This is due to the fact that only a very small portion of the cycle's time is spent with the working fluid's high temperature. As a result, extremely high working fluid temperatures can be used, increasing thermal efficiency.

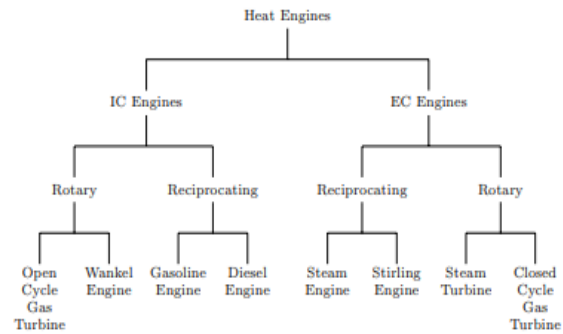


Figure 1: Representing the Classification of Heat Engine [Ftp.Idu.Ac.Id].

Additionally, internal combustion engines have a lower weight to power ratio than steam turbine power plants because they may achieve higher thermal efficiency with a moderate maximum operating pressure of the fluid in the cycle. Additionally, reciprocating internal combustion engines with negligible power output—even a few hundred watts—and reasonable thermal efficiency have been developed. The main drawback of this kind of engine is the vibration issue brought on by the reciprocating parts. Additionally, different fuels cannot be used in these engines. Only fuels that meet certain specifications, whether liquid or gaseous, can be used efficiently. These fuels cost more than average. Given the aforementioned considerations, it has been determined that reciprocating internal combustion engines are suited for use in cars, motorbikes, scooters, power boats, ships, slow-moving aircraft, locomotives, and power units with relatively modest output.

Internal and External Combustion Engines

Engines classified as external combustion engines are those in which combustion occurs outside the engine, as opposed to internal combustion engines, where combustion occurs inside the engine. For instance, the heat produced by the burning of fuel is used in a steam engine or steam turbine to produce high pressure steam that serves as the working fluid in a reciprocating engine or a turbine. The working fluid in a petrol or diesel engine is made up of the combustion products that are produced when fuel and air are burned inside the cylinder.

Principal Engineering Elements and Nomenclature

Internal combustion engines with reciprocating pistons appear to be relatively basic, but they are actually very sophisticated devices. To produce output power, hundreds of components must each successfully carry out their respective tasks. There are two different kinds of engines: compression-

ignition CI and spark-ignition SI. Now let's go through some of the critical engine parts and the terminology used to describe engines.

Engine Subsystems

Figure 2 depicts the cross section of a single-cylinder spark-ignition engine with overhead valves. The main engine parts and their roles are succinctly summarized below. The major supporting structure for the numerous components is the cylinder block. A multi-cylinder engine's cylinders are cast as a single unit known as the cylinder block. The cylinder block is attached to the cylinder head. When using water cooling or when using air cooling, the cylinder head and cylinder block are equipped with cooling fins or water jackets. Between the cylinder block and cylinder head is a cylinder head gasket. Numerous nuts or studs tightly secure the cylinder head to the cylinder block. The crankcase refers to the base of the cylinder block.

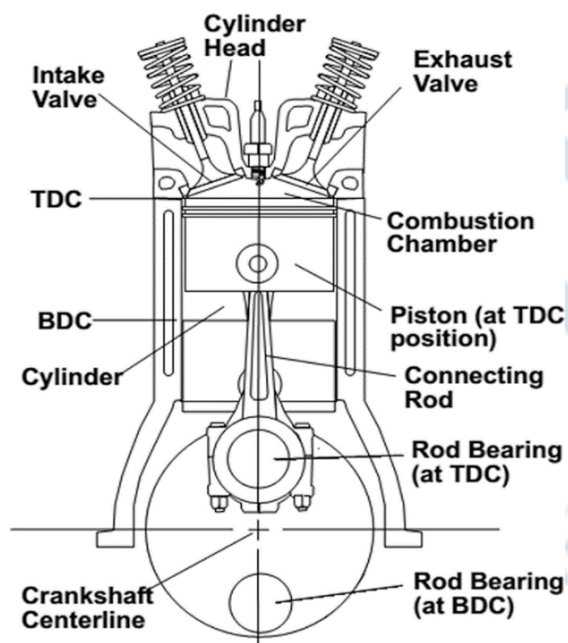


Figure 2: Representing the Cross-section of a spark-ignition engine [Springer Link].

The bottom of the crankcase has a cover called the crankcase that serves as a sump for lubricating oil. Bore or face refers to the inner surface of the cylinder block that has been precisely machined and finished to have a cylindrical shape. As the name suggests, a cylinder is a cylindrical container or space in which a piston reciprocates. The working fluid is filled into the variable volume formed in the cylinder during engine operation and is subjected to various thermodynamic processes. In the cylinder

block, the cylinder is supported. A piston is a cylindrical part that is inserted into the cylinder that serves as the combustion system's movable boundary. It provides a gas-tight space with the piston rings and the lubricant because it fits properly and snugly into the cylinder. It is the first link in the chain that connects the output shaft to the gas forces.

Combustion Chamber: The combustion chamber is the area that the cylinder head and piston top enclose in the upper half of the cylinder during the combustion process. Pressure builds up in this area of the cylinder as a result of fuel combustion and the subsequent release of thermal energy. The inlet manifold is the conduit that connects the intake system to the engine's inlet valve and is used to draw air or an air-fuel combination into the cylinder.

Exhaust Manifold: The exhaust manifold is the conduit that joins the exhaust system to the engine's exhaust valve and allows the combustion products to escape into the atmosphere.

Exhaust and inlet valves: Most valves are of the mushroom-shaped poppet variety. For controlling the charge entering the cylinder inlet valve and for releasing the combustion products from the cylinder exhaust valve, they are either provided on the cylinder head or on its side. The spark plug, which is often found on the cylinder head, is a part used in Spark Ignition SI engines to start the combustion process.

Connecting Rod: This component links the piston to the crankshaft and transfers the piston's gas forces to the latter. The little end and the big end of the connecting rod are its two ends. The big end is attached to the crankshaft by a crankpin, while the little end is connected to the piston by a gudgeon pin. The crankshaft transforms the piston's reciprocating action into useful rotary motion for the output shaft. A single cylinder engine's crankshaft contains two crank arms, as well as balance weights. The rotating system can be statically and dynamically balanced using the balance weights. A crankcase houses the crankshaft.

Piston Rings: Fitted into the slots all around the piston, piston rings create a tight seal that stops combustion gases from escaping the cylinder.

Gudgeon Pin: This device connects the piston to the tiny end of the connecting rod.

Camshaft: The two valves' opening and shutting are controlled by the camshaft, which isn't depicted in the diagram. Push rods, rocker arms, valve springs, and tappets are the related components. The ignition system is also driven by this shaft. Through timing gears, the crankshaft powers the camshaft. Integrated inside the camshaft, cams are constructed to open the valves at the right time and hold them

open for the required amount of time not indicated in the illustration.

Fly Wheel: During an engine's full cycle of operation, the net torque applied to the crankshaft changes, changing the shaft's angular velocity. The flywheel not depicted in the illustration is a wheel-shaped inertia mass that is mounted to the output shaft in order to produce a uniform torque.

Engine Principles of Operation: Four-Stroke Cycle Diesel Engine:

In four-stroke cycle engines, four strokes are used to complete two crankshaft revolutions. These are the suction, compression, power, and exhaust strokes, respectively. The piston lowering during its suction stroke. During this stroke, the exhaust valve is closed while the input valve draws only pure air into the cylinder. The rocker arm, push rod, and cam can all be used to open and close these valves. The piston rises upward during the following stroke, known as the compression stroke, during which both valves are still closed. As the piston rises, the air that has been sucked into the cylinder during the suction stroke is gradually compressed.

Typically, the compression ratio ranges from 14:1 to 22:1. 30 to 45 kg/cm² of pressure is present at the end of the compression stroke. Near the end of the compression stroke, the temperature of the compressed air in the cylinder reaches a high enough level 650–800 °C to instantly ignite any fuel that is put into the cylinder. A liquid hydrocarbon fuel, such as diesel oil, is sprayed into the combustion chamber at high pressure 140–160 kg/cm², which is higher than that present in the cylinder itself, while the piston is near the peak of its compression stroke. When the oxygen in the highly compressed air is added, this fuel then ignites and burns.

The piston completes its compression stroke during the fuel injection phase and starts to return on its third consecutive stroke, or power stroke. The heated combustion byproducts, namely carbon dioxide and the nitrogen leftover from the compressed air, expand during this stroke, pushing the piston downward. Only the cylinder's operating stroke is being used here. Near the end of the power stroke, the pressure drops from its maximum combustion value 47–55 kg/cm², which is often higher than the bigger value of the compression pressure 45 kg/cm². When the piston reaches its lowest point of travel, the exhaust valve normally opens a little earlier. On the next upward piston stroke, the exhaust gases are expelled. The exhaust valve opens at the beginning of the stroke and closes at the end. A connecting rod and crankshaft work together to transform the piston's reciprocating motion into the crankshaft's

rotating motion. The main bearings, which are installed in the crankcase, allow the crankshaft to rotate. To equal out the uneven torque produced by the reciprocating engine, the flywheel is mounted on the crankshaft.

Diesel Two-Stroke Cycle Engine

In the case of a two-stroke engine, the suction, compression, power, and exhaust strokes of the piston's cycle are only accomplished in two strokes. Due to the suction produced by the piston's upward stroke, air is pulled into the crankcase. It is compressed in the crankcase during the piston's down stroke; however, the compression pressure is typically relatively low, only high enough to allow air to enter the cylinder through the transfer port when the piston is almost at the bottom of its down stroke. Thus, the air enters the cylinder and is compressed as the piston rises, almost reaching the peak of its stroke. The compression pressure is raised to a level where the air temperature is above the fuel's self-ignition point. Just before the end of the compression stroke, a little amount of fuel is briefly delivered into the cylinder head. During the following piston downward stroke, the burned gases expand. Through the piston opening the exhaust port, these gases are allowed to flow into the exhaust pipe and reach the atmosphere.

Contemporary Two-Stroke Diesel Engine

Because the exhaust gases do not exit the cylinder during port opening, the crankcase method of air compression is inadequate. During the cylinder charging operation, air is also lost through the exhaust ports. Blowers are used to recompress the air in order to overcome these drawbacks. Through the port, this pre-compressed air enters the cylinder. Additionally, there is a mechanical exhaust valve that opens exactly before the entrance ports do.

Spark-Ignition Four-Stroke Engine

In a four-stroke engine, the cycle of operations is finished in four piston strokes or two crankshaft revolutions. There are five tasks that need to be finished throughout the course of the four strokes: suction, compression, combustion, expansion, and exhaust. The crankshaft rotates 180 degrees for each stroke, making a four-stroke cycle take 720 degrees to complete. The four strokes that make up the ideal four-stroke SI engine's cycle of operation are the suction or intake stroke, the compression stroke, the expansion or power stroke, and the exhaust stroke.

- 1. Suction or Intake Stroke:** The piston begins the suction stroke when it is at top dead center and about to descend downhill. The exhaust valve is currently in the closed position and the

inlet valve is assumed to open instantly. The charge of fuel-air mixture is drawn into the cylinder as a result of the suction produced by the piston's motion towards bottom dead center. The suction stroke of the piston terminates when it reaches bottom dead center, and the inlet valve instantly closes.

2. **Compression Stroke:** The return stroke of the piston 12 compresses the charge drawn into the cylinder during the suction stroke. Both the inlet and exhaust valves are closed during this stroke. The entire cylinder's volume of mixture has now been crushed into the clearance volume. A spark plug on the cylinder head ignites the mixture at the conclusion of the compression stroke. Burning can be roughly described as heat addition at constant volume because it is thought that in perfect engines burning occurs instantly when the piston is at top dead center. The chemical energy of the fuel is transformed into heat energy throughout the burning process, resulting in a temperature increase of around 2000 °C process 2–3. The heat released from the fuel causes a significant increase in pressure at the end of combustion.
3. **Expansion or Power Stroke:** The piston is pushed towards the BDC by the high pressure of the burned gases stroke. Both valves are in the closed position. Only during this stroke out of the four doe's power are produced. During expansion, the temperature and pressure both falls.
4. **Exhaust Stroke:** As soon as the expansion stroke is complete, the exhaust valve opens while the inlet valve stays closed. Part of the burning gases escape when the pressure drops to atmospheric pressure. The burnt gases are swept out of the cylinder virtually at atmospheric pressure by the piston, which begins to move from the bottom dead center to the top dead center stroke. When the piston reaches T DC at the conclusion of the exhaust stroke, the exhaust valve closes, leaving some leftover gases trapped in the clearance volume in the cylinder. The subsequent cycle's new charge and these remaining gases combine to create the working fluid. In a four-stroke engine, each cylinder performs the aforementioned four tasks in two crankshaft revolutions: the first during the suction and compression strokes, and the second during the power and exhaust strokes. As a result, there is only one power stroke for a full cycle, whereas the crankshaft completes two revolutions. The heat addition process 23 should be as high and

the heat rejection process 34 as low as feasible in order to increase the engine's output. As a result, one should take care while creating the ideal p-V diagram, which should accurately portray the processes.

CONCLUSION

For more than a century, transportation and power production have benefited greatly from the use of internal combustion engines IC engines. But it is becoming more and more obvious as the twenty-first century goes on that the IC engine is reaching its limits and experiencing many difficulties. IC engines' effects on the environment are one of the main issues. Carbon dioxide CO₂ and other greenhouse gases are released when fossil fuels are used in internal combustion engines ICs, which contributes to the global warming trend. Additionally, IC engines release pollutants that are harmful to air quality and human health, like nitrogen oxides NO_x and particulates.

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Application and Working Operation of Two Stroke Engine

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ABSTRACT: *An internal combustion engine known as a two-stroke or two-cycle is frequently used in smaller, lower power vehicles including scooters, dirt motorcycles, jet skis, smaller outboard motors, and lawn and garden equipment like lawnmowers and chainsaws. The ubiquitous four-stroke engines seen in automobiles share many but not all of its components with two-stroke engines, but there are also important changes that allow for differing performance and necessitate various kinds of lubrication.*

KEYWORDS: *Air Fuel, Cross Flow, Exhaust Port, High Power, Stroke Engines*

INTRODUCTION

Donald Clerk, a Scottish engineer, is credited with developing the first commercial two-stroke engine with cylinder compression; his design was patented in 1881 [1], [2]. His, however, had a separate charge cylinder, unlike the majority of succeeding two-stroke engines. The Englishman Joseph Day is generally recognized as the inventor of the crankcase-scavenged engine, which uses the space below the piston as a charging pump. German inventor Karl Benz created a two-stroke petrol engine on December 31, 1879, for which he was granted a patent in Germany in 1880. Alfred Angus Scott, a Yorkshire man who began making twin-cylinder water-cooled motorcycles in 1908, is credited with developing the first fully practical two-stroke engine. Chainsaws and motorbikes are two-stroke gasoline engines with electrical spark ignition are particularly helpful in light or portable applications. The cycle is appropriate for diesel compression ignition engines working in large, weight-insensitive applications, such as marine propulsion, railway locomotives, and electricity production, because it has the potential for a high thermodynamic efficiency. When compared to a four-stroke engine, a two-stroke engine's exhaust gases carry less heat to the cooling system, which frees up more power for the piston and, if one is present, a turbocharger [3], [4].

A two-stroke or two-stroke cycle engine is a type of internal combustion engine that performs two piston strokes up and down movements throughout a power cycle, which is finished in one crankshaft revolution. In a four-stroke engine, a power cycle lasts for two crankshaft revolutions and involves four piston strokes. In a two-stroke engine, the intake and exhaust or scavenging activities happen

simultaneously at the end of the combustion stroke and the start of the compression stroke. Due to the power being accessible in a constrained range of rotational speeds known as the power band, two-stroke engines frequently have a high power-to-weight ratio. Because they contain fewer moving parts than four-stroke engines, two-stroke engines are less expensive to produce. Two-stroke engines have been phased out of use in automobiles and motorcycles in nations and regions with strict emissions regulations. Small displacement two-stroke engines are still widely used in mopeds and motorbikes in places with laxer rules, or no regulations at all [5], [6].

Applications

An outboard engine from the British Seagull Forty series, shown from the side, has a serial number that places it in the 1954–1955 time period. When mechanical simplicity, light weight, and a high power-to-weight ratio are design considerations, two-stroke petrol engines are favored. They may run in any orientation because the oil reservoir is not gravity-dependent by combining oil and gasoline. Two-stroke engines have formerly been utilised by a number of well-known automakers, including the German DKW, Auto-Union, VEB Sachsen ring Automobile were Zwickau, VEB Automobile Eisenach, and VEB Farceur- und Jagdwaffen werk „Ernst Thiemann. In the 1970s, Japanese automakers Suzuki and Subaru followed suit. Due to more rigorous air pollution regulations, two-stroke car production was put to a halt in the West in the 1980s [7], [8].

The Trabant and Wartburg in East Germany remained in use until about 1991 in the Eastern Bloc nations. Outboard motors, small on- and off-road motorcycles, mopeds, scooters, tut-tuts, snowmobiles, go-karts, ultralight and model aero

planes are just a few minor propulsion devices that still use two-stroke engines. Pollution control laws, particularly in wealthy nations, have made it necessary to phase out their use for many of these purposes. For instance, Honda abandoned road-going versions far earlier and stopped marketing two-stroke off-road motorcycles in the United States in 2007. Two-stroke engines, which have a high power-to-weight ratio and can be used in any direction, are frequently seen in handheld outdoor power equipment like leaf blowers, chainsaws, and string trimmers. Several trucks and pieces of heavy equipment, as well as big industrial and maritime applications, use two-stroke diesel engines [9], [10].

Inlet Port with a Piston

The most typical design for compact two-stroke engines is the piston port, which is also the simplest. The piston alone controls every action by moving up and down in the cylinder and covering and revealing the ports. Yamaha formulated the fundamental ideas behind this system in the 1970s. They discovered that, in general, enlarging an exhaust port produces an increase in power that is equivalent to raising the port, but the power band does not narrow as it does when the port is elevated. A single exhaust port's width, however, is mechanically limited to around 62% of the bore diameter for a good piston ring life. Beyond this point, the piston rings protrude into the exhaust port and quickly deteriorate. In racing engines where rings are changed every few races, a maximum 70% of bore width is permissible. There is a 120° to 160° period of intake. Minimum transfer port temperature is set at 26 degrees. When the piston is at bottom dead center and the transfer ports are practically wide open, the intense, low-pressure pulse of a racing two-stroke expansion chamber can reduce the pressure to -7 psi. One of the factors contributing to two-stroke engines' high fuel consumption is the forced passage of a portion of the entering pressurized fuel-air combination across the top of the piston, where it cools the piston, and straight out the exhaust pipe. This outgoing flow is stopped by a chamber expansion with a powerful reverse pulse. The two-stroke crankcase is sealed and participates in the induction process in gasoline and hot bulb engines, which distinguishes it fundamentally from normal four-stroke engines. For scavenging, diesel two-stroke engines frequently use a Roots blower or a piston pump.

DISCUSSION

Turning Inlet Valve

A revolving member opens and closes the intake pathway. One common design, known as a disc

valve, is a slotted disc attached to the crankshaft that covers and reveals an opening in the end of the crankcase, allowing charge to enter during one phase of the cycle. This design is occasionally seen on compact motorcycles. Two cylindrical parts with proper cutouts are positioned to spin one inside the other in a different type of rotary inlet valve used on two-stroke engines. This valve only allows entry to the crankcase when the two cutouts are parallel. As in the majority of glow-plug model engines, the crankshaft itself could make up one of the members. In another variation, similar to Vespa motor scooters, the crank disc is designed to fit snugly inside the crankcase and is equipped with a cutout that, when necessary, aligns with an inlet passage in the crankcase wall.

The benefit of a rotary valve is that it makes it possible for two-stroke engines to have asymmetrical intake timing, which is not achievable with piston-port engines. The intake timing of a piston-port type engine opens and closes prior to top dead center and after at the same crank angle, while the rotary valve permits the opening to start and close earlier. Unlike piston-port or reed-valve engines, rotary valve engines can be configured to deliver power over a wider speed range or more power over a smaller speed range. No wear should occur if a piece of the rotary valve is a part of the crankcase itself, which is particularly important. Scavenging in a cross-flow

Vacuum Scavenging Cross-Flow Deflector Piston

A deflector on top of the piston in a cross-flow engine directs the fresh intake charge into the upper section of the cylinder while forcing the residual exhaust gas down the other side of the deflector and out the exhaust port. The transfer and exhaust ports are on opposite sides of the cylinder in a cross-flow engine. After the 1960s, inflow scavenging largely replaced this design, particularly for motorcycles, because it reduced the weight and exposed surface area of the piston and made it easier to achieve an effective combustion chamber shape. However, the deflector piston can still be a viable option for smaller or slower engines using direct injection. Scavenging loops. Two-stroke engines

1. Acquiring or scavenging.
2. Exhaust.
3. Compression.
4. Power or expansion.

Primary Schnuerle Machining

With this scavenging technique, the flow of fresh mixture as it enters the cylinder is directed into the combustion chamber using properly designed and placed transfer ports. After impacting the cylinder

head, the fuel/air combination travels down the combustion chamber's curve before being redirected downward. This not only stops the fuel/air combination from leaving the exhaust port directly, but it also generates swirling turbulence that increases the effectiveness, power, and economy of combustion. This method has a clear benefit over the cross-flow strategy because typically a piston deflector is not needed. Adolf Schnürle, a German creator of an early type in the mid-1920s, inspired the term Schnuerle loop scavenging, which was commonly used in that nation throughout the 1930s and extended internationally after World War II. The most typical method of transferring the fuel/air mixture utilised in contemporary two-stroke engines is loop scavenging. One of the first producers outside of Europe to use two-stroke, loop-scavenged engines was Suzuki. This operational characteristic was applied in conjunction with the expansion chamber exhaust created by Walter Kaden and MZ, a German motorcycle manufacturer.

When combined, loop scavenging, disc valves, and expansion chambers dramatically increased the power output of two-stroke engines, especially those produced by the Japanese firms Suzuki, Yamaha, and Kawasaki. In the 1960s, loop scavenging's improved power was a major factor in Suzuki and Yamaha's success in Grand Prix motorcycle racing. The ability to make the piston almost flat or slightly domed made it noticeably lighter and stronger, allowing it to withstand higher engine speeds. This was another advantage of loop scavenging. The "flat top" piston also has better thermal characteristics and is less prone to compression losses, dimensional changes, piston seizures, uneven heating, and expansion.

Based on a DKW design, SAAB produced 750- and 850-cc three-cylinder engines that used loop charging and were reasonably successful. The original SAAB 92 had a two-cylinder, relatively inefficient engine. At cruising speed, the frequency of reflected-wave, exhaust-port blocking was too low. Fuel economy was enhanced by using the asymmetrical three-port exhaust manifold found in the same DKW engine. Depending on the model year, the 750-cc basic engine put out 36 to 42 horsepower. The 750-cc version for the Monte Carlo Rally produced 65 horsepower and had a filled crankshaft for better base compression. In 1966, the SAAB Sport (a regular trim vehicle as opposed to the Monte Carlo's premium trim) offered an 850-cc version. A two-stroke engine's overall compression ratio includes base compression in part. Research presented at SAE in 2012 shows that loop

scavenging is always more effective than cross-flow scavenging.

The Two-Stroke Inflow Engine

- i. TDC (top dead center).
- ii. BDC (bottom dead center).
- iii. Intake effective scavenging, 135° – 225° ; symmetric about BDC by necessity; diesel injection typically begins at 4° before TDC.
- iv. Exhaust, Compression.
- v. Power.

The mixture or charge air in the case of a diesel engine enters the cylinder under the control of a piston at one end, and the exhaust leaves the cylinder under the control of a piston, exhaust valve, or both at the other end. The name inflow comes from the fact that the scavenging gas flow only flows in one direction. A few small marine two-stroke engines Grey Marine, a few railroad two-stroke diesel locomotives Electro-Motive Diesel, huge marine two-stroke main propulsion engines, and stationary two-stroke engines all use the valve configuration. The Junkers Jumbo 205 and Napier Deltaic are examples of opposed piston designs, which have two pistons in each cylinder that move in the opposite directions. This category includes the once-common split-single design, which is essentially a folded inflow. Inflow engines can be supercharged using a crankshaft-driven piston or Roots blower with advanced-angle exhaust timing.

Scaled-Down Piston Engine

This engine's piston is top-hat shaped, with the upper section serving as the standard cylinder and the lower section serving as a scavenger. The piston's lower half charges the adjacent combustion chamber as the units operate in pairs. The other engine elements are sump lubricated with benefits for cleanliness and dependability, but the upper area of the piston still relies on total-loss lubrication. Because skirt thicknesses can be lower, the piston's mass is only around 20% greater than that of a loop-scavenged engine's piston.

Power-Valve Mechanisms

Two-stroke power valve system, the main argument a power-valve system is used in a lot of contemporary two-stroke engines. Typically, the valves are near or in the exhaust ports. As with the Rota R.A.V.E., Yamaha YPVS, Honda RC-Valve, Kawasaki K.I.P.S., Caria C.T.S., or Suzuki AETC systems, they either change the exhaust port by blocking off the top part of the port, which changes port timing, or they change the exhaust volume, which alters the resonant frequency of the expansion chamber. An engine with improved low-speed

power without sacrificing high-speed power is the end result. Power valves, however, require routine maintenance to function properly since they are exposed to hot gas flow.

Direct Implantation

In two-stroke engines, petrol direct injection is the main topic. In two-stroke engines, direct injection offers a number of benefits. A significant issue with carbureted two-stroke engines is that some of the fuel air mixture escapes through the exhaust port unburned. Direct injection successfully solves this issue. There are two systems in use high-pressure injection and low-pressure air-assisted injection. Since the crankcase is not traversed by the fuel, a different source of lubrication is required.

Lubrication

Many two-stroke engines pressurize the air-fuel combination in the crankcase before transferring it to the cylinder. They cannot be lubricated by oil contained in the crankcase and sump, unlike four-stroke engines, because the lubricating oil would be swept up and burned together with the fuel. Oil is added to the fuel that is delivered to two-stroke engines so that it can cover the cylinder and bearing surfaces as it travels through them. Petrol to oil volume ratios range from 25:1 to 50:1. The residual oil in the combination burns together with the fuel, producing the recognizable blue smoke and odor. Two-stroke oils, which were first made available in the 1970s, are created specifically to mix with fuel and burn with the least amount of unburned oil or ash. As a result, spark plug fouling, which had previously been an issue with two-stroke engines, was significantly reduced.

Another tank of two-stroke oil may be used to lubricate more two-stroke motors. The throttle position and engine speed regulate the flow of this oil. Examples can be found in numerous two-stroke snowmobiles and the Yamaha PW80. The technique is known as auto-lube. The oil is burned in the same manner as in the premix system therefore, this is still a total-loss system. It offers slightly more effective lubrication due to the fact that the oil is improperly mixed with the fuel when burned in the combustion chamber. This lubrication method ensures proper engine lubrication, with less oil at light loads like idle and more oil at high loads like full throttle, and eliminates the user's need to mix the gasoline at each refill. It also makes the motor much less susceptible to atmospheric conditions like ambient temperature and elevation.

As the loading on the engine parts was sufficiently low, certain firms, like Bombardier, had some oil-pump designs with no oil pumped at idle to reduce

smoke levels. This is because the low levels of lubrication provided by the gasoline are sufficient for the engine parts' needs. Oil injection ultimately functions the same way as premixed petrol in that the oil is burned in the combustion chamber albeit not as completely as premix and the petrol and oil are still mixed, albeit not as well. To pump the oil from the separate tank to the carburetor or throttle body using this approach, additional mechanical components are needed. Premix lubrication is nearly often employed in situations where performance, simplicity, and/or dry weight are crucial factors. For instance, a two-stroke engine on a motocross bike prioritizes weight, simplicity, and performance. To lessen operator fatigue and risk, chainsaws and brush cutters must be as lightweight as feasible.

Compression in the crankcase if the throttle is closed while the engine is rotating quickly, two-stroke engines get oil starved. Examples are motorcycles descending steep hills and possibly slowing down gradually after reaching high speeds. The powertrain of two-stroke cars, which were common in Eastern Europe in the middle of the 20th century, was typically equipped with freewheel systems that allowed the engine to idle when the throttle was closed and necessitated the use of brakes to slow down. Large two-stroke engines, especially diesels, typically employ a four-stroke engine-like sump lubrication system. The ancillary Roots-type blower or a specialized turbocharger typically a turbo-compressor system that has a locked compressor for starting and during which it is powered by the engine's crankshaft but is unlocked for running and during which it is powered by the engine's exhaust gases flowing through the turbine is used to pressurize the cylinder instead of the crankcase.

Reversibility with Two Strokes

For the sake of this discussion, it is helpful to visualize a motorcycle, where the exhaust pipe is directed into the cooling air stream and the crankshaft typically rotates in the same plane and direction as the wheels, or forward. Since practically all four-stroke engines rotate forward, several of the issues raised above also apply to these engines which cannot reverse their direction of rotation without significant modification. The front and back faces of the piston are, respectively, its intake port and exhaust port sides they have nothing to do with the top or bottom of the piston. This is important to keep in mind. Regular gasoline two-stroke engines have been utilised to replace the lack of a reverse gear in microcaps like the Messerschmitt KR200 because they can run backward for brief periods of time and under little load without much difficulty.

The motor is stopped and restarted backward in vehicles with electric starting by turning the key in the opposite direction. Similar systems have been employed by two-stroke golf carts.

Traditional flywheel magnetos, which broke contact before top dead center using contact-breaker points rather than an external coil, performed equally well in forward and reverse due to the symmetry of the cam controlling the points. Although rotary valve engines have uneven inlet timing and perform poorly, reed-valve engines operate backward just as well as piston-controlled ones. Running many engines backward under load for any length of time has serious drawbacks, some of which are universal and apply to both two-stroke and four-stroke engines. In most situations where cost, weight, and size are important factors, this disadvantage is acknowledged. The issue arises because the back face of the cylinder, which is particularly the coolest and best-lubricated area in a two-stroke engine, is where the principal thrust face of the piston is located in forward running. Since it covers and reveals the exhaust port in the cylinder, the hottest area of the engine and where piston lubrication is at its least effective, the forward face of the piston in a trunk engine is less well-suited to be the principal thrust face.

The largest exhaust vent in the engine is located in the front wall of the cylinder, making the front face of the piston more prone to damage. Since piston skirts and rings run the risk of being forced into this port in a cross-flow engine, it is always ideal to have them pressing hardest on the opposing wall where there are just the transfer ports. However, while travelling backward, this weaker forward face experiences additional mechanical stress that it was not intended to withstand. In certain engines, the tiny end is offset to minimize thrust in the intended rotating direction, and the forward face of the piston has been made thinner and lighter to compensate. Crossheads and thrust bearings can be used to isolate the engine from end loads in order to prevent this. Reversible large two-stroke ship diesels are occasionally produced. They utilize mechanically driven valves, just as four-stroke ship engines some of which are also reversible, necessitating additional camshaft systems. Crossheads are used in these engines to isolate the under-piston space from the crankcase and reduce side thrust on the piston.

Two-Stroke Engine Operation

The two-stroke engine operates on the same thermodynamic cycle as any petrol or diesel engine. If you're unsure which of these cycles applies to you, you can read the preceding post on the Diesel cycle

and the Otto cycle. The two-stroke engine employs two strokes in total.

During the Compression Stroke: the piston's upward motion opens the cylinder's inlet port, allowing air or an air-fuel combination to enter. The mixes are compressed by the piston's continued action. A spark plug starts the power stroke by igniting the compressed air-fuel mixture.

Exhaust Stroke and Power Stroke: As the burned gas expands, the piston is put under pressure. This causes the piston to start moving downward, which is followed by the exhaust port opening. This clears the cylinder of its exhaust fumes. On a more specific level, the processes of removing exhaust gases and consuming the air-fuel mixture partially occur simultaneously. Additionally, the combination of compressed air and fuel helps remove exhaust pollutants. Crankcase scavenged two-strokes is the name given to these two-stroke engines as a result.

Two-Stroke Engine Application

These engines are frequently chosen because of their compact size and simple maintenance requirements. These engines' primary uses and benefits are:

High Power-to-Weight Ratio: In applications where weight is critical, these engines are favored due to their mechanical simplicity and high power-to-weight ratio. Work with various perspectives:

1. These engines can run in different directions thanks to the conventional fuel-based lubrication system.
2. Two-stroke engine restrictions
3. Although it is little and simple, this simplicity comes at a high price.

These two-stroke engines have the following significant drawbacks:

- i. **Low Thermal Efficiency:** By rejecting the unburned air-fuel mixture, the efficiency is decreased.
- ii. **Incomplete Combustion:** The presence of exhaust gases in the new mixture and incomplete combustion are both factors. These engines frequently produce severe vibration at high speeds.

CONCLUSION

In this chapter discussed about the two-stroke engine and its working and. There are numerous heat engines, particularly internal combustion engines. One of those engines is the two-stroke motor, which has gradually lost popularity and uses. Due to their inferior efficiency compared to their four-stroke equivalent, these engines acquired a bad reputation at a time when the globe was thriving to become sustainable. Despite all the criticism, it is still the

preferred option in places where size and weight are significant restrictions. Power is often accessible in a constrained range of rotational speeds known as the power band, which contributes to the high power-to-weight ratio of two-stroke engines.

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Application, Advantage and Disadvantage of IC Engine

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ABSTRACT: A 4-stroke engine is an IC engine that runs through a working cycle using four piston strokes. Due to the piston's upward and downward movement, it transforms the thermal energy of the fuel into productive mechanical work. As a result, it falls under the reciprocating engine category. A power cycle in a four-stroke engine is finished after two crankshaft revolutions or four piston strokes. In addition to being employed in generators and other machinery, 4-stroke engines are most frequently used in motorcycles, cars, and other vehicles. These engines are well-known for being extremely dependable and effective. They are comparatively simple to maintain and repair. Compared to 2-stroke engines, these engines are heavier.

KEYWORDS: Air Fuel, Combustion Engine, Compression Stroke, Four Stroke, Internal Combustion

INTRODUCTION

For a supermarket company, Nicolaus August Otto travelled as a sales representative. He came seen the internal combustion engine created in Paris by Belgian immigrant Jean Joseph Etienne Lenoir during his travels. Lenoir succeeded in developing a double-acting engine in 1860 that utilised lighting gas with a 4% efficiency. Only 2 horsepower were produced by the 18-litre Lenoir engine. The Philip Lebon-developed coal-based lighting gas that powered the Lenoir engine was invented in Paris. Otto discovered the effects of compression on the fuel charge while testing a Lenoir engine copy in 1861. Otto made an attempt to create an engine in 1862 to enhance the Lenoir engine's inadequate reliability and efficiency. He attempted to build an engine that would compress the fuel mixture before ignition, but he was unsuccessful since the engine could only be operated for a short period of time before exploding. Many additional engineers attempted to address the issue but were unsuccessful [1], [2].

Otto and Eugen Lange established NA Otto and Cie (NA Otto and Company), the first manufacturer of internal combustion engines, in 1864. The same year, Otto and Cie were successful in developing an atmospheric engine. In 1869, due to a lack of space, the business was relocated to the German town of Deutz, where it was given the new name Deutz Gasmotoren fabrik AG (The Deutz Gas Engine Manufacturing Company). In 1872, Wilhelm Maybach oversaw engine design while Gottlieb Daimler served as the technical director. Gunsmith Daimler had experience with the Lenoir engine. Otto and Lange were successful in developing the first

internal combustion engine by 1876, which had a much higher efficiency than any engine made up to that point because to compression of the fuel mixture before combustion [3], [4].

In 1883, Daimler and May Bach departed Otto and Ice and created the first high-speed Otto engine. They created the initial automobile with an Otto engine in 1885. The Daimler Reitwagen was the first internal combustion engine-powered car in history, using a hot-tube ignition system and the fuel known as Ligroin. Based on Otto's blueprint, it had a four-stroke engine. The first car was a four-stroke engine vehicle made by Karl Benz the following year. Otto's business, Gasmotoren fabrik Duets (GFD), created the carburetor and electric ignition in 1884. Daimler Motormen Gesellschaft was the name of the business that Daimler and May Bach founded in 1890. Currently, Daimler-Benz is that business [5], [6].

Atkinson Gas Cycle

James Atkinson created the Atkinson-cycle engine, a single-stroke internal combustion engine, in 1882. The Atkinson cycle, which is employed in several contemporary hybrid electric systems, is created to give efficiency at the sacrifice of power density. The first Atkinson-cycle piston engine was created to not infringe on any Otto-cycle engine patents and allowed the intake, compression, power, and exhaust strokes of the four-stroke cycle to happen in a single turn of the crankshaft. The Atkinson engine can achieve higher thermal efficiency than a conventional piston engine because of the special crankshaft design that allows its expansion ratio and compression ratio to differ. This engine also has a power stroke that is longer than its compression stroke. While Atkinson's original concept is little

more than a historical curiosity, many contemporary engines use unusual valve timing to simulate a shorter compression stroke/longer power stroke and so benefit from the increased fuel efficiency the Atkinson cycle can offer [7], [8].

Diesel Cycle

The Otto-cycle engine from 1876 has been technologically improved with the diesel engine. Otto had discovered in 1861 that the fuel mixture could be compressed before being ignited to boost the engine's efficiency, but Rudolf Diesel sought to create an engine that was more effective and could burn much heavier fuel. Illuminating Gas coal gas was the intended fuel for the Lenoir, Otto Atmospheric and Otto Compression engines 1861 and 1876, respectively. Diesel sought to develop an engine for small industrial businesses to have their own power source so they could compete with larger corporations and, like Otto, escape the restriction of being dependent on a municipal fuel supply. Diesel's motivation was the same as Otto's [9], [10].

Concerns for Fuel

Compressed charge engines have the issue that pre-ignition can be brought on by the compressed charge's temperature rise. This can harm the engine if it happens at the incorrect time and with too much vigor. The temperatures at which fuel may self-ignite, or flash point, varies greatly among various petroleum fractions. The design of the engine and fuel must take this into account. The chemical makeup of the fuel regulates the compressed fuel mixture's propensity to ignite early. To meet different engine performance levels, different gasoline grades exist. The fuel is changed to adjust the temperature at which it self-ignites. There are various methods for doing this. Pre-ignition is significantly more likely to happen as a result of greater compression ratio engines because the fuel mixture is pushed to a higher temperature before intentional ignition. Fuels like petrol may be evaporated more efficiently at a greater temperature, which boosts the compression engine's effectiveness. Higher expansion ratios also known as compression ratios increase the distance that the piston can push in order to produce power. An indicator of a fuel's resistance to self-ignition is its octane rating. A fuel with a higher numerical octane rating enables a higher compression ratio, which pulls more energy from the fuel and more successfully transforms that energy into usable work while also reducing pre-ignition damage to the engine. Additionally, high-octane fuel is more expensive. Petrol direct injection, or GDI, is a common feature of contemporary four-stroke

engines. An engine that uses direct injection of petrol has an injector nozzle that extends into the combustion chamber. The compression stroke, when the piston is at the top, is when the direct fuel injector injects petrol into the cylinder at a very high pressure. By definition, pre-ignition is not a problem for diesel engines. They are worried about whether or not combustion can occur. The Catani rating is a measure of how likely diesel fuel is to catch fire. Diesel fuels have a low volatility, which makes them difficult to start when cold. The most popular method for starting a cold Diesel engine is the use of a glow plug.

DISCUSSION

Limits on Power Output

A. Four-Stroke Engine

1=TDC

2=BDC

A: Compression

B. Intake

Power, C

Exhaust, D

The most air that an engine can take in will decide how much power it can produce. The size cylinder capacity, type of engine two- or four-stroke, volumetric efficiency, losses, air-to-fuel ratio, calorific value of the fuel, oxygen content of the air, and speed (RPM) all affect how much power a piston engine can produce. The speed is ultimately constrained by the lubrication and material strength. Connecting rods, pistons, and valves experience strong acceleration forces. Power loss or even engine failure can come from physical breakage and piston ring flutter at high engine speeds. When the rings vibrate vertically inside the grooves of the piston, it is known as piston ring flutter. Ring flutter weakens the seal that keeps the ring attached to the cylinder wall, reducing cylinder pressure and power. Too much engine speed prevents the valve springs from closing the valves quickly enough. Commonly known as "valve float," this can cause piston to valve contact and seriously harm the engine. The lubrication of the piston cylinder wall interface tends to degrade at high speeds. As a result, industrial engines can only have piston speeds of roughly 10 m/s.

Port Flow for Intake and Exhaust

The capacity of intake air-fuel combination and exhaust materials to pass swiftly via valve ports, which are normally found in the cylinder head, determines an engine's output power. Inconsistencies in the intake and exhaust routes, such as casting faults, can be eliminated to boost an

engine's output power, and with the use of an air flow bench, the radii of valve port turn and valve seat layout can be changed to lower resistance. Porting is a procedure that can be carried either manually or with a CNC machine. Recovery of waste heat from an internal combustion engine. Only 40–45% of the energy input can, on average, be converted by an internal combustion engine into mechanical work. Heat is one of the main forms of waste energy, and it is discharged into the atmosphere through coolant, fins, etc. The performance and/or fuel economy of the engine might be enhanced by increasing cycle efficiency if waste heat could be somehow caught and converted to mechanical energy. It has been discovered that even recovering 6% of the heat that is completely lost can significantly improve engine efficiency. In order to remove waste heat from an engine exhaust and use it further to extract some useful work while also lowering exhaust pollutants, numerous ways have been developed. As a waste heat recovery method, the Rankin Cycle, turbocharging, and thermoelectric production can all be very helpful.

Supercharging

More air being forced into the cylinder can help an engine run more efficiently by allowing for more power to be generated during each power stroke. A supercharger, a sort of air compression device that uses the engine crankshaft as power, can be used to do this. An internal combustion engine's maximum power output restriction is raised by supercharging in relation to its displacement. The supercharger is typically always operating, however some designs let it be turned off or run at different rates relative to engine speed. The drawback of mechanically driven supercharging is that some of the output power is used to power the supercharger, while power is lost in the high-pressure exhaust because the air is compressed twice and only expands once during combustion, wasting energy.

Turbocharging

A turbocharger is a supercharger that uses a turbine to be powered by the engine's exhaust gases. A vehicle's exhaust system incorporates a turbocharger to make use of the exhaust that is released. It comprises of a two-piece, high-speed turbine assembly, with one side powered by the exhaust gas outflow and the other side by the compressed intake air. The turbocharger has little impact when the engine is idling and at low to moderate speeds, the turbine creates little power from the limited exhaust volume, and the engine runs almost like a normally aspirated engine. The engine speed and throttle opening are increased when significantly more

power is needed until the exhaust fumes are enough to spool up the turbocharger's turbine and begin compressing significantly more air than usual into the intake manifold.

Thus, the operation of this turbine results in the release of additional power and speed. Because it uses exhaust pressure that would otherwise be squandered, turbocharging makes engines run more efficiently, but it has a design flaw known as turbo lag. Due to the need to quickly increase engine RPM, build up pressure, and spin up the turbo before the turbo starts to accomplish any meaningful air compression, the enhanced engine power is not immediately available. The turbo spins faster due to the greater intake volume, which also results in increased exhaust, and so on until stable high-power operation is obtained. Another issue is that the engine's mechanical components experience more heat transfer from the exhaust gas due to the higher exhaust pressure.

Ratio of Rod and Piston to Stroke

The ratio of the length of the connecting rod to the length of the piston stroke is known as the rod-to-stroke ratio. A longer rod decreases the stress forces and sideways pressure of the piston on the cylinder wall, extending the life of the engine. Additionally, the price, engine height, and weight all go up. An engine is referred to as a square engine if its bore diameter and stroke length are the same. An over square engine is one in which the bore diameter is more than the stroke length; an under square engine is one in which the bore diameter is less than the stroke length.

A valve train:

Typically, a camshaft moving at half the speed of the crankshaft opens and closes the valves. Along its length, it has a succession of cams, each of which is intended to open a valve at the proper time during an intake or exhaust stroke. The contact surface on which the cam glides to open the valve is called a tappet between the valve and the cam. As shown in the figure, many engines use one or more camshafts above a row or each row of cylinders, with each cam directly operating a valve through a flat tappet. Other engine designs place the camshaft within the crankcase, where each cam typically engages a push rod, which engages a rocker arm, which engages a valve, or where a push rod is not required, as in the case of a flathead engine. Because the overhead cam design offers the most direct path between the cam and the valve, higher engine speeds are often possible.

Valve Spacing

The little space that exists between a valve lifter and a valve stem to guarantee that the valve completely closes is known as valve clearance. Excessive clearance in engines with mechanical valve adjustment makes the valve train noisy. The valves may not close correctly if the clearance is too tiny. Performance is lost as a result, and exhaust valve overheating may occur. Typically, a feeler gauge must be used to readjust the clearance every 20,000 miles or 32,000 km. Hydraulic lifters are used by the majority of contemporary manufacturing engines to automatically correct for worn valve train parts. Lifter failure could result from dirty engine oil.

Energy Equilibrium

Otto engines have an efficiency of roughly 30%, which means that at the engine's output shaft, 30% of the energy produced by combustion is transformed into useful rotational energy, with the remaining energy being wasted as a result of waste heat, friction, and engine accessories. The energy lost to waste heat can be recovered in a variety of methods. By increasing incoming air pressure, turbocharging diesel engines is particularly successful at giving them the same performance gain as adding additional displacement. Years ago, the Mack Truck Corporation invented a turbine system that used waste heat to generate kinetic energy that was then fed back into the gearbox of the engine. BMW announced the invention of the turbo steamer, a two-stage heat-recovery system that recovers 80% of the energy in the exhaust stream and increases the efficiency of an Otto engine by 15%, in 2005. This technology is similar to the Mack system and recovers heat in two stages. In contrast, a six-stroke engine may consume up to 40% less fuel.

Modern engines are frequently purposefully designed to be a little less efficient than they could be. This is required for emission controls that lessen smog and other air pollutants, such as catalytic converters and exhaust gas recirculation. Lean burn approaches can be used by an engine control unit to combat efficiency drops. In contrast to the existing requirement of 25 mpg US 9.4 L/100 km 30.0 primp, the Corporate Average Fuel Economy in the United States requires that vehicles achieve an average of 34.9 mpg US 6.7 L/100 km 41.9 primp. New approaches to building the conventional internal combustion engine (ICE) must be taken into account as automakers strive to satisfy these regulations by 2016. Applying the Miller cycle and firing when the piston is farthest from the crankshaft, or top dead center, are two potential ways to boost fuel efficiency to comply with new regulations. This

change could drastically lower NOx emissions and fuel consumption.

Working Principle

A four-stroke engine uses four piston strokes to complete a power cycle as shown in Figure 1. The way it operates is as follows:

1. Compression stroke
2. Power stroke.
3. Exhaust stroke.
4. Intake stroke.

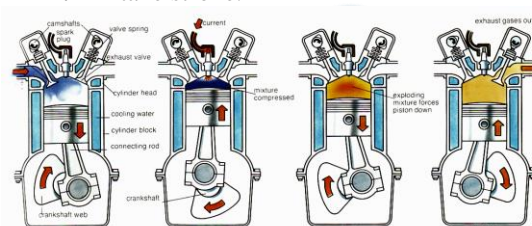


Figure 1: Representing the Working operation of Four Stroke Engine [Bike Republic].

Intake Stroke

- i. At the suction side of the compression chamber cylinder, a vacuum begins to form when the piston reciprocates from TDC towards BDC downward.
- ii. The exhaust valve closes and the inlet valve opens when a vacuum is created on the suction side.
- iii. Air and fuel begin to enter the compression chamber as soon as the inlet valve opens.

Compression Stroke

- i. The inlet valve closes and the compression stroke begins as soon as the internal pressure of the compression chamber reaches equilibrium with the external pressure.
- ii. By compressing the air-fuel combination inside the compression chamber as it rises from BDC to TDC, the piston raises both the temperature and pressure of the mixture.

Power Stroke

- i. The term combustion stroke also applies to the power stroke.
- ii. A spark plug burns the compressed air-fuel mixture as the compression stroke approaches its conclusion.
- iii. The power to propel the piston from TDC to BDC is produced as the gasoline ignites by intensifying the chemical reaction. Consequently, this stroke is known as a power stroke.
- iv. The mixture's temperature and pressure increase dramatically as a result of this

burning process. The air-fuel mixture is forced to move downward towards BCD from TDC by the air-fuel mixture as a result of a rise in pressure, which powers the crankshaft and further moves the vehicle.

- v. The inlet and exhaust valves are shut off during this procedure.

The Exhaust Stroke

- i. The power stroke is over, and the exhaust stroke begins.
- ii. The piston again rises higher during this stroke, this time from BDC to TDC.
- iii. The inlet valve is closed during this stroke, while the exhaust valve is opened. The piston forces the combustion chamber's exhaust gases outside.
- iv. The cycle is then repeated once the piston swings lower once more from TDC to BDC, sucking the air-fuel mixture. The exhaust and used gases are forced out of the cylinder by this last stroke.

Engine Cycle of a Four-Stroke

The 4-stroke engine's operational cycle is depicted in the following PV diagram (Figure 2). The steps a four-stroke engine takes to complete a working cycle are as follows:

Isobaric Process (0 to 1): The piston descends during the isobaric process, creating a vacuum inside the combustion chamber. The ambient pressure and the internal pressure of the chamber develop a pressure differential during vacuum generation. This pressure differential causes the intake valve to open, allowing the mixture of fuel and air to enter the combustion chamber.

Process of adiabatic (1 to 2): The inlet valve closes after the isobaric process is finished, and the piston rises to pressurize the fuel-air combination. The mixture's temperature and pressure increase during this operation, but its heat remains constant.

Isochoric Process (2 to 3): An adiabatic process spark plug ignition occurs at the end of the compression stroke. The air-fuel mixture is brought to a high temperature and pressure through this process, which increases its temperature and pressure. The entropy of the air-fuel mixture also rises as a result of the ignition process.

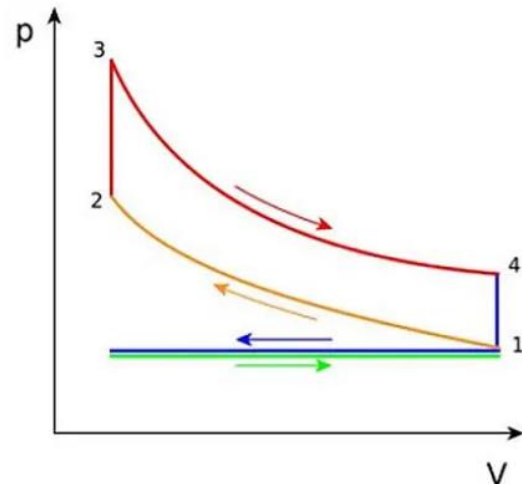


Figure 2: Representing the Engine Cycle of a Four-Stroke [Mechanical Boost].

Power Stroke (Processes 3 to 4): During this stroke, heat generated by the ignition process is used to force the piston to descend, which further drives the crankshaft. The engine's crankshaft rotates, which moves the car. As a result, this procedure is known as a power stroke.

Exhaust Phase (4 to 1): During this phase, the piston rises once again, the exhaust valve opens, and the waste heat from the combustion chamber is released. The air-fuel mixture's molecules' kinetic energy reduces when waste heat is removed. The cycle starts over when there is another pressure difference between the interior pressure of the chamber and the pressure in the atmosphere.

Four-Stroke Engine Benefits

1. **Reliability:** Compared to 2-stroke engines, these models are more dependable and efficient.
2. **Durability:** Compared to 2-stroke engines, these engines are more durable.
3. **Environmentally Beneficial:** Compared to 2-stroke engines, 4-stroke engines emit fewer hazardous emissions, making them more environmentally friendly. Heavy loads and large vehicles are best suited for these engines.
4. **Fuel Economy:** Compared to 2-stroke engines, these engines are highly fuel efficient. These operate more quietly than two-stroke engines.
5. **More Torque:** Four-stroke engines generate more torque than two-stroke engines while operating at low speeds.

6. **More Fuel-Efficient:** Compared to a two-stroke engine, this type of IC engine is more fuel-efficient. No need for additional lubrication or oil to add fuel. This engine does not need any additional lubrication or oil. Only the moving parts need to be lubricated right away.

Drawback of Four-Stroke Engines

Four-stroke engines have less power than two-stroke engines, which is a drawback:

1. **Costly:** A four-stroke engine contains a lot of components. As a result, it is more expensive than a two-stroke engine.
2. **Weight:** Compared to 2-stroke engines, these engines weigh more.
3. **Area:** They needed a sizable space to install. More piston strokes are necessary to complete a power cycle.
4. **Design:** The design of these engines is intricate.

Application of Four Stroke Engine:

1. **Automobiles:** Cars, trucks, and other motor vehicles most frequently employ 4-stroke engines. These are extremely reliable and efficient. They are frequently utilised in construction machinery like loaders, excavators, and bulldozers.
2. **Marine Engines:** The propellers on boats and ships are propelled by 4-stroke diesel engines.
3. **Garden and Lawn Equipment:** Various leaf blowers, lawn mowers, and other outdoor power equipment employ four-stroke engines. These engines start very quickly.
4. **Generators:** These engines convert fuel into energy in both stationary and portable generators.

CONCLUSION

An internal combustion engine with a four-stroke cycle uses four different piston strokes to complete one operational cycle: intake, compression, power, and exhaust. To accomplish one operating cycle, the piston must make two full passes through the cylinder. The crankshaft must turn twice (720°) in one functioning cycle. The most prevalent kind of tiny engine is one with a four-stroke cycle. In one working cycle, a four-stroke cycle engine completes five strokes intake, compression, ignition, power, and exhaust. An internal combustion (IC) engine is referred to as having four strokes also known as four cycles when the piston is rotating the crankshaft. A piston's whole travel in either direction along the cylinder is referred to as a stroke.

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Introduction of Fuel and Its Types

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ABSTRACT: Any substance that may be produced to react with other substances to release energy as thermal energy or to be used for work is referred to as a fuel. The idea was initially only applied to substances that could release chemical energy, but it has subsequently been extended to include other types of thermal energy, such as nuclear energy produced by nuclear fission and nuclear fusion. A heat engine can transform the heat energy released by fuel reactions into mechanical energy. Other times, the heat itself is prized for its ability to provide warmth, facilitate cooking or industrial activities, and produce light when combustion occurs. In a process known as cellular respiration, fuels are also utilised by living things in their cells to oxidase organic molecules and produce useful energy.

KEYWORDS: Crude Oil, Diesel Fuel, Diesel Engine, Liquid Fuel, Natural Gas

INTRODUCTION

The burning of firewood by Homo erectus almost two million years ago is the first recorded usage of fuel. Humans have only ever used fuels made from plants or animal fat for the majority of recorded human history. Since at least 6,000 BCE, metals have been melted using charcoal, a wood product. It was only replaced by coal-derived coke as European forests began to disappear around the 18th century. As a fuel for grilling, charcoal briquettes are now often utilised. In Arabic handbooks like those of Muhammad ibn Zakary Rzi, crude oil distillation was described in detail by Persian chemists. In his Kitab al-Asrar, he explained the procedure for distilling petroleum/crude oil into kerosene as well as other hydrocarbon compounds. The same time period saw the production of kerosene from bitumen and oil shale by burning the rock to release the oil, which was then refined. Rzi also described the naphtha, a kerosene lamp that burns crude mineral oil, for the first time [1], [2].

Baghdad's streets were covered in tar, which was made from petroleum that was discovered in nearby natural reserves. Oil resources were used in the 9th century in and around the contemporary Azerbaijani city of Baku. Both Marco Polo and the Arab geographer Abu al-Hasan 'Al al-MAs' recorded these fields in the tenth and thirteenth centuries, respectively. Marco Polo estimated that the yield of those wells was hundreds of shiploads. After the invention of the steam engine in the United Kingdom in 1769, coal became a more popular power source since it could be converted into chemical energy and released by combustion. Later, coal was utilised to power ships and locomotives. Gas produced from coal was utilised for street lighting in London by the 19th century. In the 20th and 21st centuries, coal has mostly been used to

produce energy, accounting for 40% of the global supply in 2005. Due to their greater concentration and adaptability compared to more conventional energy sources like water power, fossil fuels were quickly embraced throughout the Industrial Revolution. Fossil fuels have played a significant role in our modern culture, with the majority of nations using them to generate electricity. However, they are losing favor because of the consequences they have on the environment, including global warming [3], [4].

Renewable fuels, particularly alcohol-based biofuels, are currently in demand. Any substance that can react with other materials to release energy as thermal energy or for use in work is a fuel. The idea was first only applied to substances that could release chemical energy, but it has subsequently been expanded to include other heat-producing sources, such as nuclear energy produced by nuclear fission and nuclear fusion. A heat engine can transform the thermal energy produced by the combustion of fuels into mechanical energy. In other cases, the heat itself is prized for its use in cooking, heating, or industrial activities, as well as for the illumination it provides while fuel burns. In the process known as cellular respiration, which uses fuels, organic molecules are oxidized to produce usable energy in the cells of living things. Humans use a variety of chemicals, including radioactive metals, as fuel, although hydrocarbons and related organic compounds are by far the most frequent. Fuels are compared to other materials or machinery that store potential energy, such as those that quickly release mechanical or electrical energy such as flywheels, springs, compressed air, or water in a reservoir, or both. Chemical fuels are compounds that release energy when they interact with other substances, most notably when they burn [5], [6].

There are two categories for chemical fuels. Initially, based on their physical characteristics as a solid, liquid, or gas. Next, based on where they are found primary natural fuel and secondary. So, the following is a general classification of chemical fuels: The term solid fuel refers to a variety of solid materials used as fuel to generate energy and provide heating, which is typically released by combustion. Solid fuels can be in the form of wood, charcoal, peat, coal, hexamine fuel tablets, wood pellets, corn, wheat, rye and other grains. Another form of solid fuel is used in solid-fuel rocketry. Humanity has been using solid fuels to start fires for a very long time. The fuel source for the industrial revolution, used to power everything from steam engines to furnaces, was coal. Steam locomotives were also frequently powered by wood. Today, the production of power still uses both peat and coal. Due to dangerous levels of harmful emissions, several urban areas restrict or outright forbid the use of some solid fuels, such as coal. As heating technology and the availability of high-quality fuel improve, less and less other solid fuels like wood are used. The only solid fuel that is frequently utilised in some places is smokeless coal. Briquettes made from peat are used as smokeless fuel in Ireland. They can also be utilised to light a coal fire [7], [8].

Fluid Fuels

The Petrol Station

Liquid fuels are combustible or energy-producing substances that can be used to generate kinetic energy or other forms of mechanical energy. Since the fumes from liquid fuels, not the fluids themselves, are combustible, they must also conform to the shape of their container. The majority of liquid fuels that are used extensively are made from the fossilized remains of extinct plants and animals that were exposed to heat and pressure inside the Earth's crust. However, there are a number of different kinds, including liquid fuels like biodiesel, ethanol, jet fuel, and hydrogen fuel for use in automobiles. Heavy oil fractions can now be used as liquid fuels thanks to the development of emulsified fuels of oil in water like Orimulsion. The economy and transportation both heavily rely on liquid fuels. Liquid fuels typically have the ability to be handled easily and are simple to carry. Additionally, they are fairly simple to utilize for both domestic and engineering applications.

Some nations ration fuels like paraffin for domestic consumption, such as in government-funded stores in India. In that it is a blend of aliphatic hydrocarbons that are derived from petroleum, conventional diesel is similar to petrol in that regard.

Kerosene is a fuel that can be used in tiny motors, kerosene lamps and for cooking and warmth. Natural gas, which primarily consists of methane, can only be used directly as a liquid fuel in a small number of applications since it can only exist as a liquid at very low temperatures regardless of pressure. Under typical air circumstances, propane and butane, which make up LP gas, are both easily compressible gases. Despite being denser than air, burning less cleanly, and being considerably easier to compress, it has many of the same benefits as compressed natural gas (CNG). Compressed propane and LP gas are increasingly being used in motorized vehicles, where they are frequently employed for cooking and space heating [9], [10].

DISCUSSION

Fuel Type Employed

Engines are divided into two groups according to the type of fuel they utilize. Engines that use volatile liquid fuels like alcohol, kerosene, petrol and benzene. The fuel is often combined with air in a carburetor located outside the cylinder to create a homogenous charge, which is then sucked into the cylinder during its suction stroke. These engines are known as spark-ignition engines because the charge is ignited at the end of the compression stroke by an externally delivered spark. Engines that run on gaseous fuels, such as biogas, compressed natural gas (CNG), liquefied petroleum gas (LPG), blast furnace gas and compressed natural gas (CNG). The shorter igniting delay of gaseous fuels makes them superior to liquid fuels in comparison. During the suction operation, a mixture of the gas and air is delivered into the cylinder. This engine's operation is comparable to that of the SI petrol engine, which uses volatile liquid fuels. An engine that burns solid fuels like charcoal, coal powder, etc. In most cases, gas is produced outside the engine from solid fuels in a separate gas producer, and the engine operates as a gas engine. Engines that run on liquid fuels that are viscous low volatility at normal ambient temperatures, such as heavy and light diesel oils. A fuel injection system typically injects fuel into the cylinder in the form of tiny droplets at the conclusion of the compression process. When the fuel in the cylinder comes into contact with the hot compressed air, combustion of the fuel occurs. As a result, these engines are referred to as compression-ignition engines.

Motors that Use Two Fuels

Through a gas valve in the cylinder head, a gaseous fuel or a highly volatile liquid fuel is supplied alongside air during the suction stroke or the

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beginning of compression, and the other fuel a viscous liquid fuel is injected into the combustion space close to the end of the compression stroke. They are known as dual-fuel engines.

Types of Fuel

A fuel passes through a process when it is utilised, which results in a form with less energy. This indicates that while the majority of fuels are not renewable, they can still be discovered in sufficient quantities to be regarded as sustainable. For additional information on the distinction between primary flows and fuels, fuel vs flow. Primary flows, such as wind, are not considered fuels and constitute a class of primary energy supply wholly different from fuels. Nuclear fuels, biofuels, and fossil fuels are examples of primary fuels. The way that primary fuels are extracted from a natural resource is frequently treated to create something that is chemically different. For instance, fractional distillation is used to create products that are more valuable to consumers out of crude oil, a primary fuel. Diesel, kerosene, and petrol are all types of fuels; however, they differ from one another since they come from basic energy sources. As opposed to primary fuels, they are secondary fuels. These second-hand fuels, which may also be regarded as energy currencies, were processed from the form found in natural resources. In order to get the maximum energy feasible from a given amount of crude oil, secondary fuels are frequently manufactured since they are simpler for engines to burn.

Furthermore, during the fractional distillation process, fuels like methane, butane, and propane that are found mixed together in their natural resource which would be the major energy source are separated. Since hydrogen does not occur naturally in large quantities on Earth, it is regarded as an energy currency. Hydrogen is a fuel that may be chemically produced from water or methane among other sources. The energy density, price, and environmental impact of fuels vary widely. For instance, uranium has a slightly higher energy density than fossil fuels but is also significantly more expensive. Due to the differences in how each is used, it is also challenging to compare the energy density and environmental impact of fuels to main flows. The energy mix in each nation varies greatly. For a complete map showing the sources of electricity in each nation, please see electricity generation. The world's energy production is depicted in the pie chart below, with fuels represented by the graph's remaining slices.

Petroleum Fuel History

Since ancient times, petroleum has been utilised in some capacity. When the Sumerians employed it to create boats more than 4300 years ago, bitumen was first referenced. A straw and bitumen-sealed basket was referenced on the tablet describing the narrative of Sargon of Akkad's birth. Herodotus and Deodars Spicules claim that asphalt was used to build Babylon's walls and towers more than 4,000 years ago. They also mention the presence of oil pits close to Ardericca and a pitch spring on Acanthus. On the banks of the Issus River, one of the Euphrates' tributaries, it was discovered in large amounts. Petroleum was used in the higher classes of Persian society, according to ancient Persian tablets, for illumination and medicine. More than 2000 years ago, the Chinese used petroleum for many purposes. Oil was originally found, produced, and used in China in the first century BCE in its unrefined state, according to the I Ching, one of the earliest Chinese books. Furthermore, petroleum use was first documented by the Chinese in the fourth century BCE, making them the first to do so. Oil production began in China around 347 CE from wells dug with bamboo.

In Arabic handbooks like those of Muhammad ibn Zakary Rzi, crude oil distillation was a common practice among Persian chemists, and it is clearly described in these works. The tar that was used to pave Baghdad's streets was made from petroleum that was discovered in the area's natural fields. The present Baku, Azerbaijan, area saw the first oil field exploitation in the ninth century. Hundreds of shiploads of oil were produced by those wells, according to Marco Polo and the Arab geographer Abu al-Hasan 'Al al-MAs', who recorded these fields in the 10th and 13th centuries, respectively. For combustible compounds used in warfare, Arab and Persian chemists also refined crude oil. By the 12th century, Western Europe had access to distillation thanks to Islamic Spain. In Romania, where it is known as pacer, it has existed since the 13th century.

The Seneca People and other Iroquois in Western Pennsylvania began to excavate sophisticated oil pits as early as 1415–1450, measuring 4.5–6 meters deep. When the French General Louis-Joseph de Montcalm visited Fort Duquesne in 1750, he saw Seneca utilizing petroleum for ceremonial fires and as a therapeutic ointment. In 1795, there were hundreds of hand-dug wells operating in Yuanyuan, the center of a thriving oil extraction business that was noted by early British explorers to Myanmar. The first location in Europe where petroleum has been explored and exploited is reportedly Echelon. Since 1498, the still-operational Erdpech quelle has

been used, particularly for medical purposes. It is a spring where water and oil may be seen mingled together. Since the 18th century, people have been mining oil sands. In lower Saxony's Wetzlar since the 18th century, natural asphalt/bitumen has been investigated. The coal industry dominated the petroleum technologies in Wetzlar and Pechelbrunn, respectively.

Petroleum

Also known as crude oil or simply oil, petroleum is a naturally occurring yellowish-black liquid mixture made up primarily of hydrocarbons that is found in geological strata. Both naturally occurring crude oil that hasn't been refined and petroleum products made of refined crude oil are referred to by the same name: petroleum. Petroleum is a fossil fuel that is created when massive amounts of extinct creatures, primarily zooplankton and algae, are buried beneath sedimentary rock and subjected to high temperatures and pressures over an extended period of time. Drilling for oil is the main method of recovering petroleum. Following investigations of sedimentary basin analysis, structural geology, and reservoir characterization, drilling is done. Oil sands and oil shale are two more unconventional sources that have recently been exploited because of technological advancements. When oil is extracted, it is refined and separated easiest done by distillation into a plethora of products that can be used either directly or as inputs in other processes. Products range from fuels like petrol (petrol), diesel, kerosene and jet fuel to asphalt and lubricants to chemical reagents used in the production of plastics, solvents, textiles, refrigerants, paint, synthetic rubber, fertilizers, insecticides, pharmaceuticals and countless more items.

It is estimated that the world consumes roughly 100 million barrels 16 million cubic meters of petroleum per day. Petroleum is used to manufacture a wide range of goods that are necessary for contemporary life. Petroleum production has a high potential for profit and was crucial to the growth of the world economy in the 20th century. Some nations, known as oil states, gained tremendous economic and political clout as a result of controlling oil production. The environment and public health are both harmed by petroleum products. Petroleum is a significant contributor to climate change since it is extracted, refined, and burned as fuel, all of which emit significant amounts of greenhouse gases. Oil spills, contamination of the air and water, and other environmental harms are also present. Humans are affected by some of these effects in both direct and indirect ways. In addition to causing wars that are

led by states, oil has also been a source of other conflicts. As global economies migrate away from reliance on petroleum as a means of reducing climate change and shifting towards renewable energy and electrification, it is predicted that oil production will peak by 2035.

Diesel Fuel

Rudolf Diesel, a German scientist and inventor who created the compression-ignition engine in the early 1890s, experimented with diesel fuel. Diesel first rejected the idea of using a particular form of fuel, claiming that the rational heat motor's underlying principle would function with any kind of fuel in any condition of matter. However, only liquid fuels were intended for use in the initial diesel engine prototype and the first operational diesel engine. Diesel used Pechelbrunn crude oil at first, but soon switched to petrol and kerosene since it was too viscous, with kerosene serving as the primary testing fuel for the Diesel engine. Additionally, Diesel experimented with several kinds of lamp oil from different sources, as well as various kinds of petrol and lignin, all of which were successful as fuels for Diesel engines. Diesel later tried fuel oil, petrol, crude oil, paraffin oil and coal tar creosote, all of which eventually worked as well.

Shale oil served as the primary fuel for the first Diesel engines produced in 1898 in Scotland and France since other fuels were prohibitively expensive. The French Otto Society created a Diesel engine for use with crude oil in 1900, and it was displayed at both the 1911 World's Fair and the 1900 Paris Exposition. Instead of using crude oil, the engine really operated on peanut oil, and no adjustments were needed. Diesel also employed lighting gas during his initial engine tests, and he was able to create functional designs both with and without pilot injection. Diesel claims that in the late 1890s, there was neither a fine, high-quality coal dust business nor was it commercially available. Because of this, the Diesel engine was never intended to be a coal-dust engine. Diesel didn't test a coal-dust prototype with liquid fuel pilot injection and external mixture creation until December 1899. This engine performed as expected, but after a short time, piston rings failed as a result of coal dust buildup.

Diesel: Diesel engines are a particular type of internal combustion engine in which fuel ignition occurs without the use of a spark as a result of the compression of the inlet air and the subsequent injection of fuel. Diesel fuel, also known as diesel oil or historically heavy oil, is any liquid fuel specifically designed for use in a diesel engine.

Diesel fuel therefore requires favorable compression ignition properties. The most prevalent type of diesel fuel is a particular fractional distillation of petroleum fuel oil, however non-petroleum alternatives such as biodiesel, biomass to liquid (BTL), or petrol to liquid (GTL) diesel are being researched and accepted more frequently. In some academic circles, petroleum-derived diesel is sometimes referred to as petrol diesel to distinguish between these varieties. Diesel fuel is regulated in many nations. For instance, EN 590 is the standard for diesel fuel in the European Union. The Sulphur level of diesel fuel has been significantly reduced in ultra-low-sulfur diesel (ULSD). In the United Kingdom, continental Europe, and North America as of 2016, practically all petroleum-based diesel fuel was of the ULSD kind. Prior to the standardization of diesel fuel, the bulk of diesel engines were normally powered by inexpensive fuel oils. Diesel engines for watercraft continue to use these fuel oils. Diesel fuel can be used as fuel for a number of non-diesel engines, such as the Arold engine, the Sterling engine, or boilers for steam engines, despite being specifically made for diesel engines. Heavy vehicles frequently run on diesel fuel, but older engines' exhaust, in particular, can be harmful to health.

CNG, or Compressed Natural Gas

Less than 1% of the volume of CNG, or compressed natural gas, is made up of compressed methane. It is one of the newer fuel kinds for cars operating in cities, with the main goal being to cut down on pollution.

Pros

It is commonly referred to as Green Fuel and is devoid of lead and Sulphur. 540 degrees Celsius is the high auto-ignition temperature, and the flammability range is just 5%–15%. It suggests that a concentration of 5%–15% CNG in the air won't result in unintentional burning, hence ensuring safety. It is safe to use because it is kept in approved, leak-proof cylinders. It is a light gas, so if it leaks, it disperses and mixes with the air.

Cons

CNG stations are harder to find than petrol or diesel outlets. It takes up at least one-third of the boot's volume. This makes it challenging for owners of sedans or other types of vehicles to pack their belongings into the trunk. After 3 to 4 years of use, a CNG car's performance starts to degrade. After frequent use for a year, the engine's peak performance drops by 10%. It impairs fuel injector performance and hastens their drying. Long-term, it degrades engine performance. Compared to other vehicles, a CNG-powered vehicle has a poorer fuel

efficiency. Converting from traditional petrol or diesel-powered cars to CNG is expensive. It is frequently utilised in passenger cars. CNG can be used to power vehicles like the BMW 3 Series (E36) and Audi A5 2, 0 TFSI CNG. In addition, vehicles including vans, buses, trucks, and more may run on CNG.

Bio-Diesel

Biodiesel is created by mixing different oils with diesel. Tran's esterification is the process of turning waste cooking oil, animal fat, and vegetable oil into biodiesel. It is one of the greatest biofuel substitutes because it contains natural chemicals.

Pros

Beneficial to the environment and emits 11% less carbon monoxide than other fuels. It burns at a higher temperature and has lower flashpoints. This fuel may be stored easily because there is less chance of a sudden ignition. Sustainability is ensured because it is a non-toxic, renewable source of energy. aids in extending an engine's life. It functions as a solvent to get rid of the dirt built up in the engine and get it back to running at its best, avoiding regular wear and tear.

Cons

It can be used in diesel engines with or without modifications by vehicle owners. This fuel lessens reliance on oil imports from other nations. It should not be used at lower temperatures since it could gel. In colder temperatures, a fuel's paraffin component freezes and transforms into a gel-like substance, a process known as gelling. Despite the fact that bio-diesel has the benefit of cleaning the engine, this dirt gets lodged in the gasket and causes harm. Additionally, it harms an engine's rubber housings. A car powered by biodiesel has a lower fuel efficiency than other vehicles. By 1% to 2%, fuel efficiency is decreased. It costs more than petroleum. Animal fat, vegetable oil, and other materials are used to make bio-diesel. It has an impact on the availability of food and raises the cost of the specified goods, creating a food shortage. It has regional limitations because not all regions can produce a particular crop. Ideal for: Biodiesel is suitable for use in diesel-powered vehicles. Additionally, permitted to run on biodiesel are vehicles, vans, and SUVs with particular construction.

CONCLUSION

In order to produce energy through a chemical or nuclear reaction, fuel must be burned. Typically, it is employed to drive machinery, produce electricity, or power vehicles. Fuels can be solid, liquid, or gas,

and they are typically burned or interacted with oxygen to release energy in the form of heat or to perform mechanical work. Fuels can also exist in other forms as well, such as vapor or gas. Any substance that may be produced to react with other substances to release energy as thermal energy or to be used for work is referred to as a fuel. The idea was initially only applied to substances that could release chemical energy, but it has subsequently been extended to include other types of thermal energy, such as nuclear energy produced by nuclear fission and nuclear fusion.

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Air-Standard Cycles and Their Analysis

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ABSTRACT: An internal combustion engine's operating cycle can be divided into a series of distinct operations, including intake, compression, combustion, expansion, and exhaust. Because the internal combustion engine is an open system, where the working fluid enters the system under one set of conditions and exits under another, it does not run on a thermodynamic cycle. However, it's frequently possible to analyse an open cycle as if it were closed by conceiving one or more processes that would return the working fluid at the exit circumstances to its initial state.

KEYWORDS: Carnot Cycle, Constant Volume, Effective Pressure, Internal Combustion, Mean

INTRODUCTION

Intake, compression, combustion, expansion, and exhaust are the individual steps that make up an internal combustion engine's operational cycle. Since the working fluid enters the system at one set of conditions and exits at another, the internal combustion engine does not run on a thermodynamic cycle. However, by picturing one or more processes that would return the working fluid at the exit circumstances to the state of the starting point, it is frequently possible to analyse the open cycle as though it were a closed one. It is extremely difficult to analyse internal combustion engine processes accurately [1], [2]. It is helpful to examine the performance of an idealized closed cycle that closely resembles the real cycle in order to comprehend them. The air-standard cycle is one such method and is predicated on the following notions:

- i. It is assumed that the working medium is a perfect gas and that it follows the relationship $pV = mRT$ or $p = RT$.
- ii. The working medium's mass remains unchanged.
- iii. The cycle's many components can all be reversed.
- iv. It is assumed that heat is not produced by chemical processes during the cycle but rather comes from a steady source of high temperature.
- v. It is believed that some heat will be rejected during the cycle to an ongoing low temperature sink.
- vi. It is assumed that the system doesn't lose any heat to the environment.
- vii. The working medium's specific temperatures remain constant over the course of the cycle.
- viii. The physical constants of the working medium, C_p , C_v , and M , are the same as those of air under typical atmospheric

conditions. For instance, in SI units $C_p = 1.005 \text{ kJ/kg}$, $M = 29 \text{ kg/kmol}$.

$K = 1.4$ and $C_v = 0.717 \text{ kJ/kg}$.

These presumptions cause the analysis to be oversimplified, and the outcomes diverge from those of the actual engine. The highest achievable values for work output, peak pressure, peak temperature, and thermal efficiency will be based on air-standard cycles and will be very different from those of the real engine. It is frequently utilised, mostly because to how easy it is to obtain approximations of the complex processes in internal combustion engines. The numerous cycles will be discussed in this chapter, together with the formulae for work output, mean effective pressure, efficiency, etc. Additionally, a comparison between the Otto, Dual, and Diesel cycles will be done to determine which cycle is most effective given a particular set of operating circumstances [3], [4].

The Carnot Cycle

In a reversible cycle, the working medium accepts heat at a higher temperature and rejects heat at a lower temperature, as postulated by French engineer Sadi Carnot in 1824. As indicated in Figure. 1, the cycle will be composed of two isothermal and two reversible adiabatic processes. The Carnot cycle is portrayed as a benchmark for excellence, and engines can be compared to it to assess their level of excellence. It introduces the idea of maximizing output while staying within two temperature thresholds [5], [6].

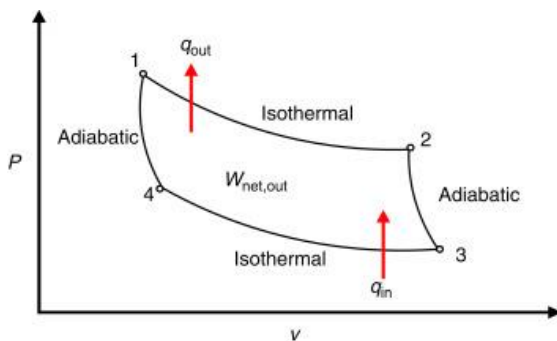


Figure 1: Representing the steps involved in the Carnot Cycle [Science Direct].

Which depicts a cylinder and piston arrangement operating without friction, the operation of an engine based on the Carnot cycle can be explained. It is assumed that the cylinder's walls are ideal insulators. The cylinder head's design enables it to function perfectly as both a heat conductor and an insulator. A high temperature source (T_3) first transfer's heat to the working media in the cylinder, which causes the working medium to expand. The isothermal process 34 in Figure 1 serves as a representation of this. The cylinder head is now sealed and functions flawlessly as insulation. Now, state 4 working medium in the cylinder is permitted to expand even more [7], [8].

DISCUSSION

Carnot Engine Fundamentals

Only cyclical machines, such as heat engines, are subject to Carnot principles, which stipulate. An irreversible heat engine's efficiency is always lower than a reversible one running between the same two reservoirs. All reversible heat engines that operate between the same two reservoirs have the same efficiency. More information on reversible and irreversible processes can be found here. The temperature of the combustion chamber must be raised in order to improve the thermal efficiency of a gas power turbine. For instance, the high-temperature gas is too much for turbine blades, which causes early fatigue [5], [9].

Carnot Theorem:

According to this theorem, no engine operating between a pair of specified temperatures can be more effective than a reversible engine operating between a pair of identical temperatures, and all reversible engines operating between a pair of identical temperatures have the same efficiency, regardless of the type of working substance. The reversible engine will always be more efficient than the irreversible one, according to the Carnot theorem. The reversible heat engine works as a heat

pump and runs on a reverse cycle. Figure 1's p-V and T-s diagrams show state 1 and the reversible adiabatic process 41 that represents it. As the cylinder head is now designed to function as an ideal heat conductor, the system is now brought into touch with a constant low temperature sink, (T_1). The working medium is compressed from state 1 to 2, which is symbolized by isothermal line 12, as a result of some heat being rejected to the sink without changing the temperature of the sink [10].

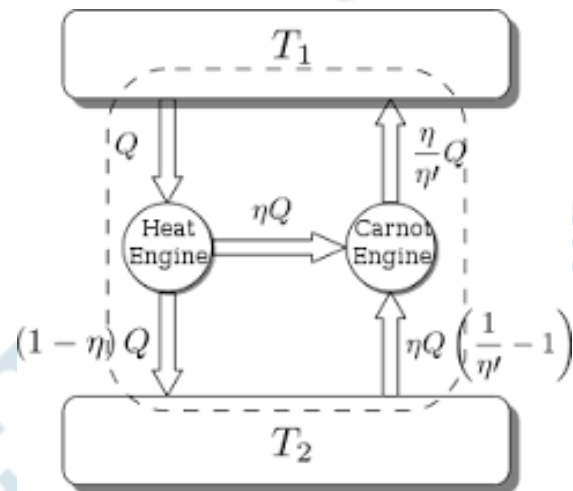


Figure 2: Representing the theorem de Carnot engine cycle [Topper].

Finally, the working medium is adiabatically compressed from state 2 to state 3, which is represented by process 23, and the cylinder head is once more made to function as a perfect insulator. The cycle is finished as a result. When the cycle is examined thermodynamically, its efficiency can be expressed as. Heat provided to the system during the cycle (Q_S) equals Carnot = Work done by the system during the cycle (W). The first law of thermodynamics states that work is equal to heat supplied minus heat rejected.

$$W = Q_S - Q_R$$

When the isothermal processes 12 and 34 are taken into account, we obtain $Q_R = mRT_1 \log V_1 V_2$.

$mRT_3 \log V_4 V_3$ $Q_S =$ Taking into account the adiabatic processes 23 and 41, $V_3 V_2 = T_2 T_3$

$$V_4 V_1 = T_1 T_4 (1 - \gamma - 1)$$

Given that $T_1 = T_2$ and $T_4 = T_3$, we obtain $V_4 V_1 = V_3 V_2$ or $V_4 V_3 = V_1 V_2 = r$ (say) (2.6), which leads to the following: Carnot = $mRT_3 \log r - mRT_1 \log r$ $mRT_3 \log r$ (2.7) = $T_3 T_1 T_3 = 1 T_1 T_3$.

The atmospheric temperature or the temperature of the cooling water is typically the lower temperature, or sink temperature, T_1 , and is thus constant. Therefore, increasing the source temperature is the only way to boost thermal efficiency. To attain the

highest level of thermal efficiency, the top temperature must be maintained as high as feasible. In comparison to other air-standard cycles, the Carnot cycle and other reversible cycles operate with the highest efficiency between two set temperatures. Despite this benefit, the work output from the Carnot cycle will be very low, making it an unsuitable foundation for the operation of an engine using a gaseous working fluid. Mean effective pressure, or pm, is the constant pressure that would hypothetically act on the piston during its expansion stroke to do the same amount of work that the real cycle would. PM is calculated mathematically as Work Output Swept Volume.

The constant, which has the unit's bar/m, is dependent on the method used to obtain the indication diagram. These formulas are frequently used to determine an internal combustion engine's performance. When the work output is the brake output, the pressure is referred to as brake mean effective pressure, or pbm, and when the work output is the indicated output, it is referred to as indicated mean effective pressure, or pim. The Carnot cycle, which is reversible, represents the maximum efficiency achievable for an engine cycle. When operating at the same temperatures, practical engine cycles have an inherent lower efficiency than the Carnot efficiency because they are irreversible. The addition of and withdrawal of from the working fluid during the cycle are two elements that affect efficiency. The working fluid is heated to its highest temperature before being added to the Carnot cycle, which results in maximum efficiency.

The Performance of the Carnot Cycle

The Carnot cycle, which is reversible, represents the maximum efficiency achievable for an engine cycle. When operating at the same temperatures, practical engine cycles have an inherent lower efficiency than the Carnot efficiency because they are irreversible. The addition of and withdrawal of from the working fluid during the cycle are two elements that affect efficiency. The working fluid is heated to its highest temperature before being added to the Carnot cycle, which results in maximum efficiency. The following steps make up the Carnot engine cycle when used as a heat engine:

1. The gas experiences reversible isothermal expansion at the hot temperature.
2. Gas isentropic expansion reversible adiabatic expansion.
3. The gas is compressed using reversible isothermal compression at the cold temperature.
4. The gas is compressed isentropic ally.

Stirling Cycle

The exceedingly low work output of the Carnot cycle results in a low mean effective pressure. Therefore, the Sterling cycle is one of the modified variants of the cycle that create higher mean effective pressure while theoretically achieving full Carnot cycle efficiency. There are two isothermal and two constant volume processes in it. At a steady temperature, heat rejection and addition occur. Figures 3 display the p-V and T-s diagrams for the Sterling cycle. Figure 3 shows that for constant volume operations, the amount of heat addition and rejection is the same. As a result, Sterling = RT3 loge V4 V3

$$RT1 \log_e V1 V2$$

$$V4 V3$$

$$RT3 \log_e$$

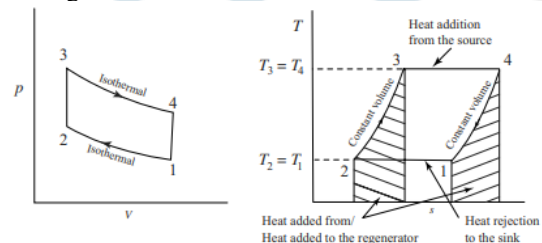


Figure 3: Representing the p-V and T-s of Stirling cycle [Ftp.Idu.Ac.Id]

But $V3 = V2$ and $V4 = V1$, therefore Sterling = $T3 T1 T3$, which is equivalent to Carnot efficiency. Earlier hot air engines employed the Sterling cycle, which was later replaced by the Otto and Diesel cycles. The design and installation of heat exchangers that can function continuously at extremely high temperatures is a significant challenge in the Sterling engine architecture. The Sterling engine has, however, staged a comeback in practical form thanks to advancements in metallurgy and intense research on this type of engine. The heat exchanger efficiency can never be 100% in real life. As a result, the Sterling cycle efficiency will be lower than the Carnot efficiency. This can be expressed mathematically as $= R (T3 T1) \log_e r RT3 \log_e r + (1) CV (T3 T1)$ where is the heat exchanger efficiency.

The Otto and Diesel cycles replaced the Sterling cycle, which had previously been employed for hot air engines. The Sterling engine's design and construction of the heat exchanger, which must function constantly at extremely high temperatures, presents significant challenges. However, the Sterling engine has made a resurgence in practical use because to advances in metallurgy and much research into this type of engine. The efficiency of the heat exchanger cannot be 100% in real life. The Sterling cycle efficiency can be expressed as $= R (T3$

$T_1 \log_e r + RT_3 \log_e r + (1 - CV) (T_3 - T_1)$ where is the heat exchanger effectiveness. As a result, it will be lower than Carnot efficiency.

Otto's Cycle

Since the Carnot cycle uses high pressure and high-volume ratios with comparatively low mean effective pressure, its fundamental disadvantage is that it is impractical. Modern spark-ignition engines are built on Nicolaus Otto's constant-volume heat addition cycle, which was first proposed in 1876. On the p-V and T-s diagrams in Figures 3 (a) and 3(b), respectively, the cycle is depicted. The processes 01 and 10 on the p-V diagram represent the suction and exhaust processes, respectively, and their effects are negated when the engine is operating at full power. When the piston goes from the bottom dead center to the top dead center, the process 1/2 depicts the isentropic compression of the air. A consistent volume of 23 heat is provided throughout the procedure. In a real engine, combustion and spark-ignition are equivalent processes. Isentropic expansion and constant volume heat rejection are represented by the processes 34 and 41, respectively.

The Diesel Engine

The maximum compression ratio in genuine spark-ignition engines is constrained by the fuel's self-ignition temperature. If air and fuel are compressed separately and then combined during combustion, this compression ratio restriction can be overcome. In such a setup, fuel can be injected into the cylinder that is filled with compressed air at a temperature greater than the fuel's self-ignition temperature. So, the gasoline doesn't need a particular equipment like an ignition system in a spark-ignition engine and starts burning on its own. These engines use heavy liquid fuels to operate. These engines are referred to as compression-ignition engines, and the Diesel cycle is the optimum cycle that they operate on. The method of heat addition is where Otto and Diesel cycles diverge. Unlike the Diesel cycle, where the heat addition occurs at constant pressure, the Otto cycle adds heat at a constant volume. The Diesel cycle is frequently referred to as the constant-pressure cycle for this reason. This phrase should be avoided as it causes misunderstanding with the Joules cycle. On the p-V and T-s diagrams in Figures 4, the Diesel cycle is depicted. As with the Otto cycle, the suction and exhaust strokes, denoted by 01 and 10, are disregarded while analyzing the diesel cycle. Here, the compression ratio, r , is the volume ratio V_1/V_2 . The cut-off ratio, abbreviated RC, is the volume ratio V_3/V_2 .

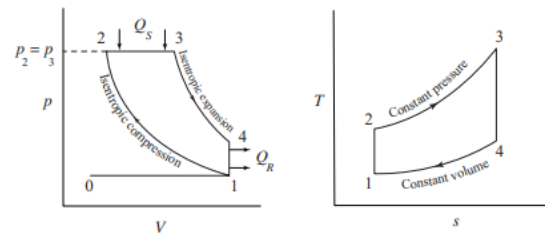


Figure 4: Representing the overview about Diesel cycle [Ftp.Idu.Ac.Id].

The Dual Cycle

Otto cycles assume constant volume combustion, whereas Diesel cycles assume constant pressure combustion. They are not real in reality. Since the chemical reactions that take place during the combustion process need time to complete, the combustion cannot occur at a constant volume. Similar to this, combustion does not take place at constant pressure in diesel engines because of the quick, uncontrolled combustion. Otto and Diesel cycles are reconciled in the Dual cycle, also known as a mixed cycle or limited pressure cycle. Depict the Dual cycle on p-V and T-s diagrams.

Otto, Diesel, and Dual Cycle Comparison

Compression ratio, peak pressure, heat addition, heat rejection, and network are major variable parameters that serve as the foundation for comparing the cycles. Some of the variable elements must be fixed in order to compare the performance of the Otto, Diesel, and Dual combustion cycles. The compression ratio, heat addition, maximum pressure and temperature, heat rejection, and network production for each of these three cycles are compared in this section. Which cycle is more effective under a specific set of operational conditions will be demonstrated by this investigation?

CONCLUSION

The ideal cycle for a heat engine is the Carnot cycle. Expansion and compression are two isothermal processes, and expansion and compression are two adiabatic processes. The engine's cylinder and piston are regarded as excellent heat-insulators however, the cylinder cover head is a capable heat conductor. The Otto cycle is a particular kind of air standard cycle that is regarded as the best cycle for the functioning of internal combustion spark ignition reciprocating engines. A portion of the heat is first delivered to the system in a dual cycle at constant volume, and the other portion is subsequently provided at constant pressure.

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Working Principle and Historical Design of IC Engine

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ABSTRACT: *In an internal combustion engine (ICE), the gasoline is ignited and burned inside the engine itself. The energy from the combustion is then partially converted into work by the engine. A stationary cylinder and a moving piston make up the engine. An internal combustion engine (ICE or IC engine) is a type of heat engine in which a fuel and an oxidizer, often air, burn together in a combustion chamber that is a vital component of the working fluid flow circuit. The expansion of the high-temperature, high-pressure gases produced during combustion in an internal combustion engine exerts direct force on a specific engine component.*

KEYWORDS: *Combustion Engines, Combustion Gases, Four Stroke, Internal Combustion, IC Engine.*

INTRODUCTION

More than 250 million highway transportation vehicles in the US rely on internal combustion engines because of their excellent durability and drivability. They can use renewable or alternative fuels, such as natural gas, propane, biodiesel, or ethanol, in addition to petrol or diesel [1], [2]. Additionally, they can be coupled with plug-in hybrid electric systems to expand the range of hybrid electric vehicles or hybrid electric powertrains to improve fuel efficiency. The fundamental chemical process of releasing energy from a fuel and air mixture is combustion, sometimes referred to as burning. In an internal combustion engine (ICE), the gasoline is ignited and burned inside the engine itself. The energy from the combustion is then partially converted into work by the engine. A stationary cylinder and a moving piston make up the engine. The piston is propelled by the expanding combustion gases, which turns the crankshaft. This motion ultimately propels the wheels of the car through the powertrain's gearing system [3], [4]. The spark ignition petrol engine and the compression ignition diesel engine are the two types of internal combustion engines currently in production. The majority of these engines have a four-stroke cycle, which requires four piston strokes to complete a cycle. The intake, compression, combustion and power stroke, and exhaust are the four separate processes that make up the cycle. Diesel engines with compression ignition and spark ignition use different fuel delivery and igniting systems. During the intake process of a spark ignition engine, the fuel and air are combined before being inducted into the cylinder. The spark ignites the fuel-air mixture after the piston compresses it,

resulting in combustion. During the power stroke, the piston is propelled by the expansion of the combustion gases. Only air is sucked into a diesel engine, where it is compressed. The gasoline is then sprayed into the heated, compressed air by diesel engines at an appropriate, calibrated rate, setting it ablaze [5], [6].

Advancements in Combustion Engines

To meet EPA emission limits, manufacturers have had to lower ICE emissions of criteria pollutants like nitrogen oxides (NO_x) and particulate matter (PM) by more than 99% during the past 30 years thanks to research and development. Additionally, research has improved ICE efficiency and performance horsepower and 0-60 mph acceleration time, assisting manufacturers in maintaining or improving fuel economy. Learn more about our advanced combustion engine research and development initiatives aimed at increasing the energy efficiency and reducing emissions from internal combustion engines [7], [8].

Uses

a car's revolving engine using a diesel generator as backup energy for land and water vehicles, such as cars, motorbikes, ships, and to a lesser extent locomotives most use diesel engines, some are electrical, but most are powered by reciprocating piston engines, reciprocating piston engines are by far the most prevalent form of propulsion. Some cars, planes, and motorbikes employ rotary engines of the Winkle design. Internal-combustion-engine vehicles (ICEV) refer to all of these. Internal combustion engines often take the form of combustion turbines or Winkle engines where high power-to-weight ratios are required. A conventional ICE, which could be a reciprocating engine, is used by powered aircraft. Instead of using jet engines,

which are forms of turbines, aero planes and helicopters can use turbo shafts. Airliners may use an additional ICE as an auxiliary power source in addition to propulsion. Many unmanned aerial vehicles have Winkle engines installed [9], [10]. Large electric generators that power electricity grids are propelled by ICEs. They are typically found in the form of combustion turbines, which have an average electrical output of about 100 MW. In combined cycle power plants, water steam is boiled and superheated using the high temperature exhaust to drive a steam turbine. Because more energy is recovered from the fuel than could be by the combustion engine alone, the efficiency is higher as a result. Energy efficiency for combined cycle power plants is between 50 and 60%. On a smaller scale, backup power or the provision of electricity to locations not connected to an electric grid is provided by stationary engines like petrol or diesel generators. Lawnmowers, string trimmers, chainsaws, leaf blowers, pressure washers, snowmobiles, jet skis, outboard motors, mopeds, and motorbikes frequently use small engines typically 2 stroke gasoline/petrol engines.

Internal Combustion Engine Types

The following are the main categories for internal combustion engines:

According to the quantity of strokes

1. Two-stroke.
2. Four-stroke.
3. Five-stroke.
4. Six-stroke engines.

Fuel Used in Petrol and Diesel Engines

1. Dual-fuel engine.
2. Operating Cycle.
3. Otto Cycle.
4. Diesel Cycle.
5. Dual Cycle.

Techniques for Cooling

1. Water-cooled engine.
2. Air-cooled engine.
3. Reciprocating engine design.

Winkle Engine Field of Application

1. Portable Aero Engine.
2. Vehicle Engine Stationary Engine.
3. Engine Configuration.
4. Spark-ignition Engine.
5. Compression-Ignition Engine.
6. Engine, Cylinder.

DISCUSSION

The combustion of a fuel takes place with the help of an oxidizer often air in a combustion chamber that is a crucial component of the working fluid flow

circuit in an internal combustion engine (ICE or IC engine). In an internal combustion engine, a component is subjected to direct force as a result of the expansion of the high-temperature and high-pressure gases produced during combustion. Typically, the force is applied to a rotor Winkle engine, a piston, turbine blades gas turbine, or a nozzle jet engine. The component is propelled across a distance by this force, which converts chemical energy into kinetic energy that is then utilised to move or power whatever the engine is connected to. Étienne Lenoir developed the first internal combustion engine that was a commercial success around 1860, and Nicolaus Otto developed the Otto engine, the first modern internal combustion engine, in 1876. Typically, when you hear the word "internal combustion engine, you're thinking of a machine with intermittent combustion, like the more well-known two- and four-stroke piston engines, as well as variations like the six-stroke piston engine and the Winkle rotary engine. Continuous combustion is used by a second class of internal combustion engines, including gas turbines, jet engines, and the majority of rocket engines, all of which operate on the same fundamental design as the one just mentioned.

Although they are of a sort so specialized that they are frequently considered as a different category, weapons like mortars and anti-aircraft cannons are also a type of internal combustion engine. In contrast, energy is given to a working fluid that does not contain, is mixed with, or is contaminated by combustion products in external combustion engines like steam or Sterling engines. External combustion engines can run on air, hot water, pressurized water, or even sodium-based liquid that has been heated in a boiler. Despite having several stationary uses, ICEs are mostly used in mobile applications and serve as the main power source for cars, boats, and other moving objects. Hydrocarbon-based fuels like ethanol, natural gas, petrol, or diesel fuel are frequently used to power ICEs. Compression ignition (CI) engines use biodiesel as a fuel, while spark ignition (SI) engines use bioethanol or ETBE (ethyl tert-butyl ether), which is made from bioethanol. Rudolf Diesel, the creator of the diesel engine, was using peanut oil to power his machines as early as 1900. Frequently, fossil fuels and renewable energy are combined. Rarely used hydrogen can be produced using fossil fuels or clean energy sources.

4-Stroke Engine

An illustration of a 4-stroke SI engine in action.
Labels:

1. Induction.
2. Compression.
3. Power Four.

Exhaust: A piston's position closest to the valves is called its top dead center (TDC), while its position furthest from them is called its bottom dead center (BDC). A piston's travel from TDC to BDC or the opposite, along with the related process, is known as a stroke. The crankshaft revolves continuously and at a fairly constant speed when an engine is running. Each piston in a 4-stroke ICE experiences 2 strokes each crankshaft revolution in the manner described below. The following are listed from TDC forward.

The Intake Valves: are open because the cam lobe is pressing down on the valve stem, also known as induction or suction (Figure. 1). In CI engines, the piston moves lower, expanding the combustion chamber's capacity and allowing air to enter in SI engines without direct injection, this allows an air-fuel mixture to enter. In any case, the charge refers to the mixture of air and fuel.

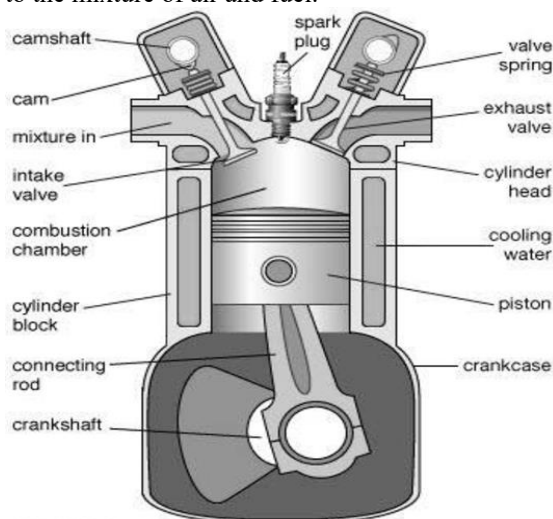


Figure 1: Representing the 4-Stroke IC Engine instruments and their components [Research Gate].

Compression: During this stroke, the piston goes upward while both valves are closed, reducing the capacity of the combustion chamber, which is at its smallest when the piston is at top dead center (TDC). As the charge is compressed, the piston exerts force on it, increasing its pressure, temperature, and density; the ideal gas law provides an approximation to this behavior. Ignition starts just before the piston reaches TDC. A high voltage pulse is delivered to the spark plug in a SI engine, which causes it to produce the spark that lends it its name and ignites the charge. The fuel injector swiftly sprays fuel into the combustion chamber in the case of a CI engine; the fuel ignites as a result of the high temperature.

Power or working stroke: As the combustion gases press against the piston, more kinetic energy is produced than is needed to compress the charge. The combustion gases expand in response to the compression stroke, which reduces temperature, pressure, and density. The exhaust valve opens as the piston draws close to BDC. Due to the remaining pressure—more than back pressure, or the gauge pressure on the exhaust port—during the blow down, the combustion gases expand permanently.

Exhaust: As the piston rises, the exhaust valve stays open, releasing the combustion gases. Because the piston does not entirely seal the combustion chamber during normal operation, a tiny portion of the combustion gases may still be present in the cylinder for naturally aspirated engines. These gases dissolve in the subsequent charge. The exhaust valve closes at the end of this stroke, the intake valve opens, and the process is repeated in the following cycle. For greater scavenging, the intake valve may open before the exhaust valve closes.

4-Stroke Engine Application

A typical kind of engine utilised in a variety of applications is the four-stroke engine, usually referred to as the internal combustion engine.

1. **Automobiles:** The majority of today's automobiles on the road are powered by four-stroke engines. Cars, trucks, motorbikes, and other motorized vehicles all use them. The engine transforms the energy released during fuel combustion into mechanical power to advance the vehicle.
2. **Power Generation:** Generators that produce electricity use four-stroke engines. These generators can be employed in a variety of locations, including residences, workplaces, building sites, and outlying regions with insufficient connectivity to the power grid.
3. **Marine Propulsion:** Four-stroke engines are frequently used to propel boats and ships. These engines supply the force required to propel the ship through the water. Both smaller boats used for enjoyment and bigger ships like cruise liners and cargo ships employ them.
4. **Lawn & Garden Equipment:** Lawnmowers, garden tractors, leaf blowers, chainsaws, and other outdoor power equipment frequently use four-stroke engines. They supply the energy required to complete jobs like cutting the grass, trimming the hedges, and picking up rubbish.
5. **Agricultural Machinery:** Tractors, harvesters, and irrigation pumps are just a few examples of the agricultural equipment that heavily utilizes four-stroke engines. They

supply the energy required for tillage, crop planting, harvesting, and other agricultural tasks.

6. **Construction Equipment:** Several pieces of construction equipment, such as excavators, bulldozers, loaders, and concrete mixers, are powered by four-stroke engines. The power and torque needed to complete heavy-duty operations in the construction sector are provided by these engines. Four-stroke engines are used in portable equipment such as pressure washers, portable generators, and compact construction tools. When mobility is crucial, these engines provide a portable and dependable power supply. Four-stroke engines are suitable for a variety of applications across many industries due to their adaptability, effectiveness, and dependability.

4-Stroke Engine's User Benefits

Compared to other engine types, the four-stroke engine has a number of benefits. Some of the main benefits are as follows:

1. **Efficiency:** Compared to two-stroke engines, four-stroke engines typically use less gasoline. They can better regulate the combustion process and have better fuel efficiency since they have separate strokes for the intake, compression, power, and exhaust.
2. **Environmental Friendliness:** When compared to two-stroke engines, four-stroke engines emit less exhaust fumes. They feature an exhaust stroke specifically designed for this purpose, which improves combustion and lessens the amount of unburned fuel and pollutants released into the environment.
3. **Longevity and Durability:** Four-stroke engines are renowned for their longevity and durability. They have a more straightforward construction with fewer moving components, which lowers the possibility of mechanical breakdowns and boosts overall reliability. They are therefore appropriate for demanding uses like industrial and automotive use.
4. **Low Noise and Vibration:** Four-stroke engines produce less noise and vibration because they rotate at fewer RPMs (revolutions per minute) than two-stroke engines. As a result, they are more suited for uses such those in residential areas and passenger vehicles where noise reduction is crucial. The ability to run on a variety of fuels, including petrol, diesel, and alternative fuels like natural gas and biofuels, makes four-stroke engines extremely

versatile. Because of their adaptability, they can be used in a wide variety of sectors.

5. **Maintenance:** Compared to two-stroke engines, four-stroke engines typically require less frequent maintenance. Four-stroke engines' independent lubrication systems avoid the need to mix oil and fuel, making maintenance easier. Four-stroke engines can generate a lot of torque and power, which makes them excellent for heavy-duty applications like those used in automobiles, construction equipment, and power generators.

Historical Design

In 1879, Donald Clerk created the first two-cycle engine. It utilised a different cylinder that served as a pump to deliver the fuel mixture to the cylinder. John Day streamlined Clerk's design in 1899, creating the kind of 2-cycle engine that is still quite popular today. Crankcase scavenging and port timing are done on day cycle engines. A pump is made from of the crankcase and the portion of the cylinder that is below the exhaust port. When the crankshaft is cranked, the piston travels from BDC upward, starting the Day cycle engine. This creates a vacuum in the crankcase or cylinder area. The fuel mixture is then fed from the carburetor into the crankcase via a rotary disc valve or a reed valve that is powered by the engine. Cast-in ducts connect the crankcase to the cylinder's intake port and the exhaust port to the exhaust pipe, respectively. The term port timing refers to the ratio of the port's height to the cylinder's length. Since the crankcase was empty, no fuel would be introduced into the cylinder during the initial upstroke of the engine. On the down stroke, the piston compresses the fuel mixture that, thanks to the addition of oil to the fuel mixture, has lubricated the piston in the cylinder and the bearings.

Although there is no consumed fuel to exhaust on the initial stroke, the exhaust is first exposed as the piston descends. The intake port, which has a duct leading to the crankcase, is revealed as the piston descends farther. Because of the pressure in the crankcase, the fuel mixture travels through the duct and into the cylinder. Early engines employed a high domed piston to delay the flow of gasoline since there is no impediment in the cylinder to prevent the fuel from moving right out of the exhaust port before the piston rises far enough to close the port. With the aid of an expansion chamber design, the fuel was afterwards resonated back into the cylinder. A spark ignited the fuel as the piston ascended almost to the top dead center. The exhaust port, where the burned fuel is discharged under high pressure, and the intake

port, where the operation has been finished and will be repeated, are initially revealed as the piston is propelled downward with force. Later engines employed performance-enhancing porting developed by the Deutz business. The system's name was Schuler Reverse Flow. All of DKW's motorcycles use this design under license. As a result, their DKW RT 125 was among the first automobiles to attain over 100 mpg.

Benefits and Drawbacks of an IC Engine

1. Internal combustion engines have advantages.
2. When compared to EC engines, these engines are lighter.
3. They are diminutive in stature.
4. They get going right away.
5. Compared to external combustion engines, they are less expensive.
6. The IC engines are user-friendly and secure.
7. Internal combustion engines are portable and small in size. These engines are perfect for usage in mobile machinery and vehicles due to their mobility features.
8. They can use a variety of fuels, including biofuels, kerosene, diesel, and natural gas.

Internal Combustion Engine Drawbacks

1. Fuels for IC engines, like petrol or diesel, are expensive.
2. Compared to EC engines, they have higher emission rates.
3. They are not the best for producing a lot of electricity.
4. Regular maintenance for this kind of engine is necessary, including oil changes, filter replacements, fuel changes, and other upkeep.
5. For them to operate effectively, sufficient cooling and lubrication are necessary.
6. Only 20% to 30% of the energy from the fuel is converted into productive work by internal combustion engines, which have low efficiency. Heat is released from the rest of the energy.
7. Because IC engines are intricate machines with several rotating parts and high levels of precision engineering, they can be more expensive to produce and operate than other types of engines.

CONCLUSION

Any of a series of devices known as internal combustion engines use fuel and oxidizer as the reactants of combustion, respectively, and the products of combustion as the working fluids. In such an engine, the heat produced during the

combustion of the oxidizer-fuel mixture, the nonrelated working fluids, serves as the engine's energy source. As a part of the engine's thermodynamic cycle, this process takes place inside the machine. Internal combustion (IC) engines produce useful work by acting on moving engine surfaces like the face of a piston, a turbine blade, or a nozzle with hot gaseous products of combustion.

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Fuel–Air Cycles and Their Analysis

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ABSTRACT: *The working fluid in a real engine is a combination of air, fuel vapor, and leftover gases from the previous cycle. Furthermore, the working fluid's specific heats are temperature-dependent and not constant. Finally, at high temperatures, certain dissociation of the combustion products is subjected. It is possible to estimate pressures and temperatures that are relatively near to the actual pressures and temperatures present inside the engine cylinder by taking into account the actual physical properties of the gases in the cylinder before and after the combustion.*

KEYWORDS: *Air Cycle, Air Fuel, Cycle Analysis, Maximum Temperature, Mixture Strength*

INTRODUCTION

The fuel-air cycle is a hypothetical cycle based on the real qualities of the cylinder's contents. The following factors are taken into account by the fuel-air cycle. One is the actual make-up of the cylinder's contents. The fluctuation in the cylinder's gases' specific heat. The consequence of separation. How the cylinder's mole concentration varies as temperature and pressure do. There are no chemical alterations in the fuel or air before burning. Top dead center is the instantaneous location of combustion. Every process is adiabatic. The air and gasoline are properly blended. The working fluid in a real engine is a combination of air, fuel vapor, and leftover gases from the previous cycle. Furthermore, the working fluid's specific heats are temperature-dependent and not constant. Finally, at high temperatures, certain dissociation of the combustion products is subjected [1], [2].

It is possible to estimate pressures and temperatures that are relatively near to the actual pressures and temperatures present inside the engine cylinder by taking into account the actual physical properties of the gases in the cylinder before and after the combustion. In the case of well-designed engines, the mean effective pressures and efficiencies predicted by this approach differ from the measured values by just a few percent. The fuel-air cycle analysis is based on the real characteristics of the working medium, namely fuel and air, and even this study makes simplifying assumptions. Nevertheless, they are more reasonable and accurate than the conditions utilised in the air-standard cycle study [3], [4].

Fuel-Air Cycles and Their Impact

By using air-standard cycle analysis, it is clear how raising the compression ratio increases efficiency. However, because the working medium was believed to be air, analysis was unable to reveal the impact of the air-fuel ratio on the thermal efficiency.

In this chapter, the existence of fuel in the cylinder is considered, and as a result, a mixture of fuel and air will be used as the working medium. By analyzing the fuel-air cycle, it will be feasible to show how the fuel-air ratio affects thermal efficiency and investigate how the peak pressures and temperatures during the cycle change depending on the fuel-air ratio. In general, it may be easier to comprehend the effects of numerous engine operating factors on the pressures and temperatures inside the engine cylinder if the analysis of fuel-air cycle [5], [6]. The following factors are considered in the fuel-air cycle analysis:

- i. The gases in the cylinders' actual composition Fuel, air, water vapor, and leftover gas are all present in the cylinder gases. During engine operation, the fuel-air ratio fluctuates, which affects the relative volumes of CO₂, water vapor, etc.
- ii. The relationship between temperature and the specific heat. With the exception of monoatomic gases, specific heats rise with temperature. As a result, temperature also affects the value of.
- iii. At high temperatures (over 1600 K), the chemical reaction between the fuel and the air is incomplete, and as a result, CO, H₂, and O₂ are present in equilibrium circumstances.
- iv. The variation in molecule counts. The ratio of fuel to air, as well as the post-combustion pressure and temperature, all affect how many molecules are present.

Along with the above-mentioned elements, the following presumptions are frequently used:

- i. Prior to combustion, neither the fuel nor the air undergoes any chemical changes.
- ii. The charge is always in chemical equilibrium after burning.
- iii. All processes are adiabatic, meaning there is no heat transfer between the gases and the

cylinder walls. The processes of compression and expansion are also frictionless.

- iv. It is thought that fluid motion inside the cylinder can be disregarded in the case of reciprocating engines. It is also assumed that
- v. The burning occurs instantly at top dead center at constant volume.
- vi. The fuel is totally vaporized and perfectly mixed with the air, with specific reference to the constant volume fuel-air cycle [7], [8].

As was already indicated, the fuel-air cycle analysis reveals the impact of variations in fuel-air ratio, inlet pressure, and temperature on engine performance as opposed to the air-standard cycle analysis, which demonstrates the general effect of just compression ratio on engine efficiency. You'll see that the compression ratio and fuel-air ratio are crucial engine factors, whereas inlet conditions are not as crucial. A good engine operates at approximately 85% of its predicted fuel-air cycle efficiency. Fuel-air cycle analysis can provide a reliable estimate of the power that can be anticipated from the real engine. Additionally, peak pressures and exhaust temperatures that have an impact on the construction and design of the engine can be predicted rather well. Thus, fuel-air cycle analysis helps to better understand how many factors affect an engine's performance.

DISCUSSION

The Makeup of Engine Gases

Throughout the engine's functioning, the air-fuel ratio varies. The composition of the gases both before and after combustion is impacted by this shift in the air-fuel ratio, specifically the amount of carbon dioxide, carbon monoxide, water vapor, etc. in the exhaust gases. In four-stroke engines, the burnt gases remaining in the clearance gap from the previous cycle come into contact with the fresh charge as it enters the engine cylinder. Depending on the engine load and speed, there are different amounts of exhaust gases in the clearance gap. This information is taken into consideration in the fuel-air cycle analysis, and the results are computed to prepare the combustion charts. However, thanks to the availability of quick digital computers, it is now possible to use the right numerical approaches to analyses how the composition of the cylinder gas affects the engine's performance. Results from the computer analysis can be produced quickly and precisely. As a result, computer analysis of the fuel-air cycle is easier to perform than human calculation.

Variable Specific Heats

With the exception of monoatomic gases, all gases exhibit a rise in specific heat with temperature. There is no exact law that governs how specific heat increases. However, the specific heat curve is almost a straight line over the typical temperature range for gases in heat engines (300 K to 2000 K), and it can be roughly stated in the form $C_p = a_1 + k_1T$.

$$CV = b_1 + k_1T$$

A_1 , B_1 , and K_1 are constants in equation.

Now, where R is the typical gas constant, we have $R = C_p - CV = a_1 - b_1$.

The specific heat increases significantly more quickly at 1500 K and can be written as $C_p = a_1 + k_1T + k_2T^2$ (3.3).

$$CV = (3.4) b_1 + k_1T + k_2T^2.$$

If the phrase T^2 is ignored, the result is the same as. C_p and CV can be calculated using a variety of expressions up to the sixth order of T (T^6). As the temperature rises, bigger fractions of heat would be needed to induce motion of the atoms within the molecules, which is the basic explanation for the rise in specific heat. Since the motion of the molecules as a whole determines temperature, the energy required to move the atoms does not cause a proportional increase in temperature. As a result, more energy is needed to raise the temperature of a unit mass by one degree at higher altitudes. By definition, this heat is the specific heat. Typically, the values for C_p and CV for air are taken as C_p equals 1.05 kJ/kg K at 300 K. At 300 K, CV equals 0.71 kJ/kg K.

C_p equals 1.34 kJ/kg K at 2000 K. 1.057 kJ/kg K for CV at 2000 K

The value of decreases with an increase in temperature since C_p and CV 's difference is constant. As a result, the final temperature and pressure would be lower than they would be if the specific heat variation was ignored during the compression stroke.

Dissociation

The dissociation process can be thought of as the high-temperature disintegration of combustion products. Dissociation can be viewed as the opposite of combustion. Heat is absorbed during dissociation, whereas it is released during combustion. While there is relatively little dissociation of H_2O in IC engines, CO_2 is primarily split into CO and O_2 . Around 1000 degrees Celsius, CO_2 begins to dissociate into CO and O_2 , and the reaction equation is $CO_2 \rightarrow CO + O_2$. Similar to this, H_2O dissociates at temperatures above 1300 °C and is represented by the formula $H_2O \rightarrow H_2 + O_2$.

A rich fuel mixture, which suppresses CO₂ dissociation by creating more CO, is an example of how the presence of CO and O₂ in the gases tends to hinder the dissociation of CO₂. On the other hand, a low fuel-air mixture's consumed gases do not dissociate. This is mostly because the temperature produced is insufficient for this phenomenon to occur. As a result, dissociation increases to its maximum extent in the burnt gases of the chemically right fuel-air mixture when high temperatures are anticipated, but decreases in leaner and richer mixtures.

Internal combustion engines' maximum temperature and pressure are reduced as a result of heat transfer to the cooling medium. The separated elements rejoin as the temperature drops during the expansion stroke, releasing the heat that was trapped during dissociation once more. However, it is too late in the stroke for the lost power to be fully recovered. The exhaust gases carry away a portion of this heat. Displays a typical curve that depicts the drop in exhaust gas mixture temperature brought on by dissociation with respect to air-fuel ratio. Chemically speaking, the maximum temperature is reached without dissociation. Proper fuel to air ratio. Dissociation occurs at its highest temperature when the combination is just a little bit rich. Even at the chemically right air-fuel ratio, dissociation lowers the maximum temperature by around 300 C. Lean mixtures and rich mixtures are distinguished easily in Figure 1.

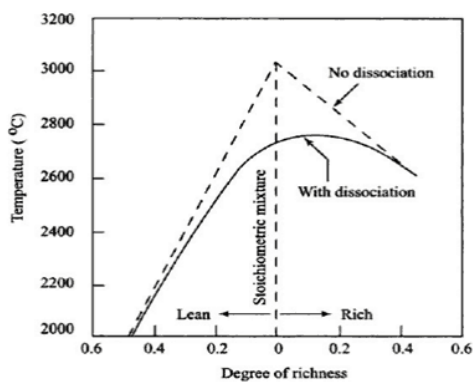


Figure 1: Representing the Effect of dissociation on temperature [Learn Match].

Dissociation is essentially the high-temperature breakdown of combustion products. Dissociation can be thought of as combustion's opposite process. Heat is absorbed during dissociation, whereas it is released during combustion. While there is relatively little H₂O dissociation in IC engines, CO₂ is primarily broken down into CO and O₂.

Around 1000 oC, CO₂ begins to dissociate into CO and O₂, and the reaction equation can be written as $CO_2=2CO+O_2+Heat$. $CO_2=2CO+O_2+Heat$ Similar to this, H₂O dissociates at temperatures higher than 1300°C, and the formula for this reaction is



A rich fuel mixture, which suppresses dissociation of CO₂ by producing more CO, is an example of how the presence of CO and O₂ in the gases tends to inhibit it. In the case of ICE, a decrease in the maximum temperature and pressure results from heat transfer to the cooling medium. The separated elements rejoin as the temperature drops during the expansion stroke, releasing the heat that was trapped during dissociation once more. However, it is too late in the stroke for the lost power to be fully recovered. The exhaust gases carry away a portion of this heat. The temperature of the exhaust gas mixes is reduced due to dissociation with regard to the A/F ratio, as shown by a typical curve in Fig. At the chemically ideal air-fuel ratio, the maximal temperature is reached without dissociation. Dissociation occurs at its highest temperature when the combination is just a little bit rich. Even at the chemically ideal A/F ratio, dissociation lowers the maximum temperature by around 300 °C. Rich mixtures and lean mixtures are clearly marked in the Figure. 1.

Figure illustrates how dissociation affects output power for a conventional four-stroke spark-ignition engine running at constant speed. When the mixture ratio is stoichiometric, the brake power output is greatest if there is no dissociation. The power loss resulting from dissociation is depicted by the shaded region between the brake power graphs. There is no dissociation when the combination is very lean. The maximum temperature rises and dissociation starts as the A/F ratio falls, or as the mixture gets richer. When the mixture strength is chemically correct, the greatest dissociation takes place. Due to incomplete combustion, the dissociation effect tends to diminish as the mixture gets richer. In a CI engine compared to a SI engine, dissociation effects are not as noticeable. This is mostly because

- i. The presence of an uneven mixture.
- ii. Extra air to guarantee full combustion.

Depicts the impact of dissociation on output power for a typical four-stroke, spark-ignition engine running at constant speed. When the mixture ratio is stoichiometric, the brake power output is greatest if there is no dissociation. The power loss resulting from dissociation can be seen in the shaded region between the brake power graphs. There is no dissociation when the combination is very lean. The

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- i. Heterogeneous mixture is present.
- ii. There is extra air to ensure complete burning.

Both of these elements work to lower the CI engine's peak gas temperature. The impact of dissociation on the p-V diagram of the Otto cycle. Dissociation causes the maximum temperature to drop, which also lowers the maximum pressure, which causes the state following combustion to be represented by 3 rather than 3. If there was no association as a result of the temperature falling during expansion, the process would be represented by 34; however, association causes the expansion to proceed down the path 34. Air-Standard and Fuel-Air Cycles Are Compared. This section discusses the causes of the differences between fuel-air cycles and air-standard cycles. The following elements contributed to the size of the difference between the two cycles:

- i. The cycle's nature (owing to presumptions).
- ii. The equivalency ratio or real F/A stoichiometric F/A.
- iii. The fuel's chemical make-up.

As the mixture gets leaner, efficiency increases as seen in Figure 2, which also compares the fuel-air cycle's mixture strength to the air cycles. As the combination gets thinner and leaner, heading towards the air-standard cycle efficiency, the efficiency ratio fuel-air cycle efficiency/air-standard cycle efficiency rises. It should be noted that this trend is present across the board for compression ratios. The mixture would tend to behave like a perfect petrol with constant specific heat at very low fuel-to-air ratios. Lean to very lean mixture cycles have a tendency to be air-standard cycles. Pressure and temperature increase during these cycles. As the pressure rises, several of the chemical reactions involved tend to be more complete. Both cycles with constant volume and constant pressure must take these factors into account. Since air is taken to be the working medium in the simple air-standard cycle analysis, the variation of thermal efficiency with mixture strength cannot be predicted. However, fuel-air cycle research predicts that as an engine's fuel mixture is enriched, its thermal efficiency would decrease.

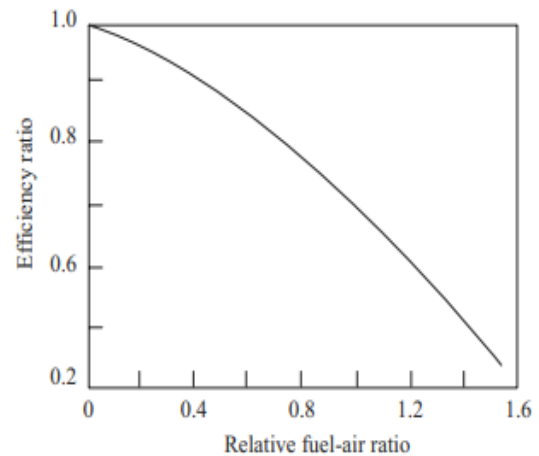


Figure 2: Relative fuel-to-air ratio impact on efficiency ratio [Ftp.Idu.Ac.Id].

The rising losses resulting from dissociation and varying specific temperatures as the mixture strength approaches chemically correct values can be used to explain this. This is due to the fact that as the mixture strength approaches chemically correct values after combustion, the gas temperature increases (Figure 3). Over-enrichment will result in incomplete combustion and a reduction in thermal efficiency. Therefore, it would seem that as the combination gets leaner, thermal efficiency would rise. However, the combustion becomes irregular and loses efficiency at a certain tilt. Therefore, the lean zone is very close to the stoichiometric ratio and contains the highest efficiency. This results in the combustion loop that may be displayed for various mixture strengths for an engine running at constant speed and at a constant throttle setting. This loop provides information on the impact of mixture strength on a given fuel's consumption [9], [10].

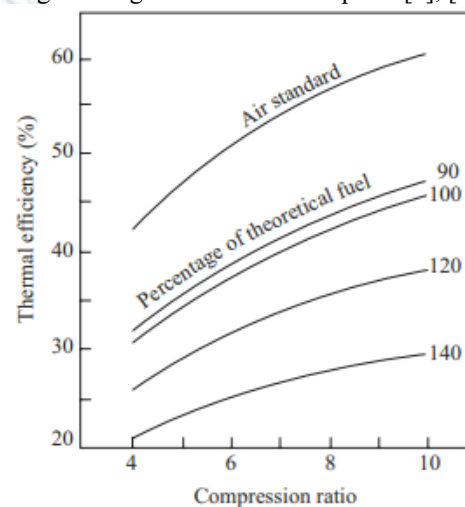


Figure 3: Effect of compression ratio and mixture strength on efficiency [Ftp.Idu.Ac.Id].

Effect of Operating Variables

Fuel-air cycle analysis helps to better understand how common engine operating factors affect the pressure and temperature inside the engine cylinder. The next parts go into the specifics. Compression Ratio 3.8.1: The efficiency of the fuel-air cycle rises with compression ratio in a similar way to the efficiency of the air-standard cycle, mostly due to the same factor a wider range of expansion work. The fluctuation of stated thermal efficiency relative to the equivalence ratio for various compression ratios. The actual fuel-air ratio to the chemically correct fuel-air ratio is known as the equivalency ratio, or.

CONCLUSION

The Rankin cycle is based on a working fluid that undergoes phase change throughout the cycle, with the heat rejection and addition zones occurring at constant temperature. The most accessible working fluid, however, is air, which is a superheated gas under typical operating circumstances. As a result, the energy is received and rejected in a series of cycles with varying temperatures. These cycles can be used to evaluate the performance of gas turbines and internal combustion engines like petrol and diesel engines. The fact that mass flows across boundaries when air and fuel enter the engines and exhaust gases exit means that internal combustion engines and gas turbines are not heat engines.

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Actual Cycles and Their Analysis

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ABSTRACT: Reciprocating internal combustion engines run on actual engine cycles, which are primarily split into two groups Spark Ignition engines and Compression Ignition engines. These classifications depend on how the air/fuel mixture charge is ignited in the cylinder. While compression ignition engines mostly follow the Diesel cycle principle, spark ignition engines work on the Otto cycle principle. A reciprocating engine may need to execute a mechanical task in two or four strokes of engine piston action, which correspond to one or two crankshaft revolutions, respectively. A reciprocating internal combustion engine's piston can be either a double-acting piston like those found in steam engines or a single-acting piston like those found in the majority of these engines.

KEYWORDS: Fresh Charge, Heat Loss, IC Engine, Ignition Engines, Loss Factor.

INTRODUCTION

In many ways, the actual cycles for IC engines are different from the fuel-air cycles and the air-standard cycles. Due to numerous losses that occur when the engine is really operating, the actual cycle efficiency is significantly lower than the air-standard efficiency. The main losses result from Specific heats changing with temperature. The combustion products' dissociation and Progressive combustion. Incomplete fuel combustion. Heat transfer into the combustion chamber's walls. Blow down at the conclusion of the exhaust process. From prior knowledge and a few straightforward engine tests, it is possible to estimate these losses, and these estimates can be applied to gauge an engine's performance [1], [2].

Analysis of Actual and Air-Standard Cycles

Internal combustion engine actual cycles diverge significantly from air-standard cycles in numerous ways. These variations are mostly caused by:

- i. The working substance being a mixture of air and fuel vapor or liquid fuel that has been finely atomized in air along with the combustion byproducts left over from the previous cycle.
- ii. The alteration in the chemical makeup of the active ingredient.
- iii. The relationship between temperature and specific heats.
- iv. The modification of the fresh charge's composition, temperature, and actual quantity brought on by the leftover gases.
- v. The gradual combustion as opposed to the immediate combustion.
- vi. The heat exchange between the working medium and external sources.

- vii. The significant exhaust blows down loss, or loss of work during the expansion stroke as a result of the exhaust valve opening too soon.
- viii. Actual engine problems like gas leaking and fluid friction.
- ix. The fuel-air cycle points i to iv. The remaining points, v through viii, are in fact what distinguish fuel-air cycles from true cycles.

The majority of the aforementioned elements work to reduce the thermal efficiency and output of the actual engines [3], [4]. The examination of the cycles, however, shows that the anticipated thermal efficiencies are not significantly different from those of the actual cycles when these parameters are taken into consideration. Out of all the aforementioned components. Time loss factor, or loss due to the time needed for combustion and the mixing of fuel and air, has the most significant influence. The heat loss factor, or the heat transferred from gases to cylinder walls. The exhaust blows down factor, which is the work lost during the expansion stroke as a result of the exhaust valve opening too soon. The next parts explore these significant losses that were not covered in the first two chapters [5], [6].

Time Loss Factor

In air-standard cycles, the addition of heat is thought to happen instantly, although in a real cycle, it happens gradually over time. The duration of the combustion is such that, in every case, some volume change occurs while it is taking place. Between the spark's ignition and combustion's completion, the crankshaft typically rotates between 30 and 40 degrees. During this time, there will be a time loss, which is referred to as the time loss factor. The result of combustion having a finite period is that the peak pressure won't occur when the volume is lowest, or when the piston is at TDC, but rather at some point

after T DC. As a result, the pressure increases from b to c during the first portion of the working stroke. If the combustion had been immediate and additional work equivalent to the area depicted hatching had been required, the state of the gases is represented by point 3. This loss of work, often known as time loss due to progressive combustion or just time losses, lowers efficiency. The amount of time it takes for a fuel to burn relies on the flame velocity, which in turn depends on the fuel type, the fuel-air ratio, the form of the combustion chamber, and its size. Additionally, the distance from the point of ignition to the other side of the combustion area is crucial [7], [8].

DISCUSSION

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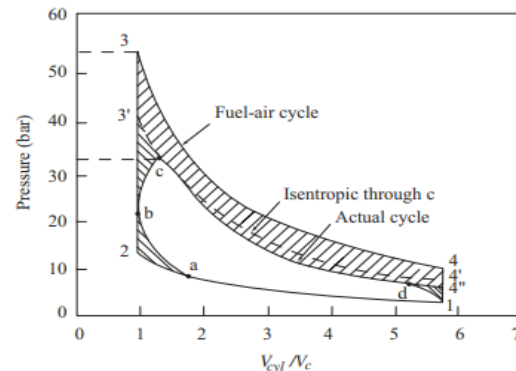


Figure 1: Representing the effect of time losses shown on p-V diagram [Ftp.Idu.Ac.Id].

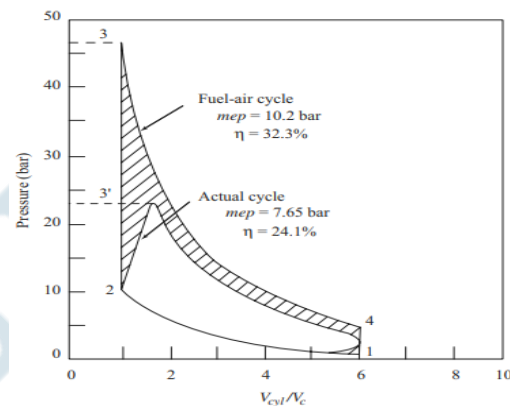


Figure 2: Representing the Spark at T DC, advance 0 [Ftp.Idu.Ac.Id].

A direct loss has occurred here. The work area is smaller and the power output and efficiency are decreased in any scenario, i.e., with or without spark advance. Since there are minimal losses on both the compression and expansion strokes, a moderate or optimal spark advance Figure 4 is the ideal compromise. Using a p-V diagram, Figure 5 illustrates the impact of spark advance on power output. As can be observed from Figure 6, the imp is drastically reduced as the ignition advance is increased, which results in a loss of power. In actual practice, it may occasionally be required to deliberately retard the spark from its ideal position in order to prevent knocking and simultaneously lower the exhaust emissions of hydrocarbons and carbon monoxide.

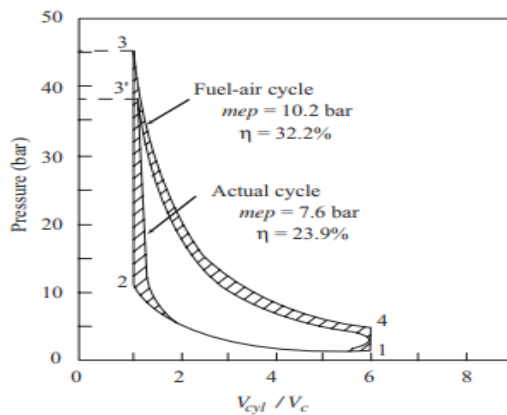


Figure 3: Representing the Combustion is finished at T DC and progress is 35 [Ftp.Idu.Ac.Id].

The time losses may be responsible for a drop in efficiency of about 5% at full throttle when the fuel-air ratio corresponds to maximum power and the best ignition advance the fuel-air cycle efficiency is decreased by about 2%. When the ignition advance is not optimal, the mixture is richer or leaner, and part throttle operations are used, the losses are higher. Due to the presence of leftover gases from the previous cycle in the cylinder's clearance volume, a completely homogenous combination of air and fuel vapor is not achievable. Additionally, there isn't much time between the production of the mixture and ignition. In these conditions, it is possible that the cylinder contains pockets of surplus fuel in one area and excess oxygen in another.

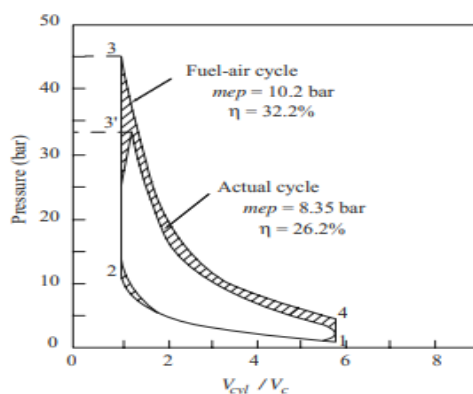


Figure 4: Representing the optimal advance 15 to 30 Degree [Ftp.Idu.Ac.Id Ic].

As a result, some fuel does not burn completely or only partially, turning to CO, leaving some fuel unused, as illustrated in Figure 7. Energy release data reveals that only around 95% of the energy is released with stoichiometric fuel-air ratios. 90% of the energy input from the gasoline is actually released in the engine. It should be emphasized that while a rich mixture is needed to use all the oxygen, a lean mixture is needed to prevent fuel waste.

Maximum efficiency would be achieved with a little leaner mixture, but a mixture that is too lean will burn slowly, increasing time losses, or it won't burn at all, wasting all of the fuel. A portion of the fuel will not receive the required oxygen in a rich mixture and will be completely lost. Additionally, the flame speed is low in combinations that are greater than 10% richer, which reduces efficiency and raises time losses. The energy released at such a late time cannot be exploited, even if the unused fuel and oxygen finally combine and burn during the exhaust stroke. Different fuel-air ratios during the suction stroke or constantly weaker mixes in some cylinders of a multicylinder engine may result from improper fuel and air mixing.

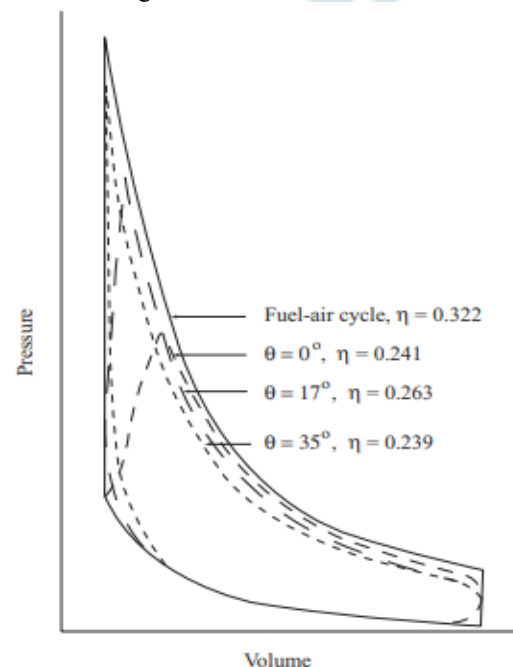


Figure 5: Power loss caused by ignition advance is depicted's p-V diagram [Ftp.idu.ac.id ic].

Heat Loss Factor

Heat from the cylinder gases passes through the cylinder walls and cylinder head and into the water jacket or cooling fins during combustion and the ensuing expansion stroke. A small amount of heat enters the piston head, passes through the piston rings, and then escapes through the cylinder wall or is absorbed by the engine lubricating oil that splashes on the piston's underside. In Fig.4.8, the p-V diagram, along with other losses, displays the heat loss. The cycle efficiency will naturally be most affected by heat loss during combustion, whereas heat loss right before the conclusion of the expansion stroke can have very little impact because it contributes very little useful work. Even under the

most ideal conditions anticipated for an air-standard cycle, only a portion of the heat lost during combustion could be turned into work equal to Q the, and the remainder would be rejected during the exhaust stroke. As a result, the heat lost during combustion does not represent a total loss. In the process of combustion and expansion, about 15% of the total heat is lost.

But a lot of it is lost too late in the cycle to have made a difference. Only approximately 20% of the heat loss could be perceived as useful work if it is completely recovered. In a Cooperative Fuel Research CFR engine, time loss, heat loss, and exhaust loss are all depicted as percentages in Figure 7. Losses are expressed as a % of the effort in the fuel-air cycle. The consequence of heat loss during combustion is to lower the maximum temperature, leading to lower specific heats. The heat loss factor accounts for around 12% of the total losses, as shown in Figure 7.

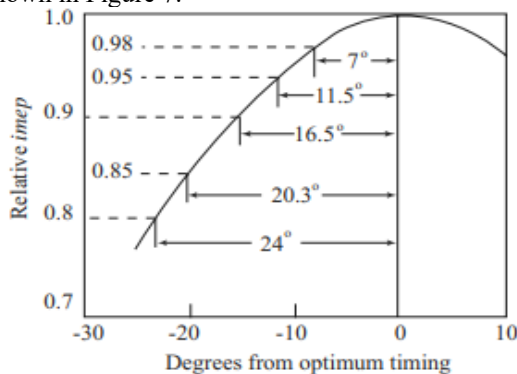


Figure 6: Representing the Ignition advance causes a power loss [Ftp.idu.ac.id ic].

Exhaust Blows Down

Depending on the compression ratio used, the cylinder pressure at the end of the exhaust stroke is around 7 bar. The piston must exert effort against high cylinder pressures if the exhaust valve is opened at bottom dead center.

Loss as Resulting from Gas Exchange Procedures

Pumping work is the difference between the work required to expel exhaust gases and the work required to introduce a fresh charge during the suction stroke. In other words, the gas exchange process also known as pumping loss results in a loss of gas as it is pumped from lower inlet pressure p_i to higher exhaust pressure p_e .

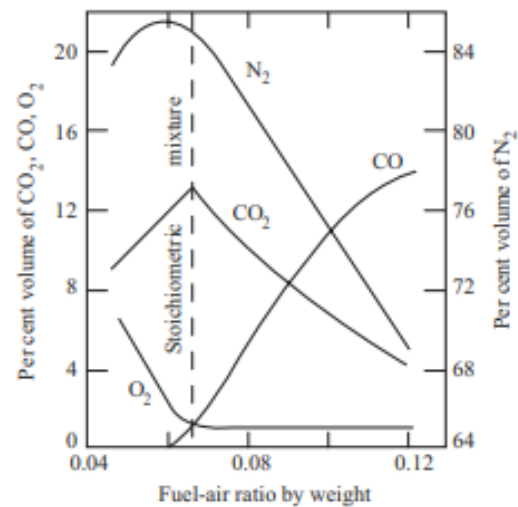


Figure 7: Exhaust gas composition for various fuel-to-air ratios [Ftp.idu.ac.id ic].

Due to the reduced suction pressure caused by throttling, the pumping loss increases at partial throttle. With speed, pumping loss also rises. The volumetric efficiency of the engine is influenced by the gas exchange procedures. The volumetric efficiency has a significant impact on the engine's performance. Therefore, it is beneficial to go into more depth about this parameter here.

Volumetric Efficiency

Volumetric efficiency, which is defined as the ratio of the volume of air actually admitted under ambient condition to swept volume, is a measure of the engine's ability to breathe. However, it can also be defined in terms of mass as the ratio between the actual mass of air drawn into the engine over the course of a given period of time and the theoretical mass that should have been drawn in over the course of that same period of time, based on the total displacement of the engine's pistons as well as the temperature and atmospheric pressure. Only naturally aspirated engines fall under the scope of the aforementioned definition. The theoretical mass of air should be computed for the supercharged engine, though, under the pressure and temperature that are present in the intake manifold. Numerous factors influence volumetric efficiency among the crucial ones are:

- i. The fresh charge's density Heat is transferred from the hot chamber walls and the hot residual exhaust gases to the fresh charge when it enters the hot cylinder, increasing its temperature. As a result, the mass of fresh charge accepted decreases, and volumetric efficiency declines. Low temperatures assuming there are no heat transfer effects and high pressure of the fresh charge boost

- volumetric efficiency because they increase density and allow for the induction of more mass of charge into a given volume.
- ii. The exhaust gas in the clearance volume: These products have a tendency to expand when the piston advances from TDC to BDC on the intake stroke, occupying a portion of the piston displacement greater than the clearance volume, and hence limit the space available for the incoming charge. These exhaust byproducts also have a tendency to elevate the temperature of the fresh charge, which lowers its density and lowers volumetric efficiency even more.
 - iii. The design of the intake and exhaust manifolds: The intake manifold should be made to bring in the freshest charge possible, while the exhaust manifold should be made to allow the exhaust products to exit easily. This indicates that both the forced-out exhaust products and the fresh charge flowing into the cylinder are subjected to the least amount of restriction possible.
 - iv. The intake and exhaust valve timing: The control of the points in the cycle at which the valves are programmed to open is known as valve timing.

CONCLUSION

Internal combustion engines encounter an open cycle with changeable composition real cycle efficiency is substantially lower than air standard efficiency owing to numerous losses occurring in the actual engine. Internal combustion engines use a four-stroke cycle, which is also known as the engine cycle. These four-stroke cycles begin with intake, then go on to compression, combustion expansion, and exhaust. Heat addition is believed to be an instantaneous process in air standard cycles, but it occurs over a specified time period in real cycles. The duration needed for combustion is such that some volume change occurs while burning is taking place in all conditions.

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Introduction about Conventional Fuels and Its Significance

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ABSTRACT: Since IC engines first appeared on the scene, fuels have been the subject of research. When fuel and oxygen combine chemically to form heat energy, the engine transforms that heat energy into mechanical energy. Since the fuel is where the heat energy comes from, understanding the different fuel kinds and their properties is crucial to comprehending the combustion process. The properties of the gasoline used have a big impact on the reliability and durability of the engine, as well as its design, efficiency, and output. Additionally, the fuel properties have a big impact on how much air pollution is produced by car engines.

KEYWORDS: Antiknock Properties, Combustion Engine, Distillation Curve, Engine Fuel, Heat Energy

INTRODUCTION

Since the creation of IC engines, the study of fuels for these engines has been conducted. The engine transforms the heat energy that results from the chemical reaction of the fuel and oxygen into mechanical energy [1], [2]. Understanding the combustion process requires a basic understanding of the many types of fuels and their properties as the heat energy is obtained from the fuel. The design, effectiveness, output, and especially the dependability and durability of the engine are all significantly influenced by the fuel's properties. Additionally, the fuel's properties are crucial in determining how much air pollution is produced by automotive engines.

Fuels

Various fuel sources, including liquid, gaseous, and even solid fuels, can be used to power internal combustion engines. The engine needs to be developed in accordance with the type of fuel that will be utilised [3], [4].

Liquid Fuels

Because of the difficulties in handling the fuel and getting rid of the solid byproduct or ash after combustion, solid fuels are currently not used in many practical applications. Solid fuels, such as finely ground coal, were tried out during the early stages of the engine development. Solid fuels are harder to handle than gaseous and liquid fuels, and they require more space to store and feed. These fuels are no longer appropriate for use in solid form due to difficulties in the design of the fuel feed systems. There are initiatives to produce gaseous or liquid fuels from charcoal for IC engines. Liquid fuels must conform to the geometry of their

container and are made of flammable or energy-producing molecules that can be used to generate mechanical energy, typically kinetic energy. Instead of the fluid itself, liquid fuels are combustible in their vapors. The majority of liquid fuels in common usage come from fossil fuels, although there are a few others that can also be categorized as liquid fuels, including ethanol, biodiesel, and hydrogen fuel for use in automobiles. Numerous liquid fuels are essential to the economy and transportation. Solid, gaseous, and liquid fuels are contrasted with one another [5], [6].

The Gaseous Fuels

Gaseous fuels are the best and present the fewest issues when used in internal combustion engines. Due to their gaseous nature, they mix more uniformly with air and do not cause the distribution or starting issues that liquid fuels do. Although gaseous fuels are best for internal combustion engines, storage and handling issues prevent their widespread usage in automobiles. As a result, stationary power plants that are close to the fuel's supply source frequently use them. While some gaseous fuels can be liquefied under pressure to reduce storage volume, this method is both expensive and dangerous. Due to the recent energy crisis, significant research is being done to upgrade the performance and design of petrol engines, which became outdated when liquid fuels were introduced [7], [8].

Fluid Fuels

Liquid fuels, which are derived from liquid petroleum, are used in the majority of modern internal combustion engines. Benzyl, alcohol, and petroleum products are the three main commercial types of liquid fuels. But as of right now, petroleum

products are the primary fuels for internal combustion engines [9], [10].

Petroleum's Chemical Structure

Petroleum as it is extracted from oil wells is mostly a combination of many hydrocarbons with various molecular structures. Small amounts of nitrogen, oxygen, and contaminants like sand and water are also present. Various hydrocarbon groups have various chemical and physical properties depending on how the carbon and hydrogen atoms are bonded together in the molecule. The majority of petroleum fuels often display the traits of the particular class of hydrocarbon that makes up the fuel's main component. A variety of hydrocarbons are created when carbon and hydrogen mix in various amounts and molecular configurations. The energy properties of hydrocarbon fuels are determined by their bonding and one of the key parameters, the carbon to hydrogen ratio. Petroleum products are divided into various groups according to the quantity of carbon and hydrogen atoms present. According to their chemical makeup, the various types of hydrocarbons have varying physical and chemical qualities that primarily affect the combustion processes and, as a result, the ratio of fuel to air needed in an engine.

DISCUSSION

To create toluene ($C_6H_5CH_3$), the base for creating the explosive molecule trinitrotoluene (TNT), a methyl group (CH_3) is added to benzene. Due to their molecular makeup, the aforementioned groups of hydrocarbons exhibit the following general traits, which are summarized below:

- i. When utilised in a SI engine, regular paraffin's have the worst antiknock properties. However, as the molecular structure becomes more compact and as the amount of carbon atoms increases, the antiknock quality improves.
- ii. In SI Engines, the aromatics provide the best knock resistance.
- iii. The optimum fuels for CI engines are regular paraffin's, while aromatic fuels are the least preferable.
- iv. The boiling temperature rises as the molecular structure's number of atoms does. As a result, fuels with smaller molecular sizes tend to be more flammable.
- v. Because hydrogen has a larger heating value than carbon, the heating value typically rises as the ratio of hydrogen to carbon atoms in the molecule does. As a result, aromatics have a lower heating value than paraffin's.

Process of Petroleum Refining

Methane and ethane are the major gases and impurities found in crude petroleum, which is derived from oil wells. Other impurities include water and solids. The fractional distillation method separates the crude oil into different types of fuel oil, such as petrol, kerosene, and fuel oil. This method is based on the observation that as the molecular weight of different hydrocarbons increases, so do their boiling points. The first process involves passing the petroleum through a separator to remove the gases and produce what is known as natural petrol. Following the vaporization of the liquid petroleum at temperatures of $600\text{ }^\circ\text{C}$ in a still, the vapor is entered at the base of the fractionating tower.

The vapor is pushed to flow through trays of liquid fuel maintained at various temperatures via a labyrinth-like system of plates that direct the vapors upward. While compounds with lower boiling points travel up to higher levels where they are condensed in trays at the proper temperature, those with higher boiling points condense out at lower levels. The top fraction, known as straight-run petrol, is typically obtained in a range of boiling temperatures together with the other fractions, such as kerosene, diesel oil, and fuel oils. Figure 1 provides more information. Some of these fractions can be transformed through a variety of procedures into compounds with higher demand. The following are some of the primary refinery processes:

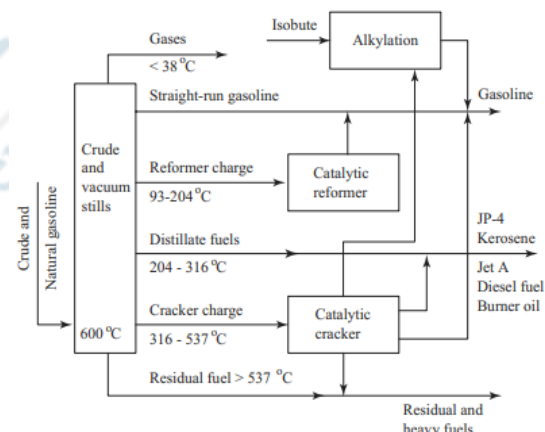


Figure 1: The process of refining petroleum depicts in the diagram [Ftp.Idu.Ac.Id Ic].

- i. Cracking is the process of dissolving large, intricate hydrocarbon molecules into smaller ones. Large hydrocarbon molecules are subjected to high temperatures and pressure during thermal cracking, where they are broken down into smaller molecules with lower boiling points. Thermal cracking

occurs at a higher pressure and temperature than catalytic cracking, which uses catalysts. The naphthenic are catalysed into olefins, paraffin's, and olefins, which are needed to make petrol. Compared to thermal cracking, catalytic cracking provides better antiknock properties for petrol.

- ii. Hydrogenation is the process of adding hydrogen atoms to specific hydrocarbons at high pressures and temperatures in order to create molecules that are more aesthetically pleasing. It is frequently used to change unstable molecules into those that are stable.
- iii. Olefins, the unsaturated by-products of cracking, are transformed into heavier and more stable molecules through polymerization.
- iv. In the presence of a catalyst, alkylation combines an olefin and an is paraffin to create a branched chain is paraffin. For instance, isobutylene and isobutane Alkylation is-octane
- v. Isomerization modifies the relative positions of the atoms within a hydrocarbon molecule without altering the molecule's molecular formula. For instance, isomerization is used to change n-butane into isobutene so that it can be alkylated. Another example is the conversion of n-pentane and n-hexane into iso paraffins to raise the knock rating of highly volatile petrol.
- vi. The ends of a straight chain molecule are joined by cyclization to create a ring compound that belongs to the naphthenic family.
- vii. Aromatization is a process that is comparable to cyclization, with the difference being that the end result is an aromatic molecule.
- viii. Reformation, a sort of cracking procedure, is used to turn low antiknock quality stocks into petrol with a higher-octane rating. The overall volume of petrol is unchanged.
- ix. Blending is the process of combining various products in an appropriate ratio to produce a product of the desired quality.

Important Engine Fuel Qualities

Fuels used in IC engines need to have a few fundamental characteristics in order for the engines to function properly. The crucial characteristics of fuels for both SI and CI engines are addressed in this section.

SI Engine Fuels

Today's SI engines often use petrol, which is typically a blend of a number of low-boiling

paraffin's, naphthenic, and aromatics in variable amounts. We'll go over a few of the crucial characteristics of petrol below. One of the key characteristics of petrol that determines whether it is suitable for use in a SI engine is volatility. Since petrol is a blend of several hydrocarbons, its volatility is influenced by its fractional makeup. The standard method for determining a fuel's volatility is to distil it in a specific apparatus under atmospheric pressure while its own vapour is present. It is measured how much of the mixture boils out at a specific temperature. The temperatures at which 10, 40, 50, and 90% of the volume evaporates, as well as the temperature at which the fuel's boiling process comes to an end, are the typical points. The fractional distillation curve of petrol for both winter and summer grade petrol are shown in Figure. 2. The American Society for Testing Materials (ASTM) has standardised the procedure for calculating volatility, and the graphical display of the test results is typically referred to as the ASTM distillation curve. Along with the distillation curve, the more significant features of engine fuel volatility are covered in detail.

For a smooth start of the engine, a specific portion of the petrol should vaporize at room temperature. As a result, the distillation curve's region between 0 and 10% that was boiled off has comparatively low boiling temperatures. The temperature will gradually rise to operating temperature as the engine warms up. For the optimal warm-up, low distillation temperatures are preferred across the span of the distillation curve. Low distillation temperatures are preferred in the engine operating range to achieve good petrol vaporization. By lowering the number of liquid droplets in the intake manifold, improved vaporization tends to result in both a more uniform distribution of fuel to the cylinders and better acceleration characteristics. Dilution Liquid fuel in the cylinder reduces the amount of lubricating oil present, which degrades the lubrication's quality and tends to harm the engine through increased friction. Additionally, the lubricating oil may be diluted by the liquid petrol, weakening the oil coating between the rubbing surfaces. The upper part of the distillation curve should have sufficiently low distillation temperatures to ensure that all the petrol in the cylinder is vaporized by the time combustion begins, obviating the need for this possibility.

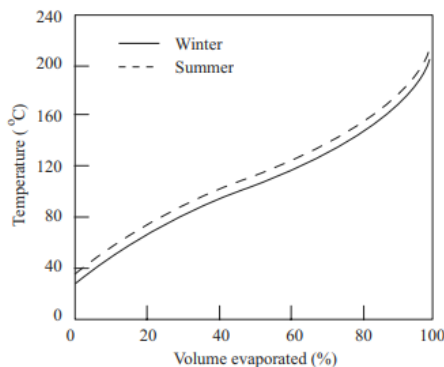


Figure 2: illustrates typical petrol distillation curves [Ftp.Idu.Ac.Id].

By creating a vapour lock in the fuel channels, a high rate of gasoline vaporization might interfere with carburettor metering or potentially block the fuel flow to the engine. This property necessitates the presence of hydrocarbons with relatively high boiling temperatures across the whole spectrum of distillation. The desired distillation temperatures must be compromised because this requirement is incompatible with the other requirements desired in (a), (b), and (c). Abnormal burning or detonation in a SI engine combustion chamber results in a very high rate of energy release, and high temperatures and pressures inside the cylinder reduce the engine's thermal efficiency. Therefore, the fuel's properties should be such that they resist the propensity to cause detonation; this trait is known as the fuel's antiknock property. The self-ignition properties of a fuel's mixture determine its antiknock property, which varies significantly depending on the fuel's chemical make-up and molecular structure. The best SI engine fuel will typically be the one with the strongest antiknock property because this enables the use of greater compression ratios, which considerably increases the engine's thermal efficiency and power production.

When fuel is stored, reactive hydrocarbons and impurities have a propensity to oxidise and produce both liquid and solid gummy substances. Cracked petrol containing unsaturated hydrocarbons is the worst culprit, while petrol containing hydrocarbons from the paraffin, naphthenic and aromatic families form little gum. High gum content petrol will lead to operating issues such sticking valves and piston rings, carbon build-up in the engine, gum build-up in the manifold, clogged carburettor jets and expanding of the valve stems, cylinders and pistons. Increases in oxygen content, temperature, exposure to sunlight, and metal contact all result in an increase in gum production. As a result, petrol specifications set limits on both the fuel's gum content and

inclination to gum up while being stored. Hydrocarbon fuels may contain free sulphur, hydrogen sulphide, and other sulphur compounds, all of which are undesirable for a number of reasons. Fuel sulphur is a corrosive component that can destroy fuel lines, carburettors, and injection pumps. It will combine with oxygen to generate sulphur dioxide, which, when it comes into contact with water at low temperatures, may result in the formation of sulphurous acid. Sulphur's low ignition temperature means that its presence can lower the self-ignition temperature, which can lead to knock in a SI engine.

Fuels for CI Engines

- i. **Knock Characteristics:** In CI engines, knock happens when there is a delay between the time of fuel injection and the time of actual combustion. Fuel builds up in the combustion chamber as the ignition lag lengthens, and when combustion actually occurs, an abnormal quantity of energy is quickly released, creating an excessive rate of pressure rise and a loud knock. As a result, a good CI engine fuel should ignite more quickly and have a short ignition lag. Additionally, ignition lag makes it difficult for CI engines to start, warm up, and produces exhaust smoke. The best fuel today, as measured by the cetane rating, will have a cetane rating that is sufficiently high to prevent undesirable knock.
- ii. **Volatility:** In order to create effective mixing and combustion, the fuel must be sufficiently volatile in the working temperature range.
- iii. **Engine Starting Qualities:** The fuel ought to make it simple to start the engine. In order to meet this condition, the fuel must have a high cetane rating and adequate volatility to easily ignite a flammable mixture.
- iv. **Smoking and Odour:** Neither smoke nor odour in the engine exhaust should be encouraged by the fuel. Good volatility is typically the first requirement to make sure good mixing and hence complete combustion.
- v. **Viscosity:** CI engine fuels should be able to pass through the fuel system and strainers even at the engine's coldest operating temperatures.
- vi. **Corrosion and Wear:** Neither before nor after combustion, the fuel shouldn't produce corrosion or wear on engine parts. These specifications are based on the amount of sulphur, ash, and residue in the fuel.

- vii. Fuels Rating:** Normally, the antiknock properties of fuels are rated. The rating of fuels is done by defining two characteristics for petrol and diesel oil, respectively, called Cetane number and Octane number. In this section, the rating of fuels for both SI and CI engines is covered.
- viii. SI Engine Fuel Rating:** Fuel for spark-ignition engines must have a high level of resistance to knocking. Based on their chemical make-up, these fuels' capacity to resist knock varies greatly. For comparing the antiknock properties of the various fuels, a reliable rating system has been developed. Other operating parameters, such as fuel-air ratio, ignition time, dilution, engine speed, combustion chamber shape, ambient conditions, compression ratio, etc., affect the likelihood to knock in the engine cylinder in addition to the chemical properties of the hydrocarbons in the fuel. Therefore, the engine and its operating variables must be set at standard values in order to calculate the knock resistance characteristic of the fuel.

It is common practice to compare a SI engine fuel's antiknock property to a blend of iso-octane (C₈H₁₈) and regular heptane (C₇H₁₆) to assess the antiknock value of the fuel. Chemically, iso-octane is a very good antiknock gasoline, hence it is arbitrarily given an octane value of 100. On the other hand, normal heptane (C₇H₁₆) receives a rating of 0 octane number due to its extremely low antiknock properties. The proportion of iso-octane by volume in a blend of iso-octane and regular heptane that precisely matches the fuel's knocking intensity in a typical engine under a set of standard operating conditions is known as the Octane number fuel. Iso-octane can be combined with some substances, such as tetraethyl lead, to provide fuels with higher antiknock qualities (octane numbers exceeding 100). Tetraethyl lead's antiknock effectiveness declines with increasing lead content in the fuel for the same amount of lead added.

Additionally, compared to the identical unit at the lower end of the scale, each octane number in the higher region of the octane scale will generate a stronger antiknock impact. For instance, an increase in octane from 92 to 93 produces a stronger antiknock effect than an equivalent rise in octane from 32 to 33. Due to this non-linear variance, a new scale was created, and its units called Performance Number the estimated relative engine performance. CI Engine Fuel Rating The knock resistance of compression-ignition engines is influenced by chemical properties as well as by the engine's

operating and design parameters. As a result, the knock rating of a diesel fuel is determined by contrasting it with primary reference fuels while it is operated in a specific engine under predetermined conditions. The reference fuels are alpha methyl naphthalene (C₁₁H₁₀) and normal cetane (C₁₆H₃₄), both of which have arbitrary cetane numbers of 0 and 100, respectively.

When combustion is carried out in a typical engine under predetermined operating conditions, the cetane number of a fuel is defined as the volume percentage of normal cetane in a mixture of normal cetane and -methyl naphthalene that has the same ignition characteristics as the test fuel. Given that the key governing factor for the first auto ignition in the CI engine is ignition delay, it makes sense to assume that knock should be directly connected to the fuel's ignition delay. Diesel oil's knock resistance feature can be increased by sparingly adding substances like ethyl nitrate, amyl nitrate, or ether. A typical single-cylinder engine, such as a CFR diesel engine or a Ricardo single-cylinder variable compression engine. The test fuel is initially used in the engine running under the predetermined circumstances. An ideal fuel-to-air ratio is achieved by adjusting the fuel pump delivery. A 13-degree injection advance is achieved by adjusting the injection timing as well. The ignition delay can be raised or lessened by adjusting the compression ratio until combustion starts at T DC. The test fuel experiences an igniting delay of 13 degrees after this point is discovered.

CONCLUSION

The term conventional energy simply refers to energy sources that are fixed in nature, such as coal, gas, and oil. So conventional energy is also referred to as a non-renewable form of energy. Increased greenhouse gas emissions and other environmental harm result from their use. Conventional gasoline/diesel cars make up the majority of the cars on the road today. Petrol stations serve as the cars' fueling locations. Their engines run on petrol. Because of how quickly petrol or diesel burns inside the engine, they release more GHGs. This thermal energy that fuels provide is utilized for a variety of reasons, including cooking, heating, and a variety of industrial and manufacturing processes. At times, we use an engine to transform heat energy into mechanical energy. As an example, we utilize gasoline to power our automobiles.

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A Basic Approach on Alternate Fuels

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ABSTRACT: Gaseous fuels like hydrogen, natural gas, and propane, alcohols like ethanol, methanol, and butane, vegetable and waste-derived oils, and electricity are all examples of alternative fuels. These fuels can be utilised in dedicated systems that only burn one fuel or in systems that combine them with other fuels, including conventional petrol or diesel, as in hybrid-electric or flexible fuel vehicles. Some cars and engines have been manufactured with alternate fuels in mind. Others are modified to run on an alternate fuel by changing the engine controls and fueling system from their initial design.

KEYWORDS: Alcohol Fuel, Compression Ratios, Diesel Engines, Diesel Fuel, IC Engine

INTRODUCTION

It is predicted that crude oil and petroleum-based products would become extremely expensive and scarce in this century. The fuel efficiency of engines is increasing daily and will keep doing so. The need for fuel has begun to be determined by the massive growth in the number of automobiles, though. In the near future, petrol and diesel will become more expensive and scarcer. In the future decades, alternative fuel technology will spread due to rising demand and the depletion of fossil resources [1], [2]. Some IC engines have been running on non-gasoline or diesel oil fuel for all these years. They have, however, been extremely few in number. Some developing nations are attempting to adopt alternative fuels for their automobiles due to the high cost of petroleum goods. Concern over the emission issues with petrol and diesel engines is another factor driving the development of alternative fuels for IC engines. The enormous number of automobiles, when combined with other air-polluting technologies, is a significant factor in the global problem with air quality. There have been significant advancements made in lowering emissions from car engines [3], [4].

If a 35% improvement occurs over a number of years, it should be noted that the world's automotive population grows by 40% at the same time, negating the benefit. There have been numerous efforts made to improve the net level of exhaust filtration. However, more advancements are required to reduce the rising levels of air pollution brought on by the automotive population. The requirement to import a sizable portion of crude oil from other nations, which control the larger oil reserves, is a third justification for the development of alternative fuels. Currently, a variety of alternative fuels are being utilised sparingly in automobiles. Quite frequently, fleet vehicles such as taxis, delivery vans, and utility company trucks have been employed for testing.

This facilitates comparison with equivalent gasoline-fueled vehicles and makes filling up these vehicles easier [5], [6]. Engines that are utilised with alternative fuels are modified versions of engines that were created for use with petrol. As a result, they are not the best design for the other fuels. Maximum performance and efficiency from these engines can only be realized via years of intensive research and development. However, until the fuels are approved as suitable for a significant number of engines, it is challenging to justify the research and development [7], [8].

On the market, there are now a few diesel engines. A small amount of diesel fuel is injected at the right time to ignite both fuels, together with methanol or natural gas. Since they are being used in such small amounts, the majority of alternative fuels are quite expensive. The price of many of these fuels is 164 IC. If the amount of their utilization reaches the same order of magnitude as petrol, engines will be used significantly less. Manufacturing, distribution, and marketing expenses would be lower. Lack of distribution locations service stations where the gasoline is made available to the general population is another issue with alternative fuels. Unless there is a substantial network of service stations where fuel for that automobile can be purchased, the people will be hesitant to buy an automobile. On the other hand, until there are enough vehicles to make them lucrative, it is challenging to justify constructing a network of these service stations. A few distribution terminals for some of these fuels, including propane, natural gas, LPG, and methanol, have been established in a few cities. The process of switching from one main fuel type to another will be drawn-out, expensive, and even traumatic. The numerous alternative fuels will be covered in the sections that follow [9], [10].

Possible Replacements

There are three categories for fuels: solid, liquid, and gaseous. The majority of liquid fuels used in modern autos are petroleum-based. The usage of gaseous fuels like CNG and LPG is however common. It is noteworthy to remember that even solid fuels like coal, slurry, and charcoal were attempted in the beginning. The numerous alternative fuels for IC engines are covered in detail in the sections that follow.

Solid Fuels

For IC engines, solid fuels are no longer useful. Some of the earlier attempts are described in this section for historical context. Before petroleum-based fuels were mastered in the second part of the 1800s, a variety of alternative fuels were tested and employed in IC engines. One of the fuels Rudolf Diesel utilised when constructing his engine was coal dust diluted with water. Early diesel engines used water to disperse fine coal particles, which were then injected and burned. Many experimental engines have been created during the past century that use this fuel, even though it was never widely used as fuel. On this fuel technology, considerable work is still being done today. The reduction in average coal particle size has been this fuel type's most significant advancement.

The typical particle size in 1894 was in the range of 100 ($1 = 1 \text{ micron} = 106 \text{ m}$). Between 1940 and 1970, this was lowered to around 70, and by the early 1980s, it had been further decreased to about 10. A typical slurry has a bulk composition of roughly 50% coal and 50% water. The abrasiveness of the solid particles in this fuel is a significant issue since it causes piston rings and injector wear. Given its widespread availability, coal is a desirable fuel. Other forms of utilization, however, appear more practical when used as motor fuel. These include the coal being liquefied or gasified. Due to World War II, petroleum products were extremely difficult to come by in the late 1930s and early 1940s, particularly in Europe. The German army consumed almost all petrol products, leaving no fuel for private automotive use.

DISCUSSION

Although this caused the civilian population some discomfort, it did not prevent them from driving their cherished cars. People with initiative in several nations, primarily Sweden and Germany, created a system of operations they power their cars with solid fuels like coal, wood, or charcoal.

Liquid Fuels

Because they are convenient to store and utilize, liquid fuels are chosen for IC engines. Have a calorific value that is generally good. Among liquid fuels, the primary Alcohol is an alternate.

Beverage

Because they may be derived from several sources, alcohols are a desirable alternative fuel. Sources that are both natural and man-made. Methyl alcohol, or methanol, and the two types of alcohols that seem the most promising are ethanol and ethyl alcohol. The benefits of using alcohol as a fuel include:

- i. It can be acquired from a variety of natural and artificial sources.
- ii. The fuel has a high-octane number and anti-knock index numbers. More than 100. High-octane fuel allows engines to operate more effectively by higher compression ratios are used. Alcohols burn more quickly.
- iii. In comparison to petrol, it generates fewer overall emissions.
- iv. Alcohols produce more moles of exhaust gases when they are burned, which offers the expansion stroke greater pressure and power.
- v. The high latent heat of vaporization (hf_g) makes it colder intake method. This increases the engine's volumetric efficiency and lowers the amount of work that must be put into the compression stroke.
- vi. The fuel's Sulphur content is low for alcohols.

Alcohol's Drawbacks

Alcohol's drawbacks as a fuel include:

- i. Alcohols' calorific value, or the amount of energy they contain, is low. That is almost half the gasoline. Thus, roughly twice as much alcohol will be consumed. As petrol must be burnt to provide the engine with the same amount of energy. Using an engine with the same power and thermal efficiency, twice as the amount of fuel that would need to be purchased and the potential distance Driving distance would be cut in half with a given fuel tank volume. Automobiles additionally, distribution centers would need twice as much storage. Capacity, twice as many storage facilities, and four times as much pipes, twice as many tank trucks, and storage at the service station, etc.
- ii. Despite alcohol's decreased energy content, engine power for every displacement would roughly be the same. This is as a result of for alcohol, a lower air-fuel ratio is required.

Alcohol has oxygen in it, therefore because stoichiometric combustion needs less air. One can burn more fuel.

- iii. Alcohol combustion results in more aldehydes being produced in the exhaust with the same amount of air. If as If petrol and alcohol were drunk in equal amounts, aldehyde emissions be a significant issue with exhaust emissions.
- iv. Alcohol corrodes copper, brass, aluminum, rubber, and many polymers significantly more than gasoline does. This limits the in some ways. Design and production of engines that will consume this fuel. Fuel Gaskets, metal engine parts, lines, and tanks can all degrade. Long-term alcohol usage can lead to broken gasoline lines and the requirement for unique fuel tank, etc. Metals are severely corroded by methanol.
- v. Due to low vapor pressure and evaporation, it has poor cold weather starting characteristics. Alcohol-fueled engines typically struggle in commencing at degrees below ten degrees Celsius. Frequently a little bit of petrol is added to alcohol fuel, which significantly enhances starting in cold weather. However, the necessity of doing this significantly lessens any alternative energy.
- vi. Alcohols often have poor ignition properties. Alcohols have very imperceptible flames, which is harmful. When working with fuel. This risk can be eliminated with a modest amount of petrol.
- vii. Because of the low vapor pressure in storage tanks, there is a risk of fire. Storage tanks may experience air leaks that result in flammable mixtures. Due to the low flame temperatures, there will be fewer NOx emissions. However, it takes more time to adjust to the lower exhaust temperatures that occur. Bring the catalytic converter's operating temperature up to efficiency.
- viii. A lot of people find the strong alcohol smell to be highly repulsive. Headaches and lightheadedness have been reported after refueling a car. In fuel delivery systems, vapor lock is a potential problem.

Methanol

Methanol is one of the fuels being studied as an alternative to petrol. Most promising and has undergone extensive research and development. Pure methanol and different methanol-to-gasoline ratios in mixes have undergone years of thorough testing in engines and vehicles. M85, which is

composed of 85% methanol and 15% petrol, and M10 (methane at 10% and petrol at 90%). The test results' information, which includes pure petrol (M0) and other fuels are compared for performance and pollution levels. Pure (M100) methanol. Certain intelligent flexible fuel or variable fuel engines are capable of employing any arbitrary methanol and petrol mixture varying from pure petrol to pure methanol. It uses two gasoline tanks, and the two fuels can be injected into the engine at varied flow rates, passing by means of a mixing chamber. Using data from intake and exhaust sensors, when the engine is finished, the electronic monitoring system (EMS) makes the appropriate adjustments for the air-to-fuel ratio, ignition, injection, and valve timing. For the fuel mixture that is being used. One issue with fuels made of petrol and alcohol is the propensity for any water to combine with the alcohol.

This occurs because the alcohol produces a non-homogeneous mixture when it locally separates from the petrol. Due to the significant variances in the air-fuel ratios of the two fuels, this results in the engine running erratically. There are numerous sources of methanol, both renewable and fossil. These include landfills, coal, oil, natural gas, biomass, wood and even the water. Nevertheless, any source that necessitates substantial manufacturing or the cost of the fuel goes up due to processing. When utilizing M10 fuel, an engine's emissions are comparable to those produced by gasoline. The biggest benefit (and drawback) of utilizing this fuel is the 10% less petrol is used. M85 fuel causes a discernible drop in performance. In the exhaust emissions of HC and CO. However, NOx and other pollutants have increased. A significant rise in formaldehyde emissions (by about 500%). Some dual-fuel CI engines use methanol. By itself, methanol is not a Because of its high-octane rating, CI engine fuel is recommended, although if only a small amount When utilised for ignition, diesel oil can be used successfully. That is really attractive for underdeveloped nations as methanol is frequently obtainable from a far more affordable source than diesel fuel.

Alcohol

Several nations have long utilised ethanol as a fuel for automobiles. The globe. Early in the 1990s, Brazil was undoubtedly the top user. About 93% ethanol was used as fuel in 5 million cars. For a variety of Gasohol, a mixture of petrol and alcohol, has been sold at service stations in the United States. Petrol contains 10% ethanol and 90% petrol. As The creation of systems utilizing mixes of petrol and

ethanol keeps going. The crucial mixtures are E85 and ethanol. E10 and 85% ethanol. E85 is essentially a 15% alcohol fuel. To solve some of pure alcohol's issues such as frigid temperature, petrol was added. E10 decreases petrol consumption without any the car's engine has to be modified. The development of flexible-fuel engines tested that can run on any ethanol-to-gasoline ratio. The use of ethanol produced using ethylene or through the fermentation of sugar and cereals. Lots of it is made from cellulose, corn, sugar beets, and even sugar cane. Ethanol's current price is high because of the manufacturing and processing is necessary. If greater quantities of this fuel were used, this would be decreased. Used. A food-fuel competition, however, would result from extremely high output. With increased prices for both as a result. Studies indicate that at the moment in crops produced in the United States to make ethanol use more energy ploughing, planting, harvesting, fermenting, and delivery require more energy than what is in the finished item. This negates a key benefit of using an alternative, fuel. While it emits more HC than methanol, ethanol emits less than petrol.

Alcohol for SI Engines

Alcohols exhibit a stronger anti-knock property than petrol. As a result, Engine compression ratios between 11:1 and 13:1 are possible with an alcohol fuel. 169 Alternative Fuels usual. Compression ratios for petrol engines today typically range from 7:1 to 9:1. Much too little alcohol when pure. Alcohol produces fewer emissions in a properly constructed engine and fuel system. Exhaust emissions that are hazardous. The thermal energy in alcohol is around half that in water. Petrol per liter. Alcohol has a lower stoichiometric air fuel ratio than for kerosene. A carburetor or fuel injector will give the correct fuel-air mixture. To allow for more fuel flow, fuel channels should be doubled in size. Unlike petrol, alcohol does not easily vaporize. It has a substantially higher latent heat of vaporization. Starting cold weather is impacted by this. Unless the alcohol if it liquefies inside the engine, it won't burn correctly. Hence, the engine could in an exceptionally chilly climate, starting is challenging or even impossible. To prevail petrol is added to the engine until it fires up and heats up. Up. When alcohol is introduced after the engine has warmed up, it will swiftly vaporize. And entirely, burning normally. Even while operating normally, additional Alcohol may require more heat to thoroughly vaporize. Alcohol burns. Roughly half as fast as petrol. As a result, ignition timing needs to be altered. To provide a greater spark progress. As a

result, the alcohol that burns slowly will have more time to build up the pressure and power inside the cylinder. Moreover, given that alcohols are corrosive, fuel systems require materials that are Corrosive in character.

Petrol Reformulated for SI Engine

Regular petrol that has been slightly modified and contains additives to help lower engine emissions is known as reformulated petrol. Chemicals in the Fuel comprise detergents, additives for deposit control, metal deactivators, oxidation and corrosion inhibitors, and others. Alcohols are combined with oxygenates, such as methyl tertiary-butyl ether (MTBE), to produce a mixture with 1-3% oxygen. By mass. This will aid in lowering the CO content of the exhaust. Benzene levels, Reduced aromatic and high-boiling constituents as well as a decrease in vapor pressure. Engine deposits are known to cause emissions thus, cleaning chemicals are comprised. Various additives clean fuel injectors, carburetors, and some people clean the intake valves, but they rarely clean the other parts. The good news is that both new and old gasoline-powered engines can don't modify this fuel; use it as is. The drawback is that only moderate as emissions are reduced, costs rise and petroleum use declines goods are not significantly diminished.

Water-Gasoline Blend for SI Engines

Along with the advancement of the spark-ignition engine, a desire to boost compression ratio for better fuel efficiency. The octane quality of some fuels has occasionally been a barrier to this economic improvement. The petrol on hand. Water was suggested as a solution to get around this restriction a knock-inhibitor. The addition of water to petrol slows the burning process and lowers the cylinder's gas temperature likely prevents detonation. There have also been reports of decreased engine combustion chamber deposits when the intake charge included water. Considering nitric oxide emissions

IC Engines

Significant decreases were noted. In contrast, adding water likely results in an increase in hydrocarbon emissions. And last, regarding carbon monoxide water addition appears to have no impact on emissions. Only a tiny fraction Work has been done using an emulsion with the addition of water to fuel as opposed to separately. Emulsion might improve atomization, do away with the requirement for a separate tank, and boost fuel safety. However, there could be an issue with water and fuel separation.

Alcohol for CI Engines

There are several ways to use alcohol in diesel engines.

- i. Diesel fuel solutions with alcohol.
- ii. Alcohol diesel emulsions.
- iii. Fumigation with alcohol.
- iv. Dual fuel injection.
- v. Alcohols can ignite on the surface.
- vi. Alcohols ignited by sparks.
- vii. Alcohols with chemicals that help with ignition.

Alcohols with a high self-ignition temperature include methyl and ethyl. Hence, to self-ignite them, very high compression ratios (25-27) will be necessary. Since the more advantageous approach avoids making the engine exceedingly heavy and expensive. Uses them for dual fuel operation. Alcohol is carbureted or injected into the inducted air in the dual fuel engine. Air. Alcohols have a high self-ignition temperature, hence using the standard 16 to 18:1 diesel compression ratio will not cause combustion. A little earlier a tiny amount of diesel oil is injected towards the conclusion of the compression stroke into through the typical diesel pump and spray nozzle into the combustion chamber. The quick ignition of the diesel oil starts burning in the alcohol air. Combination as well. A variety of techniques are used to introduce alcohol into the intake in various ways. They are the carburetor, vaporizer, micro fog unit, and pneumatic spray nozzle. Fuel injector, etc. There are differences in the fineness of the fuel and air mixture. For the methods listed above. Alcohol injection into the combustion chamber has also been tried as a technique. Following the injection of diesel fuel. This method of alcohol injection stays away from alcohol. Cooling the cylinder charge to a point where the ignition is put in danger. Fuel for diesel engines. However, this design necessitates two full, independent tank, fuel pump, injection pump, and injectors are all part of the fuel system. The majority of the heat release in the dual fuel engines stated above occurs during powered by the provided alcohol, which is lit by a pilot spray of diesel. Injection of oil. The dual fuel engine's performance is affected by the alcohols possess the following qualities.

Various Fuels

Alcohols have a lower calorific value than diesel, hence they have a greater in order to generate the same amount of electricity, and a certain amount of alcohol must be used. Output. However, they demand less air for combustion, therefore as a result, the mixture has nearly the same amount of energy.

They have been latent the temperature and pressure towards the conclusion, as well as the heat of vaporization, are very high. A result of their evaporation, fall down in compression. So, if the alcohol level injected diesel won't be able to ignite if the induction rate surpasses a certain threshold. And as a result, the engine won't run. All of the dual fuel systems previously mentioned share the fundamental flaw of requiring two various fuels and related parts. Alcohols have a because strong propensity to pre-ignite in SI engines; this characteristic was recently used by starting the ignition using a heated surface in a compression ignition engine.

Diesel with Spark-Assisted

It is anticipated that petroleum gasoline will eventually be available as broad cut fuel. Alcohols and fuel made from coal will also be used as an alternative. The low cetin value of these fuels is one of their noteworthy properties. As a result, the emergence of these fuels could pose a danger to the continued usage of diesel engines because these engines heavily rely on fuels with high cetin values. Diesel engines have many benefits in terms of performance, particularly in terms of their great thermal efficiency. From the perspective of energy conservation, this benefit will be quite significant in the future. With this in mind, a major Japanese firm has created a spark-assisted diesel engine that can run on future low cetin gasoline without losing its diesel engine characteristics. By adding a spark plug to the precombustion chamber, they have modified the normal 19:1 compression ratio precombustion chamber type diesel engine. The commercial CDI, multistoried kind of igniter is employed. Two spark-assisted transit vehicles have been used in studies by the California Energy Commission.

Parallel tests were conducted on both buses and their diesel-powered equivalents. Road performance for each pair of buses was substantially comparable when comparing diesel and methanol fuel. The methanol buses' exhaust is cleaner, and smoke and stink are gone. Methanol significantly reduces particles and nitrogen oxides. One of the substitute fuels for diesel engines is vegetable oil. Vegetable oil has a higher viscosity than diesel, though. It must be reduced as a result to enable effective atomization in engines made to use diesel fuel. Otherwise, the engine will eventually suffer harm from incomplete combustion and carbon buildup. To distinguish vegetable oil from biodiesel, some writings label it waste vegetable oil (WVO), straight vegetable oil (SVO), or pure plant oil (PPO). Metals are negatively impacted by the free fatty acids in WVO.

WVO has an impact on copper and its alloys, such as brass. FFAs strip zinc and zinc plating, and they also have an impact on tin, lead, iron, and steel. Aluminum and stainless steel are typically unaffected. More than 12 billion liters of wasted vegetable oil were produced annually in the United States as of 2010, primarily from industrial deep fryers used in potato processing plants, snack food companies, and fast-food establishments. Nearly 1% of US oil use might be offset if all 12 billion liters could be gathered and used to replace the energy equivalent of petroleum. It should be emphasized that using used vegetable oil as fuel puts some other uses of the resource in competition. This affects the cost of the fuel and raises the price of the input used for the other purposes.

CONCLUSION

Fuels made from sources other than petroleum are referred to as alternative fuels, non-conventional fuels, and advanced fuels. Gaseous fossil fuels like propane, natural gas, methane, and ammonia are examples of alternative fuels, as are biofuels like biodiesel, bio alcohol, and fuel obtained from waste, as well as other renewable fuels like hydrogen and electricity. These fuels can assist in decarbonization and pollution reduction efforts by replacing more carbon-intensive energy sources like petrol and diesel in the transportation sector. Alternative fuels have also been demonstrated to lower emissions that aren't carbon-based, like the release of nitrogen dioxide and nitric oxide in the atmosphere, as well as Sulphur dioxide and other dangerous chemicals in the exhaust. This is crucial in fields like mining where it's easier for harmful gases to build up.

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Application, Advantages and Disadvantages of Liquid Fuel

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ABSTRACT: *Due to their convenience in storage and comparatively high calorific value, liquid fuels are favored for IC engines. Alcohol serves as the primary substitute in the liquid fuel category. In this chapter discussed about the liquid fuel that is used in to the combustion for starting an engine. addition to having to conform to their container, liquid fuels are combustible or energy-producing molecules that can be used to generate mechanical energy, most commonly kinetic energy. The economy and transportation both heavily rely on liquid fuels. In comparison to solid and gaseous fuels, liquid fuels are described.*

KEYWORDS: *Alcohol Related, Calorific Value, Dual Fuel, Harmful Alcohol, Liquid Fuels.*

INTRODUCTION

Since liquid fuels are simple to store and have a respectable calorific value, they are favored for IC engines. The primary substitute in the liquid fuel category is alcohol. As combustible or energy-producing molecules, liquid fuels must conform to the geometry of their container in order to produce mechanical energy, typically kinetic energy. Liquid fuels are combustible not in their fluid form but in their vapor's. Although most liquid fuels in common usage are made from fossil fuels, there are a few others that fall under this category, including biodiesel, ethanol, and hydrogen fuel for use in automobiles [1], [2].

Alcohol

Alcohol is a psychoactive drug with addictive qualities that has been used for ages in many different cultures. Alcohol abuse has serious social and economic repercussions as well as a high illness burden. Alcohol abuse can hurt not just oneself, but also friends, family, co-workers, and complete strangers. More than 200 diseases, injuries, and other health issues are caused by alcohol use [3], [4]. Alcohol consumption is linked to an increased risk of acquiring serious no communicable diseases such liver cirrhosis, several malignancies, and cardiovascular diseases as well as mental and behavioural disorders, including alcohol dependency. Unintentional and intentional injuries, such as those caused by car accidents, acts of violence, and suicide, account for a sizable amount of the disease burden linked to alcohol intake. Younger age groups tend to experience fatal alcohol-related injuries. There is evidence linking problematic drinking to the occurrence or

consequences of infectious diseases including HIV and TB [5], [6].

Foetal alcohol syndrome (FAS) and pre-term birth difficulties can arise from a pregnant woman drinking alcohol. At both the individual and societal levels, numerous variables that influence alcohol consumption levels, patterns, and the scope of alcohol-related issues in populations have been found. Level of economic development, culture, social norms, accessibility to alcohol, and the introduction and enforcement of alcohol legislation are all societal factors. For a given level and pattern of drinking, adverse health effects and social consequences are greater for poorer societies. Age, gender, familial situation, and socioeconomic level are all personal aspects. Although there is no one risk factor that predominates, a person is more likely to experience alcohol-related difficulties as a result of alcohol usage if they have more vulnerabilities. Alcohol drinking causes more health and societal problems for less wealthy people than for more wealthy people. The total amount of alcohol drunk and drinking habits, particularly those habits linked to bouts of heavy drinking, play a significant role in how alcohol intake affects chronic and acute health consequences [7], [8].

Alcohol-related harm is frequently brought on by the circumstances surrounding drinking, particularly when alcohol intoxication is involved. Drinking alcohol can affect not only the occurrence of diseases, accidents, and other health conditions, but also their results and how they change over time. As well as amounts and patterns of alcohol intake, there are gender disparities in mortality and morbidity caused by alcohol use. Men account for 7.7% of all fatalities worldwide from alcohol-related causes, compared to women who account for 2.6% of all deaths. Male drinkers used 19.4 litres of pure alcohol

on average per person in 2016 whereas female drinkers consumed 7.0 litres on average. Easing the burden of harmful alcohol usage Governments can lessen alcohol-related health, safety, and socioeconomic issues by developing and putting into place the right regulations. Action on proven effective and economical techniques is encouraged at the policy-making level [9], [10]. These consist of: Alcohols are an attractive alternate fuel because they can be obtained from both natural and manufactured sources. Methanol and ethanol are two kinds of alcohols that seem most promising. The advantages of alcohol as a fuel are:

- i. It can be obtained from a number of sources, both natural and manufactured.
- ii. It is a high-octane fuel with anti-knock index numbers of over 100. Engines using high-octane fuel can run more efficiently by using higher compression ratios. Alcohols have higher flame speed.
- iii. It produces fewer overall emissions compared to gasoline.
- iv. When alcohols are burned, it forms more moles of exhaust gases, which gives higher pressure and more power in the expansion stroke.
- v. It has high latent heat of vaporization which results in a cooler intake process. This raises the volumetric efficiency of the engine and reduces the required work input in the compression stroke.
- vi. Alcohols have low sulphur content in the fuel.

Disadvantages of Alcohol as a Fuel

Alcohols have a low energy content or in other words the calorific value of the fuel is almost half. This means that almost twice as much alcohol as gasoline must be burned to give the same energy input to the engine. With equal thermal efficiency and similar engine output usage, twice as much fuel would have to be purchased, and the distance which could be driven with a given fuel tank volume would be cut in half. Automobiles as well as distribution stations would require twice as much storage capacity, twice the number of storage facilities, twice the volume of storage at the service station, twice as many tank trucks and pipelines, etc. Even with the lower energy content of alcohol, engine power for a given displacement would be about the same. This is because of the lower air-fuel ratio needed by alcohol. Alcohol contains oxygen and thus requires less air for stoichiometric combustion. More fuel can be burned with the same amount of air.

DISCUSSION

The WHO places a strong emphasis on the creation, collection, and dissemination of scientific data on alcohol use, dependence, and associated health and social implications. It also emphasizes the development, implementation, and assessment of cost-effective interventions for hazardous alcohol use. International agreement that reducing harmful alcohol use and the accompanying health and social costs is a public health priority is reflected in the Global plan to decrease the harmful use of alcohol, which WHO Member States adopted in 2010. The Strategy offers recommendations for 10 recommended policy options and interventions for national action to reduce the harmful use of alcohol, as well as the key elements for global action to support and supplement actions at the national level. The Strategy offers guidelines for action at all levels. A new set of enabling and targeted recommended actions to lessen the harmful use of alcohol is provided by the update of the evidence on the cost-effectiveness of policy options and interventions implemented in the context of the Global action plan for the prevention and control of non-communicable diseases 2013-2020. Increased taxes on alcoholic beverages, comprehensive restrictions on exposure to alcohol advertising across all media, and restrictions on the availability of retail alcohol are among the most economically advantageous measures, or so-called best buys. The demand for global information on alcohol consumption, alcohol-attributable and alcohol-related harm, as well as related policy responses, has significantly increased due to growing awareness of the impact of alcohol consumption on global health and an increase in international frameworks for action.

WHO has created the Global Information System on Alcohol and Health (GISAH) to dynamically convey data on levels and patterns of alcohol consumption, alcohol's effects on health and society, and all levels of policy responses? To reduce harmful alcohol use in accordance with the SDG 2030 agenda's goals and the WHO Global Monitoring Framework for Non-communicable Diseases, countries must work together, there must be effective global governance, and all pertinent stakeholders must be appropriately involved. The harmful effects of alcohol on one's health and society can be lessened by effective collaboration. It can be obtained from a number of sources, both natural and manufactured. Alcohol combustion increases the number of aldehydes in the exhaust. Aldehyde emissions would be a significant exhaust pollution issue if alcohol fuel was consumed on a par with petrol.

Alcohol corrodes copper, brass, aluminum, rubber, and many polymers significantly more than gasoline does. This imposes some limitations on the development and production of engines using this fuel. Long-term alcohol usage can cause fuel tanks, gaskets, and even metal engine parts to degrade leading to fractured fuel lines, the need for a special fuel tank, etc. Metals are severely corroded by methanol. Due to low vapor pressure and evaporation, it has poor cold weather starting characteristics. Engines running on alcohol typically have trouble starting at low temperatures (below 10 °C). Alcohol fuel is frequently mixed with a tiny amount of petrol, which significantly enhances cold-weather starting. But doing so significantly lessens the appeal of any alternative fuel.

Alcohols often have poor ignition properties. Alcohols have very imperceptible flames, which is harmful when working with fuel. This risk can be eliminated with a modest amount of petrol. Because of the low vapor pressure in storage tanks, there is a risk of fire. Storage tanks may experience air leaks that result in flammable mixtures. Due to the low flame temperatures, there will be fewer NO_x emissions. Though it takes longer to heat the catalytic converter to an effective operating temperature as a result of the lower exhaust temperatures, a lot of people find the strong alcohol smell to be highly repulsive. When refueling an automobile, people have reported experiencing headaches and lightheadedness. In fuel delivery systems, vapor lock is a potential problem.

Methanol

The ancient Egyptians utilised a variety of chemicals, including methanol, which they derived by paralyzing wood, in their embalming procedure. Robert Boyle, however, created pure methanol for the first time in 1661 by distilling boxes. Later, it was referred to as pyroxylin spirit. The elemental makeup was identified in 1834 by French chemists Jean-Baptiste Dumas and Eugene Penlight. Additionally, they created the name methylene for organic chemistry by combining the Greek words methyl and hl. Methylene denoted a radical that was 1 atom of carbon and 14% hydrogen by weight. This should be CH₂, but because it was believed at the time that carbon's atomic weight was only six times that of hydrogen, the formula was given as CH. As a result, they referred to wood alcohol as hydrate de methylene hydrate since they believed the formula to be C₄H₈O₄ = (CH)₄(H₂O)₂. Around 1840, the word methyl was created by back-forming the word methylene, and it was subsequently used to denote methyl alcohol. In 1892, the International

Conference on Chemical Nomenclature abbreviated this to methanol. Methyl is the source of the suffix -eel, which is used to designate carbon groups in organic chemistry.

The first method for synthesizing methanol was introduced by French chemist Paul Sabatier in 1905. This procedure implied that hydrogen and carbon dioxide may react to create methanol. A method to turn synthesis gas a mixture of carbon monoxide, carbon dioxide, and hydrogen into methanol was created by German chemists Alwen Metcath and Mathias Pier, who were employed by Badische-Anilin & Soda-Fabrik (BASF), and they were granted a patent. The BASF procedure, according to Bolzano and Magenta, was first used at Leona, Germany, in 1923. With a zinc/chromium oxide catalyst, the operating conditions included high temperatures between 300 and 400 °C and pressures between 250 and 350 atm. The method used a chromium and manganese oxide catalyst under exceptionally demanding circumstances, including pressures ranging from 50 to 220 atm and temperatures up to 450 °C. The US patent 1,569,775 (US 1569775) was applied for on September 4, 1924, and it was granted to BASF on January 12, 1926. The introduction of catalysts (often copper) able to operate at lower pressures has improved the efficiency of modern methanol production. ICI created the contemporary low-pressure methanol (LPM) method in the late 1960s, however the technological patent has long ago expired US 3326956.

Methanol, also known as M-Staff, was a fuel utilised in a number of German military rocket designs during World War II. It was also combined roughly 50/50 with hydrazine to create C-Staff. The 1970s oil crisis brought attention to the usage of methanol as a motor fuel. Over 20,000 methanol 'flexible fuel vehicles' (FFV) that could run on either methanol or petrol had been introduced in the U.S. by the middle of the 1990s. Additionally, throughout much of the 1980s and the beginning of the 1990s, modest amounts of methanol were a blend in petrol fuels marketed in Europe. By the late 1990s, automakers had stopped producing methanol FFVs and had turned their focus to ethanol-powered vehicles. Although the methanol FFV programmer was technically successful, interest in methanol fuels was waning as a result of rising methanol prices in the mid- to late-1990s during a period of falling petrol pump prices. Mobil created a method in the early 1970s for turning methanol into petrol fuel. Methanol became a predecessor to the feedstock chemicals acetic acid and acetic anhydride between the 1960s and the 1980s. These procedures include

the Captiva, Tennessee Eastman, and Monsanto acetic acid production techniques. The simplest aliphatic alcohol, methanol is an organic compound having the chemical formula CH_3OH a methyl group connected to a hydroxyl group, commonly written as ME OH . It is also known as methyl alcohol and wood spirit. It has a pronounced alcoholic fragrance resembling that of ethanol, and it is a colorless, flammable liquid that is light, volatile, and volatile. Methanol was previously primarily created by the destructive distillation of wood, hence the name wood alcohol. Nowadays, industrial methanol production primarily involves hydrogenating carbon monoxide. A methyl group and a polar hydroxyl group are joined to form methanol. It is manufactured in excess of 20 million tons annually and serves as a precursor to a variety of more specialized chemicals as well as formaldehyde, acetic acid, methyl tart-butyl ether, methyl benzoate, anisole, and other common chemicals.

Methanol is one of the most promising fuels being investigated as a substitute for petrol and has undergone extensive study and development. For many years, various percentages of pure methanol and mixes of methanol and petrol have been thoroughly tested in engines and vehicles. M85 (85% methanol and 15% petrol) and M10 (10% methanol and 90% petrol) are the most popular blends. These tests' performance and emission results are contrasted with those of pure petrol (M0) and pure methanol (M100). Any random blend of methanol and petrol, from pure methanol to pure petrol, can be used in some intelligent flexible fuel (or variable fuel) engines. The engine can receive fuel from either of two fuel tanks at different flow rates after passing through a mixing chamber. The electronic monitoring system (EMS) adapts to the appropriate air fuel ratio, ignition timing, injection timing, and valve timing for the fuel mixture being utilised using data from sensors in the intake and exhaust. Alcohol's propensity to interact with any water in gasoline-alcohol combinations as fuel is one issue. This results in a non-homogeneous mixture as the alcohol locally separates from the petrol. Due to the significant variances in the air-fuel ratios of the two fuels, this results in the engine running erratically.

There are numerous sources of methanol, both renewable and fossil. Coal, oil, natural gas, biomass, timber, landfills, and even the ocean are examples of these. However, the cost of gasoline increases with any source that needs considerable manufacture or processing. The emissions from an engine running on M10 fuel are comparable to those from a

gasoline-powered engine. The biggest benefit of utilizing this fuel is the 10% drop in petrol consumption. There is a discernible reduction in HC and CO exhaust emissions while using M85 fuel. However, formaldehyde emissions have increased significantly by about 500% and NO_x emissions have increased. Some dual-fuel CI engines use methanol. Due to its high-octane number, methanol is not an acceptable CI engine fuel on its own, but it can be used successfully if a tiny amount of diesel oil is used for ignition. This is extremely appealing to underdeveloped nations because methanol is frequently available for a much less money than diesel fuel.

Ethanol

For many years, ethanol has been utilised as motor fuel in a number of nations around the world. Early in the 1990s, Brazil was undoubtedly the top user. 5 million cars and trucks used fuel that was 93% ethanol. Gasohol, a mixture of petrol and alcohol, has been sold at petrol stations in the US for a while. Petrol contains 10% ethanol and 90% petrol. The development of systems that use petrol and ethanol combinations is ongoing, just like with methanol. The mixtures E85 (85% ethanol) and E10 are noteworthy examples of mixture combinations. In essence, E85 is alcohol fuel with 15% petrol added to get rid of some of the issues with pure alcohol such as cold starting and tank flammability. E10 reduces petrol consumption without requiring any changes to car engines.

We are testing flexible-fuel engines that can run on any ethanol-to-gasoline ratio. Either ethylene or the fermentation of cereals and sugar can be used to make ethanol. Corn, sugar beets, sugar cane, and even cellulose are used to make a large portion of it. Ethanol now costs a lot because of the production and processing steps needed. If more of this fuel were utilised, this would be decreased. High production would, however, lead to a food-fuel price war and increased costs for both. According to some research, the energy used to cultivate, plant, harvest, ferment, and transport ethanol-producing crops in the United States today is greater than the energy contained in the finished product. One of the main arguments for utilizing alternative fuel is defeated by this. While it emits more HC than methanol, ethanol emits less than petrol.

Alcohol for SI Engines

Alcohols exhibit a stronger anti-knock property than petrol. Engine compression rates of between 11:1 and 13:1 is so typical when using an alcohol fuel. The compression ratio of modern petrol engines is typically 7:1 or 9:1, which is far too low for pure

alcohol. Alcohol emits fewer damaging exhaust emissions in an engine and fuel system that are appropriately constructed. Per liter, alcohol has about half the thermal energy of petrol. Alcohol has a lower stoichiometric air fuel ratio than petrol. A carburetor's or fuel injector's fuel passageways should be doubled in area to allow for increased fuel flow in order to deliver the correct fuel-air mixture. Unlike petrol, alcohol does not easily vaporize. It has a substantially higher latent heat of vaporization. Starting cold weather is impacted by this. Alcohol won't burn correctly if it liquefies in the engine. As a result, in severely cold climates, the engine may be difficult or even impossible to start. In order to get around this, petrol is added to the engine up until it starts and heats up. Alcohol injected after the engine has warmed up will swiftly and totally vaporize and burn normally. To totally vaporize alcohol, more heat may need to be provided even during regular operation. Alcohol burns at a rate that is roughly half that of petrol. As a result, the ignition timing needs to be adjusted in order to produce additional spark advance. As a result, the alcohol that burns slowly will have more time to build up the pressure and power inside the cylinder. Furthermore, as alcohols are naturally corrosive, corrosion resistant materials are needed for the fuel system.

Petrol Reformulated for SI Engine

Regular petrol that has been slightly modified and contains additives to help lower engine emissions is known as reformulated petrol. Oxidation inhibitors, corrosion inhibitors, metal deactivators, detergents, and deposit control additives are some of the additives found in fuel. Alcohols and oxygenates, such as methyl tertiary-butyl ether (MTBE), are combined to contain 1-3% oxygen by weight. This will aid in lowering the CO content of the exhaust. The concentrations of benzene, aromatic, and high-boiling substances are decreased, along with the vapor pressure. Cleaning chemicals are present because engine deposits influence emissions. Some additives clean intake valves, fuel injectors, and carburetors, but they frequently do not clean other parts of the vehicle. The good news is that this fuel may be used in any gasoline-powered engine, regardless of age, without needing to be modified. The usage of petroleum products is not significantly decreased, costs go up, and there is only a small reduction in emissions.

Water-Gasoline Blend for SI Engines

Increasing the compression ratio for better efficiency and fuel economy has been a goal of spark-ignition engine development. The octane rating of the petrol that is readily accessible has

occasionally been a barrier to this improvement in economy. It was suggested that using water as an antiknock additive would get over this restriction. The addition of water to petrol slows down combustion and lowers gas temperature in the cylinder, likely suppressing detonation. There have also been reports of decreased engine combustion chamber deposits when water was added to the intake charge. Nitric oxide emissions from 170 IC Engines experienced substantial reductions. In contrast, adding water likely results in an increase in hydrocarbon emissions. Finally, adding water appears to have no impact on carbon monoxide emissions. With the addition of water via an emulsion with the fuel rather than independently, relatively little effort has been done. Emulsion might improve atomization, do away with the requirement for a separate tank, and boost fuel safety. The separation of the water and fuel could be a concern, though.

Alcohol for CI Engines

- i. Alcohol/diesel fuel mixes are one method of employing alcohol in diesel engines.
- ii. Alcohol diesel emulsions.
- iii. Alcohol surface ignition.
- iv. Alcohol fumigation.
- v. Dual fuel injection.
- vi. Alcohols ignited by sparks.
- vii. Alcohols with chemicals that help with ignition.

Alcohols with a high self-ignition temperature include methyl and ethyl. To self-ignite them, very high compression ratios (25-27) will be necessary. The preferable approach is to use them in dual fuel operation because doing so would make the engine incredibly heavy and expensive. Alcohol is carbureted or injected into the inducted air in a dual fuel engine. Alcohols have a high self-ignition temperature, hence using the standard 16 to 18:1 diesel compression ratio will not cause combustion. A little amount of diesel oil is sprayed into the combustion chamber through the standard diesel pump and spray nozzle just before the compression stroke ends. When the diesel oil ignites easily, the alcohol and air combination also begin to burn. A variety of techniques are used to introduce alcohol into the intake in various ways. They are the fuel injector, carburetor, vaporizer, pneumatic spray nozzle, and micro fog unit. The degree of fineness in the fuel and air mixture varies for the procedures mentioned above. Alcohol injection into the combustion chamber following diesel fuel injection has also been tested. This method of alcohol injection prevents the alcohol from cooling the

cylinder charge to a point where it could compromise the ignition of the diesel fuel. Nevertheless, this design necessitates the use of two full, independent gasoline systems, each with its own tank, fuel pump, injection pump, and injectors. In the dual fuel engines discussed above, the provided alcohol accounts for the majority of the heat release, and it is ignited by a pilot spray of diesel oil injection. The characteristics of alcohols that affect the dual fuel engine's performance are:

Various Fuels

Alcohols have a lower calorific value than diesel oil; as a result, more alcohol must be used to provide the same amount of power. However, because they require less oxygen for burning, the mixture's energy content is essentially the same. Since their latent heat of vaporization is quite high, their evaporation causes the temperature and pressure at the conclusion of compression to decrease. As a result, the engine won't run if the alcohol induction rate goes beyond a certain threshold since the injected diesel won't be able to ignite. The fundamental drawback of all the dual fuel systems mentioned above is the requirement for two distinct fuels and related components. Alcohols have a high propensity to pre-ignite in SI engines; lately, a compression ignition engine exploited this trait by using a hot surface to trigger ignition.

CONCLUSION

In this chapter discussed about the liquid fuel and its uses. Because of its high energy density, widespread availability, and established infrastructure, liquid fuel continues to be an essential source of energy for engines. Liquid fuels continue to power many different forms of transportation, such as cars, planes, and ships, despite the growing interest in alternative fuels and electric propulsion. Energy storage and portability are two key benefits of liquid fuels like petrol, diesel and aviation turbine fuel. They can have smaller gasoline tanks because of their high energy density, which enables cars to go farther between fill-ups. Furthermore, liquid fuels are widely accessible because to the already-existing fuel distribution network, providing options for refueling that are dependable and practical.

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Biodiesel Properties, Advantages and Effect on Environment

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ABSTRACT: A type of diesel fuel called biodiesel is made of long-chain fatty acid esters and is obtained from either plants or animals. It is typically created using the Trans esterification process, which involves chemically interacting lipids like animal fat, soybean oil, or some other vegetable oil with an alcohol to produce a methyl, ethyl, or propyl ester. Biodiesel is a drop-in biofuel, which means it is compatible with existing diesel engines and distribution infrastructure, in contrast to the vegetable and waste oils used to power converted diesel engines. However, as most engines cannot run on pure biodiesel without modification, it is commonly blended with petroleum fuel normally to less than 10%.

KEYWORDS: Diesel, Diesel Engine, Heating Oil, Petroleum Based, Vegetable Oil

INTRODUCTION

A type of diesel fuel called biodiesel is made of long-chain fatty acid esters and is obtained from either plants or animals. It is typically created using the Trans esterification process, which involves chemically interacting lipids like animal fat, soybean oil, or some other vegetable oil with an alcohol to produce a methyl, ethyl, or propyl ester. Biodiesel is a drop-in biofuel, which means it is compatible with existing diesel engines and distribution infrastructure, in contrast to the vegetable and waste oils used to power converted diesel engines [1], [2]. However, as most engines cannot run on pure biodiesel without modification, it is commonly blended with petroleum fuel normally to less than 10%. Blends of biodiesel can be used as heating oil as well. As a mono-alkyl ester, biodiesel is what the US National Biodiesel Board describes it as. Patrick Duffy performed the first Trans esterification of a vegetable oil in 1853, four decades before the first working diesel engine. Older methods for producing lamp oil were patented, but they weren't made public in peer-reviewed journals. On August 10, 1893, in Augsburg, Germany, Rudolf Diesel's primary model, a single 10-foot iron cylinder with a flywheel at its base, powered itself for the first time using only peanut oil. The 10th of August has been designated as International Biodiesel Day in honor of this occasion [3], [4].

Contrary to popular belief, Diesel did not intend for his engine to run on peanut oil. According to Diesel's papers, at the Paris Exhibition in 1900 Exposition Universally, the Otto Company displayed a small Diesel engine that, at the French government's request, ran on arachnid (earth-nut or pea-nut) oil. The engine operated so smoothly that only a few

people were aware of it. The engine was built to run on mineral oil, but it was used on vegetable oil without any modifications. The Arachnid, or earth-nut, which grows in large quantities and is easily cultivable in their African colonies, was once considered by the French government as a potential source of power. Later, Diesel himself performed analogous experiments and seems to be in favor of the concept. In a 1912 lecture, Diesel stated that the use of vegetable oils for engine fuels may seem insignificant today, but such oils may become, in the course of time, as important as petroleum and the coal-tar products of the present time. Despite the extensive usage of diesel fuels made from petroleum, there was interest in using vegetable oils as fuel for internal combustion engines during the 1920s, 1930s, and later during World War II in a number of nations. During this time, it was reported that countries like Belgium, France, Italy, the United Kingdom, Portugal, Germany, Brazil, Argentina, Japan, and China tested and employed vegetable oils as diesel fuels. Due to the higher viscosity of vegetable oils compared to petroleum diesel fuel, which causes poor fuel atomization in the fuel spray and frequently causes deposits and coking of the injectors, combustion chamber, and valves, several operational issues have been recorded. Vegetable oil was heated, mixed with ethanol or diesel fuel obtained from petroleum, paralyzed, and cracked in an effort to solve these issues [4], [5].

A patent for a Procedure for the transformation of vegetable oils for their uses as fuels (fry Proceed de Transformation d'Huiles Vegetables en Vie de Leer Utilization Come Carburant) was awarded to G. Chavannes of the University of Brussels (Belgium) on August 31, 1937. Patent No. 422,877 in Belgium. In order to separate the fatty acids from the glycerol by substituting short linear alcohols for the glycerol,

this patent describes the alcoholics (commonly referred to as trans esterification) of vegetable oils using ethanol (and mentions methanol). This seems to be the first account of what is currently referred to as biodiesel production. This is a duplicate of patented processes used to create lamp oil in the 18th century, and in certain cases, it's possible that antique, historical oil lamps served as inspiration. More recently, in 1977, a Brazilian scientist named Expedite Parented created the first industrial biodiesel production technology and filed for a patent on it. International standards have designated this process as biodiesel, giving it a standardized identity and quality. The automotive industry has not approved any other potential biofuels. As of 2010, Parent's business, Tec bio, is collaborating with NASA and Boeing to certify bioerosion also known as bio-kerosene, another creation of the Brazilian scientist [6], [7].

In 1979, South Africa began conducting studies on the utilization of Tran's esterified sunflower oil and its refinement to diesel fuel specifications. The procedure for creating fuel-quality, engine-tested biodiesel was finished and made public on a global scale by 1983. The South African Agricultural Engineers provided the technology to an Austrian firm called Gasworks, which built the first biodiesel pilot plant in November 1987 and the first industrial-scale facility with a capacity of 30,000 tons of rapeseed per year in April 1989. Numerous European nations, notably the Czech Republic, Germany, and Sweden, opened plants throughout the 1990s. The manufacturing of biodiesel fuel also known as dieter from rapeseed oil was started locally in France. This fuel is blended with conventional diesel at a level of 5% and with diesel used by some captive fleets such as public transit at a level of 30%. Truck engines from Renault, Peugeot, and other manufacturers can be used with up to that amount of partial biodiesel; trials with 50% biodiesel are currently underway. The domestic production of biodiesel began in other countries at the same time period; by 1998, the Austrian Biofuels Institute had recognized 21 nations with commercial biodiesel projects. Across Europe, many regular service stations now sell 100% biodiesel [8], [9].

DISCUSSION

The term biodiesel refers to a diesel engine fuel made from long-chain alkyl ethyl, methyl, or propyl esters and derived from vegetable or animal fat. The typical process for making biodiesel involves chemically interacting lipids such as vegetable oil or animal fat with an alcohol. Standard diesel engines are intended to burn biodiesel. It differs from the

waste and vegetable oils used as fuel in diesel engines. You can use biodiesel by itself or in a combination with Petro-diesel. Additionally, biodiesel can be utilised as a low-carbon substitute for heating oil. For use in the retail diesel fuel industry, blends of conventional hydrocarbon-based diesel and biodiesel are the most frequently distributed products. The quantity of biodiesel in any fuel mix is expressed using a metric known as the B factor across the majority of the world[10].

100% biodiesel and 0% petroleum-based diesel are labelled B100, 20% biofuel and 80% petroleum-based diesel are labelled B20, 5% biodiesel and 95% petroleum-based diesel are labelled B5, and 2% biodiesel and 98% petroleum-based diesel are labelled B2. Blends with less than 20% biodiesel can be utilised in diesel engines without any modifications or with very modest ones. Although pure biodiesel (B100) can be utilised, it may need to be modified in order to avoid maintenance and performance issues. In most diesel engine injection pumps, biodiesel and Petro diesel can be combined in any ratio. Where Petro-diesel has been utilised, biodiesel has been proven to dissolve deposits of residue in the fuel lines. As a result, if a rapid switch to 100% biodiesel is made, fuel filters may become clogged with debris. Therefore, it is advised to replace the fuel filters on heaters and engines soon after making the initial transition to a biodiesel blend. Everywhere in the world, biodiesel use has increased since 2005.

Manufacturing

Tran's esterification of the feedstock, such as vegetable oil or animal fat, is a typical method for producing biodiesel. For carrying out this Trans esterification reaction, the following techniques are typically used:

- i. Common batch process.
- ii. Supercritical processes.
- iii. Ultrasonic procedures.
- iv. Microwave techniques.

Tran's esterified biodiesel is made up chemically of a combination of mono-alkyl esters of long chain fatty acids. The most popular type creates methyl esters using methanol converted to sodium methoxide. These are also known as FAMES, or Fatty Acid Methyl Ester. Although ethanol can be used to create an ethyl ester, it is the least expensive alcohol that is currently accessible. It is frequently referred to as FAEE, or Fatty Acid Ethyl Ester. Higher alcohols like butane and isopropanol have also been applied. The cold flow characteristics of the resultant ester are enhanced by higher molecular weight alcohols. It comes at the expense of a less

effective Trans esterification reaction, though. The basic oil is transformed into the necessary esters through a production method called lipid Tran's esterification. Any free fatty acids (FFAs) present in the base oil are either esterified producing additional biodiesel or transformed to soap and eliminated from the process.

After this processing, biodiesel, unlike plain vegetable oil, has combustion characteristics that are extremely close to those of petroleum diesel and may take its place in the majority of present applications. Glycerol is created as a by-product of the Trans esterification process. Around the world, research is being done to utilize this glycerol as a chemical building block. This unpurified glycerol must typically be cleaned, usually by vacuum distillation. This uses a lot of energy. Glycerol that has been refined to a purity of at least 98% can subsequently be used directly or processed into other goods. Biodiesel can be made from a variety of oils. These consist of:

Virgin Oil Source: Currently, rapeseed and soybean oils are the most popular. Atrophy, mustard, flax, sunflower, palm oil, coconut, hemp, and other crops including mustard, flax, sunflower, and field pennycress are further sources of it.

Straight Vegetable Oil or Unadulterated Plant Oil: The capacity of a given economy's agricultural sector should be the only factor limiting the production of vegetable oils for use as fuels. However, doing so reduces the supply of pure vegetable oil for other uses. Animal tallow: Tallow, lard, yellow grease, chicken fat, and by-products of the synthesis of Omega-3 fatty acids from fish oil are a few examples. A by-product of meat production is animal fat. Although raising animals or fishing for fish just for their fat would not be effective. However, a small portion of Petro-diesel use might be replaced by biodiesel production from animal fat that would have otherwise been wasted.

Algae: This can be cultivated without taking up land that is currently utilised for growing food by using waste materials like sewage. Many people think used cooking oil is the best source of oil to make biodiesel, but since there isn't nearly as much of it as there is petroleum-based fuel burnt for transportation and house heating, this idea might not be practical anytime soon.

Properties

Depending on the manufacturing process and the feedstock used to produce the fuel, the color of biodiesel can range from clear to golden to dark brown. The qualities of the fuel as a result are altered as well. In general, biodiesel has a high boiling

point, a low vapor pressure, and is only mildly miscible with water. Biodiesel's flash point, which can reach 130 °C (266 °F), is far higher than petroleum diesel's, which may only reach 52 °C (126 °F). Petro diesel has a density of about 0.85 g/cm³, whereas biodiesel is about 0.88 g/cm³. Biodiesel has a calorific value of roughly 37.27 MJ/kg. This is 9% less Petro diesel than usual Number 2. More so than the manufacturing technique, variations in biodiesel energy density rely on the feedstock used. These variances, however, are lower than for Petro diesel. It's been asserted that biodiesel has better lubricity and more complete combustion, which increases engine output and partially offsets Petro diesel's higher energy density. Biodiesel has promising lubricating properties and cetin ratings compared to low Sulphur diesel fuels, and it frequently acts as an additive to ultra-low-sulfur diesel (ULSD) fuel to help with lubrication.

Biodiesel also contains almost no sulfur. Despite lacking Sulphur compounds that in petroleum diesel fuel provide much of the lubricity. Biodiesel Higher lubricity fuels may lengthen the useable life of high-pressure fuel injection machinery that depends on the lubrication provided by the fuel. This could contain high pressure injection pumps, pump injectors also known as unit injectors, and fuel injectors, depending on the engine. In most injection pump diesel engines, biodiesel can be used in pure form (B100) or combined with petroleum diesel at any concentration. The tight manufacturing restrictions for new extreme high-pressure (29,000 psi) common rail engines are B5 or B20, depending on the manufacturer. Although natural rubber gaskets and hoses in vehicles (mostly those made before 1992) tend to wear out naturally and were likely already replaced with FKM, which is nonreactive to biodiesel, biodiesel will still degrade them due to its different solvent properties. Petro diesel-using fuel lines have been reported to accumulate residue that biodiesel has been known to dissolve. As a result, if a sudden switch to pure biodiesel is made, fuel filters may become clogged with debris. Therefore, it is advised to replace the fuel filters on heaters and engines soon after making the initial transition to a biodiesel blend.

Distribution

The usage of biodiesel has increased in the US since the Energy Policy Act of 2005 was passed. By 2010, every transport fuel sold in the UK must include 5% renewable fuel, according to the UK's Renewable Transport Fuel Obligation. This effectively means 5% biodiesel (B5) in road diesel. Use of vehicles and support from manufacturers in 2005, Chrysler at the

time a division of DaimlerChrysler shipped Jeep Liberty CRD diesels with 5% biodiesel blends into the European market, demonstrating at least some acceptance of biodiesel as a legal diesel fuel additive. In 2007, DaimlerChrysler stated that if biofuel quality in the US could be standardized, it would enhance warranty coverage to 20% biodiesel mixes. According to a statement published by the Volkswagen Group, several of its cars are compatible with B5 and B100 manufactured from rapeseed oil and the EN 14214 standard. No warranty will be voided if the designated biodiesel kind is used in its vehicles.

Mercedes Benz forbids diesel fuels with a biodiesel content of more than 5% (B5) due to worries about production shortcomings. The Mercedes-Benz Limited Warranty will not cover any harm brought on by the use of such unapproved fuels. The city of Halifax, Nova Scotia, made the decision to upgrade its bus system in 2004 so that the fleet of city buses could run exclusively on a biodiesel based on fish oil. The city initially experienced some mechanical problems, but after several years of improvement, the whole fleet had been successfully converted. McDonald's of the UK stated in 2007 that it will begin making biodiesel from the used cooking oil left over from its restaurants. Its fleet would be powered by this fuel. The 2014 Chevy Cruze Clean Turbo Diesel will be rated for up to B20 a blend of 20% biodiesel and 80% normal diesel when it leaves the factory. Compatibility with biodiesel.

Properties

Compared to today's reduced Sulphur Petro-diesel fuels, biodiesel has greater lubricating qualities and substantially higher cetin ratings. Contrary to SVO/PPO, the addition of biodiesel extends the life of high-pressure fuel injection system equipment by reducing fuel system attrition. Compared to Petro-diesel, which has a calorific value of roughly 37.27 MJ/kg, biodiesel has a lower calorific value. This is 9% less Petro diesel than usual. The feedstock utilised and not the manufacturing process has a greater impact on variations in biodiesel energy density. These variances, however, are still fewer than for Petro diesel. According to some claims, biodiesel offers better lubricity and more thorough combustion, which increases engine performance and somewhat offsets the higher energy density of petroleum-based diesel. Depending on the feedstock used for manufacture, biodiesel is a liquid that ranges in color from golden to dark brown. It has a high boiling point, a low vapor pressure, and it is insoluble in water. Compared to petroleum diesel (64°C) and petrol (45°C), biodiesel has a much

higher flash point (> 130°C). Petro diesel's density (0.85 g/cc) is 0.85 g/cc lower than that of biodiesel. Because the Sulphur compounds in petroleum diesel account for the majority of the lubricity, biodiesel is frequently added as a lubricating aid to Ultra-Low Sulphur Diesel (ULSD) fuel.

Effects on the Environment

The resurgence of interest in biodiesel has brought to light a number of negative environmental repercussions related to its use. These might entail lower levels of pollution, deforestation, greenhouse gas emissions, and the rate of biodegradation. In comparison to fossil diesel, biodiesel made from soy oil typically results in a 57% decrease in greenhouse gases, and biodiesel made from waste grease results in an 86% reduction, according to the EPA's Renewable Fuel Standards Programmed Regulatory Impact Analysis, published in February 2010.

Present-Day Study

Finding better crops and increasing oil yield are still being researched. To produce enough oil to totally replace the use of fossil fuels at the current yields, enormous tracts of land and freshwater would be required. To meet the country's existing transportation and heating needs, the production of rapeseed or soybeans would need to take up two-thirds of the US's total land area. With the additional bonus that the meal left over after the oil has been pressed out can operate as an efficient and biodegradable insecticide, specially developed mustard types can generate a respectably high oil yield and are highly helpful in crop rotation with cereals.

Advantage of Biodiesel

1. As an alternative fuel, biodiesel, which is made from renewable resources including vegetable oils, animal fats, and used cooking grease, has many benefits. The following are some of the main benefits of biodiesel:
2. Plant oils, one of the renewable feedstock's used in the production of biodiesel, may be grown and refilled. Biodiesel helps lessen our reliance on non-renewable resources and promotes a more sustainable energy future, in contrast to fossil fuels, which are limited and cause climate change through carbon emissions.
3. When compared to diesel derived from petroleum, biodiesel has a substantially lower carbon impact. During combustion, it releases fewer greenhouse gases, like as carbon dioxide (CO₂) and particulates. So, by lowering smog and hazardous emissions, utilizing biodiesel can

aid in decreasing climate change and improving air quality.

4. The use of biodiesel in current diesel engines or its blend with petroleum diesel can be done without requiring any significant changes to the infrastructure or engines. Because it may easily be integrated into the current transportation infrastructure without requiring major investments or adjustments, it is a practical and convenient choice.
5. Biodiesel contains excellent lubricating qualities that can help engines last longer and wear on important parts less. Additionally, it has a higher cetin rating, which increases combustion efficiency and contributes to a smoother, quieter, and possibly more fuel-efficient engine running.
6. Energy security is improved by biodiesel because it broadens our fuel options and lessens our reliance on foreign oil. Because it can be made domestically from locally accessible feedstock's, biodiesel supports regional economic growth and lessens sensitivity to changes in the world oil markets.
7. The production and distribution of biodiesel can boost economic activity and provide up job opportunities in the manufacturing, transportation, and agricultural industries. It helps local farmers and promotes the growth of a strong and sustainable biofuel industry.

Application

Automobiles: The majority of biodiesel is consumed by mixing it in various ratios with petroleum-based diesel for use in diesel-powered vehicles. For their diesel engines, several car manufacturers advise various blends with varying percentages of biodiesel. The manufacturers' warranties on the vehicles may be voided if the owners don't utilize the biodiesel-Petro-diesel mixtures they've been advised to use. The manufacturers of the vehicles also specified the type of biodiesel that may be used in their diesel engines and, in certain circumstances, the raw materials used to make the biodiesel. Halifax, in the Canadian province of Nova Scotia, first permitted the use of fish oil-based biodiesel in its city buses in 2004. It took years to perfect the biodiesel consumption so that the buses could run efficiently on it. The conversion of the entire bus fleet was successful. In the European market during the year 2005, Daimler Chrysler first permitted the use of 5% biodiesel in the engines of the then-new Jeep model. For standardized biodiesel quality in the USA, the business said it intended to expand the warranty

coverage to 20% biodiesel mixes. The engines of Volkswagen vehicles can now use diesel fuels comprising B5 and B100 biodiesel blends manufactured from rapeseed oil and compliant with European Standard EN 14214. Only the B5 blend, or fuel containing only 5% biodiesel, is permitted for use in Mercedes Benz engines. The business has expressed worries regarding production shortcomings, or the non-uniform manufacture of biodiesel due to a variety of circumstances, including the raw materials used.

Following standardization, official advice and rules, and the availability of biodiesel in the relevant nations, many businesses with their own fleets of vehicles have begun employing a variety of biodiesel mixes in their vehicles. Some enterprises even make biodiesel from the used cooking oil left over from their operations and use it as fuel for their fleets of vehicles. For instance, McDonald's of the UK declared in 2007 that it would begin making biodiesel from the leftover cooking oil left over from its restaurants. This biodiesel would then be used to power its fleet of vehicles. The big oil firms in India either create biodiesel in-house or even purchase the fuel from other producers spread out around the nation. For use in diesel-powered cars, this biodiesel is blended with petroleum-based fuel and sold to customers at retail petrol stations. Regulations and the costs of biodiesel production determine the blending percentages.

Railways: Coal was initially the fuel for railway engines. However, with the introduction of diesel engines, the majority of railways began using diesel engines and fuel made of petroleum-based diesel. When it was practical, several of these railway engines were later converted to electricity-driven engines to pull the rakes. Numerous railway engines are still powered by diesel engines for a variety of reasons. Many of these have undergone modifications over the past ten years or more to accept biodiesel, either in pure form (B100) or in mixtures with diesel derived from petroleum.

A British railway operating firm called Virgin Trains West Coast asserted to have operated the nation's first biodiesel-powered train in 2007. They employed a B20 blend, which is a mixture of 80% petroleum-based diesel and 20% biodiesel. In the same year, Green Fuels Ltd.'s 100% biodiesel fuel (B100) was used for the first time by the British Royal Train. Since then, the firm has been running its trains in the UK entirely on biodiesel. The move to biodiesel could, however, compete with the electrification of the railway rails. In 2008, a short-line railway in Eastern Washington, United States, operated a train using a fuel mixture called B25,

which contained 25% biodiesel and 75% petroleum-based diesel. In 2009, Disneyland announced that it would stop using soybean oil in favor of biodiesel derived from its own leftover cooking oils.

For its diesel railway engines, Indian Railways began using fuel mixes that contain 5% biodiesel (B5). The conversion of the Indian railway engines to run on biodiesel was also the subject of a report. However, the majority of railway tracks are now electrified, which has increased the employment of electric locomotives. In some instances, CNG-powered locomotive engines are also in use. All of Indian Railways' diesel engines will likely be replaced with electric ones in the not-too-distant future. Additionally, being developed for use in powering trains are fuel cells and solar energy. To power their trains, several other emerging nations still use biodiesel mixes and diesel with a petroleum base.

Aviation – Aircraft: Over the past ten years, biodiesel that complies with aviation fuel regulations has been created and utilised to power a variety of aircraft types. Solaced, a renewable jet fuel made from algae, was used by United Airlines' Boeing 737-800 to fly the first commercial flight in 2011 utilizing a microbial produced biodiesel. Jet fuel was made up of 60% petroleum-based jet fuel and 40% Solaced biodiesel. The aircraft left Houston and touched down in Chicago, both in the United States. Since 2016, KLM, a Dutch airline, has used biofuel manufactured by Altair Fuels on flights departing from Los Angeles, California. For use in aero planes, a specific grade of biodiesel is now being created using a variety of source materials, including leftover cooking oils. SAF, or sustainable aviation fuel, is the name given to it nowadays. A number of businesses have begun producing this SAF biodiesel and have contracts in place to deliver it to different commercial airlines all around the world.

Ocean Liners and Sea Trains: Ocean-going ships have run on biodiesel. Additionally, the ocean ships have diesel engines that can run on mixtures of biodiesel and petroleum-based diesel. This application also makes use of a variety of mixtures. Newer fuels, like hydrogen, are currently being studied and used for ocean-going ships on a significant scale due to the trend towards using cleaner fuels. This pattern might cause biodiesel in this application to suffer.

Oil Heating: Boilers for homes and businesses can be heated with biodiesel by blending it in various ratios with petroleum-based heating oil. For certain heating applications, these mixtures or blends have been standardized in a variety of ratios. The

biodiesel blends used for heating differ slightly from those used for transportation. They are taxed differently as well. One such heating oil biodiesel blend is Bio heat, a registered trademark in the United States and Canada that combines biodiesel with conventional heating oil made of petroleum. According to ASTM 396, mixtures of up to 5% biodiesel with heating oil derived from petroleum are comparable to pure heating oil derived from petroleum. The majority of the time, mixes up to 20% biodiesel content are utilised as heating oil. The use of biodiesel blends as heating oil has the potential to harm various rubber components in furnaces, much like it does to diesel engines. To ensure the continuous operation of these furnaces, care must be taken to regularly inspect and replace those parts. To guarantee the use of a minimum quantity of biodiesel (about 2%) in heating oils, some government bodies in a number of nations have implemented laws and rules.

Energy Production

The usage of biodiesel in power generators is appropriate. Systems for generating backup power that use biodiesel as fuel come in a variety of shapes and sizes. These generators run entirely on pure biodiesel (B100), which eliminates the petroleum-based diesel's byproducts that cause pollution, ozone, and Sulphur emissions. Therefore, using these power generators that burn biodiesel results in a significant decrease in carbon monoxide and particulate matter. Thermal power production units have run on biodiesel as a fuel. Many people and organizations have started making biodiesel from various locally accessible raw materials in rural places, particularly in India. Although the produced biodiesel is of varying quality, it is used in power generators to supplement the region's limited electrical supply.

Agrotechnical Equipment: Tractors, agricultural processing equipment, and other agricultural machinery are all powered by biodiesel. The main purpose of the biodiesel usage is to run the diesel engines that drive these devices. In these applications, suitable biodiesel mixes with petroleum-based fuel are used. This end-use is more common in rural locations where biodiesel made from locally accessible raw materials is readily available but petroleum-based diesel is hard to come by.

CONCLUSION

A sustainable, biodegradable fuel made domestically from vegetable oils, animal fats, or used restaurant grease is called biodiesel. The Renewable Fuel

Standard's biomass-based diesel and total advanced biofuel requirements are both satisfied by biodiesel. Biodiesel is different from renewable diesel. In its unblended form, biodiesel is a liquid fuel also known as B100, pure, or neat biodiesel. Biodiesel is used to power compression-ignition engines, same like petroleum diesel. The physical features of biodiesel are listed in the table below. The blend of biodiesel, the feedstock, and the properties of petroleum diesel all affect how well biodiesel performs in cold climates. In general, blends with lower biodiesel percentages function better in cold climates. In cold conditions, No. 2 diesel and B5 up to 5% biodiesel typically function similarly. Some of the chemicals in both No. 2 diesel and biodiesel crystallize at very low temperatures.

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An Introduction about Gaseous Fuel and its Significance

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ABSTRACT: *The optimum fuels for IC engines are gaseous fuels since the physical delay is so little. However, as fuel displaces an equivalent amount of air, the engines could not be very efficient in terms of volume. There aren't many gaseous fuels that can be utilised in place of conventional fuels. Four categories of gaseous fuels can be distinguished: natural gas, producer gas, water gas, and coal gas. Natural gas can be extracted from the earth in its natural state through drilling tube wells, much like oil wells. Methane (CH₄) and hydrogen are the two main components that produce heat in it. When burned properly, it is the least expensive and most effective fuel; nevertheless, it needs a lot of air to burn, thus special burners are needed.*

KEYWORDS: *Gaseous Fuels, Hydrogen Air, IC Engine, Methane CH₄, Natural Gas*

INTRODUCTION

Four categories of gaseous fuels can be distinguished: natural gas, producer gas, water gas, and coal gas. Natural gas can be extracted from the earth in its natural state through drilling tube wells, much like oil wells. Methane (CH₄) and hydrogen are the two main components that produce heat in it [1], [2]. When burned properly, it is the least expensive and most effective fuel; nevertheless, it needs a lot of air to burn, thus special burners are needed. In carefully designed furnaces, producer gas is produced by driving air across a bed of glowing coal or coke. Carbon monoxide (CO), which constitutes roughly 28 to 30 percent of its total heat ingredient, is its primary component. However, it also contains some carbon dioxide and around 63% nitrogen from the air, which considerably dilute the gas and lower its calorific intensity. Because of its low cost, purity, and consistency of temperature, it is often utilised as fuel [3], [4].

About one-third of the heat value of carbon is released during the conversion process to carbon monoxide, making the gas extremely hot. A much higher temperature is attained if it is immediately led through short flues into the combustion chamber and burned with air than if it is allowed to cool first. This heat loss is generally prevented in modern gas generators by adding steam and air to the incandescent coal. The steam splits into hydrogen and oxygen, and the latter gas reacts with the carbon to make more carbon monoxide. As these gases mix with the producing gas, their calorific intensity is increased. The coal is inserted at, falls onto the step grate (B, B), and is ignited by air coming through the holes while steam is injected from the pipe, and the

gas created escapes through, in the Siemens gas producer. The pit, which is kept closed other than for cleaning, receives the ashes as they fall through the grate into it. The fuel is brought to incandescence by air being blasted through the blast pipe while the coal is resting on a bed of ashes [5], [6].

By way of the pipe, the created gas is released. The crank at turns the grate, and at, the ashes spill over the edge of the grate. On the rotating bottom, the bed of ashes is always maintained at a depth of roughly three feet. Across the blast pipe, which has a hood to disperse them across the fuel, air and steam from the pipe are introduced. Every producer gas plant makes use of the regenerative heating system. One example of this sort of heating is the Siemens regenerative boiler. The material that has to be heated is placed on the furnace's hearth. The checker work is the loosely stacked fire-brick that fills the four pathways B, C, D, and E. The hot gases from the furnace flow through and heat two checkerboards, such as (B) and (C), on their way to the chimney. The flow of furnace gases transforms into (D) and (E) when heated enough, via which they exit to the chimney. Following that, fuel gas is transported through the hot tube to the furnace, where it mixes [7], [8].

As a result, (A) is significantly hotter than it would be if the air and gas had arrived at (A) cool. After some time, the dampers are turned, forcing the gas to pass through (E) and the air through (D), while the combustion byproducts pass via (B) and (C) to the chimney. As (B) and (C) are being thusly cooled, (D) and (E) are being heated by the furnace gases. As a result, the procedure involves an alternation of heating the checker work on one side and releasing heat to the gas or air on the other. The combustion gases are occasionally passed through flues with narrow tubes, through which the gas and air are

passing to the furnace, in the opposite direction to that taken by the fire gases, as ashes and soot frequently clog the spaces between the bricks of the checker work. Over 30% carbon monoxide and over 63% nitrogen can be found in blast furnace waste gases. These gases are primarily used for heating close to the furnaces. Water gas is occasionally used as a fuel, but it is more frequently used as an ingredient in illuminating gas. It is created by blowing steam over incandescent anthracite coal or coke, and contains small amounts of nitrogen, oxygen, and carbon dioxide as well as about 45% each of carbon monoxide and hydrogen.

Smog or smoke. It has a calorific value per cubic meter of roughly 3000 C. Anthracite has a higher output than coke, which yields roughly 1.13 cubic meters of water gas per kilogram me. A burst of air is used to ignite the fuel, and during this phase of the process, the heat is typically lost. The air is cut off and the steam is turned on when it reaches a temperature of white hot; this causes decomposition, as indicated by the first reaction above. The steam is switched off as soon as the temperature drops below 1000° C. and the air blast is activated until the coal is once again white hot. Steam and air are thus continuously blown out in alternation. Although the generator gas created by the air blast is occasionally retained and used, it is wasted when producing lighting gas. This water gas is enriched with naphtha to make lighting gas. Bituminous coal is distilled in retorts to produce coal gas. It has significant amounts of hydrogen and marsh gas roughly 40% of each as well as trace amounts of illuminating hydrocarbons from the C_nH_{2n} and C_nH_{2n-2} series, carbon monoxide, carbon dioxide, nitrogen, and oxygen. It only plays a little role in household stoves and as a fuel source for petrol motors.

DISCUSSION

Since physical delay is practically nonexistent, gaseous fuels are ideally suited for IC engines. The engines may, however, have low volumetric efficiency since fuel displaces air in an identical amount. Few gaseous fuels can be utilised as substitutes for conventional fuels. The sections that follow have a detailed discussion of them. Numerous car companies have created prototype or modified engines that run on hydrogen fuel. As a fuel for IC engines, hydrogen offers the following benefits: Low emission levels. The exhaust essentially contains no CO or HC because the fuel has no carbon. Most emissions would be made up of NO_x, N₂, and water.

1. The presence of fuel. Making hydrogen can be accomplished by a variety of processes, such as water electrolysis.
2. Fuel spills into the environment do not constitute pollution.
3. A liquid that is kept has a high energy content per volume. This would result in a significant increase in vehicle range for a given fuel tank capacity, however consider the following.
4. Using hydrogen as a fuel has a number of drawbacks, including:
5. The need for large, heavy fuel tanks, both inside the car and at the petrol stations. Storage options for hydrogen include compressed gas and cryogenic liquid. It would need to be kept under pressure at a very low temperature if it were to be stored as a liquid, necessitating a fuel tank with superior thermal insulation. A vessel with a high pressure and a small volume would be needed for gas phase storage.
6. A detonation could occur and refueling would be difficult.
7. Insufficient engine volumetric efficiency. Any time a gaseous fuel is utilised in an engine, the fuel will displace some of the input air, which will lead to lower volumetric efficiency.
8. With today's technology and supply, fuel would be expensive.
9. High NO_x emissions as a result of a hot flame.
10. A rotary Winkle engine that runs on hydrogen fuel has been modified by the automaker Mazda. The design of the fuel intake was done with the utmost care because hydrogen fuel ignites quite easily. Metal hydride fuel storage is employed by this test vehicle.

Hydrogen Engines

Another alternative fuel for IC engines is hydrogen. Numerous countries conducted in-depth investigations. The production of hydrogen from water, a potentially abundant raw material, and the principal byproduct of its combustion, water, make it one of the most desirable fuels for IC engines. Both as a gas and as a liquid, hydrogen has an extremely low density. As a result, although having a high calorific value on a mass basis, ethanol has a liquid energy density that is just one-fourth that of petrol. It is less dense as a gas than air, and it has a heating value per unit volume that is less than one third that of methane. One of its main drawbacks is this. None of the storage options for hydrogen are as practical as those for petrol. It must be kept as a compressed gas, a liquid in cryogenic containers or an absorbed state as metal hydrides.

Wide restrictions for hydrogen's ignition exist. This has the distinct advantage of enabling hydrogen-powered spark ignition engines to run with barely any throttling. The comparable petrol air mixture burns seven times more quickly than the stoichiometric hydrogen air mixture. This is also a significant benefit for IC engines, resulting in faster engine speeds and increased thermal efficiency. Despite having a high self-ignition temperature, hydrogen only needs a little amount of energy to ignite. In SI engines, it is hence extremely susceptible to resignation and backlash. Hydrogen has a slightly lower adiabatic flame temperature than petrol, but because of the quick burning, relatively little heat is lost to the environment, leading to high, immediate local temperatures. High levels of nitric oxide are formed as a result. There are three ways that hydrogen can be used in SI engines:

- i. Through manifold induction.
- ii. Through the addition of hydrogen directly into the cylinder.
- iii. By adding fuel to the mix.

Cold hydrogen is injected into the manifold using a valve-controlled route in the manifold introduction of hydrogen process. This lessens the possibility of a back flash. Resignation and back flash are the two elements that restrict the engine's ability to produce power. Additionally, compared to liquid hydrocarbon fuels, the energy content of an air hydrogen mixture is lower. Hydrogen is kept in liquid form in a cryogenic cylinder for direct input into the cylinder. This liquid is sent via a pump to a tiny heat exchanger, where it is heated to produce cold hydrogen gas. This machine also handles the hydrogen metering. In addition to lowering NO_x production, the cold hydrogen aids in the prevention of resignation. In SI engines, hydrogen can also be used as an additional fuel to petrol. In this system, petrol and hydrogen are introduced together, compressed, and ignited by a spark. Figures 1 and 2, respectively, show the arrangements for liquid hydrogen storage and specifics of hydrogen induction into the SI engine cylinder.

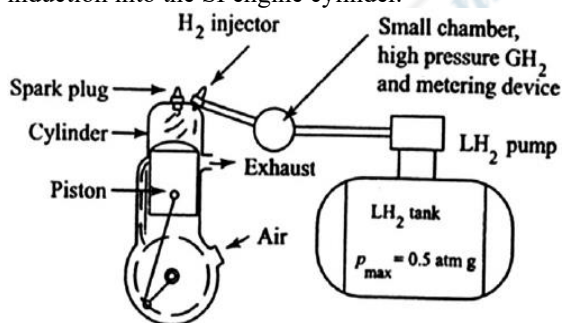


Figure 1: Gaseous hydrogen injection and liquid hydrogen storage system [Research Gate].

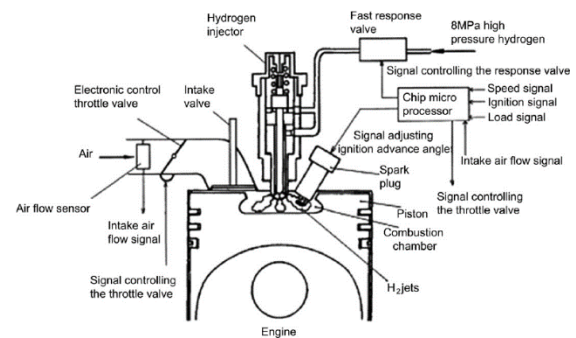


Figure 2: Spark-ignition engine with hydrogen induction [Research Gate].

Hydrogen may be used in diesel engines in two different ways. By mixing hydrogen with air and lighting the combination, which is employing the dual fuel mode, with a spray of diesel oil. The limiting conditions include when the amount of diesel is insufficient to generate an effective ignition, which is known as failure of ignition, and when the hydrogen air combination is too rich, causing an intolerably violent combustion. A wide range of diesel to hydrogen proportions can be tolerated between these limitations. Investigations have shown that substitution of hydrogen for diesel fuel above a specific point result in a severe rise in pressure. Directly injecting hydrogen into the cylinder after compression. Since hydrogen has a very high self-ignition temperature, surface ignition occurs when the gas spray impinges on a hot glow plug inside the combustion chamber. Another option is to fuel an engine with a relatively lean mixture of hydrogen and air, and then add the majority of the hydrogen near the conclusion of the compression stroke. Hydrogen needs to be handled with extreme caution because it is a fuel that is very reactive. To stop flash back from entering the storage tank, flash black arresters must be placed between the engine and the container.

Petrol

Oil and gas containing sand strata can be found in numerous locations in the earth's interior at varying depths. The gas typically comes out of the oil well naturally and is under a lot of pressure. The gas must first be removed from any entrained sand if it is to be used in an engine close to the well. Casing head gas is the name given to the natural gas extracted from oil wells. Usually, the treatment is done to recover the petrol. After that, it is referred to as dry gas. To be used as fuel, it is transported into the pipeline networks. The creation of natural petrol is possible using natural gas. Methane makes up the majority of natural gas (60–95%), with trace amounts of other hydrocarbon fuel components.

From location to location and from time to time, the makeup varies greatly. But often have a high concentration of methane (CH₄) and a low concentration of ethane (C₂H₅). It also contains varying concentrations to significant amounts. It is kept as liquid natural gas (LNG) at pressures of 70 to 210 bar at a temperature of around 160°C, or as compressed natural gas (CNG) at pressures of 16 to 25 bar. In an engine system with a single-throttle body fuel injector, it performs best as fuel. This results in a longer mixing period, which is what this fuel requires. Governmental organizations and the business sector continue to test CNG in different sized cars. Hydrogen may be used in diesel engines in two different ways.

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makeup varies greatly. But often have a high concentration of methane (CH₄) and a low concentration of ethane (C₂H₅). It also contains varying concentrations of N₂, CO₂, He, and traces of other gases. Its Sulphur level varies, ranging from almost none (sweet) to significant amounts (sour). It is kept as liquid natural gas (LNG) at pressures of 70 to 210 bar at a temperature of around 160°C, or as compressed natural gas (CNG) at pressures of 16 to 25 bar. In an engine system with a single-throttle body fuel injector, it performs best as fuel. This results in a longer mixing period, which is what this fuel requires. Governmental organizations and the business sector continue to test CNG in different sized cars.

Natural Gas's Benefits

Natural gas has several benefits as a fuel.

The fuel is excellent for SI engines due to its high-octane rating of roughly 110. The higher flame speed and higher compression ratio that engines may operate at are both results of the high-octane number. Low engine emissions. Less aldehydes compared to methanol. Fuel is pretty commonplace throughout the world. Coal can be used to make it, but the process is quite expensive [9], [10].

Natural Gas Disadvantages

Natural gas has the following drawbacks as an engine fuel. Low engine performance due to low energy density. Due to the fuel being gaseous, the engine's volumetric efficiency is low. A sizable pressurized fuel storage tank is required. A pressurized gasoline tank raises certain safety questions. Variable fuel characteristics. Refueling is a drawn-out procedure. Some extremely large stationary CI engines run on a fuel mix of diesel and methane. More than 90% of the fuel is methane, making it the primary fuel. High-pressure pipelines deliver it as a gas to the engine. For ignition, a tiny amount of high-grade, low-sulfur diesel fuel is used. The end result is an engine that runs very cleanly. Large ships might also benefit from these engines as power sources, however high-pressure gas pipes are not ideal for use on ships.

Well drilling is a method of obtaining petroleum and natural gas. As is already common knowledge, hydrocarbons make up crude petroleum. It has some water, sulphur, and other impurities in it. When petroleum and natural gas are combined, a highly volatile liquid is created. Natural petrol is the name of this substance. The petrol condenses as the mixture of petroleum and natural gas cools. Compressed Natural Gas (CNG) is the name given to the natural gas after it has been compressed. Just like LPG, CNG is used to power vehicles. The LPG

fuel feed system and the CNG fuel feed system are comparable. Petrol-powered vehicles can be converted to run on compressed natural gas using CNG conversion kits. These kits include auxiliary components including the mixer, converter, and other crucial components needed for conversion. Liquefied petroleum gas (LPG), Oil and gas wells provide propane and butane. They are also byproducts of the refinement of petroleum. There are two types of LPG utilised for car engines. The first is butane, and the second is propane. A propane and butane mixture are occasionally utilised in automotive engines as liquid petroleum gas. In place of petrol, liquid petroleum gases are used as fuel. They are frequently utilised in vehicles, autos, and buses. Petroleum gases are compressed into liquid form and then cooled. The pressure tanks used to store this liquid are sealed.

LPG's Benefits and Drawbacks

The potential for liquefied petroleum gas as a substitute fuel for IC engines is greater. The benefits and drawbacks of utilizing LPG are as follows:

- i. LPG has a lower carbon content than petrol. Even while LPG-powered vehicles only emit slightly less nitrogen compounds every kilometer, they produce 50% less carbon monoxide. Therefore, using LPG greatly reduces emission.
- ii. LPG and air combine in all conditions.
- iii. In multi-cylinder engines, all cylinders can get a consistent mixture.
- iv. (iv)There is no crankcase dilution because the fuel is in the form of a vapor.
- v. If an automobile engine has a high compression ratio (10:1), propane can be used in it.
- vi. LPG exhibits strong antiknock properties.
- vii. Its heat energy is roughly 80% that of petrol, but its high-octane rating makes up for the engine's lack of thermal efficiency.
- viii. Using LPG as fuel results in a 50% cost reduction.
- ix. The engine's lifespan might be increased by 50%.

Disadvantages

Typically, engines are built to ingest a certain volume of the fuel and air mixture. As a result, at full throttle, LPG will result in 10% fewer horsepower for the given engine. LPG has a somewhat greater ignition temperature than petrol. Consequently, operating on LPG could result in a 5% reduction in valve life. A strong cooling system is essential, as the LPG vaporizer relies on engine coolant to generate heat that turns liquid LPG into gas. The use of heavy

pressure cylinders to store LPG results in an increase in the weight of the vehicle. Liquid petroleum gas needs to be fed using a particular mechanism. Future LPG Vehicle Scenario The public will be enticed to utilize LPG-powered automobiles by cost savings, an extended engine life, and lower emissions.

LPG's potential If the system is improved in the ways listed below, vehicles are bright. LPG cylinders are currently used in vehicles in numerous nations. An issue with the cylinders is their weight. The power used to pull these cylinders behind the cars wastes some energy. However, the majority of LPG vehicles in developed nations use LPG tanks. The tank typically fits snugly into the spare wheel's area and is the same size as the spare. Such LPG tanks should be used in LPG-powered vehicles instead of cylinders. Efforts should be made to increase the number of LPG filling stations in practical locations so that LPG tanks may be conveniently filled. Safety devices must be implemented to stop accidents brought on by gas cylinder explosions or gas line leaks.

Petrol (LPG) Fuel Feed System

For a lot of years, propane has been tested in commercial vehicles. With around 60% less CO, 30% less HC, and 20% less NOx emissions than petrol, it is a good high octane SI engine fuel. Propane is kept under pressure as a liquid and fed to the engine via a high-pressure line, where it is vaporized. Therefore, appropriate and sufficient safety measures must be implemented before installing the gasoline feed system in the vehicle. Otherwise, there is always a chance that gas leakage will result in fire hazards. Because it is a gaseous fuel, it has a reduced volumetric efficiency in engines. Depicts the fuel feed system for LPG. The storage tank for this gasoline supply system is located in the car's back. The converter-regulator, the LPG propane carburetor, and the vacuum filter fuel lock are located on the system's front side.

Liquid LPG is forced via the fuel pipe line to the converter in the fuel feed system. The converter transforms the liquid into vapor. As LPG transforms from a liquid to a gas, the temperature drops significantly. It should be kept from freezing within the converter while using LPG. This keeps the gas from freezing by passing engine cooling water very close to the converter. LPG fumes travel to the propane LPG carburetor. The engine receives the air-fuel combination from this carburetor for burning. Currently, LPG auto conversion kits are offered for converting gasoline-powered vehicles to LPG gas-powered vehicles. These kits include extra parts required for converting gasoline-powered

vehicles to LPG-powered vehicles, such as the converter, pressure regulator, mixer, and other LPG vapor/air mixing components.

CONCLUSION

Natural gas, producer gas, water gas, and coal gas are the four categories into which gaseous fuels can be categorized. Similar to petroleum wells, drilling tube wells are used to extract natural gas from the earth, which is already in its formed state. Methane (CH₄) and hydrogen are its primary heat-producing components. When properly burned, it is the most affordable and effective fuel; nevertheless, it needs a lot of air to be burned, thus special burners are needed. Producer gas is created in specially designed furnaces by blowing air through a bed of glowing coal or coke. Carbon monoxide (CO), which makes up roughly 28 to 30 percent of it, is the primary source of heat in it. However, it also contains some carbon dioxide and around 63% nitrogen from the air, which considerably diminish the calorific intensity of the gas and greatly dilute it. It is widely utilised as fuel since it is affordable, clean, and consistently produces the desired temperature.

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Carburetion and Principle of Carburetion

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ABSTRACT: Typically, liquid fuels that are volatile are used in spark-ignition engines. A homogenous mixture is typically not formed at the inlet manifold since the preparation of the fuel-air mixture is done outside the engine cylinder. In this chapter discussed about the carburetion and the principle of the carburetion. Even during the suction and compression processes, fuel droplets that are in suspension continue to evaporate and mix with air. Spark-ignition engines place a lot of importance on the preparation of the mixture. Carburetion's main goal is to supply a combustible mixture of gasoline and air in the precise amount and quality needed for the engine to run effectively in all circumstances.

KEYWORDS: Air Fuel Ratio, Combustion Chamber, Engine Cylinder, Fuel Air, Homogenous Mixture

INTRODUCTION

Typically, liquid fuels that are volatile are used in spark-ignition engines. A homogenous mixture is typically not formed at the inlet manifold since the preparation of the fuel-air mixture is done outside the engine cylinder. Even during the suction and compression processes, the fuel droplets that are still in suspension continue to evaporate and mix with the air. Spark-ignition engines place a lot of importance on the preparation of the mixture. Carburetion's main goal is to supply a combustible mixture of gasoline and air in the precise amount and quality needed for the engine to run effectively in all circumstances [1], [2].

Carburetion is the act of creating a combustible fuel-air combination by combining the right amount of fuel with air prior to admission to the engine cylinder, and the tool that does this task is known as a carburetor. Volatile liquid fuels are typically used in spark-ignition engines. The preparation of the fuel-air combination takes place outside the engine cylinder, and the input manifold is typically not where the homogenous mixture is fully formed [3], [4]. Even during the suction and compression processes, fuel droplets that are still in suspension continue to evaporate and mix with the air. For spark-ignition engines, the preparation of the mixture is critical. The goal of carburetion is to supply a combustible mixture of gasoline and air in the necessary quantity and quality for the engine to run effectively under all circumstances [5], [6].

Carburetion Affecting Factors

The carburetion process is affected by a number of variables, including

- i. Engine speed.
- ii. Fuel vaporization properties.
- iii. Air temperature.
- iv. Carburetor design.

Due to the high-speed nature of current engines, there is relatively little time for mixture formation. For instance, the amount of time an engine has for mixture induction during the intake stroke at 3000 rpm is just about 10 milliseconds (ms). The time allowed is barely 5 ms when the speed reaches 6000 rpm. In order to achieve high quality carburetion (i.e., a mixture with a high vapor content), the air stream velocity at the fuel injection site must be raised. By placing a venturi portion in the air's passage, this is accomplished. At the venturi's throat, which is its narrowest cross section, the main metering jet releases its fuel. The presence of highly volatile hydrocarbons in the gasoline is another aspect that guarantees good grade carburetion quickly. Therefore, effective carburetion, especially at high engine speeds, requires the gasoline to have proper evaporation characteristics, as indicated by its distillation curve. The effectiveness of carburetion is significantly influenced by the temperature and pressure of the surrounding air [7], [8].

A more homogenous mixture is created by increased vaporization of fuel the fraction of fuel vapor increases with increase in mixture temperature. However, while the air-fuel ratio remains unchanged, an increase in atmospheric temperature causes the engine's power output to fall because there is less mass flow into the cylinder, or, in other words, there is less volumetric efficiency. The uniform distribution of mixture to the different engine cylinders is greatly influenced by the design of the carburetor, the intake system, and the combustion chamber. Under various engine running situations, only properly designed carburetor components can guarantee the supply of the desired composition of the mixture [9], [10].

Mixtures of Air and Fuel

Typically, an engine is run at a variety of speeds and loads. To achieve this, the engine cylinder should receive the right air-fuel mixture. Three different sorts of mixes are created by combining fuel and air. Chemically sound mixture, rich mixture, and lean mixture, respectively. A stoichiometric mixture is one that contains just the right amount of air to allow the fuel to burn completely. For instance, 15.12 kilograms of air are needed to totally burn 1 kg of octane (C₈H₁₈). Therefore, the C₈H₁₈ A/F ratio is 15.12:1, which is typically rounded to 15:1. The numerical value of this chemically sound mixture will only slightly alter between various hydrocarbon fuels.

It is always calculated using the chemical formula for full combustion of a certain fuel. Complete combustion occurs when all of the fuel's carbon and hydrogen are transformed into CO₂ and H₂O, respectively. A mixture is referred to as rich if it includes less air than the stoichiometric need for instance, an A/F ratio of 12:1, 10:1, etc. Lean mixtures, such as those with an A/F ratio of 17:1, 20:1, etc., contain more air than the stoichiometric need. However, combustion in a SI engine will only take place within a specific range of A/F ratios in a homogenous mixture. A ratio that is either too rich or too lean to support flame propagation is outside of this range. this useful A/F ratio ranges from around 9:1 (rich) to 19:1 (lean). According to the needs of the engine, the carburetor should give an A/F ratio, and this ratio must fall within the combustible range.

DISCUSSION

Requirements for Mixture at Various Loads and Speeds

The air-fuel ratio that an engine runs at has a big impact on how well it performs. Think about an engine with a variable A/F ratio that is running at full throttle and constant speed. The typical curves shown in Figure 1 indicate that under these circumstances, the A/F ratio will affect both the power output and the brake specific fuel consumption, and the mixture with the maximum output on the curve is referred to as the best power mixture with an A/F ratio of approximately 12:1. The best economy blend is the one that corresponds to the least point on the bsfc curve. The A/F ratio is roughly 16 to 1. It should be noted that the best economy mixture is slightly leaner than the chemically right mixture while the best power mixture is significantly richer than the latter. Figure

1 shows how a SI engine's power output and bsfc vary as a function of the air-fuel ratio.

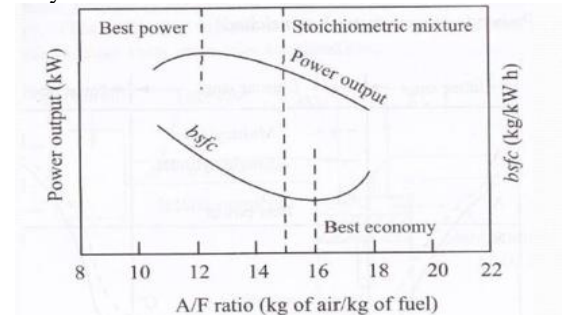


Figure 1: Representing how a SI engine's power output and bsfc vary as a function of the air-fuel ratio [Slide Player].

The operation shown in Figure 1 is at full throttle. At part throttle, the ideal A/F ratios for power and economy are not exactly the same as those at full throttle. The optimal fuel metering mechanism would be a simple two position carburetor if the A/F ratios for the best power and economy are constant over the full range of throttle action and if the influence of other factors is ignored. When maximum performance is sought, a carburetor of this type could be tuned for the best power mixture, and when fuel efficiency is the main concern, for the best economy mixture. The solid horizontal lines X-X and Z-Z denote these two configurations, respectively. However, actual engine requirements once more prevent the implementation of such a straightforward and practical system. The section that follows talks about these specifications. Under typical circumstances, it is preferable to operate the engine with the highest possible economy mixture, or roughly a 16:1 air-to-fuel ratio. Rich mixture, or a 12:1 air-to-fuel ratio, is needed for rapid acceleration and for optimum power

Requirements for Air-Fuel Mixture in Automotive Engines

The actual requirements for an air-fuel mixture in an automobile engine differ greatly from the ideal circumstances covered in the preceding section. The carburetor must supply mixtures that closely match the overall form of the curves ABCD for single cylinder engines and A B C D for multicylinder engines in Figure 2, which depicts a typical need for an automotive engine. To properly fulfil the varied engine requirements, the carburetor must be constructed. There are three general ranges of throttle functioning, as shown in Figure 2. The demands placed on the automobile engines vary in each of them. In order to meet these needs, the carburetor must be able to provide the necessary air-fuel ratio.

Idling mixture must be enhanced. Cruising mixture needs to be lean. High Power must be enriched combination. An engine that runs with no load and a nearly closed throttle is said to be idling. As shown by point A in Figure 2, the engine needs a rich mixture when it is idle. This is caused by the exhaust gas dilution of the fresh charge caused by the pressure conditions that already exist in the combustion chamber and the intake manifold. Typical values that exist when the engine is idling. Regardless of the throttle position, the exhaust gas pressure at the end of the exhaust stroke does not deviate significantly from the value. The mass of exhaust gas in the cylinder at the conclusion of the exhaust stroke tends to stay fairly constant throughout the idling range since the clearance volume is constant. Due to the extremely small opening of the throttle, however, the amount of fresh charge introduced during idling is far lower than that during full throttle operation. As a result, under idle conditions, a substantially greater quantity of exhaust gas is mixed with the fresh charge.

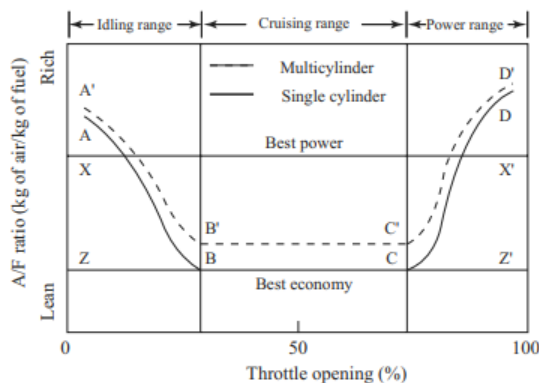


Figure 2: illustrates anticipated carburetor performance to meet engine needs ic engine book [Ftp.Idu.Ac.Id].

Additionally, due to air flow limitation, the pressure in the intake manifold is significantly below atmospheric with nearly closed throttle. Exhaust gases initially flow backward into the intake manifold when the intake valve opens due to the pressure difference between the combustion chamber and the intake manifold. These exhaust gases are pulled back into the cylinder with the new charge as the piston descends on the intake stroke. As a result, exhaust gas dilutes the final mixture of fuel and air in the combustion chamber more. This exhaust stream tends to prevent the contact between fuel and air particles, which is essential for combustion. As a result, there is inefficient combustion and a loss of power. Therefore, it is required to enrich the air-fuel mixture in order to give additional fuel particles. This richening

improves combustion by raising the likelihood of fuel and air particle interaction.

Power Spectrum

For the following reasons, the engine needs a richer mixture when operating at peak power, as shown by the line CD. To offer the best power: Since high power is required, it makes sense to go from the economy cruising range settings to the mixture that will create the most power, or to a level that is close to the optimal power mixture, typically in the 12:1 range. To avoid exhaust valve and nearby region overheating: High power requires more heat to be transferred away from sensitive regions, like those near the exhaust valve, due to the larger amount of gas moving through the cylinder at higher temperatures. The temperature of the flame and the cylinder are lowered by enriching the mixture. This lessens the cooling issue as well as the propensity for exhaust valve damage at high power.

The mass of the charge is lower and there is less of a tendency to burn the exhaust valve in the cruising range. In an automotive engine, a knocking indicator is present in the form of an audible sound, and the operator can reduce the demands placed on the engine by releasing the throttle or downshifting. A complicated and expensive system for enrichment for this purpose is not economically possible, despite the fact that various methods of richening at high power are typically included. Additionally, automotive engines typically operate well below full power. The complexity and cost of installing an aircraft engine are justified by the need to enhance power during takeoff. Therefore, Figure 2 provides a better representation of the carburetor needs for a typical engine. The needs for an automobile engine are similar in the cruising and idling ranges, but they tend to be lower or less rich in the power rang displays a more accurate engine demand curve for autos. The requirements after the throttle is opened wide and the load is further raised are seen in the section of the curve from D to E.

The Carburetion Principle

The suction produced by the piston's descent draws both gasoline and air into the engine cylinders and through the carburetor. This suction results from a rise in the cylinder's volume and a corresponding fall in the gas pressure inside of this chamber. The air flows into the chamber as a result of the pressure differential between the atmosphere and the cylinder. Air entering the combustion chamber of the carburetor picks up fuel that is released via a tube. A tiny hole called a carburetor jet is present in this tube and is open to the air flow. The pressure head or difference in pressure between the float chamber and

the venture's throat, as well as the size of the tube's outlet, determine how quickly fuel is released into the atmosphere. The suction action must be strong and the nozzle output relatively tiny in order for the fuel extracted from the nozzle to be completely atomized. The pipe in the carburetor that carries air to the engine is engineered to have a restriction in order to provide a powerful suction. Due to the increase in flow velocity at this limitation known as the throat, a suction action is produced.

The venture or throat of the carburetor is where the fuel jet's terminus is situated. The air must pass through a much smaller flow region since the channel is narrower in the core. The air will move at its fastest speed via the tube's narrowest point because the same amount of air must pass through all of the tube's points. The suction will increase proportionally as the area decreases due to the higher air velocity. As was previously indicated, the fuel discharge jet's entrance is often situated where suction is greatest. This often lies directly below the venture tube's narrowest point. In this area, the air coming from the venture tube and the petrol spray from the nozzle combine to create a combustible combination that travels through the intake manifold and into the cylinders. A minor portion of the fuel vaporizes while the majority of it is atomized. The rate of fuel evaporation is sped up by increased air velocity at the venture's throat. The higher air velocity at the venture throat alone cannot fully address the challenge of creating a mixture of sufficiently high fuel vapor-air ratio for effective starting of the engine and for uniform fuel-air ratio in different cylinders in case of multicylinder engine.

The suction produced by the piston's descent draws both gasoline and air into the engine cylinders and through the carburetor. This suction results from a rise in the cylinder's volume and a corresponding fall in the gas pressure inside of this chamber. The air flows into the chamber as a result of the pressure differential between the atmosphere and the cylinder. Air entering the combustion chamber of the carburetor takes up gas released from a tube. The air passage is exposed to a tiny opening on this tube known as the carburetor jet. The pressure head or difference in pressure between the float chamber and the venture's throat, as well as the size of the tube's outlet, determine how quickly fuel is released into the atmosphere. The suction action must be strong and the nozzle output relatively tiny in order for the fuel extracted from the nozzle to be completely atomized.

The pipe in the carburetor that carries air to the engine is engineered to have a restriction in order to

provide a powerful suction. Due to the increase in flow velocity at this limitation known as the throat, a suction action is produced. To reduce throttling losses, the restriction is constructed in the shape of a venturi. The venture or throat of the carburetor is where the fuel jet's terminus is situated. The venture tube's geometry is depicted. The air must pass through a much smaller flow region since the channel is narrower in the center. The air will move at its fastest speed via the tube's narrowest point because the same amount of air must pass through all of the tube's points. The suction will grow proportionately as the region shrinks because the air will move more quickly. As was already indicated, the fuel discharge jet's hole is often lopped where the suction is greatest. This often lies directly below the venture tube's narrowest point. In this area, the air coming from the venture tube and the petrol spray from the nozzle combine to create a combustible combination that travels through the intake manifold and into the cylinders.

A little portion of the fuel will be vaporized while the majority of it is atomized. An increase in air velocity at the venture's throat speeds up fuel evaporation. The higher air velocity at the venture throat alone cannot fully address the challenge of obtaining a mixture of sufficiently high fuel vapor-air ratio for effective starting of the engine and for uniform fuel-air ratio indifferent cylinders in case of multi-cylinder engine. The inlet manifold is upstream of the carburetor. When fuel is supplied within the carburetor, atmospheric air enters the device often via an air cleaner, travels through the inlet valve, and then enters the combustion chamber. Although certain high-performance engines have employed numerous carburetors, the majority of engines have a single carburetor shared by all of the cylinders. Based on the Bernoulli principle, the carburetor draws more gasoline into the airstream as engine speed increases because the static pressure of the intake air decreases. The majority of the time, when a driver presses the throttle pedal, less fuel enters the engine with the exception of the accelerator pump. Instead, the amount of fuel pulled into the intake mixture grows together with the airflow through the carburetor.

Since a carburetor is a fluid dynamic device, the pressure reduction in a venture tends to be proportional to the square of the intake airspeed, which is the fundamental drawback of basing a carburetor's operation on Bernoulli's Principle. The fuel flow tends to be proportionate to the pressure difference since the fuel jets are considerably smaller and the fuel's viscosity is the key factor

limiting fuel flow. Therefore, at lower speeds and reduced throttle, jets that are designed for full power sometimes starve the engine. Usually, this has been rectified by employing several jets. It was fixed in SU and other variable jet carburetors by adjusting the jet size. A crucial design factor is the carburetor's orientation. Updraft carburetors, where air enters from below and leaves via the top, were utilised with older engines. Side draught carburetors and downdraft carburetors primarily in the United States started to be used more frequently starting in the late 1930s.

CONCLUSION

Volatile liquid fuels are typically used in spark-ignition engines. The preparation of the fuel-air combination takes place outside the engine cylinder, and the input manifold is typically not where the homogenous mixture is fully formed. Even during the suction and compression processes, fuel droplets that are still in suspension continue to evaporate and mix with the air. For spark-ignition engines, the preparation of the mixture is critical. The goal of carburetion is to supply a combustible mixture of gasoline and air in the necessary quantity and quality for the engine to run effectively under all circumstances.

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Working Operation of Simple Carburetor and Its Advantages

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ABSTRACT: Carburetors are really complicated. Let's first comprehend how a basic or simple carburetor function in order to give an air-fuel combination for cruising or typical range at a single speed. Later, more mechanisms will be added to support the many unique requirements, including acceleration, variable load and speed operation, beginning, and idling. The main fuel jet, venture, jet tube, throttle valve, float chamber, and other components make up a simple carburetor. In the float chamber where a float is present. Fuel is delivered from the fuel tank via the strainer and into the float chamber with the aid of a fuel pump. This entire unit can be referred to as a simple carburetor.

KEYWORDS: Basic Carburetor, Discharge Nozzle, Float Chamber, Fuel Injection, Main Discharge

INTRODUCTION

Carburetors are really complicated. Let's first comprehend how a basic or simple carburetor function in order to give an air-fuel combination for cruising or typical range at a single speed. Later, more mechanisms will be added to support the many unique requirements, including acceleration, variable load and speed operation, beginning, and idling [1], [2]. A straightforward carburetor's details. The basic components of a carburetor are a float chamber, throttle valve, choke, fuel discharge nozzle, and metering orifice [3], [4]. The level of petrol in the float chamber is kept constant by the float and a needle valve mechanism. The float descends, opening the fuel supply valve and allowing fuel to enter if the level of fuel in the float chamber is below the desired level. The float closes the gasoline supply valve once the intended level has been reached, preventing further fuel flow from the supply system. Either the atmosphere or the side of the venture upstream are where the float chamber is vented. Air is pulled through the venture during the suction stroke. Ventura, as previously mentioned, is a tube with a decreasing cross-section and a small neck area [5], [6].

The venture tube, sometimes referred to as the choke tube, is formed in such a way as to provide the least amount of resistance to the air flow. The air velocity rises as it moves through the venture, peaking at the venture throat. As a result, the pressure drops until it is at its lowest. Fuel is delivered from the float chamber to a discharge jet, the tip of which is situated in the venture's throat. Fuel is released into the air stream as a result of the carburetor depression, or pressure difference between the float chamber and the venture's throat. The fuel discharge

is influenced by the discharge jet's size, which is selected to provide the necessary air-fuel ratio. When the throttle is fully open, the pressure at the throat is typically between 4 and 5 cm Hg and seldom more than 8 cm Hg. The liquid level in the float chamber is kept at a level just below the discharge jet's tip to prevent fuel from overflowing through the jet. The nozzle tip is what we refer to as here [7], [8].

The petrol engine is quantity controlled, which means that to change the amount of power output at a specific speed, one must change the amount of charge delivered to the cylinder. This is accomplished by placing a throttle valve typically a butterfly valve after the venture tube. As the throttle is closed, less air passes through the venture tube and less air-fuel mixture are fed to the cylinder, resulting in a decrease in power output. More air passes through the choke tube as the throttle is opened, which increases the amount of mixture fed to the engine. This boosts the engine's production of power. A basic flaw with a simple carburetor of the sort mentioned above is that it only supplies the necessary A/F ratio at one throttle position. Depending on how far the throttle is opened, the mixture is either richer or leaner at the other throttle locations. The air flow varies with the throttle opening, which results in a specific pressure difference between the venture throat and the float chamber [9], [10].

The fuel flow via the nozzle is regulated by the same pressure differential. As a result, the velocity of flow of fuel and air varies similarly. At the same time, when air flow increases, the pressure at the venture throat falls but the pressure of the fuel does not change. When a result, a basic carburetor produces an increasingly rich mixture when the throttle is opened further. The performance of a basic

carburetor is mathematically analyzed in the part after that. 7.9.3 Carburetor Dimensions The diameter of the venturi tube in millimeters and the size of the jet in hundredths of millimeters are typically used to describe a carburetor's size. Under a head of 500 mm pure benzoyl, the calibrated jets have a stamped number that indicates the flow in milliliters per minute. The pressure difference ($p_1 - p_2$) for a venturi of 30 to 35 mm size (with a jet size that is one-sixteenth of venturi size) is around 50 mm of Hg. The velocity at the throat is approximately 90 to 100 m/s, and the venturi Cd's coefficient of discharge is typically 0.85.

Key Components of a Carburetor

The following components make up a carburetor:

- i. A fuel strainer.
- ii. A float chamber.
- iii. The main fuel metering and idling nozzles.
- iv. The choke and throttle.

The following sections provide a brief discussion of the various aspects described above.

The Fuel Strainer

There is a chance that the nozzle may clog during extended engine operation because the gasoline must travel via a small nozzle exit. By placing a fuel strainer at the entrance to the float chamber, the petrol is filtered in order to prevent potential nozzle blockage by dust particles. The strainer is made of a thin wire mesh or another kind of cylindrical or conical filtering device. The strainer is typically detachable so that it can be removed and completely cleaned. A strainer plugs or a compression spring holds it in place.

The Float Chamber

In a carburetor, the float chamber's job is to provide fuel to the nozzle at a constant pressure head. This is made feasible by keeping the fuel level in the float bowl constant. A carburetor's float is used to regulate the amount of fuel in the float chamber. To ensure the proper amount of fuel flow and to stop fuel leaking from the nozzle when the engine is not running, this fuel level must be kept just slightly below the discharge nozzle outlet holes.

DISCUSSION

The main fuel jet, venturi, jet tube, throttle valve, float chamber, and other components make up a simple carburetor. In the float chamber where a float is present. Fuel is delivered from the fuel tank via the strainer and into the float chamber with the aid of a fuel pump. This entire unit can be referred to as a simple carburetor Figure 1.

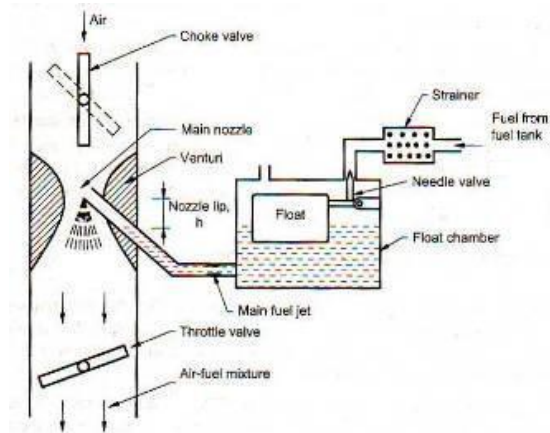


Figure 1: Representing the overview about Simple Carburetor [Scholar Express].

Working Operation of Simple Carburetor

We will gradually discover how this carburetor operates

1. As is common knowledge, a basic carburetor has an air-open float chamber. It keeps the float chamber at atmospheric pressure.
2. The fuel pump transfers fuel from the external fuel tank to the float chamber. Utilizing the strainer, this fuel from the fuel tank is filtered to remove any solid particles.
3. The primary nozzle, which is a component of the jet tube, is now supplied with fuel from the float chamber. The main fuel jet is responsible for this fuel flow from the float chamber to the main nozzle.
4. Through the choke valve, the engine draws air from the atmosphere. As this air flows through the venturi, the cross-sectional area at the venturi's throat is reduced.
5. As a result, there is a decrease in pressure at the main nozzle and an increase in air velocity.
6. The combination of fuel and incoming ambient air is brought about by the pressure difference between the main nozzle and the float chamber.
7. The engine fuel is partially vaporized by the venturi after which it is completely evaporated by the heat of the combustion chamber and cylinder walls.
8. Because gasoline engines are quantity controlled, carburetors are only installed in gasoline engines.
9. When the throttle valve at the bottom of the jet tube is opened, more air may pass through the venturi tube and be given to the engine in greater quantities, which results in the engine producing more power.

The power of the engine decreases when we close the throttle valve due to reversible action. **What does the simple carburetor's nozzle lip do?**

The end of the main nozzle is kept just above the level of fuel in the float chamber to prevent fuel from overflowing from the nozzle. Nozzle lip refers to the level difference between the main nozzle's tip and the fuel level in the float chamber. The diagram above shows the nozzle lip level. In this carburetor, the air-fuel ratio is entirely dependent on where the throttle valve is positioned. Additionally, when engine speed rises, the air-fuel ratio falls. The main drawback or limitation of a basic carburetor is that it produces a powerful mixture when the speed is too low, which interferes with the mixture's ability to ignite. Simple carburetors are only used in stationary, small engines; they are not employed in any modern engines.

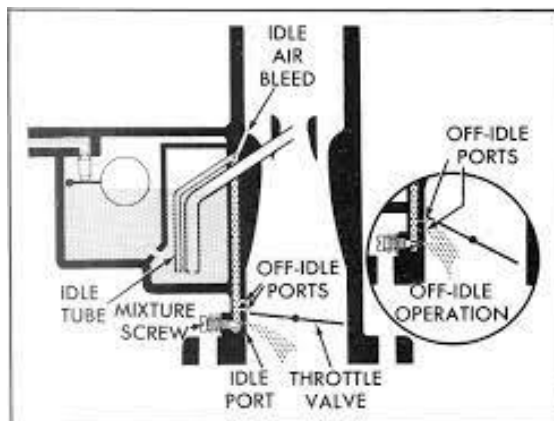


Figure 2: Representing the Main metric ideal system [The Carburetor Doctor].

Primarily Used Metering and Idle System

Fuel feed for cruising and full throttle activities is controlled by the carburetor's primary metering mechanism (Figure 2). There are three main units in it. The main discharge nozzle, the fuel metering orifice, the main discharge nozzle, and the tube leading to the idling system are all examples of fuel-related components. The main discharge nozzle, the fuel metering orifice, the main discharge nozzle, and the tube leading to the idling system are all examples of fuel-related components. The main discharge nozzle, the fuel metering orifice, the main discharge nozzle, and the tube leading to the idling system are all examples of fuel-related components. The primary metering system has three functions: proportioning the fuel-air mixture, lowering pressure at the discharge nozzle exit; and restricting airflow at maximum throttle. For idling and low speed operation, SI engine-equipped cars need a rich mixture (Figure 1). A schematic diagram of a

carburetor is shown in Figure 2, with the primary metering and idling system highlighted. A typical air fuel ratio for idling is around 12:1.

Most current carburetors have a specific idling system built into them in order to give such a rich mixture while idling. As depicted in Figure 2, this consists of an idle fuel channel and idle ports. When the vehicle engine is started, idled, or operating at a very low speed, this system becomes active. It ceases to function when the throttle is opened above 15% to 20%. The extremely little air that enters the engine when the throttle is almost closed or only slightly open causes very little depression at the venturi's throat, which is insufficient to draw any fuel from the nozzle. However, the gasoline rises in the idling tube and is discharged through the idling discharge port, directly into the engine intake manifold, due to the extremely low pressure created on the downstream side of the throttle by the suction stroke of the piston. A little amount of air is also sucked because of the low pressure created by the idling air bleed. The idle air-bleed combines air with the petrol pulled from the float chamber, aiding in its vaporization and atomization as it travels along the idle tube. When the engine is not running, the air bleed also stops the gasoline in the float chamber from being drained off through the idle route owing to syphon action. The suction pressure at the idle discharge port is insufficient to pull the gasoline through the idling route when the throttle is opened and the engine is operating in the idling range of operation. And the idle system stops functioning. After that, the primary air flow picks up and the operating range is defined.

The Throttle and the Choke:

Starting becomes more challenging when the car is left still for an extended amount of time during the cool winter months, possibly overnight. As was already said, a highly rich mixture is needed to start combustion at low cranking speeds and intake temperatures. Sometimes a 9:1 air-to-fuel ratio is needed. The primary cause is that a significant portion of the fuel may continue to be liquid and suspended in air even inside the cylinder. Fuel-vapor and air must be combined at a ratio that can support combustion in order to start combustion.

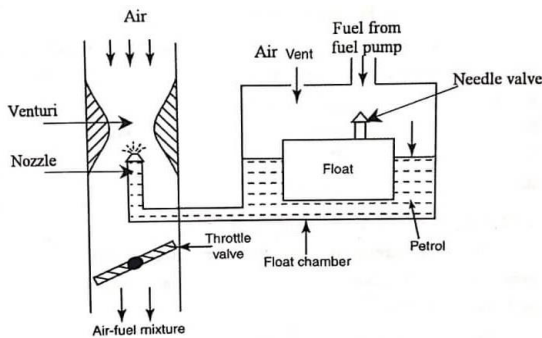


Figure 3: Representing the throttle and the choke [Theteche].

It should be remembered that the fuel's vapor percentage is likewise very small at very low temperatures, and this creates a combustible mixture that starts combustion. So, a highly rich mixture needs to be provided. The use of a choke valve is the most common way to supply such a mixture. As depicted in Figure 3, this straightforward butterfly valve is situated between the carburetor's inlet and the venturi's throat. The venturi throat experiences a significant pressure reduction when the choke is partially closed, which would ordinarily be caused by the volume of air moving through the venturi throat. In order to keep the ratio of evaporated fuel to air in the cylinder within the combustible limits, the very large depression at the throat inducts a significant amount of fuel from the main nozzle and creates a very rich mixture. The choke valves may occasionally be spring-loaded to prevent excessive choking and massive carburetor depression after the engine has started and attained the appropriate speed. By using a thermostat, it is possible to automate this choke's operation, closing it while the engine is cold and turning it off when it warms up after starting.

The throttle valve, which is situated on the downstream side of the venturi, is used to regulate the speed and output of an engine. The amount of mixture delivered to the cylinders decreases as the throttle is closed farther, increasing the barrier to the mixture's flow in the route. Because there is less mixture present, the pistons receive a weaker impulse, which in turn lowers engine output. The engine's output rises as the throttle is increased. The engine speed often increases as the throttle is opened. However, given that the load on the engine also plays a role, this is not always the case. For instance, depending on the steepness of the hill and the length of the throttle opening, opening the throttle when the motor vehicle is starting to climb a hill may or may not improve the vehicle speed. To put it briefly, the throttle is just a way to modify the

amount of charge entering the cylinder in order to control the engine's output (Figure 3).

Advantages of Simple Carburetor

1. Carburetors have the benefit of being simpler to maintain than fuel injection systems.
2. Cheap to produce.
3. High reliability is possible with simple construction.
4. Simple to change or fix.
5. Low cost per displacement unit in comparison to fuel injection methods.
6. Better performance over 3,000 meters in elevation.
7. They are comparatively straightforward machines.
8. Aspiration helps in the mechanical mixing of the fuel that enters the intake apertures of a carburetor. While fuel injection relies on valve timing at the throttle position, which means there are five times more moving parts while requiring less pressure overall, this approach has the benefit of short reaction times.
9. For various engine conditions, they can be modified to deliver the ideal air/fuel combination.
10. Under light load conditions, they can offer good fuel efficiency.
11. A fuel injection system's complex, turbulent fuel-air mix in the intake system is avoided with a carburetor.

Carburetor Drawbacks

1. Incomplete combustion of the air/fuel combination results in lower fuel efficiency when compared to fuel injection systems.
2. More sensitive to changes in height
3. When the engine is under a strong load, they are not as effective as fuel injection systems.
4. In chilly weather, they can be challenging to start.
5. The needle takes a while to respond to changes in air demand or momentum, which causes jetting issues during acceleration.
6. Compared to fuel injectors, there is more fuel waste.

CONCLUSION

A carburetor is a mechanism that draws air from the environment, combines fuel from the fuel tank in a specified ratio, and then suctions the mixture to power the engine. Carburetion, or the carburetion process, is the act of sucking and blending the air-fuel ratio in a specified ratio in accordance with the state of the engine's operation. Carburetors are really

complicated. Let's first comprehend how a basic or simple carburetor function in order to give an air-fuel combination for cruising or typical range at a single speed. Later, more mechanisms will be added to support the many unique requirements, including acceleration, variable load and speed operation, beginning, and idling. The basic components of a carburetor are a float chamber, throttle valve, choke, fuel discharge nozzle, and metering orifice.

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Introduction to Compensation Technologies and Some Components

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ABSTRACT: *A simple carburetor has a propensity to gradually richen the mixture as the throttle opens. The needs of the engine cannot be supplied by the main metering system on its own. In order to deliver a mixture with the necessary air-fuel ratio, several compensating mechanisms are typically installed in the carburetor along with the main metering system. As was already noted, automatic compensating mechanisms are present in modern carburetors to maintain the proper mixture proportions at greater speeds. The metering system of the carburetor is determined by the type of compensating mechanism utilised. The following sections provide a brief discussion of the various compensating devices' operating principles.*

KEYWORDS: *Air Bleed, Air Flow, Compensatory Jet, Float Chamber, Jet Tube*

INTRODUCTION

A vehicle on the road must operate under various loads and speeds. The state of the roads is crucial. One may only be able to use between 25% and 60% of the throttle, especially on city streets [1], [2]. The carburetor must be able to provide a 16:1 air-to-fuel mixture that is almost consistent under such circumstances. However, a simple carburetor has a propensity to gradually richen the mixture as the throttle opens. The needs of the engine cannot be supplied by the main metering system on its own. In order to deliver a mixture with the necessary air-fuel ratio, several compensating mechanisms are typically installed in the carburetor along with the main metering system. There are several compensatory mechanisms in use. The crucial ones are

- i. The air-bleed jet.
- ii. The compensating jet.
- iii. The emulsion tubes.
- iv. The back suction control mechanism.
- v. The auxiliary air valve.

As was already noted, automatic compensating mechanisms are present in modern carburetors to maintain the proper mixture proportions at greater speeds [3], [4]. The metering system of the carburetor is determined by the type of compensating mechanism utilised. The following sections provide a brief discussion of the various compensating devices' operating principles.

Jet With Air-Bleed

A typical modern down-draught carburetor's air-bleed system. As an example of how it works. As was evident, it has an air-bleed into the primary nozzle [5], [6]. This bleed is known as a confined air-bleed jet because the air flow through it is

constrained by an aperture. The main jet and the air bleed jet will be filled with fuel when the engine is not running. When the engine first cranks, fuel first enters through the main and air bleed jets (A). Only air begins to enter the air bleed as the engine accelerates, mixing with gasoline at B to create an emulsion of air and fuel. As a result, the fluid stream that has combined with the liquid to form an emulsion has very little viscosity and surface tension. As a result, the fuel flow rate is increased and more fuel is sucked at low suction. For the whole power range of an engine's operation, a fairly consistent mixture ratio can be maintained by properly designing a whole size at B compatible with the entry hole at A.

When the air-bleed system's fuel flow nozzle is positioned in the center of the Venturi, the Venturi and air-bleed nozzle are both subjected to the same engine suction, producing roughly the same fuel-air mixture over the whole operating power range. The Compensating Jet The idea of a compensating jet device is to gradually open the throttle while making the mixture leaner [7], [8]. This approach incorporates a compensatory jet in addition to the main jet. Connected to the compensation well is the compensating jet. Similar to the main float chamber, the compensating well is similarly vented to the atmosphere. Fuel is delivered to the compensating well through a limiting aperture from the main float chamber. Fuel delivery through the compensating jet reduces as a result of the fuel level in the compensating well falling with an increase in air flow rate. Thus, as the primary jet gradually increases the mixture's richness, the compensatory jet gradually makes the mixture slimmer. The total of the two tends to maintain the fuel-air mixture more or less constant. The compensatory jet curve

and the main jet curve are essentially reciprocals of one another.

Emulsion Tubes

In contemporary carburetors, the mixture correction is attempted using air bleeding. One such setup is depicted, where the primary metering jet is maintained at a level that is approximately 25 mm below the fuel level in the float chamber. As a result, it is also known as submerged jet. At the bottom of a well, there is a jet. The well's sides have holes in them. The Figure 1 shows that these holes are in contact with the atmosphere. The initial fuel level in both the well and the float chamber is the same. The pressure at the venturi throat drops when the throttle is opened, allowing fuel to be sucked into the airstream. This causes the center tube's holes to be gradually revealed, boosting air-fuel ratios or decreasing mixture richness when all holes are eventually exposed. The main jet produces normal flow. Through these apertures in the well, air is taken in, the gasoline is emulsified, and the 7.1.4 Back Suction Control Mechanism is activated [9], [10].

The details of the rear suction control device are. In this apparatus, a sizable vent line equipped with a control valve connects the top of the fuel combustion chamber to air entrance. The top of the fuel float chamber is connected to the venturi throat by another line that has a tiny opening. The vent line is unrestrained, the float chamber pressure (p_1) is atmospheric, and the throat pressure (p_2) is present when the control valve is fully open. As a result, (p_1 p_2) is the pressure difference pressing on the orifice. If the valve is shut, the pressure in the float chamber and the venturi throat will be equal, preventing fuel from flowing. The control valve can be properly adjusted to create the necessary pressure differential in the float chamber. As a result, the 7.11.5 Added Valve.

To further comprehend the concept presents a simplified image of an auxiliary valve device. When the engine is not running, atmospheric pressure, or p_1 , is acting on the top of the auxiliary valve. As the load increases, the vacuum at the venturi throat grows while the throat pressure, p_2 , falls. The valve is raised by the pressure difference (p_1 p_2) in opposition to the spring's tension. And as a result, more air is let in, keeping the mixture from getting too rich. Depicts an auxiliary port used in a downdraft carburetor in accordance with 7.11.6. When the butterfly valve is opened, more air flows through this port, which lowers the venturi's airflow. This implies that p will be significantly smaller. Fuel is drawn less as a result. This technique

was widely used in aero plane carburetors to make up for the reduction in air density at high altitudes.

DISCUSSION

As many air bleed holes as may be required to carry out the compensatory function are drilled radially into the jet tube in the image, two pairs of diametrically opposed holes are indicated. A deep thimble-shaped cap with an open lower end is screwed into the carburetor body's lower end and placed over the jet carrier E. There is a tiny hole in the otherwise closed upper end of the cap, which is secured to the upper end of the jet tube to hold it in place. A tubular upward extension of the jet carrier, which forms a well between it and the jet tube, is positioned concentrically between the jet tube and cap. There is a space between the upper end of this extension tube and the inner face of the higher end of the cap, allowing air entering through the two radial holes near the cap's base to flow up and over into the well's top. The jet tube and well are filled with fuel to level XX when the engine is started. The level of the fuel in the well decreases when the throttle is opened and the well empties.

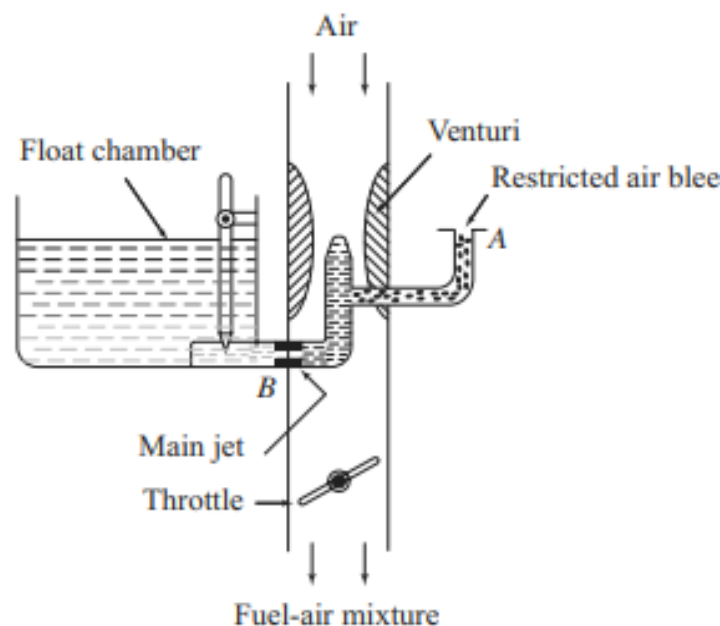


Figure 1: A typical modern carburetor's air-bleed principle [Ftp.Idu.Ac.Id].

This gradually exposes the air bleed holes C, which weakens the depression over the upper end of the jet tube and counteracts the propensity for the mixture to become more enriched. Air starts to emulsify and bubble through the fuel in the jet tube at the same time, which speeds up the evaporation process. High degrees of depression cause air jets to shoot through the bleed tube's holes and into the fuel column,

further emulsifying it. Rotary valves and air bleed mechanisms are employed in some Weber downdraught carburetors. The latter are activated by the throttle controls and serve to enrich the mixture as full throttle is approached by decreasing the air supply. A fuel/air mixture that is given to a spark-ignition internal combustion engine is made and controlled by carburetors. Typically, carburetors feature a barrel or throat tube that is connected to a manifold that feeds one or more engine cylinders. Through a manifold, a single or multi-barrel carburetor can feed a number of cylinders, or one carburetor can be provided for every cylinder. The carburetor's throat contains a throttle that regulates the overall volume of the fuel/air combination. It might be a sliding valve or a butterfly valve. Air and fuel passageways are connected within the carburetor's area near the throat.

These can be arranged in one or more circuits, which are collections of channels that ultimately release fuel/air combinations into the throat. A mixture might be released into the throat between the cylinders and the throttle, for instance, by an idle circuit. The engine's idle speed is governed by this circuit while the throttle is essentially closed. On the other side of the throttle, a main circuit might discharge into the throat. This circuit regulates engine speed and power while the throttle is open. These circuits have an opening that connects the throat to a gasoline reservoir that is supplied by a fuel tank via a metering valve. The passage typically travels through a region, commonly referred to as an emulsion tube that has tiny holes connected to outside air via other passages. The purpose of the air bleed, or connection to ambient air, is to aerate the fuel by creating an emulsion of it with air. Because aerated gasoline flows more easily down the carburetor throat and vaporizes more quickly as it travels to the cylinders, it is favorable. The size of the air bleed tube has an impact on how much aeration occurs. A small aperture or other obstruction frequently restricts the airflow. Since air pressure, temperature, and humidity all affect air flow, the size of the limitation is typically a compromise. These limitations can occasionally be replaced with other sizes to provide for things like high altitude operation.

There are several patents that deal with employing accessories attached to traditional carburetors to control the air through an air bleed. Usually, they alter the air flow to optimize engine performance using a range of sensors and schemes. Direct fuel injection began to replace conventional carburetors in passenger cars from 1980, and the switch was complete by around 1990. The majority of

motorbike engines still do not use fuel injection due to the absence of emission control regulations. The two fuel circuits mentioned above, as well as a pilot circuit that operates most efficiently between idle and around three eighths wide-open throttle, are all common features of motorcycle carburetors, which often use slide valves. They are known as air bleeds when the air input ports to these three circuits are located inside the barrel. They very slightly bleed air from the air passing down the barrel. Although the terms are sometimes confused, air jets are the word used when the air intake apertures are located outside the barrel. Any air intake inlet that supplies air to emulsify the fuel is referred to in this context as an air bleed or air jet. The barrels of motorcycle carburetors are also enlarged by removable tubes known as velocity stacks.

The Compensating Jet

The idea of a compensating jet device is to gradually open the throttle while making the mixture leaner. This approach incorporates a compensatory jet in addition to the main jet, as seen in Figure 2. Connected to the compensation well is the compensating jet. Similar to the main float chamber, the compensating well is similarly vented to the atmosphere. Fuel is delivered to the compensating well through a limiting aperture from the main float chamber. Fuel delivery through the compensating jet reduces as a result of the fuel level in the compensating well falling with an increase in air flow rate. Thus, as the primary jet gradually increases the mixture's richness, the compensatory jet gradually makes the mixture slimmer. The total of the two tends to maintain the fuel-air mixture more or less constant. The compensatory jet curve and the main jet curve are essentially reciprocals of one another. Some carburetors have a compensating jet, especially the ones used in older or simpler internal combustion engines (IC engines). Its goal is to support maintaining the ideal fuel-air ratio under various engine operating circumstances. Although fuel injection systems are used in most modern engines, the compensating jet provides advantages in some situations. Here are some potential benefits of a carburetor's compensating jet component:

Compensation for Altitude and Atmospheric Changes: Based on changes in altitude and atmospheric conditions, the compensating jet assists in adjusting the fuel flow rate. The fuel-air mixture is impacted by the less dense air that occurs as altitude rises. In order to make up for the lower air density at higher altitudes, the compensating jet makes sure that the proper amount of fuel is delivered.

Improved Engine Performance: The compensating jet aids in optimizing engine performance by maintaining the correct fuel-to-air ratio. Under various operating conditions, it makes sure the engine gets the right quantity of fuel, resulting in smoother acceleration, greater throttle response, and increased overall engine economy. Fuel starvation or excessively rich fuel circumstances, which can happen as a result of changes in engine load or speed, are prevented by the compensating jet. It enables the carburetor to modify the fuel delivery rate to satisfy the needs of the engine, ensuring steady and dependable fueling under a variety of operating circumstances.

Design Simplified: Compared to fuel injection systems, the compensating jet and other carburetor parts may help to create a design that is simpler and possibly more cost-effective. Carburetors are useful for some applications where simplicity and economic considerations are crucial because they are typically less complex and simpler to operate and repair. It's crucial to remember that compared to fuel injection systems, carburetors, especially those with compensating jets, have several disadvantages. In extremely dynamic or demanding operating situations, they could have less accurate control over the fuel-air mixture. In comparison to contemporary fuel injection technologies, they may also be less efficient in terms of fuel economy and pollutants.

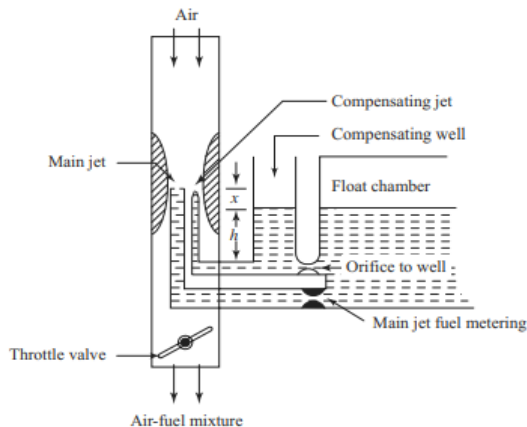


Figure 2: Representing the various Compensating Jet [Ftp.Idu.Ac.Id].

However, the compensating jet element in a carburetor can offer benefits by assisting in maintaining the proper fuel-air mixture under a variety of operating conditions in some applications where simplicity, cost-effectiveness, and ease of maintenance are prioritized over advanced fueling control. With the right amount of fuel ejected from the nozzle, the necessary air-fuel mixture may be made. This technique is exclusively used in big carburetors. Added Valve to further comprehend the

concept, presents a simplified image of an auxiliary valve device. When the engine is not running, atmospheric pressure, or p_1 , is acting on the top of the auxiliary valve. As the load increases, the vacuum at the Ventura throat grows while the throat pressure, p_2 , falls. The valve is raised by the pressure difference ($p_1 - p_2$) in opposition to the spring's tension. And as a result, more air is let in, keeping the mixture from getting too rich.

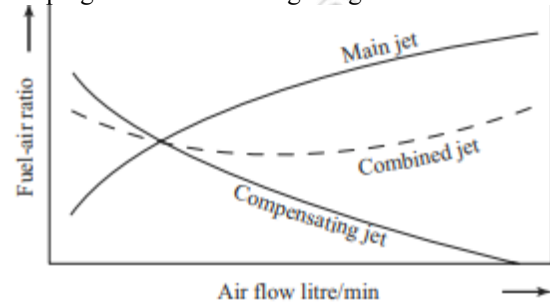


Figure 3: Effect of compensating device on fuel-air ratio [Ftp.Idu.Ac.Id].

Advanced Carburetors' Additional Systems

Extra than the aforementioned compensatory devices, modern carburetors typically use a few extra technologies to suit vehicle demands. The following sections provide explanations of the specifics of the various systems. an SI engine will occasionally continue to run after the ignition is turned off for a very little time. The term dieseling after running or run-on refers to this occurrence. Dieseling could occur for one or more of the reasons listed below:

- i. High idle speed on the engine.
- ii. Carbon deposits cause a compression ratio increase.
- iii. Low or insufficient octane rating
- iv. An overheated engine.
- v. A high heat range for the spark plug.
- vi. Improper toluene fuel-air mixture adjustment during idling.
- vii. The throttle sticking.
- viii. The need for an engine tune-up.
- ix. (ix)Oil entering the cylinder is item

Some contemporary cars have ant diesel systems to stop dieseling systems. A solenoid valve-operated idle circuit is present in this system. When the ignition key is turned on, current passes through the solenoid valve's solenoid coil, producing force. The needle valve is pulled by this force, which also opens the path for the sluggish mixture. The magnetic force vanishes when the ignition is switched off. The solenoid valve's spring then forces the needle valve back to its initial closed position. By doing this, the

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engine is shut off from the sluggish mixture passage. Emissions of hydrocarbons are decreased.

Richer Coasting System

Some contemporary cars use the richer coasting system. The wheel will accelerate the engine quickly when the accelerator foot is abruptly released when the automobile is moving at a high speed. As a result, incomplete combustion is caused by an excessive increase in the vacuum in the inlet manifold and the combustion chamber. The richer coaster system is made to get around this issue by giving the intake manifold the right mixture for optimum combustion. For stable combustion, this system has a chamber attached to the intake manifold. Intake manifold vacuum rises when the throttle valve is closed to slow down. The coasting valve opens as a result of the membrane being pulled by the vacuum that has been introduced to the chamber. Following that, air is added to the fuel in the float chamber, which is then mixed at the coasting fuel jet before being drawn into the intake manifold.

System for Acceleration Pumps

The phenomenon of acceleration is fleeting. The mixture needed is extremely rich, and it must soon and very swiftly become rich in order to accelerate the vehicle and, as a result, its engine. There are instances in automobile engines where it is important to accelerate the vehicle. This calls for the engine to provide more power in a hurry. The air flow increases in direct proportion to a quick opening of the throttle. The fuel flow does not, however, increase proportionally to the increase in air flow due to the liquid fuel's inertia. This causes a momentarily lean mixture, which misfires the engine and temporarily lowers power output. All contemporary carburetors come with an acceleration device to avoid this situation. One such gadget is depicted in a simplified schema.

The pump includes a spring-loaded plunger that handles the issue with the throttle valve opening rapidly. As the plunger travels into the cylinder, more gasoline is forced into the Ventura throat. The spring moves the plunger back when the throttle is partially open. Additionally, a mechanism prevents gasoline in the pump cylinder from leaking past the plunger or through some holes into the float chamber when the valve is slowly opened or from being driven through the jet. In some carburetors, the mechanical linkage system is replaced by a setup where the pump plunger is supported by the manifold vacuum. Rapid throttle opening reduces this vacuum, which causes a spring to press the plunger downward and flow fuel through the jet. When operating at maximum power, which is

between 80% and 100% load, a richer air-fuel ratio of around 12 to 14 is needed, and an air-fuel ratio of about 12 is anticipated.

CONCLUSION

A vehicle on the road must operate under various loads and speeds. The state of the roads is crucial. One may only be able to use between 25% and 60% of the throttle, especially on city streets. The carburetor must be able to provide a 16:1 air-to-fuel mixture that is almost consistent under such circumstances. However, a simple carburetor has a propensity to gradually enrich the mixture as the throttle opens. The needs of the engine cannot be supplied by the main metering system on its own. In order to deliver a mixture with the necessary air-fuel ratio, certain compensating devices are typically included to the carburetor along with the main metering system.

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Application and Working Operation of Economizer

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ABSTRACT: A mechanical device called an economizer lowers the amount of energy used to cool a data center or other structures. It is sometimes referred to as an HVAC economizer since it is integrated into the building's heating, ventilation, and air conditioning (HVAC) systems. In order to better regulate indoor temperatures and increase energy efficiency, an economizer uses outside air. At the maximum power range of operation from 80% to 100% load, richer air-fuel ratio of about 12 to 14 is required and at the maximum power, an air-fuel ratio of approximately 12 i.e. expected. An economizer is a valve which remains closed at normal cruise operation and gets opened to supply.

KEYWORDS: Air Side, Data Center, Feed Water, Flue Gases, Heat Exchanger

INTRODUCTION

A mechanical device called an economizer lowers the amount of energy used to cool a data center or other structures. It is sometimes referred to as an HVAC economizer since it is integrated into the building's heating, ventilation, and air conditioning (HVAC) systems. In order to better regulate indoor temperatures and increase energy efficiency, an economizer uses outside air. Economizers are frequently used in data centers to lessen the need for cooling equipment like chillers, compressors, or air conditioners in computer rooms. Economizers can aid in the transition to a greener computing environment by reducing the energy usage and costs of a data center. Economizers can increase the lifespan of cooling equipment, resulting in further cost savings. Depending on the economizer type, they may even increase the quality of indoor air [1], [2].

Although their use is not just confined to colder places, economizers are most advantageous in cooler climates. In warmer climates, an economizer can assist in lowering energy use in the cooler months or at night when temperatures drop. An economizer can be quite helpful as long as the ambient temperature and humidity are within the acceptable range. Government agencies and professional associations, like ASHRAE, offer advice and best practices for using economizers. A common industry standard for designing energy-efficient buildings is the ANSI/ASHRAE/IES Standard Energy Standard for Buildings except Low-Rise Residential Buildings. The air-side and fluid-side economizers' two fundamental types are likewise covered in the standard [3], [4].

An Air-Side Economizer: What Is It?

An air-side economizer is included into the ventilation system of a building. The economizer draws colder outside air into the data center and directs warmer inside exhaust air outside when the outside temperature and humidity are within the desired levels. To attain the ideal temperature and humidity level, an economizer may in some cases combine exhaust air with outside air, such as in extremely cold climates. The use of an air-side economizer necessitates meticulous monitoring and control of humidity levels. A data center should normally aim to keep its relative humidity between 40% and 60%. However, an environment with significant levels of copper and silver corrosion may require to maintain a lower relative humidity, maybe below 50%, per a 2020 report from ASHRAE. An automatic control system for modifying the ventilation dampers that control airflow is part of an air-side economizer. If the economizer is in a data center, it typically includes has filters to remove particles that could damage the equipment [5], [6]. The duct that connects the outside environment to the data center inside has these filters installed in i.e.

Fluid-Side Economizer: What Is It?

The cooling of the data center is done differently with a fluid-side economizer. It uses that air to reduce the temperature of the coolant in the building's cooling system rather than pumping outside air directly into the structure. By using this technique, the coolant's temperature can be lowered without as much mechanical refrigeration. In fluid-side economizers, water, glycol mixes, and refrigerants are often utilized fluids [7], [8]. This sort of economizer is frequently referred to as a water-side economizer due to the utilization of water. Data

centers with chilled water plants that are air- or water-cooled frequently employ fluid-side economizers. They utilize the cooling tower's evaporative cooling capability, allowing the economizer to replace the chiller and, consequently, lowering energy expenditures and usage. In order to lower coolant temperatures, water-side economizers frequently run at night or in the winter. They do this by taking advantage of the colder surrounding air [9], [10].

Because they don't pump outside air into the data center, fluid-side economizers have an advantage over air-side economizers. As a result, there is no need to maintain air filtering systems, upgrade ventilation systems, or regulate humidity levels. However, air-side economizers may operate in a larger temperature range, enabling them to be utilized more frequently and in more places. When operating at maximum power, which is between 80% and 100% load, a richer air-fuel ratio of around 12 to 14 is needed, and an air-fuel ratio of about 12 is anticipated. A valve known as an economizer closes during normal cruising operation and opens to supply a rich mixture during full throttle operation. During running at full throttle, it controls the extra fuel supply. The word economizer is a bit deceptive. It is most likely named an economizer because it doesn't impede when cruising and an economy mixture is supplied. It should be referred to as a power enrichment system. The skeleton of a metering rod economizer system. Only when the throttle is opened beyond a certain threshold does it permit a substantial opening to the main jet. The metering rod can have steps or be tapered.

DISCUSSION

Boilers

Economizers are heat exchange components found in boilers that heat fluids, typically water, up to but typically not over the boiling point of that fluid. Because they may employ enthalpy in fluid streams that are heated but not hot enough to be used in a boiler, economizers hence its name can recover more usable enthalpy and increase the efficiency of the boiler. They are a boiler accessory that conserves energy by heating the feed water cold water used to load the boiler using the exhaust gases from the boiler. Steam boilers take a lot of energy to heat the feed water to boiling, turn it into steam, and occasionally superheat the steam over saturation temperature. When the highest temperatures close to the combustion sources are used for boiling and superheating, the efficiency of heat transfer is increased. Meanwhile, the temperature of the feed

water entering the steam drum is raised by using the residual heat of the cooled combustion gases exhausting from the boiler through an economizer. The leftover heat from the combustion products will be recovered by a condensing economizer, either indirect contact or direct contact.

Depending on the need for make-up water and process water, a series of dampers, an effective control system, and a ventilator allow all or some of the combustion products to travel through the economizer. While preheating the feed water to the boiling temperature, the temperature of the gases can be decreased from the fluid's boiling point to barely over that of the incoming feed water. In comparison to low pressure boilers, high pressure boilers often feature bigger economizer surfaces. To enhance the heat transfer surface on the combustion gas side, economizer tubes frequently incorporate protrusion resembling fins. Boiler combustion efficiency has, on average, increased through time from 80% to more than 95%. Boiler efficiency is directly related to the effectiveness of the heat produced. Two important factors in assessing this efficiency are the proportion of surplus air and the temperature of the combustion products.

The burners require an extra air flow in order to function since natural gas combustion requires a specific amount of air in order to be completed. The quantity of water steam produced during combustion depends on how much natural gas is consumed. Additionally, the amount of extra air affects how the dew point is measured. Different natural gas combustion efficiency curves are correlated with the temperature of the gases and the amount of extra air. For instance, the efficiency will be 94% if the gases are cooled to 38 °C and there is 15% extra air. The sensible and latent heat in the steam condensate included in the flue gases can thus be recovered for the process by the condensing economizer. An alloy of stainless steel and aluminum is used to create the economizer. Water flows through the finned tubes while the gases pass through the cylinder. 11% of the water present in the gases is condensed.

History

On the outskirts of Launceston, in the boiler house of the Kill faddy Board Mills, lies one of the two original 1940s Green's Economizers. The boilers of stationary steam engines were improved with the help of the first effective economizer design. Edward Green patented it in 1845, and since then it has been referred to as Green's economizer. The boiler's exhaust fumes flowed via a series of vertical cast iron tubes that were joined to a tank of water above and below. The arrangement in an economizer

is the opposite of that typically, though not always, observed in a boiler's fire tubes; there, hot gases typically pass through tubes that are submerged in water, whereas in an economizer, water passes through tubes that are surrounded by hot gases. While both are heat exchange devices, an economizer uses some of the heat energy that would otherwise be lost to the atmosphere to heat the water and/or air that will go into the boiler, saving fuel. In a boiler, burning gases heat the water to produce steam to drive an engine, whether a piston or a turbine. The mechanical scraping device, which was required to keep the tubes clear of soot deposits, was the most effective aspect of Green's economizer design.

In the decades that followed Green's invention, nearly all stationary steam engines were eventually equipped with economizers. Even though they are typically not operated, several restored stationary steam engine sites still contain their green's economizers. The Clay mills Pumping Engines Trust in Staffordshire, England, is one such preserved site that is currently repairing one set of economizers and the related steam engine that powered them. Another instance of this is the British Engineer in Brighton & Hove, where the boiler economizer for Number 2 Engine is in service, along with the little stationary engine that goes with it. A third location is the Coldharbour Mill Working Wool Museum, where the Green's economizer is still operational and has the steam engine's Pollitt and Wig Zell drive shafts.

Building and Operation of a Boiler Economizer

The construction and use of the boiler economizer are easy and uncomplicated. The water is fed to the economizer at room temperature by a horizontal inlet pipe at the bottom of the device. At the top of the economizer, there is another horizontal pipe inserted. The bottom and top pipes, which are two horizontal pipes, are connected by a number of vertical pipes. An output valve is installed on the top horizontal pipe to feed hot water to the boiler. The vertical pipes of the economizer receive flue gases from the boiler furnace. The flue gases transfer residual heat to the water through the surface of the vertical pipe as the water rises through the vertical pipes to the top horizontal line. The economizer uses the heat from the flue gases to warm the water before it enters the boiler to make steam. Ash particles will be present in the flue gas and will be deposited on the vertical pipe surfaces. If extra care is not taken, a heavy coating of soot will build up on the surfaces, blocking the passage of heat into the water.

Each vertical pipe has a scraper attached to it that is moved up and down by a chain pulley system in

order to remove the soot. The soot is scraped all the way to the bottom of the economizer, to the soot chamber. Then, soot is collected from the soot chamber. This is how a boiler economizer works. This boiler economizer is of the most basic design. An economizer in a thermal power plant warms fluids or recovers heat from the flue gases, which are the combustion product, before they are expelled out the chimney. The combustion exhaust gases produced by power plants are called flue gases, and the main components of these gases include nitrogen, carbon dioxide, water vapor, soot, and carbon monoxide. The preheated boiler feed water is then used to create superheated steam using the recovered heat. Since we are essentially collecting waste heat and channeling it to where it is needed, we are lowering fuel usage and optimizing the process as a result. All pulverized coal-fired boilers must have an air pre-heater, which can economically recover the heat from exhaust flue gases.

Generating Units

Principal Concept: Feed water Heater Economizers, which are descendants of Green's original concept, are still used in modern boilers, like those in coal-fired power plants. They heat the condensate from turbines before it is pumped to the boilers, and in this sense, they are frequently referred to as feed water heaters. In a combined cycle power plant, economizers are frequently employed as a component of the heat recovery steam generator (HRSG). Water moves via an economizer, a boiler, and a superheated in an HRSG. The economizer also protects the boiler from being flooded with liquid water that, given the flow rates and boiler design, is too cold to boil. Economizers are frequently used in steam power plants to transfer surplus heat from boiler stack gases to the boiler feed water. This increases the temperature of the boiler's feed water, which lowers the energy input required and, in turn, lowers the firing rates required to maintain the boiler's rated output. If care is not taken in their design and material selection, economizers can lower stack temperatures, which can lead to condensation of acidic combustion gases and catastrophic equipment corrosion damage.

Working Operation

A thermal power plant's economizer has the job of recovering part of the heat that is lost in the flue gases that go up the chimney and using it to heat the boiler's feed water. It is merely a heat exchanger with a long heating surface like Fins or Gills, hot flue gas on the shell side, and water on the tube side. Thermal power plants must size their economizers for the amount and temperature of flue gas, the highest

pressure drop that can flow through the stack, the type of fuel used in the boiler, and the amount of energy that needs to be recovered. The steam that is created when the water is boiled in the steam boiler is then passed to the turbines after being superheated. The condensed water is then pre-warmed first in the feed water heater and then in it before being re-fed in the boiler after the exhausted steam from the turbine blades has been passed through the steam condenser of a turbine. It is positioned in the flue gas passageway between the chimney's entrance and the boiler's exit. In this, there are numerous thin-walled tubes with small diameters sandwiched between two headers. The flue gases typically flow in a counter flow direction outside the tubes.

HVAC

The HVAC (heating, ventilation, and air conditioning) system of a building can use an air-side economizer to conserve energy by using cool outdoor air to chill the interior space. The energy efficiency of air conditioning with outside air is higher than air conditioning with recirculated air when the temperature of the outside air is lower. Free cooling is the term for the part of the air-side economizer control system when the quantity of enthalpy in the air is acceptable and no extra conditioning of the air is required when the outside air is both sufficiently chilly and dry depending on the environment. In cold and moderate areas, air-side economizers can lower HVAC energy costs while also possibly enhancing indoor air quality, but they are frequently inappropriate in hot and humid climates. Economizers can be employed in areas that encounter a variety of weather systems with the right settings.

A water-side economizer can use water chilled by a wet cooling tower or a dry cooler also known as a fluid cooler to cool buildings without running a chiller when the dry- and wet-bulb temperatures of the outside air are low enough. The water-side economizer is not a real thermodynamic cycle, despite the fact that it has historically been referred to as the strainer cycle. Additionally, a plate-and-frame heat exchanger is frequently added between the cooling tower and chilled water loops in place of the fouling-causing process of passing cooling tower water through a strainer before reaching the cooling coils. The effective operation of the air- and water-side economizers depends on adequate controls, valves, dampers, and maintenance.

Applications

Applications include waste heat recovery, cogeneration, and preheating boiler feed water.

Waste heat boilers and other heating and cooling apparatus also use economizers. Waste heat is heat that is generated by industrial machinery and processes but has no practical use. Economizers are typically utilised with waste heat boilers that have water tube or shell designs. Gases from incinerators, gas turbines, furnaces, and diesel exhaust can all be used in such waste heat devices.

Specifications

Understanding industrial boilers, closed vessels that employ fuel sources like electricity or oil to heat water or produce steam for heating and humidification, is necessary when choosing economizers. Economizers are heat exchangers that heat a fluid, usually water, up to its boiling point but typically not above it in boilers.

Features

Economizers exploit the enthalpy in fluid streams that are hot but not hot enough to use in a boiler, hence their name. Economizers can recover more enthalpy in this method, increasing the boiler's efficiency. Enthalpy, or heat content, can be used in thermodynamics and molecular chemistry to determine how much useful work can be extracted from a closed system under constant pressure. The cold water that is used to fill an industrial boiler can be preheated by economizers using the exhaust gases from the boiler. Economizers can act as feed water heaters or feed water pre-heaters in this manner. Open and closed feed water heaters are the two different styles.

The Boiler Economizer's Purpose

The Heat Transfer Principle underlies the operation of the Economizer in a boiler. Normally, heat transfer takes place from hot to cold conditions. In the case of boilers, the water that needs to be warmed is cold, while the flue gases or exhaust from the boiler outlet are hot. As a result, an increase in feed water temperature is caused by the temperature difference between water and flue gases. Economizers can be either smoke tube type or water tube type of design, depending on the type of operations. The smoke tube type has water on the shell side and flue gases inside the tubes. With contrast, with a water tube type, flue gases are on the shell side and water is inside the tube. In a thermal power plant, an economizer recovers some of the heat lost in the flue gases that are expelled up the chimney and uses it to warm the boiler's feed water. It is merely a heat exchanger with a long heating surface, like Fins or Gills, hot flue gas on the shell side, water on the tube side, and that's all there is to it.

Benefits of Saving Money:

1. The following are economizers' benefits.
2. It boosts the boiler's effectiveness.
3. It cuts down on gasoline usage.
4. As a result of flue gas emissions, heat losses are decreased. Due to a reduced temperature differential, the temperature of flue gases, which ranges between 370°C and 540°C, decreases thermal strains in the boiler.

Economizers' Classification

Economizer is divided into a number of categories. According to Construction Based on construction, the economizers are divided into two groups. Those are Straight-Tube Economizers. Economizers of the Gilled Tube Type. Straight-Tube Economizers. The exterior of the economizer tubes is kept clean and soot-free by soot scrapers that move up and down the tubes. If not, economizer efficiency will drop and heat transfer resistance will rise.

Economizers of the Gilled Tube Type

To improve the heat transmission surfaces, rectangular gills are installed on the bare tube walls. Boiler Economizer Concept for Waste Heat Recovery Steam is often produced from water using boilers. Water is transformed into steam by the transfer of sensible and latent heat. Sensible heat is the quantity of heat needed to raise the temperature of water at constant pressure without changing its liquid state, whereas latent heat is the quantity of heat needed to convert water from a liquid to a vapor at constant temperature and pressure. If no heat recovery equipment is put after the boiler, there are large losses from the boiler because its exhaust is normally in the 200°C to 250°C range. If the exhaust gases that are leaving the boiler at such high temperatures are routed via the economizer to provide the sensible heat needed to heat the feed water to raise its temperature, the heat load on the boiler will be lowered to a larger extent.

Refrigeration Economizer Configurations

Many displays take advantage of this concept by allowing the refrigeration cycle to function as an economizer. This type of system design necessitates specialized knowledge in the field, as well as dexterity and durability in the production of some of the equipment. Oil drags, pressure drop, and electric valve control must all be handled carefully.

System in Two Stages

The number of pressure handlers fitted in the cycle may need to be doubled for two tiered systems. The two different thermal expansion valves (TXV) and

the two distinct stages of gas compression are shown in the diagram.

Boosters and Two Tiered Systems

If two different gas compressors operating in serial display operate together to create the compression, the system is said to be in a two-staged setup. A typical booster installation consists of a two-stage system that receives fluid to cool the first compressor's output before it reaches the input of the second compressor. Both compressors' interstates receive fluid from the liquid line, which is typically managed by solenoid valves, expansion, and pressure.

Setup of a Sub Cooled Booster

A sub cooling heat exchanger (SHX) in a sub cooled booster offers sub cooling for the condensed liquid line. An expansion valve that expands and controls the amount of refrigerant entering at the interstate is present in a typical two-staged cycle of this type. The fluid entering the interstate will tend to evaporate as it expands, causing a drop in temperature overall and cooling the suction of the second compressor when it mixes with the fluid being released by the first compressor. A heat exchanger may be used in this type of setup between the expansion and the interstate, which would allow the second evaporator to provide refrigeration as well, albeit at a lower temperature than the primary evaporator for example, to create air conditioning or preserve food. If the refrigerant entering the interstate flows through a sub cooling heat exchanger that sub cools the main liquid line entering the main evaporator of the same system, the two-stage system is said to be set up in a booster display with sub cooling.

CONCLUSION

A type of heat-exchanging equipment is an economizer. Mechanical devices known as economizers are created to conserve energy or carry out a specific task, like preheating a process fluid. Depending on the size of the boiler and the available space, an economizer may be built directly on the boiler close to the flue gas output in a vertical, horizontal, or cylindrical configuration. It is a heat exchanger mechanism in a steam boiler that, in a thermal power plant, heats fluids or recovers leftover heat from the combustion byproduct, i.e., flue gases, before being expelled by the chimney. Flue gases, which are produced at power plants as combustion exhaust gases, primarily contain nitrogen, carbon dioxide, water vapor, soot, carbon monoxide, etc. Thus, as implied by the device's name, the

economizer in thermal power plants is utilised to streamline the production of electrical energy.

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Application of Mechanical Injection Systems

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ABSTRACT: *The most crucial element in the operation of CI engines is the fuel-injection system. The efficiency of the fuel-injection system has a big impact on how well an engine performs in terms of power production, economy, etc. it was injected system must carry out the crucial task of starting and managing the combustion process. Carburetion and fuel injection both serve the same general objective, which is to prepare the combustible charge. However, in fuel-injection, the fuel speed at the point of delivery is larger than the air speed to atomize the fuel. In contrast, in carburetion, fuel is atomized by processes relying on the air speed greater than the fuel speed at the fuel nozzle.*

KEYWORDS: *Combustion Chamber, Effective Stroke, Each Cylinder, Fuel Injection, Fuel Pump*

INTRODUCTION

The most crucial element in the operation of CI engines is the fuel-injection system. The efficiency of the fuel-injection system has a big impact on how well an engine performs in terms of power production, economy, etc. The crucial task of starting and managing the combustion process falls on the injection system. Carburetion and fuel injection both serve the same general objective, which is to prepare the combustible charge. However, in fuel-injection, the fuel speed at the point of delivery is larger than the air speed to atomize the fuel. In contrast, in carburetion, fuel is atomized by processes relying on the air speed greater than the fuel speed at the fuel nozzle [1], [2]. Towards the end of the compression stroke, the fuel is delivered into the combustion chamber and atomized into very small droplets. Due to heat transmission from the compressed air, these droplets vaporize and combine with the air to generate fuel. The temperature rises above the fuel's self-ignition point as a result of continuing heat transfer from hot air to the fuel. As a result, the fuel spontaneously ignites, starting the combustion process. As the name suggests, mechanical fuel injection is a sort of fuel delivery system that predominantly uses mechanical techniques to inject fuel into an engine [3], [4]. It came first electronic fuel injection was created to provide a more reliable way to supply fuel.

What is the Mechanism of Mechanical Fuel Injection?

Early mechanical fuel systems usually relied on a complex fuel pump to deliver pulses of high-pressure fuel to the injector feeding each cylinder. When the injector received the fuel, it would automatically open as a result of the pressure, spraying fuel into the combustion chamber or intake.

This engine-powered pump would have a number of pistons that were controlled by cams to provide the needed fuel burst. There were not many changes that could be made to the mixture aside than adjusting the output from each piston. As a result, the mixture frequently wasn't appropriate for the engine's load or operating circumstances. Later automobile systems depended on more sophisticated metering components, often known as "fuel distributors," that could more precisely control fuel supply based on airflow entering the engine. The necessary fuel pressure would be produced by a mechanical or electric pump and would then be sent at the appropriate time by the metering unit to the necessary injector [5], [6].

Use Mechanical Fuel Injection in Diesel Engines?

Diesel engines were the first to take advantage of mechanical fuel injection because the method made it simple to directly inject the right amount of high-pressure fuel into the combustion chamber, which a distinctive characteristic of diesel engine is functioning. Mechanically injected diesels are frequently seen, especially in older industrial applications. The majority of modern vehicle diesels still rely on a primary high-pressure pump, or in some cases, a cam-pressurized injector, to produce the necessary fuel pressures while using systems with electronically controlled injectors [7], [8].

Various Mechanical Fuel Injection Methods

At the intake manifold's entrance, there may be one or many injectors. Fuel is sprayed into the combustion chamber without any intermediaries. Fuel is pumped into the intake manifold's inlet runners. Manufacturers switched to electronic fuel injection for what reason? Systems for mechanical fuel injection cost a lot since their components need to be accurately machined. The systems' range of changes was frequently constrained, making it difficult for them to supply the proper quantity as

engine load and environmental circumstances varied. As a result, customers could occasionally encounter poor running and high fuel usage. The systems' high fuel pressures could result in dangerous difficulties, which would be expensive to replace. The pumps and injectors could also wear out, which could result in similar problems and necessitate their rebuilding by a professional [9], [10].

Mechanical fuel injection systems were phased out by the majority of companies that built petrol engines as electronic fuel injection systems improved and became more affordable, and stricter emissions rules required more precise fuel delivery. An overview of the mechanical fuel injection industry the idea was first proposed in the early 1900s for use in diesel engines, but it was quickly transferred to petrol engines as engineers looked for ways to more accurately supply gasoline after the development of a throttle and various metering devices. In World War II, mechanical injection gained popularity as the demands placed on fuel systems in combat aircraft quickly outpaced carburetors' capabilities. As performance demands rose after the war, mechanical fuel injection systems were eventually integrated into race cars, then road automobiles. For instance, Bosch mechanical direct injection was used in the Mercedes-Benz 300 SL when it was first released in commercial form in 1954. Others, such as Chevrolet, Porsche, Alfa Romeo, and Triumph, did the same.

DISCUSSION

The injection system's performance must satisfy the following criteria for the engine to function correctly and perform well:

- i. Accurate fuel injection per cycle metering. Due to the extremely small quantities of gasoline being handled, this is particularly important. Metering mistakes could significantly alter the output from what is desired. As the engine's speed and load needs change, so should the amount of gasoline be metered.
- ii. Proper fuel injection timing during the cycle to achieve optimum power while guaranteeing fuel efficiency and clean burning.
- iii. Properly controlling the injection rate to ensure combustion results in the optimum heat-release pattern.
- iv. Proper fuel atomization, which produces very small droplets.
- v. A suitable spray pattern to facilitate quick fuel and air mixing.
- vi. The combustion chamber's fuel droplets are distributed evenly.
- vii. To provide multi-cylinder engines with identical amounts of metered fuel to each cylinder.
- viii. No lag at the start or conclusion of injection, i.e., to prevent fuel dribbling into the cylinder.

Injection System Classification

In a diesel or constant-pressure cycle engine, the cylinder is only pressurized with air before a fuel injection system injects fuel into it. Air or a mechanical device is utilised to generate the necessary pressure for atomizing the fuel. The injection systems can so be categorized as follows:

- i. Systems for injecting air.
- ii. Systems for solid injection.

System for Air Injection

In this technique, compressed air is used to force fuel into the cylinder. Due of the need for a large, multi-stage air compressor, and this method is rarely employed nowadays. As a result, the engine weighs more and the generation of brake power is further diminished. Good fuel and air mixing, which leads to a greater mean effective pressure, is one perk of the air injection technology. Using fuels with a high viscosity that are less expensive than those used by engines with solid injection systems is another advantage. These benefits are counterbalanced by the need for a multistage compressor, rendering the air-injection system obsolete.

Solid Injection System

Without the need of compressed air, liquid fuel is pumped into the combustion chamber directly with this technique. As a result, it is also known as a solid injection system or an airless mechanical injection system. Individual pump and nozzle systems, unit injector systems, common rail systems, and distributor systems are the different categories of solid injection systems. The following components make up the majority of all the systems mentioned above. The fuel system consists of the following components fuel tank, fuel feed pump to transfer fuel from the main fuel tank to the injection system, injection pump to measure and pressurize the fuel for injection, governor to ensure that the amount of fuel injected is in accordance with variation in load, injector to take the fuel from the pump and distribute it in the combustion chamber by atomizing it into fine droplets, and fuel filters to

Figure 1 depicts a typical setup of several parts for the solid injection system used in a CI engine. Fuel from the fuel tank initially passes through the coarse

filter before being drawn into the plunger feed pump, where a very little increase in pressure occurs. The dust and grime are then removed from the fuel as it enters the fine filter. Fuel enters the fuel pump after passing through the fine filter, where it is pressurized to around 200 bar and then injected into the engine cylinder via the injector. Any injector spillover is returned to the fine filter. For system safety, a pressure relief valve is also included. The components mentioned above enable the aforementioned functions. The sole difference between the different solid injection systems types detailed in the following sections is how the components stated above are operated and controlled.

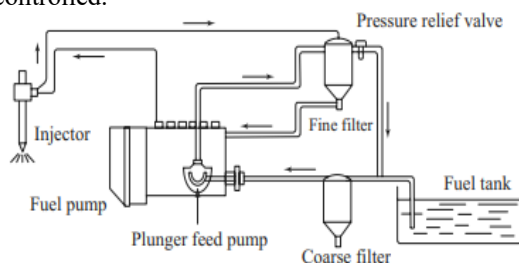


Figure 1: Typical fuel supply system for a CI engine [Ftp.Idu.Ac.Id].

Individual Pump and Nozzle System

Figure 2 depicts the specifics of the particular pump and nozzle system. One pump and one injector are available for each cylinder in this arrangement. For each cylinder in this setup, a separate metering and compression pump is offered. In Figure 2, the pump is located near to the cylinder, while in Figure 2, they are grouped together. The high-pressure pump plunger is moved by a cam, which creates the fuel pressure required for the injector valve to open at the proper moment. The effective stroke of the plunger determines how much fuel will be injected.

System Injector Unit

The unit injector system, shown in Figure 2, combines the pump and injector nozzle into a single housing. One of these unit injectors is installed in each cylinder. By use of a low-pressure pump, fuel is pushed up to the injector, where it is injected into the cylinder by a rocker arm when it is necessary. The effective stroke of the plunger controls how much gasoline is injected. According to Figure 2, the pump and injector can be combined into a single unit.

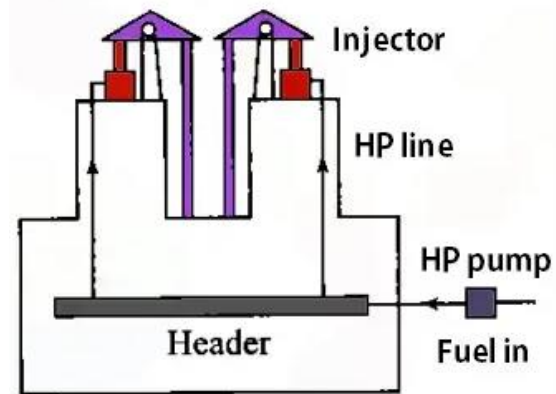


Figure 2: Individual pump and nozzle with separated pumps [Extra Design].

System for Common Rail

A HP pump delivers fuel to a fuel header at high pressure in the common rail. Fuel is forced to each of the nozzles found in the cylinders by the header's high pressure. A mechanically driven by a push rod and rocker arm valve permits the gasoline to enter the correct cylinder through the nozzle at the appropriate time. The fuel header pressure must be sufficient to allow the gasoline to enter and disseminate in the combustion chamber, which is what the injector system was intended for. By changing the length of the push rod stroke, the amount of gasoline entering the cylinder may be controlled. Fuel is delivered at high pressure to a header, where it is metered by injectors one is assigned to each cylinder. Figure 2 provides an illustration of the system's specific.

System of Distributors

A distributor system is depicted schematically in Figure. The pump that pressurizes the fuel in this system also meters and timing it. The fuel pump measures the necessary amount of fuel and then delivers it at the proper moment to a revolving distributor so that it can be delivered to each cylinder. For the pump, there are exactly as many injections strokes every cycle as there are cylinders. Figure 3 provides information about the system in detail. A uniform distribution is automatically achieved because each pump contains a single metering element. Additionally, the cost of the fuel-injection system falls to a figure that is less than two-thirds of the cost of an individual pump system. Various fuel-injection systems are contrasted.

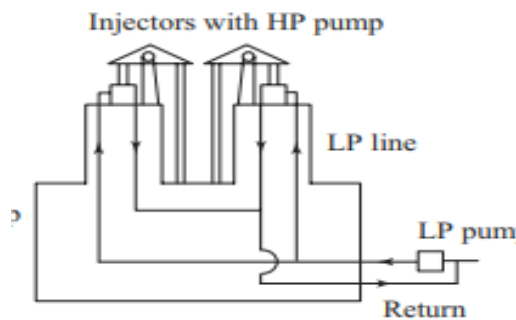


Figure 3: Representing the Unit injector system [Ftp. Idu. Ac. Id].

Distributor System

A distributor system's schematic is depicted in Figure 3. The pump in this system that pressurizes the gasoline also measures and times it. After measuring out the necessary amount, the fuel pump delivers it to a revolving distributor at the appropriate moment for delivery to each cylinder. For the pump, the number of cylinders is equal to the number of injections strokes every cycle. Figure 4 provides the system's specifics. Each pump contains a single metering element; therefore, a consistent distribution is automatically assured. Furthermore, the price of the fuel-injection system drops to a level that is less than two-thirds of the price of an individual pump system. Provides a comparison of various fuel-injection methods.

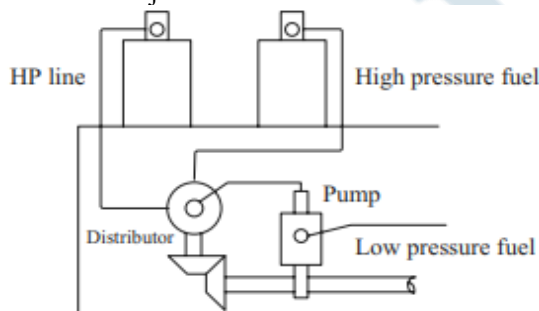


Figure 4: Representing the components of Distribution System [Ftp. Idu. Ac. Id].

Importance Pump

The primary goals of a fuel-injection pump are to properly measure the amount of fuel to be delivered at high pressure (between 120 and 200 bar) at the precise moment to each cylinder's injector. There are two different types of injection pumps Jerk type pumps, Pumps with distributors

Jerk Type Pump

It is made up of a barrel and a reciprocating plunger. The camshaft drives the plunger. provides an illustration of the jerk pump's operating concept. A typical plunger is depicted in a sketch. The plunger

inside the barrel is schematically depicted. Fuel is constantly accessible close to port A and is under low pressure. The plunger's rotating movement along its axis is accomplished by rack D, while its axial movement is accomplished by the cam shaft. The aperture through which fuel is fed to the injector is designated as Port B. At this point, a check valve that is spring-loaded is used to close it. The fuel is poured into the barrel above port A when the plunger is below it. The fuel will exit through port C when the plunger rises and closes port A. This is because in order to pass through port B, it must overcome the check valve's spring force. Thus, it exits via port C, which is the simpler route.

At this point, the rack turns the plunger, which causes port C to close as well. The check valve's orifice B, which leads to the injector, is the sole place where gasoline can escape. This marks the start of injection as well as the plunger's effective stroke. The injection continues until port C is revealed by the helical indentation on the plunger. Now that the aperture B is closed by the check valve, the fuel will exit easily through C. The effective stroke comes to an end as fuel injection stops. Therefore, the axial distance travelled between the times port A is closed off and uncovered is the effective stroke of the plunger. The same series of events take place. But in this instance, port C is exposed earlier. The effective stroke is therefore shorter. It is crucial to keep in mind that although if the plunger travels the same axial distance during each stroke, the length of the effective stroke and the amount of fuel injected are determined by how long the plunger is rotated by the rack. The Bosch fuel-injection pump is a typical illustration of this sort of pump.

This pump simply has one pumping element, and a rotor distributes the fuel to each cylinder. The rotor has two sets of radial holes each corresponding to the number of engine cylinders spaced at various heights in addition to a central longitudinal tunnel. While the second set is connected to delivery lines going to the injectors of the individual cylinders, the first set is connected to the pump inlet via a central tube. When the pump plungers move apart, the fuel is pulled from the inlet port into the central rotor passage. Fuel is provided to each cylinder in turn whenever the radial delivery tube in the rotor aligns with the delivery port for any cylinder. The tiny size and low weight of this type of pump are its main features. In a CI engine, the fuel delivery is unaffected by the characteristics of the injection pump and the air intake. While air intake decreases with speed, fuel delivery from a pump increases with speed. As a result, we over fuel.

CONCLUSION

The design of the injection system has a significant impact on the performance of diesel engines. In actuality, the best fuel injection system designs were directly responsible for the most significant advancements made in diesel engines. Although the system's primary function is to provide fuel to a diesel engine's cylinders, how that fuel is delivered has an impact on the engine's performance, emissions, and noise levels. The diesel fuel injection system supplies fuel under extremely high injection pressures, in contrast to its spark-ignited engine equivalent. This suggests that in order to function for longer periods of time that meet the engine's durability goals, the system component designs and materials should be chosen to resist larger stresses.

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An Introduction to Electronic Injection Systems in IC Engine

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ABSTRACT: *The electronic fuel injection system's goals were to limit the number of harmful particles released from the exhaust pipe and optimize fuel usage. Carbon buildup in mechanical components like the fuel system, injectors, or intake valves is prevented by this device. Several factors such as speed, temperature, pressure, etc. are taken into consideration by the electronic injection system when you are driving to control the entry of the air-fuel mixture into the combustion chamber. The gasoline travels to the manifold's end and builds up there, as can also be seen. The mixture entering into the end cylinders is enriched as a result. The leanest mixture, however, is delivered to the middle cylinders, which are in close proximity to the carburetor.*

KEYWORDS: *Cold Start, Electronic Control, Fuel Injection, Injection System, Port Injection*

INTRODUCTION

A significant year in the history of IC engine development was 1903. The Wright Brothers made their first flight powered by a SI engine during that year, using a gear pump to inject fuel into the intake ports. Soon after, in 1906, Santos Dumont of Brazil completed his first plunger-powered flight over Europe. The well-known Grade End dekker, who flew 13 kilometers in 1909, also used petrol injection in a two-stroke engine in which the crankcase pressure of each cylinder served as the injection pressure for the fuel. As a result, the idea of an injection system for petrol engines is not new. For automotive engines, a continuous, measured supply of the gasoline-air combination is required to ensure smooth operation [1], [2].

In a petrol injection system, an injector injects fuel into the intake manifold or close to the intake port. The injector receives the petrol from the pump and atomizes it finely before spraying it into the air stream. The mixing of petrol with the air stream in this situation is superior to carburetion. The flow of the mixture into the intake manifold is controlled by the position of the throttle valve in both the carburetor and the fuel injection system, as can be seen. The use of petrol injection in vehicles is common today. Petrol injection systems have been released by reputable manufacturers like Ford, Daewoo, Fiat, Mitsubishi, Honda, and others [3], [4].

Why Injects Gasoline?

It is challenging to achieve homogeneity of mixture strength in each cylinder of a multicylinder engine with a carburetor. An intake manifold of a multicylinder engine with a typical pattern of

mixture distribution is shown in Figure 1. The intake valve of cylinder 2 is open, as can be seen. As a result, the air-gasoline combination is delivered to the individual cylinders in varied amounts and richness. The port injection system can address this issue, known as misdistribution, by injecting the same volume of fuel into each intake manifold. Therefore, the creation of injection systems for petrol engines is critical. By using petrol injection, misdistribution can be greatly reduced and each cylinder can receive the same richness of the air-gasoline mixture [5], [6].

Injection System Types

One category for the fuel injection system is:

- i. Direct injection of petrol into the cylinder.
- ii. Timed and continuous port injection.
- iii. Manifold injection.

The aforementioned fuel injection systems can be divided into single-point and multi-point injection groups. One or two injectors are installed inside the throttle body assembly in the single point injection system. The intake manifold's center or a single point receives fuel sprays. Throttle body injection is another name for single point injection. Each engine cylinder has a separate injector with multipoint injection. This method uses many injection points for fuel. This is more typical and is frequently referred to as a port injection system. As was already noted, there are two types of petrol fuel injection systems that can be employed in spark-ignition engines. continuous injection and timed injection. Systems for continuous injection a rotary pump is typically used in this arrangement. A fuel line gauge pressure of roughly 0.75 to 1.5 bar is maintained by the pump. A nozzle in the manifold directly behind

the throttle plate is where the system injects the fuel [7], [8].

Fuel is fed into an engine that has a supercharger at the supercharger's entrance. Depending on the load and speed, the Electronic Control Unit (ECU) decides when and how long to inject fuel. When the engine is running at its top speed, this system's fuel delivery pump sends fuel at a low pressure of roughly 2 bar. The system's additional components include a nozzle and a fuel metering or injection pump. At pressures ranging from 16 to 35 bar, the nozzle injects gasoline into the combustion chamber, the manifold, or the cylinder head port. A timed injection system typically injects fuel at the start of the suction stroke. Injection starts during maximum power operation after the exhaust valve closes and typically ends after BDC. Manifold injection is inferior to direct in-cylinder injection and is never preferred. In this situation, higher volumetric efficiency can be attained by using both low and high volatile fuels. However, it was found that when the car is utilised for daily transportation, direct injection results in oil dilution during the repeated warm-up stages [9], [10].

DISCUSSION

Injection System Components

The goals of the fuel injection system are to precisely measure, atomize, and evenly distribute the fuel throughout the cylinder's air mass. The engine's load and speed requirements must be met while maintaining the necessary air-fuel ratio. A variety of fuel injection system components are necessary to carry out the aforementioned activities; their respective functions are listed below.

1. Fuel is moved from the fuel tank to the injector by the pumping element. This comprises any required pipework, filters, etc.
2. Measuring and supplying gasoline at the rate necessary for the load and speed conditions of the engine.
3. A mixing element that atomizes the fuel and combines it uniformly with air.
4. Metering control feature modifies the rate of metering in line with the load and engine speed.
5. modifies fuel-to-air ratio in accordance with load and speed requirements.
6. A distributing component that evenly distributes the metered gasoline throughout the cylinders.
7. Timing control, which regulates when the fuel-air mixing process starts and stops.

8. Ambient control accounts for variations in air or fuel pressure and temperature that could have an impact on the system's various components.

Electric Fuel Injection System

To precisely measure and inject the proper amount of fuel into the engine cylinders, modern petrol injection systems employ engine sensors, a computer, and solenoid-operated fuel injectors. Electrical and electronic equipment are used by these systems, also known as electronic fuel injection (EFI), to monitor and manage engine operation. Electrical signals in the form of current or voltage are received by an electronic control unit (ECU) or the computer from numerous sensors. The injectors, ignition system, and other components connected to the engine are then operated using the stored data. As a result, the car gets better mileage and emits less unburned gasoline as pollutants. The following are examples of typical sensors for an electronic fuel injection system:

Exhaust Gas or Oxygen Sensor: determines the air-fuel ratio by measuring the amount of oxygen in the engine exhaust. The air-fuel ratio has an impact on sensor output voltage. The engine temperature sensor measures the coolant temperature, and using this information, the computer changes the mixture's strength to the rich side for cold starting.

Air Flow Sensor: regulates the amount of fuel by monitoring the mass or volume of air moving into the intake manifold.

Air Inlet Temperature Sensor: used to fine-tune the mixture strength, this sensor measures the temperature of ambient air entering the engine. In order to optimize the mixture flow for engine speed and acceleration, the throttle position sensor detects the movement of the throttle plate. The engine intake manifold vacuum is monitored by the manifold pressure sensor, which allows the mixture strength to be changed in response to variations in engine load. The camshaft position sensor measures the engine's camshaft and crankshaft rotation to determine the time and pace of injection. A knock sensor is a microphone-style sensor that can be used to detect ping or resonance noise and delay the ignition. In an EFI, the fuel injector is merely a fuel valve. The injector remains closed when it is not powered up by spring pressure, which prevents fuel from entering the engine. The injector armature is drawn to the magnetic field created when the computer sends the signal through the injector coil. Following that, fuel shoots into the intake manifold. The length of time that each injector is powered on and kept open is indicated by the injector pulse

width. Based on the information from the various sensors, the computer chooses and regulates the injector pulse width. The computer will detect a wide-open throttle, high intake manifold pressure, and high inlet air flow when the engine is fully loaded. Then, in order to enrich the mixture and enable the engine to produce more power, the ECU will widen the injector pulse width. The ECU will reduce the pulse width while the engine is idle or under low load to keep the injectors closed for a longer period of time. As a result, the air-fuel combination will be leaner, improving fuel efficiency. A cold start injector is also part of the electronic fuel injection system. When the engine is cold, this additional injector pours fuel into the center of the engine's intake manifold. It does the same task as a carburetor choke. In extremely cold temperatures, the cold start injector ensures simple engine startup.

Benefits of the EFI System

The spark ignition engine with an EFI system has the following advantages over a carburetor unit:

- i. An increase in volumetric efficiency brought on by the intake manifolds' relative low resistance, which will result in lower pressure losses. It virtually eliminates the need for manifold heating and removes the majority of carburetor pressure losses.
- ii. Because the gasoline is injected into or close to the cylinder and does not need to flow through the manifold, manifold wetness is removed.
- iii. Because fuel atomization is independent of cranking speed, starting will be simpler.
- iv. The engine will be less likely to knock with improved atomization and vaporization.
- v. The throttle plate no longer develops ice.
- vi. Less volatile fuel can be used because distribution is not dependent on vaporization.
- vii. Even when the vehicle is in different locations, such as turning, travelling on grades, uneven roads, etc., the variation in the air-fuel ratio is essentially minor.
- viii. Because the injection unit's position is less important, the engine can be built lower.

EFI System Benefits

The following are some drawbacks of the EFI system expensive maintenance costs, challenging service and potential sensor malfunction.

Mafia (Multiplied Fuel Injection) System

The Multi-Point Fuel Injection (MPFI) system's primary goal is to deliver the right proportion of

petrol and air to the cylinders. These systems operate using one of two fundamental configurations: port injection or throttle body injection.

Port Injection

The injector is positioned next to the intake port on the side of the intake manifold in the port injection arrangement (Figure 1). Inside the intake manifold, the injector sprays petrol into the air. It takes some time for the petrol and air to blend evenly. After passing through the intake valve and into the cylinder, this petrol and air mixture.

Injector System for the Throttle Body

The simplified drawing of the throttle body injection system also known as single point injection is shown in Figure 1. The volume of air entering the intake manifold is controlled by the throttle valve in this throttle body, which is identical to the carburetor's throttle body.

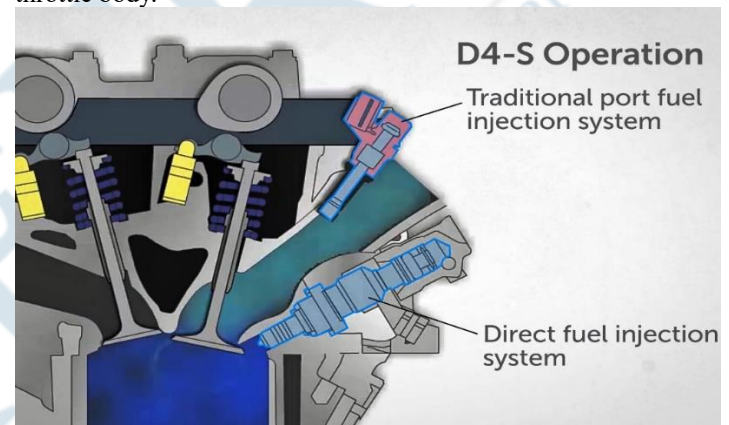


Figure 1: Representing the Port Injection system [Engine Labs].

A little amount of space is left above the throttle body's throat for an injector. The intake manifold is where petrol and air are mixed after being sprayed into the air by the injector. This mixture then reaches the intake manifold after passing via the throttle valve. Fuel-injection systems can be timed or continuous, as was before indicated. Petrol is pulsed out of the injectors in the timed injection system. Petrol is continually sprayed from the injectors in a continuous injection system. Both the throttle-body injection system and the port injection system can be continuous or pulsed systems. The amount of petrol injected in either method is based on the engine's speed and power requirements. MPFI systems are divided into two categories in some literature: both L-MPFI and DMPFI.

D-MPFI System

The manifold fuel injection system is called the D-MPFI system. In this style, the intake manifold

vacuum is initially detected. Additionally, it detects the air volume based on its density. A block diagram of how the D-MPFI system operates. The manifold pressure sensor measures the intake manifold vacuum as air enters the intake manifold and transmits the information to the ECU. The speed sensor also communicates to the ECU data on engine rpm. To control the amount of petrol supply for injection, the ECU in turn sends commands to the injector. The gasoline and air combine as the injector sprays fuel into the intake manifold, and the combination then enters the cylinder-MPFI System a port fuel-injection system is the L-MPFI system. The engine speed and the amount of air that actually enters the engine control the fuel metering in this type. This process is known as air-mass or air-flow metering. The block diagram of an L-MPFI system. The air flow sensor detects the volume of air entering the intake manifold and transmits data to the ECU. In a similar manner, the speed sensor informs the ECU of the engine speed. To control the amount of petrol supply for injection, the ECU processes the data it has received and provides the relevant commands to the injector. When injection occurs, the petrol and air combine, and the resulting mixture enters the cylinder.

Mpfi System Functional Divisions

Three distinct functional components make up the MPFI system:

- i. The electronic control system.
- ii. The fuel system.
- iii. The air induction system.

The sections that follow provide descriptions of these functional categories.

MPFI-Electronic Control System

The ECU receives signals from the sensors that track the intake air temperature, oxygen levels, water temperature, starter signal, and throttle position. The air-flow sensor informs the ECU about the volume of intake air through its signals. The engine speed is communicated through the ignition sensor. To regulate the amount of gasoline to be injected, the ECU processes all of these signals and communicates the proper instructions to the injectors. The cold start injector, which is a component of the fuel system, is operated by the ECU when the cold-start injector timing switch needs to be off.

MPFI-Fuel System

The MPFI-fuel system is depicted as block diagrams. The fuel pump supplies fuel in this system. The cold start injector timing switch controls the cold start injector during startup. The

air-fuel mixture is enhanced by the cold start injector, which injects fuel into the air intake chamber. The fuel pressure is controlled by the pressure regulator. The ECU sends commands to the injectors, which then inject fuel into the intake manifold.

System for MPFI-Air Induction

The block diagram in depicts the MPFI-air induction system. The air intake chamber and intake manifold receive the correct volume of air thanks to the air cleaner, air flow meter, throttle body and air valve. Just the right amount of air is provided to ensure full combustion. Electric control system The Electronic Control Unit (ECU), which controls how long the injectors operate, is the essential component of the electronic control system. Additionally, a starter timing switch is present, which manages how the cold start injector functions when the engine is starting. The operation of the gasoline pump is controlled by a circuit opening relay. A resistor is also included, and it stabilizes the injector's operation.

Electronic Control Unit (ECU)

The electronic control unit (ECU) of the electronic control system receives signals from the sensors and chooses the injectors' opening time, which also controls the injection volume.

Cold Start Injector

It's typically difficult to start an engine when it's cold. A richer mixture is needed to start a cold engine. The objective of the cold start injector is to start the engine with more fuel. The main injector, the air valve, and the cold start injector. The power for the cold start injector's internal solenoid valve, which opens and closes to allow for fuel injection, is supplied by a battery. Not too much fuel should be injected. Therefore, a timing switch regulates the length of the injection time. An electric heating coil and a bimetal element make up the timing switch. The starter motor turns the engine while it is cold. The cold start injector now adds gasoline to the mixture to enrich it. Fuel is also injected simultaneously by the main injector. Only the main injector will inject gasoline into the cylinder when the engine is hot, and the cold start injector will cease to operate.

Timing of Injection

Take a look at a four-cylinder engine's cylinder. Each cylinder's inlet manifold receives fuel injection at a different time. The timing at which fuel is injected into the inlet manifold is referred to as the injection timing. Here is a description of how this four-

cylinder engine's one cylinder's injection timing works. During the exhaust stroke of one cylinder, the piston rises from BDC (Bottom Dead Centre) to TDC (Top Dead Centre). When the piston is just about to reach TDC At roughly 60 crank angles before TDC during this exhaust stroke, gasoline is injected into this cylinder's inlet manifold. In the air intake chamber, the air and gasoline being injected mix. As a result, the air-fuel mixture is produced. The intake valve opens at the start of the suction stroke, allowing the air-fuel mixture to be drawn into the cylinder. The timing of the fuel injection inside the inlet manifolds of the other three cylinders varies depending on the firing order. The ECU determines the proper injection time for each cylinder in this four-cylinder engine, and the air fuel mixture is made accessible at each suction stroke. The injection valve is kept open for a longer period of time by the ECU in order to meet the operational conditions. For instance, the injection valve will be opened for a longer period of time to feed the engine with more fuel when the car is accelerating. The Applications Consist of

1. Modern control software is incorporated into the EFI System to meet performance, fuel consumption, and pollution requirements.
2. Additionally, the system uses Smart Ignition technology to regulate the ignition system, giving OEMs the freedom to achieve best-in-class fuel efficiency.

The benefits of an electronic fuel injection system include:

1. Improved volumetric performance of the engine.
2. Wetting of the manifold is eliminated by direct fuel injection into the cylinder.
3. High fuel atomization even at low cranking speeds because atomization is independent of cranking speed.
4. Enhanced atomization and vaporization results in less knocking
5. Ice buildup at the throttle plate is stopped
6. Low volatility fuels can be used since distribution is unaffected by vaporization.
7. Good engine performance is achieved since the change in the fuel/air ratio is nearly nonexistent.
8. Due to the injection unit's position not being as crucial, the height of the engine can be reduced.
9. Problems with the Electronic Fuel Injection System.

The drawbacks include:

1. Expensive upkeep.
2. Service challenges.
3. The potential for some sensors to malfunction.

CONCLUSION

Electronic technology was first applied to cars in 1965. 30–40% of a vehicle's cost is made up of electronic components. To maximize power and efficiency, electronics and computers are utilised in automobiles. EFI systems made use of a range of sensors to measure things like temperature, gas pressure, the position of the throttle valve, and airflow rate. The Electronic Control Unit (ECU), which is effectively a computer, receives the data from the sensors. To attain the highest possible levels of power, efficiency, and emissions, this ECU analyses data and regulates injectors and other components.

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A Brief Introduction on Ignition System

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ABSTRACT: Heat engines employ ignition devices to start combustion by lighting the fuel-air mixture. In internal combustion engines with spark ignition (like petrol engines), the ignition system produces a spark to ignite the fuel-air mixture right before each combustion stroke. An ignition system is often only used during startup for gas turbine engines and rocket engines. Diesel engines do not have an ignition system since they employ compression ignition to ignite the fuel-air mixture utilizing the heat of compression. For easier starting in cold weather, they typically incorporate glow plugs that pre-heat the combustion chamber.

KEYWORDS: Contact Breaker, Combustion Chamber, High Voltage, Ignition System, Spark Plug.

INTRODUCTION

In spark-ignition engines, as compression ratio is lower, and the self-ignition temperature of gasoline is higher, for igniting the mixture for the initiation of combustion an ignition system is a must. The electrical discharge produced between the two electrodes of a spark plug by the ignition system starts the combustion process in a spark-ignition engine. This takes place close to the end of the compression stroke. The high temperature plasma kernel created by the spark, develops into a self-sustaining and propagating flame front. In this thin reaction sheet certain exothermic chemical reactions occur [1], [2]. The function of the ignition system is to initiate this flame propagation process. It must be noted that the spark is to be produced in a repeatable manner viz., cycle-by-cycle, over the full range of load and speed of the engine at the appropriate moment in the engine cycle. By implication, ignition is merely a prerequisite for combustion [3], [4]. Therefore, the study of ignition is a must to understand the phenomenon of combustion so that a criterion may be established to decide whether ignition has occurred [5], [6].

Although the ignition process is intimately connected with the initiation of combustion, it is not associated with the gross behaviour of combustion. Instead, it is a local small-scale phenomenon that takes place within a small zone in the combustion chamber. In terms of its simplest definition, ignition has no degree, intensively or extensively. Either the combustion of the medium is initiated or it is not. Therefore, it is reasonable to consider ignition from the standpoint of the beginning of the combustion process that it initiates. The fuel-air combination is ignited by ignition systems in heat engines to start combustion. When a fuel-air mixture is to be burned, a spark is produced by the ignition system right before each combustion stroke in internal

combustion engines with spark ignition such as petrol engines. Typically, the ignition system is only used at startup for gas turbine engines and rocket engines. Diesel engines do not have an ignition system since compression ignition ignites the fuel-air mixture using the heat created during compression. Glow plugs are typically installed in them to pre-heat the combustion chamber and help with cold weather starting [7], [8]. The ignition magneto and trembler coil systems used in early automobiles were replaced by distributor-based systems, which were introduced in 1912. In the latter half of the 20th century, electronic ignition systems first employed in 1968 became widely used, and during the 1990s, coil-on-plug variations of these systems have also been popular. Older spark-ignition engines like gasoline engines use ignition magnetos, often known as high-tension magnetos. High voltage pulses are produced for the spark plugs using a magneto and a transformer. High-tension refers to high-voltage in older terminology [9], [10]. Induction coil ignition systems predominantly supplanted ignition magnetos, which were prevalent in early 20th-century automobiles. The majority of engines without batteries nowadays, including those in lawn mowers and chainsaws, are the only ones that use ignition magnetos. In spite of the fact that a battery is available, it is also employed in contemporary piston-engine aircraft [citation needed] to keep the engine from being dependent on an electrical system. Digital systems additionally, see ignite through capacitor discharge. Delco-Remy tested the first electronic ignition a cold cathode type in 1948, and Lucas unveiled a transistorized ignition in 1955, which was utilised in Formula One engines from BRM and Coventry Climax in 1962. The Autolytic Electric Transistor 201 and Tung-Sol EI-4 (thyatron capacitive discharge) were the first EI products available in the aftermarket that year. On some 1963 models, Pontiac was the first automaker

to offer an optional EI, the breaker less magnetic pulse-triggered Dielectronic, which was also offered on some Corvettes. Also in 1963, Hyland Electronics in Canada produced the first fully solid-state (SCR) capacitive discharge ignition. Following a fleet test in 1964 and the introduction of optional EI on some models in 1965, Ford equipped the Lotus 25s entered at Indianapolis the following year with breaker less systems that were designed by FORD. The GT40s raced by Shelby American, Holman and Moody, and other teams used this electronic system. The Mark II-GT Ignition and Electrical System, Publication #670068, was presented by Robert C. Hole from Ford Motor Company at the SAE Congress in Detroit, Michigan, from January 9–13, 1967. Earl W. Meyer of Chrysler started working on EI in 1958, and he didn't stop until 1961. As a result, Chrysler used EI in 1963 and 1964 on their NASCAR hems. The 1965 release of Presto-O-Lute's CD-65, a capacitance discharge (CD)-based product, boasted an unprecedented 50,000-mile warranty. This is different from the Presto-O-Lite system, which wasn't CD-based, was introduced on AMC products in 1972 and became standard equipment for the 1975 model year. When it came to 1967 model-year Oldsmobile, Pontiac, and GMC vehicles, a comparable CD player was optional and offered by Delco in 1966. The Motorola breaker less CD system was introduced in 1967 as well. The Delta Mark 10 capacitive discharge ignition, which was marketed as an aftermarket option and first appeared in 1965, is widely regarded as the most well-known electronic ignition. The Jaguar XJ Series 1, Chrysler, Ford and GM, and the Fiat Dino were the next production vehicles to come standard with EI. The Fiat Dino was the first to do so in 1968. For its factory Super Stock Coronet and Belvedere drag racers, Dodge and Plymouth used Presto-O-Lute's "Black Box" ignition amplifier, which was designed to lighten the pressure on the distributor's breaker points during high rpm runs. The amplifier had a duct that brought in outside air to cool it, and it was mounted on the internal side of the firewall of the automobiles. The distributor and spark plugs are still there, just like in the mechanical system, and that is all. Greater reliability and longer service intervals result from the absence of moving parts compared to a mechanical system. The paired cylinders are 1/4 and 2/3 on other methods use an on-crankshaft mounted magnetic crank angle sensor instead of the distributor as a timing device to start the ignition when it should.

DISCUSSION

Energy Requirements for Ignition

The total enthalpy required to cause the flame to be self-sustaining and promote ignition, is given by the product of the surface area of the spherical flame and the enthalpy per unit area. It is reasonable to assume that the basic requirement of the ignition system is that it should supply this energy within a small volume. Further, ignition should occur in a time interval sufficiently short to ensure that only a negligible amount of energy is lost other than to establish the flame. In view of this last-mentioned condition, it is apparent that the rate of supply of energy is as important a factor as the total energy supplied. A small electric spark of short duration would appear to meet most of the requirements for ignition. A spark can be caused by applying a sufficiently high voltage between two electrodes separated by a gap, and there is a critical voltage below which no sparking occurs. This critical voltage is a function of the distance between the electrodes, the fuel-to-air ratio, and the gas pressure. Further factors that affect the amount of energy needed include how the voltage is increased to the critical level, as well as the arrangement and health of the electrodes. The law of conservation of energy is followed by an igniting process. It can therefore be thought of as a balance of energy between:

- i. Energy supplied by an external source.
- ii. Energy created by chemical reaction.
- iii. Energy dissipated to the environment by thermal conduction, convection, and radiation.

The Energy of the Spark and Duration

To start the combustion process in a homogeneous mixture inside the cylinder, a spark with energy of around 1 mJ and a duration of a few microseconds would be sufficient. But in reality, things aren't quite as they should be. The voltage needed to spark an object depends significantly on the pressure, temperature, and density of the mixture between the spark plug electrodes. Therefore, during the whole range of engine operation, the spark energy and duration must be of sufficient order to start combustion under the worst circumstances anticipated in the area of the spark plug. Typically, reliable ignition is produced if the spark energy reaches 40 mJ and the duration is longer than 0.5 ms. The loss of electrical energy through the deposits on the spark plug electrodes may stop the spark discharge if their resistance is high enough.

Ignition Spark and Spark

The insulation between the electrodes dissolves under the high voltage created by the ignition system, current flows in the discharge phenomena, and an electrical spark is formed between the spark plug's ground electrode and centre electrode. This spark energy causes the compressed air and fuel mixture to ignite and burn. This discharge is incredibly complicated and lasts for only a very short time approximately 1/1000 of a second. The spark plug's job is to consistently produce a powerful spark between the electrodes precisely at the designated moment in order to create the trigger for gas mixture combustion.

Playing a Video

00:00 To advance one second, press the Left/Right arrow keys; to advance ten seconds, press the Up/Down arrows. 00:00 Ignition Electrical spark ignition happens as a result of the spark activating the fuel particles between the electrodes to discharge, causing a chemical reaction (oxidation), igniting the reaction's heat source, and forming the flame core. A flame core eventually forms as a result of this heat activating the nearby air-fuel mixture, spreading the combustion to the immediate surrounds. The flame core is extinguished and combustion ceases, nevertheless, if the quenching effect between the electrodes the work of the electrodes absorbing the heat and extinguishing the flame is greater than the heat generating action of the flame core. A dependable ignition can be anticipated with a wide plug gap since the flame core is larger and the quenching effect is lessened. However, if the plug gap is too great, a high discharge voltage is required, the coil's performance limits are reached, and discharge is rendered impossible.

System of Initiation

In theory, a typical ignition system should deliver a large enough voltage across the spark plug electrodes to influence the spark discharge. In addition, it needs to, under all working circumstances, provide the spark with the necessary energy to light the combustible mixture nearby the plug electrodes. It should be noted that the ideal spark timing for a particular engine design depends on the mixture's composition, inlet manifold pressure, and engine speed. To provide the spark with the right energy and duration at the right time, these considerations should be made into the design of a traditional ignition system. An electrical circuit with an air gap has a high resistance because air is a poor conductor of electricity. But when a high voltage is supplied between a spark plug's

electrodes, a spark is created across the gap. The spark-ignition system is used when such a spark is generated to ignite a homogenous air-fuel mixture in the combustion chamber of an engine. According to how the primary energy required to run the circuit is made accessible, the ignition systems are divided into the following categories: Battery ignition systems and magneto ignition systems are two examples.

Ignition System Requirements

For an engine to perform properly, the ignition system must run smoothly and consistently. These are the specifications for such an ignition system:

- i. At the appropriate time, it should produce a strong spark between the electrodes of the plugs.
- ii. It ought to work well across the entire engine speed range.
- iii. It must be portable, efficient, and dependable in use.
- iv. It needs to be portable and simple to maintain.
- v. It should be affordable and simple to use.
- vi. The radio and television receivers inside a car shouldn't be affected by the high voltage source's interference.

System for Battery Ignition

A battery ignition system is used by the majority of modern spark-ignition engines. In this method, a 6- or 12-volt battery is used to provide the spark-producing energy. A battery ignition system can be built in a huge variety of ways. It is based on the type of ignition energy storage and the ignition performance needed for the specific engine. This is due to the fact that an ignition system is only one component of the internal combustion engine, which is the engine's heart and does not function entirely independently. Therefore, it is crucial that the ignition system is adequately suited to the engine. Battery ignition systems are seen in passenger cars, light trucks, some motorbikes, and big stationary engines. Figure 1 depicts the specifics of a six-cylinder engine's battery ignition system.

The following are the system's main parts. Batteries, ignition switches, ignition coils, contact breakers, capacitors, distributors, ballast resistors, spark plugs, and distributors are among the components. The ignition coil's secondary side is where the final four components are located, whereas the primary side is where the first three components on the list above are located. The following sections provide a brief description of each component's specifics. Battery a storage battery is utilised to supply the electrical energy needed for ignition. A dynamo

powered by the engine charges it. It is able to transform chemical energy into electrical energy thanks to electro-chemical processes. The battery needs to be mechanically sturdy to resist the stresses that it is frequently put through. A battery can have a trouble-free life of up to two years with basic care and attention.

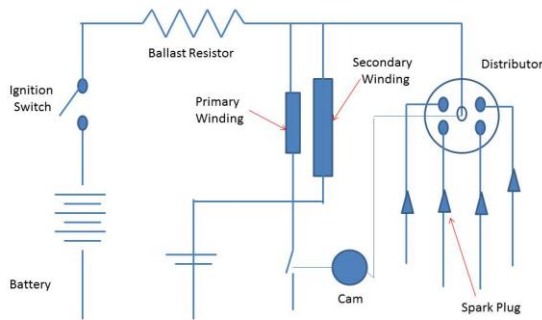


Figure 1: Battery ignition system for a six-cylinder engine [Mech 4 Study].

Each cell in a lead acid battery has a nominal potential of 2 volts when completely charged, and the cells are connected in series. There are three of these cells in a six-volt battery and six in a twelve-volt battery. Six cells are linked in series to create a 12-volt battery, and the positive of one cell is connected to the negative of the next. The lead acid battery and the alkaline battery are the two types of batteries used in spark-ignition engines. The former is used in light-duty commercial vehicles, whereas the latter is utilised in heavy-duty commercial vehicles. The primary winding of the ignition coil is linked to the ignition switch battery by a ballast resistor and an ignition switch. The ignition system can be turned on or off with the aid of the ignition switch.

Ballast Resistor

To control the primary current, a ballast resistor is offered in series with the primary winding. The purpose of this is to avoid damage to the spark coil from overheating in the event that the engine stalls while the breaker is closed or is run for an extended period of time at low speed. This coil is formed of iron wire, and iron has the feature that, when a certain temperature is exceeded, its electrical resistance increases very quickly. Therefore, if the primary current flows virtually constantly, the ballast coil reaches a temperature higher than where this abrupt increase in resistance occurs. The added resistance in the primary circuit keeps the primary current at a safe level. This resistor is bypassed when starting from scratch in order to increase current flow in the primary circuit.

Ignition Coil

In a traditional ignition system, the ignition coil serves as the source of ignition energy. This coil stores the energy in its magnetic field and transfers it to the proper spark plug at the appropriate time in the form of an ignition pulse via the high-tension ignition cables. The ignition coil's job is to boost the battery's 6 or 12 volts to a high voltage that will cause an electric spark to occur between the electrodes of the spark plug. The primary and secondary windings of the ignition coil are two insulated conducting coils that are connected to a magnetic core made of soft iron wire or sheet. The secondary coil is made up of around 21,000 turns of 38–40-gauge enamelled copper wire that is well insulated to handle the high voltage. It is wound close to the core, with one end linked to the secondary terminal and the other grounded to either the primary coil or the metal case.

The primary winding, which is situated outside the secondary coil, is typically made of 200–300 turns of 20-gauge wire, resulting in a resistance of about 1.15. The ends are attached to terminals outside. Since the primary coil is looped over the secondary coil, it is easier to dissipate heat in the primary than in the secondary. When constructed, the entire equipment is housed in a metal container and creates a tidy, little unit. The thickly insulated terminal block, which holds three terminals, is located on top of the coil assembly. The two ends of the primary are connected to the two smaller terminals (Figure. 2), which are often labelled SW (switch wire) and CB (contact breaker). One end of the secondary winding is linked to the distributor's moulded cover's central high-tension connection, as shown in Figure 2. The primary is attached to the other end. This central terminal is linked to the central terminal of the distributor by an external high-tension cable.

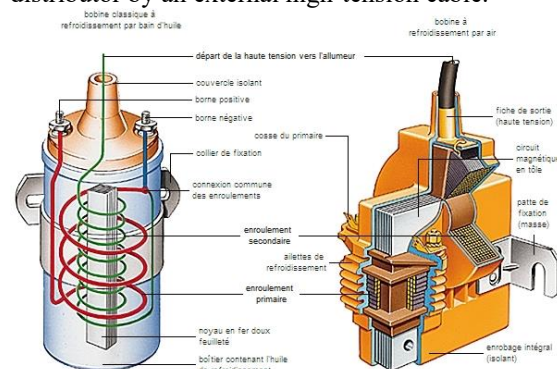


Figure 2: Representing the Ignition coil, section [Laro usse]

Contact Breaker

This is a mechanical tool used to establish the ignition coil's primary circuit. It primarily comprises of a fixed metal point that another metal point bears on while being supported by a spring-loaded pivoting arm. Each point has a circular flat face with a diameter of roughly 3 mm and is made of one of the hardest metals, usually tungsten. While the arm to which the moveable contact point is attached is electrically insulated, the permanent contact point is earthed by being mounted to the contact breaker assembly's base. When the points are closed, current flows; when they are open, the circuit is broken and the current flow ceases. The pivoted arm typically has a heel or rounded portion of hard plastic connected in the middle.

This heel bears on the cam that is pushed by the engine. The points are consequently pushed apart and the circuit is disrupted each time the cam passes beneath the heel. When the points aren't separated by the action of the cam, the spring-loaded pivoting arm holds them together, closing the primary circuit. The contact breaker points' condition and setting are crucial. Throughout their service life, the points endure a tremendous amount of hammering. Depending on the state of the points, uneven wear may call for prefacing or replacement. At 3000 revolutions per minute, an eight-cylinder engine needs 12000 sparks every minute, or 200 sparks per second. The motion of the breaker arm must be kept to a minimum to achieve a positive spark and the breaker arm must be made extremely light if the breaker is to perform satisfactorily at this speed.

Capacitor

The construction of the ignition capacitor follows the same straightforward design principles as all other electrical capacitors: two metal plates are arranged face to face and separated from one another by an insulating substance. Although the insulation is frequently just air as in the case of air capacitors, it typically consists of some high-quality insulating material that is appropriate for the specific technical requirements. Because of space constraints, this material must be as thin as possible while still being able to withstand electrostatic stresses without being damaged. The actual metal plates are typically replaced with metal foil or metallic layers that are produced by evaporation on the insulating substance. These tiny strips, for instance, which are made of two strips of aluminium foil and several layers of particular capacitor paper, are rolled up in a solid roll to conserve space. The entire roll is first saturated in an oily or waxy substance to improve the insulating characteristics of the paper, and then

it is placed inside a metal shell for protection against moisture, outside physical contact, and damage. Contacts are then bonded to the two metal strips.

Distributor

The distributor's job is to deliver the ignition surges to each spark plug in the right order and at the right times. Depending on whether an engine has four, six, or eight cylinders, four, six, or eight ignitions are produced for each distributor shaft spin. Because we typically only wish to use one ignition circuit, the use of a distributor significantly simplifies a battery ignition system. Due to the absolute necessity for the distributor to operate in synchronism with the crankshaft, the contact breaker and the spark advance mechanism are merged with the distributor as a single unit. The brush type and the gap type are the two categories of distributors. In the former, the carbon brush carried by the rotor arm slides over metallic segments embedded in the distributor cap's moulded insulating material, creating an electrical connection between the secondary winding of the coil and the spark plug. In the latter, the electrode of the rotor arm passes close to, but does not actually contact, the segments in the distributor cap.

The electrodes won't experience any noticeable wear with the latter kind of distributor. The distribution unit also comprises of a number of additional auxiliary units. A speed-sensitive device, also known as a governor, is located in the lower portion of the housing and its job is to advance the spark as engine speed increases. The contact breaker assembly, which is located above this unit, can be turned to alter the spark timing. The high-tension distributor is positioned in the housing's upper portion. Additionally, it transports the vacuum ignition governor, which delays the spark as the engine's load rises. When the rotor presses a spark plug, the contact breaker opens, allowing high-tension electricity to flow through the rotor and brass segment of the distributor and through high-tension wiring to the relevant spark plug. Naturally, the firing order of the engine will determine the order in which the spark plugs are attached to the distributor head.

Spark Plug

The spark plug gives the two electrodes a suitable spacing across which the high voltage discharges to produce a spark and ignite the combustible mixture inside the combustion chamber. Essentially, a spark plug is made up of a steel casing, an insulator, and two electrodes. The core electrode, to which the ignition coil's high-tension supply is linked, is highly insulated with porcelain or other ceramic materials. When the plug is put on the engine's

cylinder head, the other electrode, which is soldered to the plug's steel shell, automatically grounds. To endure the intense erosion and corrosion they are subjected to in use, the electrodes are typically composed of high nickel alloy. The insulator and centre electrode tips are both in contact with the combustion gases. As a result, the high thermal and mechanical loads cause the insulators to have a propensity to shatter. Moisture and unusual surface deposits can also have a significant negative impact on some insulators. The heat must transfer from the insulator to the steel shell that is in contact with the relatively cool cylinder head in order to cool the electrodes and so prevent resignation since the central electrode and the insulator are exposed to the high temperatures of the combustion gases.

CONCLUSION

Amazingly, the internal combustion engine has been developing for more than a century. With each passing year, automakers are able to eke out a little bit more economy or a little bit less emissions, so it keeps evolving. The end result is a machine that is immensely complex and remarkably dependable the mechanics of the engine and many of its components, such as the fuel system, cooling system, camshafts, turbochargers, and gears, are covered in other HowStuffWorks articles. It may be argued that the ignition system, with a precisely timed spark, is where it all comes together. We'll study about ignition systems in this post, starting with spark timing. Then, we'll examine all the parts, such as spark plugs, coils, and distributors that contribute to the spark. Finally, we'll discuss systems that omit the distributor in favor of solid-state components.

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An Overview on Spark Plug Used to Operate Engine

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ABSTRACT: A spark plug is an electrical component that attaches to the cylinder head of some internal combustion engines and uses an electric spark to ignite compressed aerosol fuel. In this chapter discussed about the spark plug and its application and working operation. In order to create a spark gap inside the cylinder, spark plugs have an insulated center electrode connected by a thickly insulated wire to an ignition coil or magneto circuit on the outside. A spark plug is a petrol engine component that provides an electric current inside the cylinder. They are a key component of the gasoline engine; without them, it would not operate.

KEYWORDS: Cylinder Head, Center Electrode, Cold Spark, Heat Range, Spark Plug

INTRODUCTION

In order to produce a spark and ignite the combustible mixture inside the combustion chamber, the high potential discharges across the two electrodes of the spark plug must pass through an appropriate gap provided by the spark plug. Essentially, a spark plug is made up of a steel casing, an insulator, and two electrodes. The core electrode is tightly insulated with porcelain or other ceramic materials. It receives a high-tension supply from the ignition coil. When the plug is put on the engine's cylinder head, the other electrode, which is soldered to the plug's steel shell, is automatically grounded. To endure the extreme erosive and corrosive conditions to which they are subjected during usage, the electrodes are typically composed of high nickel alloy. The tips of the insulator and core electrode are in contact with the combustion gases. As a result of the significant thermal and mechanical loads, insulators have a propensity to shatter. Additionally, anomalous surface deposits and moisture have a significant negative impact on some insulators [1], [2].

The heat must transfer from the insulator to the steel shell that is in contact with the relatively cool cylinder head in order to cool the electrodes and so prevent resignation since the central electrode and the insulator are exposed to the high temperature of the combustion gases. On the high-tension electrode's tip's relative working temperature range [3], [4]. The quantity of heat transferred, which in turn depends on the length of the heat transfer channel from the tip to the cylinder head and the amount of surface area exposed to the combustion gases, determines the operating temperature [5], [6]. As seen in, a cold plug has a smaller area exposed to

the combustion gases and a shorter heat transmission path than a hot plug. The kind of spark plug used in an engine depends on the needs of that particular engine. Each engine maker chooses the kind of plug cold or hot that is suitable for his particular engine. When travelling, a spark plug may operate at a suitable temperature, but when idling, it may run so cool that anomalous 304 IC Electrodes are liable to become fouled by engine deposits [7], [8]. These deposits could be made of hard shiny carbon from excessive lubricating oil that penetrates the combustion chamber or soft dull carbon from incomplete combustion. Temperatures above 340 °C will cause incomplete combustion carbon deposits to burn off, however excess oil carbon deposits from lubricating oil require temperatures above 540 °C to burn.

A spark plug may run excessively hot at high speeds and lead to resignation if it runs hot enough at idle speeds to prevent carbon deposits. When a spark plug operates at a temperature greater than 800 °C, resignation frequently occurs. To get a good spark plug that will perform satisfactorily over the entire engine operating range, a compromise must be made. A damaged spark plug is almost usually a significant cause of engine issues like misfiring and resignation. An internal combustion engine uses a spark plug to create a spark that ignites the fuel and air combination in the combustion chamber. The spark plug receives high-voltage energy as part of the engine's ignition system produced by an ignition coil in contemporary engines and delivered by a spark plug wire, which it utilizes to produce a spark in the tiny gap between the positive and negative electrodes. The engine's behavior is significantly influenced by the timing of the spark, and the spark plug typically fires just before the combustion stroke starts[9], [10].

Although the spark plug was created in 1860, it wasn't until the ignition magneto was created in 1902 that its use really took off. Diesel engines typically do not utilize spark plugs since they employ compression ignition instead of spark ignition. The shell, insulator, center electrode, and side electrode sometimes referred to as ground strap are the four basic components of a spark plug. Typically, sintered alumina (Al_2O_3), a strong ceramic substance with excellent dielectric strength, is used to make the bulk of the insulator. The spark plug housing in marine engines is frequently made of metal that has been twice dipped in zinc chromate. Since a spark plug enters through the combustion chamber's wall, it must also be a part of the seal that keeps the high-pressure gases inside the chamber from escaping. Electrodes a single-side electrode design a two-side electrode design Through an internal wire, the central electrode is joined to the terminal. The cathode, from which the electrons are ejected, is set up as the central electrode.

This is because thermionic emission principles dictate that it is simpler to eject electrons from a hotter surface, and the center electrode is typically the area of the plug that is the hottest. The central electrode's pointed tip additionally strengthens the electrical field, which boosts the emission of electrons. Only wasted spark systems employ the side electrode as the cathode since it requires up to 45 percent more voltage despite being colder and blunter. The high-nickel steel side electrode is hot forged or welded to the side of the metal shell. Up to four side electrodes may surround the core electrode in a spark plug. Since the spark moves to a closer ground electrode as the spark gap widens as a result of electric discharge degradation, many side electrodes often offer longer life. The drawback of having many side electrodes is that a shielding effect could happen for each electrode, which would result in a less efficient burn and more fuel being used. Gap width.

Ignition Plug Gauge

The spark plug gaps the distance between the spark plug's tip and the core electrode is crucial to how well a spark plug performs. Car engines commonly have spark plug spacing between 0.6 and 1.8 mm (0.024 and 0.071 inches). In comparison to older engines that used breaker point distributors and carburetors, modern engines that use solid-state ignition systems and electronic fuel injection often have greater gaps. Smaller plug gap sizes typically produce sparks more reliably, but the spark may not be strong enough to ignite the fuel-air mixture. A stronger spark will result from a greater plug gap

size, but not necessarily for example, at high RPM. For platinum and iridium spark plugs, gap modification is not advised due to the possibility of harming a metal disc soldered to the electrode.

DISCUSSION

Spark plugs are little components that are frequently taken for granted while being crucial to the operation of your car. Your automobile most likely wouldn't be able to run at all without correctly functioning spark plugs. In this blog, our automotive specialists will go deeper into the significance of spark plugs, what they perform, and how to determine when it might be time to replace them.

What Perform Spark Plugs?

Your engine is a really amazing device that uses petrol as an energy source to create movement. Yet how does it accomplish this? Internal combustion, a theory, is the solution. Your car's fuel must be released for your engine to convert it from a source of potential energy into a source of kinetic energy, and it does this through combustion. This process is initiated by the engine cycle. During the engine cycle, your valves fill your cylinder with a combination of fuel and air that, when combined, is extremely explosive. This combination is compressed to a very small volume when the piston in your engine rises, releasing even more potential energy. When the piston is at its highest point of compression, your engine ignites this mixture with a tiny spark, causing an explosion that pulls the piston back downward and turns the crankshaft in your engine to generate the power that propels your automobile forward.

The spark that ignites the air/fuel mixture and causes the explosion that propels your engine into action is provided by your spark plugs. These tiny, straightforward plugs produce an electrical arc between two leads that are not in contact but close enough to allow current to pass through the space between them. Your ignition system includes your spark plugs as well as the electrical and timing components that power them Your spark plugs are typically comprised of materials that are incredibly resilient and can sustain countless explosions before wearing out or needing to be changed. However, it is true that corrosion and explosions over time result in weaker or smaller sparks, which reduces your engine's performance and may also cause misfiring or failure to fire.

When Should I Replace My Spark Plugs?

Today, the majority of manufacturers will fit "extended-life" spark plugs at the factory. They cost

just a little bit more than a regular plug, and customers love the sound. Spark plugs with an extended lifespan are typically rated for a lifespan of about 100,000 miles. The space between the two leads starts to grow as these plugs near the end of their useful lives, which causes decreased efficiency and the potential for misfiring. If your plug corrodes to the point where it breaks, the interior of your engine cylinder could sustain catastrophic damage, necessitating a very costly repair. In order to find out how long a lifespan your spark plugs are rated for, make sure to examine the owner's manual for your car. Never let the spark plugs' rated mileage exceed that.

Methods for Choosing New Spark Plugs

Make sure you choose the proper spark plugs for your car when choosing new ones. Most auto parts stores should be able to assist you with this, and you'll have a wide variety of models to pick from. Although they often cost more than counterparts made of less expensive metals like copper, those made of precious metals like platinum or iridium typically have superior wear resistance.

Conditions for a Good Spark Plug

1. It must be dependable at high gearbox voltage, or up to 40,000 V.
2. It must be able to effectively insulate even at 1000 °C temperatures and must prevent arcing and flashover.
3. It must be able to withstand thermal shock caused by hot exhaust gases and cold intake mixes.
4. It must create a gas- and pressure-tight seal with the combustion chamber.
5. It must be able to withstand fluctuating pressures of up to about 100 bar.
6. For dependable installation, it should have a high mechanical strength.
7. By insulator tip and electrodes, it must have good thermal conduction.
8. Spark erosion, combustion gases, and residue resistance are required.
9. It must be able to stop the accumulation of deposits on the insulator.

Spark Plug Component Parts

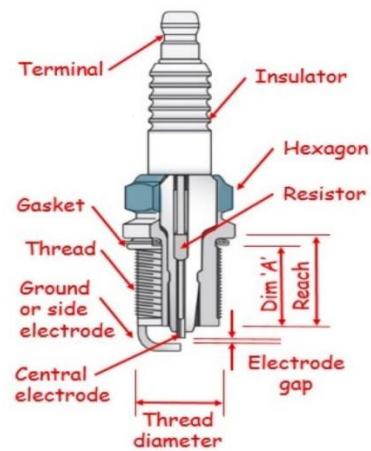


Figure 1: Representing the Spark Plug Parts [Angri Racing Academy].

1. **Plug Terminal:** This is where the high-tension cable that comes from the distributor cap is connected. The middle electrode receives the high voltage through it.
2. **Ceramic Insulator:** It is an insulator formed of ceramic aluminum oxide. At up to 40,000 Volts, it disconnects the core electrode from the earth. To avoid flashover, it can be produced in simple form or with profiles.
3. **Metal Body:** It is a steel body with precision-rolled threads for a tight fit and simple assembly and disassembly. By transmitting heat to the cylinder head, it serves as the cylinder head's electrical ground and aids in cooling the plug.
4. **Central Electrode:** It is built of nickel-based alloys and has an encapsulated copper core. The central electrode may be made of platinum or iridium depending on the type. The secondary winding of the distributor distributes the high voltage to the central electrode.
5. The ground electrode is soldered to the SP's metal body. Together with the central electrode, it creates a spark route. It is constructed of alloys based on nickel with titanium or iridium reinforcing.
6. **Sealing Washer:** It facilitates heat dissipation and seals with the cylinder head. It is extended into the combustion chamber. Insulator tip. It has a stronger impact on the spark plug's thermal rating. The distance between the ground electrode and the center electrode is known as the electrode gap. The electrode is essential for spark production. The plug cannot provide a strong enough spark to ignite the gasoline and could misfire if the proper spacing is not given.

The primary types of spark plugs are as follows:

1. Plugs made of copper.
2. Gold-plated spark plugs.
3. Igniters made of iridium.
4. Platinum-double spark plugs.
5. Spark plugs plus tar.
6. Metallic spark plugs.

1. Copper Spark Plugs No:

These spark plugs are incredibly popular today. These spark plugs use a nickel alloy at the working end of the electrode, as their name suggests, and a copper core. Due to the comparatively low melting temperatures of nickel and copper, these electrodes' edges quickly wear away, leaving a gap as small as 10,000 miles. This causes ignition wires and coils to be damaged, misfires, and decreased fuel efficiency. This spark plug is used in the majority of high-performance vehicles because of its effectiveness.

2. Spark Plugs Platinum:

Due to its high melting point, the platinum disc found at the electrode tip of platinum spark plugs often serves as protection for the electrode edge. Because platinum is a precious metal, these spark plugs are highly pricey. These spark plugs are hotter than copper type and have a long life, usually 30–40 thousand miles. Because platinum metal is so tough, spark plug durability is boosted. Spark plugs made of platinum are more resistant to carbon accumulation.

3. Iridium Spark Plugs No:

Iridium spark plugs can last around 25% longer than platinum-type spark plugs because it is a more robust metal than platinum and has a higher melting point. Iridium spark plugs are utilised when the engine has to produce a lot of power. They perform better for engines because they have smaller electrode tips, which improves spark efficiency. They offer a very effective spark, ensuring full combustion and smooth operation of the engine. They are less likely to be used because of their high cost.

4. Spark Plugs Rated in Double Platinum:

They resemble platinum spark plugs in every way except for the fact that both of their electrodes the ground electrode and the center electrode—are made of platinum.

5. Spark Plugs Plus tar:

Plus tar plugs go above and beyond the advantages of normal plugs by incorporating an internal capacitor that stores and releases energy. This results in a high-intensity electrical pulse.

A small amount of the gaseous air-fuel mixture is transformed using pulsed electrical technology into a highly excited plasma that ignites and burns quickly. This provides an engine that is more potent and reliable, starts up quickly, idles smoothly, produces more torque, consumes less gasoline, and emits less.

6. Silver Spark Plugs No:

Silver metal is used to create spark plugs, as the name suggests. They are renowned for their superior heat conductivity and good performance. This kind of spark plug is frequently used in motorcycle and high-performance antique cars from Europe. Despite their benefits, they are not very durable and have a short lifespan. Based on the quantity of heat dissipation and operating temperature, these spark plugs are further divided into two types. Below is a description of them a warm spark plug.

7. A Cold Spark Plug. Spark Plug is Hot:

A plug's heat range specifies the maximum temperature at which it will operate. It speaks to the spark plug's capacity to transfer heat from the insulator's firing tip to the engine's cooling system. The distance over which heat is transported determines the temperature that a spark plug will reach. The plug will run hotter if the path of least resistance is longer than it is short. Compared to cold spark plugs, which have a shorter path of heat travel and operate cooler, this type of spark plug has a longer path of heat travel and operates hotter. 2nd unused spark plug In order to prevent overheating, these spark plugs are utilised in heavy-duty engines that run continuously at high speeds. The carbon builds up on the insulator surrounding the center electrode when a plug runs too cold. This carbon will be burned off by a hotter operating plug, preventing its development.

Because the electrode burns away more quickly at high temperatures, a hot plug will wear out more quickly. The insulator may develop a white or greyish color and appear blistered if a plug runs too hot. A hot spark plug is necessary for low-speed, medium-duty engines working in chilly operating conditions. A factory that makes plugs in a variety of heat ranges is necessary because different engines functioning in different environments require a plug with a specified heat range. To meet the particular condition of the engine, medium hot and medium cold spark plugs are also offered in addition to hot and cold spark plugs. By allowing the plug tip to be cooled by the incoming charge at high engine speeds, the projected core nose kind of plug allows the plug to run hotter at low speeds, hence increasing the heat range. The spark plug is a fairly

straightforward device that performs a number of crucial but distinct tasks. First and foremost, it essentially produces a fake lightning strike inside the engine's cylinder head, or combustion chamber. To spark and "light the fire" inside the combustion chamber's-controlled chaos, it sends an incredibly high amount of electrical energy. At the spark plug, the voltage can range from 20,000 to more than 100,000 volts.

Heat-Efficient Spark Plugs:

The spark plug doesn't maintain combustion, even though it starts the spark that does. It does assist in transferring heat from the combustion chamber into the cylinder head's water jacket. The heat range of a spark plug determines its capacity to disperse heat from the combustion chamber. The temperature of the spark plug's firing end must be kept at a level that is both high enough to avoid fouling and low enough to avoid pre-ignition. Manufacturers of spark plugs refer to this as "thermal performance." The quantity of energy transferred from the ignition system through the spark plug has nothing to do with its thermal performance, often known as its heat range. The region in which the spark plug performs thermally is known as the spark plug heat range.

Compared To Hot Spark Plugs, Cold Spark Plugs:

Spark plugs that are cold typically have a limited heat flow channel. As a result, the rate of heat transfer is extremely rapid. Additionally, the tiny surface area of the short insulator nose on cold spark plugs prevents a significant amount of heat absorption. Hot spark plugs, on the other hand, have a longer insulator nose and a longer heat transmission channel. As a result, the surrounding cylinder head and subsequently the water jacket experience substantially slower heat transmission. The spark plug's heat range needs to be carefully chosen in order to produce the best thermal performance. If the heat range is off, you could have major issues. The normal firing end temperature ranges from 900 to 1,450 degrees. Carbon fouling is possible below 900 degrees. Overheating becomes a problem above it.

Increased Spark Plug Voltage:

The spark plug is connected to the high voltage produced by the ignition coil during operation either by an electronic or conventional distributor. The spark plug's center electrode and ground electrode experience a voltage differential as electricity leaves the coil. The spark plug cannot ignite right away because of the spark plug gap, which also contains an insulating mixture of air and gasoline. The spark

plug's gap can be breached and ignited when the voltage rise reaches about 20,000 volts. You can hear a distinct click when a spark plug is correctly grounded to fire after being removed from the cylinder head. You can see the spark if it's sufficiently dark. The spark you see resembles a tiny kind of lightning, and the click you hear is virtually a tiny thunderclap. The spark plug's high heat output inside the combustion chamber causes a tiny fireball to form inside the gap. In theory, complete combustion occurs in the cylinder as the fireball or combustion kernel grows.

How Spark Plugs Are Made

Spark plugs' construction might not be as straightforward as you might imagine. They are actually precise pieces of machinery. We are able to give you a thorough analysis of all the plug features because of the people at Champion Spark Plug. Remember that the great majority of spark plugs include comparable but not always identical architecture. The real appearance of several of the aforementioned spark plug features can be seen in the photographs that follow. Look them up. Insulator ribs help to improve the grip of the rubber spark plug boot against the plug body and offer further defense against secondary voltage or spark flashover. The body of the insulator is made of ceramic aluminum oxide. Using a high-pressure, dry moulding method, this spark plug component is produced. The insulator is molded, then heated in a kiln to a temperature over steel's melting point. A component with remarkable dielectric strength, high thermal conductivity, and excellent shock resistance is the end product of this procedure.

Insulator: Aluminum oxide ceramic is used to mound the insulator body. Using a high-pressure, dry moulding method, this spark plug component is produced. The insulator is molded, then heated in a kiln to a temperature over steel's melting point. A component with remarkable dielectric strength, high thermal conductivity, and excellent shock resistance is the end product of this procedure. The spark plug insulator is shown by the pointer. It is made of ceramic made of aluminum oxide, as was already explained. The outside surface is ribbed to enhance protection from spark flashover and to give the spark plug boot grip.

Hexagon: A socket wrench's contact point is the hexagon. In the industry, the hex size is essentially uniform and is typically correlated with spark plug thread size.

Shell: Using a unique cold extrusion method, the steel shell is manufactured with perfect tolerances.

For the shell manufacturing of some spark plug models, a steel billet (bar stock) is used.

Plating: Almost always, the shell is plated. This increases toughness and offers resistance against corrosion and rust. A specialized cold extrusion technique is used to produce the steel shell to specific tolerances, or in other specialized situations, the steel billet is machined. You can install or remove the plug with a socket wrench thanks to the hexagonal pattern that was machined onto the shell.

Spark Plug Gaskets: Some spark plugs have them, while others are "gasket less." The spark plug gasket is made of folded steel, which offers a smooth surface for sealing. Spark plugs without gaskets have a tapered seat shell that seals thanks to a tight tolerance built into the spark plug itself. Spark plug threads are often rolled rather than cut. This complies with the requirements outlined by the SAE and the International Standards Association.

Ground Electrode: Although there are many distinct ground electrode designs and shapes, they are typically made of nickel alloy steel. The ground electrode needs to be resistant to both chemical and spark erosion under extremely high temperatures.

Centre Electrode: Centre electrodes have to be made of a specific alloy that is impervious to chemical and spark corrosion. Remember that the temperature inside the combustion chamber might change, sometimes drastically. These conditions must be met by the center electrode. With a pointer, spark plug ground electrode and center gap the gap between the ground electrode and the center electrode is referred to as the spark park electrode. A unique alloy that is resistant to spark erosion and chemical corrosion must be used to make center electrodes.

Insulator Nose: Although there are many different shapes and sizes of insulator noses, they all need to be able to shed carbon, oil, and fuel deposits at low speeds. The insulator nose is typically cooled at higher engine speeds to lower temperatures and prevent electrode corrosion.

CONCLUSION

The spark plug is a fairly straightforward device that performs a number of crucial but distinct tasks. First and foremost, it essentially produces a fake lightning strike inside the engine's cylinder head, or combustion chamber. To spark and light the fire inside the combustion chamber's-controlled chaos, it sends an incredibly high amount of electrical energy. At the spark plug, the voltage can range from

20,000 to more than 100,000 volts. A spark plug is an electrical component that attaches to the cylinder head of some internal combustion engines and uses an electric spark to ignite compressed aerosol fuel. In order to create a spark gap inside the cylinder, spark plugs have an insulated center electrode connected by a thickly insulated wire to an ignition coil or magneto circuit on the outside.

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Application and Advantages of Battery Ignition System

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ABSTRACT: The battery ignition system is used in automobiles with internal combustion engines to ignite the fuel in the spark plug. In this article, we will examine the battery ignition system's definition, components, operation, benefits, and applications. In order to power the spark, plug and generate sparks that burn the air-fuel mixture in the SI engine system, a battery ignition system is utilised. To supply electricity, a battery ignition system includes a 6- or 12-volt battery charged by an engine-driven generator, an ignition coil to increase voltage, a device to interrupt current from the coil, a distributor to direct current to the correct cylinder, and a spark plug projecting into each cylinder.

KEYWORDS: Battery Ignition, Breaker Point, Contact Breaker, High Voltage, Ignition Coi

INTRODUCTION

A spark-ignition engine uses a particular kind of ignition system called a battery ignition system to power the spark plug and create the spark necessary to burn the air-fuel combination in the combustion chamber. The engine-driven generator alternator charges a 6- or 12-volt battery that powers the ignition system. Battery ignition system is the name given to the device that powers the ignition system. These ignition systems are frequently seen in light commercial vehicles, buses, and trucks in addition to trucks. Its purpose is to spark, enabling the burning of fuel. Understanding the Internal Combustion Engine Cooling System [1], [2].

Battery Ignition System Components

The battery ignition system's components and the many uses to which they are put are listed below:

Rechargeable Lead-Acid Batteries: Rechargeable lead-acid batteries, which store electrical energy, are the battery type employed in this system. The battery powers the ignition system by supplying current to it as soon as the engine is running [3], [4].

Ballast Resistor: The ballast resistor regulates the current that flows through the primary winding and is a part of the battery ignition system. It is made of iron, which has the ability to rapidly increase electrical resistance by slightly raising temperature. The extra resistance opposes the current flow that regulates the ignition coil's temperature. This aids in maintaining the temperature over an extended period of time and is connected in series with the primary winding to control the current there [5], [6].

Full Screen: Understanding High Voltage Transformers. Their Parts, Operation, and Applications [7], [8].

Ammeter: The ammeter is used to gauge the system's current flow. All ignition systems have an ignition switch that is used to turn the system on and off.

Ignition Coil: An ignition coil is used to switch the system's voltage stage from low to high. Additionally, it is employed in the spark plugs spark creation process. The component is made up of two electrical windings, known as the primary and secondary winding, and a magnetic core or soft wire. The primary winding contains 200–300 turns because the magnetic field is produced by the current passing through it. The secondary winding, however, is made up of 21000 twists of 40-gauge wire. It is insulated to resist high voltage. Learn more about how an automotive engine's charging system functions [9], [10].

Contact Breaker: The primary circuit to the ignition coil is added to and broken by the contact breaker. In other words, when it's closed, current can flow through the ignition coil; when it's open, it stops. The capacitor is identical to a standard electric capacitor. Typically, it uses an electric field to store electrical energy. Two metal plates are separated from one another in the component by an air gap. It is made of an insulating substance. The capacitor's role in the system is to keep an arc from forming across the breaker point. If there isn't a capacitor in the system, induced voltage will result in an arc at the breaker point. This might be quite harmful.

Distributor: There are two different types of distributors: brush type and gap type. It also serves a crucial function in the system by timing the delivery of ignition pulses to each spark plug in turn.

Spark Plug: The spark plug is the electrical component responsible for lighting the fuel-air mixture and causing the explosion.

Working Theory: The battery ignition system functions in a manner that is very similar to that of other kinds of ignition systems. Because it uses a 6- or 12-volt battery that is charged by the engine-driven generator alternator, it is considerably simpler to understand. The system's ignition coil boosts voltage and has a mechanism to stop the coil's current flow. Spark plugs project current into each cylinder, whereas distributors direct current to the appropriate cylinder. Well, all of these have already been mentioned. The primary winding of the coil is where the current leaves the battery and travels to the interrupting component before returning to the battery. In earlier cars, the breaker points a switch with tungsten contacts to slow erosive wear is where current is interrupted. These points are opened and closed by the cam lobe rotation, and when the breaker point is closed, electricity flows through the ignition coil's primary winding. In more recent cars with electronic ignition systems, a reflector has taken the place of the breaker point. It is a magnetic pulse distributor that generates timed electric signals to regulate the current to the ignition coil's primary winding. The following is information on the electronic ignition system:

DISCUSSION

The ignition coil in the battery ignition system is the source of the ignition energy. This coil stores the energy in its magnetic field and transfers it to the proper spark plug at the instant of ignition the firing point in the form of a surge of high voltage current the ignition pulse. We also refer to the ignition coil as an inductive storage device since energy storage in the magnetic field is based on an inductive process. Figure 1 is the schematic diagram of a typical battery ignition system for a four-cylinder engine. The ignition coil, as previously mentioned, is made up of two wire coils: the primary winding, L1, which has a few twists of heavy copper wire, and the secondary winding, L2, which has several turns of fine copper wire. Each coil is isolated from the other and is wound around the other. The laminated iron core, around which the primary and secondary windings are wound, strengthens the magnetic field and, as a result, increases the amount of energy stored.

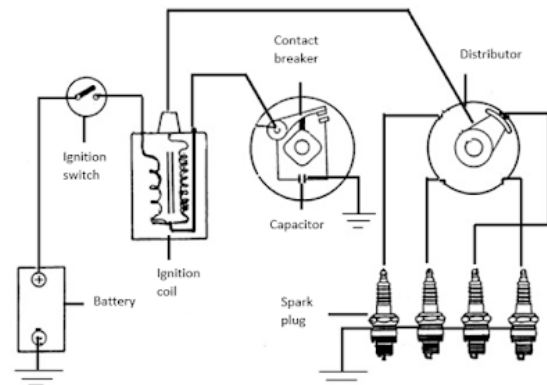


Figure 1: Representing the components of the Battery Ignition system [Auto Expose].

An engine-driven generator that charges a 6- or 12-volt battery, an ignition coil to boost voltage, a device to cut off current from the coil, a distributor to direct current to the right cylinder, and spark plugs that extend into each cylinder are all components of a battery ignition system. The primary winding of the coil serves as the conduit for the flow of current from the battery to the interrupting component and back to the battery. Breaker points, a switch with tungsten contacts to prevent erosion, were used in older cars to interrupt the primary current. A breaker cam, a revolving device having a lobed surface one lobe for each cylinder, rotated at half engine speed to open and close the points. The primary winding of the ignition coils experienced current flow when the breaker points were closed. The interrupting component in early 1960s-era electronic ignition systems is a reflector, a magnetic pulse distributor that generates timed electric signals that are amplified to regulate the current to the ignition coil's primary winding.

These devices typically decrease ignition maintenance and boost engine performance. Wire is wound around an iron core in the main winding. A secondary winding comprising numerous more turns of finer wire linked to the distributor is placed over this. A magnetic field is produced by current passing through the primary coil. The circuit is broken and current stops when the breaker cam opens the breaker points or the reflector sends its signal. When the magnetic field collapses, a much larger voltage is produced in the secondary winding and directed towards the distributor. A finger that is rotating inside the distributor does so at half engine speed. It comes into contact with contacts that are connected to various cylinders as it rotates. Rotation is timed so that the piston is almost at the top of the compression stroke and a high voltage has just been induced in the secondary winding of the ignition coil

when the finger touches the contact for a particular cylinder. Thus, the spark plug gap is exposed to a high voltage.

A center electrode embedded in insulating ceramic makes up the spark plug. A metal shell with threads around the outside screws into a hole on the top of the cylinder. Overhanging the end of the center electrode is a ground electrode that protrudes from the shell. There is a tiny space between the two electrodes, measuring .015-.040 in 0.38-.102 cm. A spark jumps the gap and ignites the mixture of petrol and air at roughly 8,000 volts. A vacuum advance causes the spark to ignite earlier at modest throttle openings above idle, whereas a centrifugal advance cause it to ignite earlier at high engine speeds. The positive terminal post of the storage battery is linked to one end of the primary winding via the ignition switch, while the other end is grounded by the contact breaker. The contact breaker and ignition capacitor are linked in parallel. The secondary winding is connected to the contact breaker on one end and to the center electrode of the spark plug on the other end through the distributor and high-tension ignition cables. The primary winding of the coil is linked to the positive terminal post of the storage battery when the ignition switch is engaged. A current known as primary current flows when the primary circuit is closed using the breaker connections.

A magnetic field is created in the core when this current passes through the primary coil, which is wound over a soft iron core. Every time an ignition discharge is required, the breaker points are set up to be opened by a cam that is powered by the engine shaft. The current that had been flowing through the points now flows into the condenser, which is linked across the points, when the breaker points open. The primary current decreases and the magnetic field collapses as the condenser charges up. The primary winding experiences a voltage due to the field collapsing, which charges the condenser to a voltage far greater than battery voltage. The magnetic field and primary current are then reversed as the condenser discharges into the battery. The secondary winding of the ignition coil experiences an extremely high voltage due to the magnetic field's quick collapse and reversal in the core. The secondary winding is made up of numerous rounds of extremely tiny wire wound around the same core as the primary winding.

The distributor, a rotating switch found in the secondary or high-tension circuit of the ignition system, directs the high secondary voltage to the appropriate spark plug. The high primary voltage brought on by the collapse of the magnetic field

surrounding the primary winding would result in an arc across the breaker points if a condenser were not utilised in the primary circuit. The primary current and magnetic field would not drop quickly enough to produce the high secondary voltage without the arc burning and destroying the points. Note that the distributor only decides the order in which the spark plugs fire; the timing of the spark is controlled by the crank angle at which the breaker points open. Rotating the plate that holds the breaker points in relation to the cam can change the ignition timing. As a result, if the plate is moved in the direction of the camshaft's rotation, ignition will be delayed.

Working Operation

The battery ignition system operates on a similar premise to other kinds of ignition systems. Because it uses a 6- or 12-volt battery that is charged by the engine-driven generator alternator, it is considerably simpler to understand. The system's ignition coil boosts voltage and has a mechanism to stop the coil's current flow. Spark plugs project current into each cylinder, whereas distributors direct current to the appropriate cylinder. Well, all of these have already been mentioned. The primary winding of the coil is where the current leaves the battery and travels to the interrupting component before returning to the battery. In earlier cars, the breaker points a switch with tungsten contacts to slow erosive wear is where current is interrupted. These points are opened and closed by the cam lobe rotation, and when the breaker point is closed, electricity flows through the ignition coil's primary winding. In more recent cars with electronic ignition systems, a reflector has taken the place of the breaker point. It is a magnetic pulse distributor that generates timed electric signals to regulate the current to the ignition coil's primary winding. The following is information on the electronic ignition system: The secondary winding is made up of numerous additional turns of finer wire connected to the distributor and is placed above the primary winding, which is formed of wire wound around an iron core.

The magnetic field is produced as the primary winding's current flows across it. The circuit is interrupted and stopped when the breaker cam opens the breaker points or the reflector sends its signal. The secondary winding is subsequently induced with a higher voltage that is directed to the distributor as the magnetic field falls. A finger that is rotating at half the engine speed is located in the distributor. A high voltage is created in the secondary winding of the ignition coil as a result of this rotation making contact with each of the contacts for a certain cylinder. The piston is also

about to complete its compression stroke at that point. Additionally, the spark plug gap is impressed with a high voltage. A ceramic insulator surrounds the center electrode of the spark plug. The outer part is a metal shell with threads that is put onto the top hole of the cylinder. Overhanging the end of the center electrode, the ground electrode protrudes from the shell. The spacing between the two electrodes is only 0.015-0.040 in. 0.38-1.02 cm wide. The fuel-air combination is ignited when a spark jumps the gap at a voltage of roughly 8,000 volts. The following are the main components of a battery ignition system:

1. Battery.
2. Ballast Resistor.
3. Ammeter.
4. Ignition Switch.
5. Primary Winding.
6. Contact Breaker.
7. Capacitor Secondary.
8. Winding Distributor.
9. Spark Plug.

Here, a lead-acid rechargeable battery is being used. It serves as a source of ignition power by storing electrical energy. The dynamo, which is powered by the engine, charges the battery. The Ignition switch is connected to the battery. The primary winding's current is regulated by the ballast resistor. It is made of iron, which has the ability to rapidly increase electrical resistance by rising temperature up to a specific point. The extra resistance blocks the current flow that regulates the ignition coil's temperature. It aids in maintaining the temperature over an extended period of time, and in this case, it is connected in series with the primary winding to control the current there. An ammeter is a measurement tool used to gauge current. The ignition switch is a crucial component of the system. It is used to turn the system ON and OFF.

Ignition Coil: The fundamental component of the battery ignition system, an ignition coil steps the voltage from low to high and is responsible for spark plug spark generation. A magnetic core, sometimes known as a soft wire, and two electrical windings, referred to as the primary and secondary windings, make up an ignition coil.

Primary Winding: There are 200–300 turns in the primary winding. A magnetic field is produced by the primary winding's current flow.

Secondary Winding: There are 21000 turns of 40-gauge wire in the secondary winding. It is protected from high voltage by insulation. One end is attached to the PW, as shown in the diagram, while the other ends are connected to the distributor.

Contact Breaker: As the name contact breaker suggests, this device adds and breaks the principal circuit that connects the ignition coil. When the contact breaker is closed, current flows in an ignition coil; when it is open, flows cease. You can see that it's near in the diagram above. It is comparable to an electric capacitor. Typically, it uses an electric field to store electrical energy. In a capacitor, an insulating substance called air separates two metal plates from one another. The major purpose of the capacitor is to stop an arc from forming across the breaker point; if it were not connected to the primary circuit, the generated voltage would instead result in an extremely dangerous arc across the breaker point.

Distributor: In this system, the distributor also has a crucial function to play. It delivers ignition pulses at precisely the right time to each sequence of spark plug insertions. Two separate categories of distributors exist. Both brush type and gap type spark plugs are attached to the distributor, as can be seen in the diagram. The air-fuel combination in the system begins to burn when a spark is introduced into the system by the spark plug. SI Engine uses a spark plug.

Work of the Battery Ignition System

The primary circuit closes and current begins to flow when the ignition switch is switched ON. The magnetic field surrounding the coil's soft iron core is created by this current. When the breaker points open, electricity begins to flow through the condenser; when it points closed, current begins to flow via the contact breaker. When a current flows through a condenser, the magnetic field shrinks, the primary current decreases, and the condenser charges. The primary winding experiences a current that is induced by this shift in the magnetic field and flows in the same direction as the primary current. It raises the condenser voltage to a level far higher than the battery voltage, interrupting the battery's current flow. The condenser discharges into the battery as a result of these activities. Now switch the primary current's direction, and the magnetic field causes a high voltage to be induced in the secondary winding. Currently, a high-tension wire transmits a high voltage to the distributor. As seen in the diagram, an ignition harness connects the distributor to the spark plug.

A Battery System's Key Benefits

1. More power is output thanks to the battery system.
2. Also excellent is fuel efficiency.
3. This system has no moving parts.

4. Better combustion because here more than 90% of the air-fuel mixture burns, compared to 70 to 75% in other conventional systems.
5. Good spark quality is also present.
6. Have a strong spark.
7. Compared to other ignition system types, this one required less maintenance.
8. Offers a strong spark even when the engine is running slowly or during initial starts.
9. Increase your power output.
10. Good fuel efficiency.
11. Nothing in the system is moving.
12. Better combustion because it can burn over 90% of the fuel, as opposed to conventional ignition systems that can only burn 70% to 80% of the fuel-air mixture.

It also has certain drawbacks, including the following:

1. Comparatively speaking to other electrical ignition systems, battery ignition requires greater area.
2. When the battery is drained, the engine won't start, which makes the battery crucial for producing the spark.
3. When the spark plug is fouled, it won't spark.
4. Arcing and contact breaker point pitting result in increased maintenance.
5. It takes up more room.
6. The battery alone will need periodic repair.
7. Spark intensity was reduced, which resulted in decreased efficiency.
8. For the spark to be produced, the battery must be used.
9. The system won't function if the spark plug and battery are defective.
10. There might be a need for more upkeep.
11. First, the SI Engine uses a battery system to generate sparks for the spark plugs.
12. Light commercial vehicles and contemporary cars with SI engines also use it.

CONCLUSION

An engine-driven generator that charges a 6- or 12-volt battery, an ignition coil to boost voltage, a device to cut off current from the coil, a distributor to direct current to the right cylinder, and spark plugs that extend into each cylinder are all components of a battery ignition system. The primary winding of the coil serves as the conduit for the flow of current from the battery to the interrupting component and back to the battery. Breaker points, a switch with tungsten contacts to prevent erosion, were used in older cars to interrupt the primary current. A breaker cam, a revolving device having a lobed surface one lobe for each cylinder, rotated at half engine speed

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Application of Magneto Ignition System

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ABSTRACT: *A battery is not necessary for the Magneto Ignition system. For the primary, it produces its own voltage. Mopeds, scooters, wheelers, motorcycles, stationary engines, and reciprocating aviation engines all use magneto ignition systems extensively. The Magneto Ignition System's definition, parts composition, operating principle, benefits, drawbacks, and application have all been covered in this article. The Magneto Ignition System uses the magneto to generate electricity, which is then used to power the engine. Without the aid of a battery or ignition coil, the system has a current producing unit of its own.*

KEYWORDS: *Coil Ignition, Current Flow, High Voltage, Ignition System, Magneto Ignition*

INTRODUCTION

The magneto ignition system uses electricity that is obtained from the magneto and is then used to power the engine. Without the help of a battery or ignition coil, the system is equipped with a current-generating device [1], [2]. Due to the low-speed efficiency of the system, which necessitates the generation of the necessary electric current by the Magneto, the starting quality of the spark is poor; but, as the engine speed increases, the high-intensity spark will cause it to improve. The Magneto Ignition system's principal elements or parts are as follows:

1. Magnets that spin.
2. First Winding.
3. Second Winding.
4. Contact Breaker.
5. Distributor Ignition Switch.
6. Spark Plug or Fixed Armature.
7. Condenser Breaker-points.

Working

A high voltage is generated in the secondary circuit and the flux of the primary circuit is altered. With a camera's assistance. The coil ignition system and the magneto ignition system's variations in breaker current with speed [3], [4]. An armature that has several hundred twists of enameled copper wire wound onto it for the primary coil and thousands of turns of thin insulated wire wound onto it for the secondary coil revolves in a permanent magnetic field. The magnets should continue to rotate while the armature remains motionless. By doing this, the electric generator's laws-based primary voltage is produced, which is then converted into a very high secondary voltage precisely along the lines specified in the battery igniting system. As can be observed, the current produced by the magneto is fairly tiny because to the low cranking rpm at startup [5], [6]. As we increase the current, the engine speed will

rise. As a result, Magneto constantly has trouble starting, and occasionally a different battery is required. Because magneto is better at high speeds, it is frequently used in aircraft engines, sports and racing cars, etc. A unique kind of ignition system called a magneto has a built-in electric generator that supplies the system with the energy it needs to function [7], [8].

With the exception of the spark plug, it is placed on the engine and replaces all of the coil ignition system's parts. A magneto does not require a battery as an external energy source because it may generate a very high voltage when the engine rotates it. Depicts a schematic representation of a high-tension magneto ignition system. The high-tension magneto combines the windings to produce the primary voltage and to step it up; as a result, a separate coil is not necessary to provide the voltage needed to operate the spark plug. Rotating armature or rotating magnet types of magnetos are both possible. In the first form, the primary and secondary windings of a stationary magnet serve as the armature, which rotates between their poles. In the second type, the magnet revolves and the windings are kept stationary. There is also application for a third kind of magneto known as the polar inductor type. Both the magnet and the windings are stationary in the polar inductor type magneto, but the voltage is produced by flipping the flux field with the aid of soft iron polar projections, or inductors [9], [10].

The coil ignition system and the magneto ignition system both operate on the exact same principles. A cam is used to modify the flux in the main circuit, which results in the secondary circuit producing a high voltage. Depicts how the breaker current varies with speed for the coil ignition system and the magneto ignition system. As can be observed, the magneto's current output is rather low during startup due to the low cranking rpm. The flow of current likewise increases as engine speed rises. As a result,

starting a magneto is never easy, and occasionally a second battery is required. Because the magneto performs best at high speeds, it is frequently employed in aviation engines, sports and racing cars, etc. The battery ignition system is more expensive but extremely dependable in comparison. The battery ignition system is generally preferable to the magneto system in automotive engines due to the magneto system's poor starting characteristics.

Magneto ignition systems, however, are preferred in two-wheelers due to their light weight and ease of maintenance. The main drawback of the high-tension magneto ignition system is that there is a significant risk of engine misfire due to leakage in the wiring, which carries a very high voltage. The high-tension wires need to be adequately insulated to prevent this. To get around this issue, the low-tension magneto system was developed. In the low-tension magneto system, a brush contact is used in place of the distributor to replace the secondary winding, which limits the secondary voltage to a value of roughly 400 volts. Using a step-up transformer, the high voltage is produced. All of these modifications confine the high voltage current in a small area of the ignition system wiring, eliminating any potential leaks, etc. The comparison of the battery and magneto ignition systems is provided.

DISCUSSION

Simple magnetos permanent magnet electrical generators can generate relatively low voltage electricity, but they are unable to provide the high voltages needed by spark plugs found in the majority of current engines apart from diesel engines. A second component of an ignition magneto is an electrical transformer, which raises the voltage of the electricity while lowering the output current. The point spacing initially causes the voltage across the primary coil to arc across the points when the points start to open. The primary coil's leakage inductance stores energy, which is absorbed by a capacitor positioned across the points. This capacitor also reduces the primary winding voltage's rising time, allowing the points to fully open.

An electrical transformer is created by winding a secondary coil with many more turns than the primary onto the same iron core. The turn's ratio is the amount of turns in the secondary winding divided by the number of turns in the primary winding. A proportionate voltage is induced across the secondary winding of the coil when the primary coil is voltage. The primary and secondary coil's turn's ratio is chosen so that the voltage across the secondary reaches a very high value, which is

sufficient to cause an arc across the spark plug gap. Since the secondary winding typically has 100 times as many spins as the primary winding, as the voltage on the primary winding increases to several hundred volts, the voltage on the secondary winding increases to several tens of thousands of volts.

Booster Coil, Induction Vibrator, and Impulse Coupling

Points open and close quickly in a vibrator coil, producing pulsating DC, or interrupted battery current. R1, the vibrator's points V1, and coil L2 all receive electricity from the battery. The vibrator points V1 are opened by the energized coil L2, interrupting the current flow through L2. The vibrator points V1 near once more as the magnetic field surrounding L2 shrinks. Once more, L2 experiences current flow, and the vibrator points on V1 are open. Continuously carrying out this technique results in a "shower of sparks. Components of a booster coil. The booster coil may produce a string of sparks on its own and is independent of the magneto. A magnetic field surrounding the primary coil created by current flow attracts the movable contact point, breaking the circuit. A spring then causes the mobile contact point to return to the fixed contact point. This repeats the process of setting up the current flow.

The magneto produces low voltage at low rpm, making engine starting more challenging. As a result, some magnetos contain an impulse coupling, a mechanical connection that resembles a spring that connects the engine's drive shaft to the magneto drive shaft, and which winds up and let's go at the appropriate times to spin the magneto shaft. A spring, a hub came with flyweights, and a shell are all used in the impulse coupling. While the driving shaft is kept still, the magneto's hub rotates as the spring tension increases. The flyweights are released when the magneto is supposed to ignite as a result of the body touching the trigger ramp. This enables the spring to unwind, speeding up the rotation of the revolving magnet and the magneto, which in turn causes a spark to fly.

History

The first practical high-tension magneto was created in the late 1890s by English engineer Frederick Richard Simms in partnership with German engineer Robert Bosch and his team of Arnold Shriner, Young Rill, and Got lob Honed. On the Zeppelin in 1900, the Gottlieb Daimler engines were equipped with Bosch magneto ignition. The German Mercedes 35 hip race car from 1901 was the first automobile to use magneto ignition. Other automobiles made by Benz, Mors, Turcat-Mery, and Nessler followed.

Ignition magnetos eventually found their way into the majority of automobiles, both in low voltage systems which ignited spark plugs using secondary coils and high voltage systems which ignited spark plugs directly, much like induction coil ignition. When batteries became widely used in automobiles, ignition coils mainly supplanted ignition magnetos because they can produce a high-voltage spark even at low speeds, which makes starting easier.

Modern Ignition Systems

The chief drawbacks of breaker-operated ignition systems are the decreasing time available to build up the primary coil's stored energy and the lowering voltage available as engine speed increases due to limits in the breaker system's current switching capabilities. Another drawback is that the high current demand on the breaker points exposes them to both electrical and mechanical degradation, resulting in short repair intervals. The amount of current that the breaker points must switch determines how long they will last. Electronic circuits are used in current ignition systems to solve the aforementioned issues with conventional ignition systems. The breaker points and condenser can be replaced with transistors, which have the ability to interrupt circuits with reasonably high currents. Ford Motor Company released one of the first iterations in 1963, and many more are still in use today. The following two types are frequently used in modern cars. Capacitive discharge ignition system (CDI system). Transistorized coil ignition system (TCI system). The following sections go into further information about these ignition systems.

Transistorized Coil Ignition (TCI) System:

The conventional ignition systems are quickly being replaced in automotive applications by transistorized coil ignition systems, which offer a greater output voltage and use electronic triggering to maintain the necessary timing. High energy electronic ignition systems are another name for these systems. The benefits of these include reduced ignition system maintenance and decreased component wear, improved dependability, prolongation of spark plug life, enhanced lean mixture ignition. Figure 1 displays the circuit schematic of a transistorized coil ignition system. A magnetic pulse generating system that monitors the position of the distributor shaft and transmits an electrical pulse to an electronic control module takes the role of the contact breaker and cam assembly of the conventional ignition system.

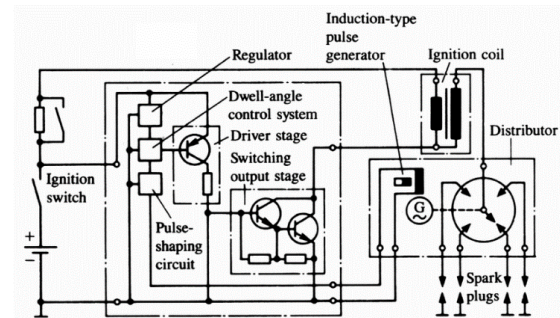


Figure 1: An induction pulse generator with transistorized coil ignition (TCI) system schematic [Research Gate].

Method of Dwell-Angle Control

1. Generator of pulses using induction.
2. Ignition coil.
3. Pulse shaping circuit.
4. Vehicle stage.
5. Changing the output stage.
6. Distributor.

Spiking Plugs

Figure 1 an induction pulse generator with transistorized coil ignition (TCI) system schematic the module interrupts the primary coil's current flow, creating a high voltage in the secondary winding that is distributed to the spark plugs like in a typical breaker system. The timing circuit in the control module closes the primary circuit later, allowing the primary circuit current to build up for the subsequent cycle. There are many different kinds of pulse generators that can start the ignition system's electronic circuit. The most common type of magnetic pulse generator is one in which a gear-shaped iron rotor powered by the distributor shaft revolves past the pole of a stationary magnetic pickup. The number of cylinders is the same as the number of teeth on the rotor. A permanent magnet creates the magnetic field. Each time a rotor tooth passes a magnet pole, the magnetic field strength initially increases and then declines, which is coupled to the pickup coil wound on the magnet and results in a voltage signal proportional to $d \cdot dt$. As the rotor tooth moves into alignment and the pickup coil voltage quickly reverses and goes through zero, the electronic module responds by turning off the primary circuit coil current to produce the spark. After this voltage reversal, the electronic module uses the growing section of the voltage waveform to determine when to turn on the primary coil current for the subsequent ignition pulse.

Working

The engine moves the armature. Primary windings cut the magnetic field's lines of force when the armature rotates, causing induced current to flow through the primary circuit. The primary circuit is abruptly opened by a contact breaker as the primary current reaches its maximum value in each direction, causing the current to collapse. This causes the secondary winding to generate a very high voltage, which jump-starts a brief spark at the spark plug gap. High tension wires are used in a distributor, which supplies the spark plug with current. Condensers are used to increase current in secondary circuits and reduce arcing at breaker points. The current must be distributed to various spark plugs in multi-cylinder engines using a distributor and rotor.

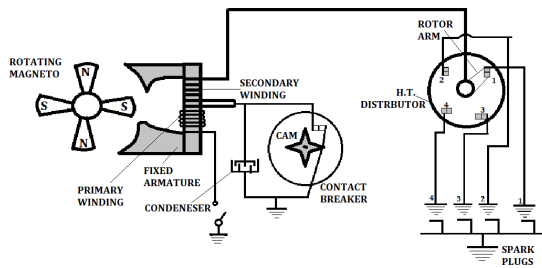


Figure 2: Representing the Magneto Ignition System [The Mechanical Engineering].

Magneto

In the Magneto Ignition System, the magneto is the energy-producing source as shown in Figure 2. A magneto is often a tiny electrically powered generator. The engine rotates the magneto, which generates the voltage. The amount of voltage that the system produces will increase with higher spin. The magneto is a source of energy generation in and of itself, hence it does not require any other power source, such as a battery, to start it. It has two different kinds of winding. Both a primary and secondary binding are present. Additionally, magneto has three different versions depending on how the engine rotates.

1. Rotational type armator.
2. Revolving type of magnet.

Type of Polar Inductor

While in the magnet rotating type, the armature is stationary and the magnets are spinning around it, the armature in the armature rotating type rotates between the stationary magnets. The magnet and windings are both stationary in the polar inductor type, but the voltage is produced by flipping the flux field with the aid of soft iron polar projections, or inductors.

Dissertator

The multi-cylinder engine uses the same distributor that is used in the magneto ignition system. It is used to control the spark in the right order in the spark plug in a multi-cylinder engine. The spark plugs receive an equal distribution of the ignition's surge. Two categories of distributors exist.

1. Distributor with carbon brushes.
2. Distributor with gaps.

In the carbon brush type, the carbon brush is placed into the distributor cap or molded insulating material and is carried by the rotor arm as it slides over the metallic segment. This facilitates the spark plug's electrical connection. The rotor arm's distributor electrode is close to the distributor cap in the gap type, but there is no contact, therefore the electrode doesn't wear out.

Spark Plug

Two electrodes in the spark plug utilised in this ignition system are separated from one another. It receives a high voltage that creates a spark that ignites the oil-based combustion mixture in the cylinder. It uses an insulator and a steel shell as the electrode. The outer steel shell, which is grounded and insulates both of them, and the supply of the ignition coil are both linked to the central electrode. The steel shell where the spark is produced has a little air gap between it and the central electrode. The core electrode, which is comprised of a high nickel alloy that can tolerate high temperatures and resistances since it is close when the spark is formed.

Capability

The capacitor utilised in the Magneto Ignition System is a straightforward electrical capacitor composed of two metal plates that are far spaced apart from one another by an insulating substance. The air is often utilised as insulation, but for a specific technical need, some high-quality insulation material is used.

The Magneto Ignition System's Operation

1. This ignition system's operation is similar to that of a coil or battery ignition system, with the exception that it generates energy using a magneto instead of a battery. Here are the scenarios that take place in it.
2. When the system's engine fires up, it aids in the magneto's rotation and generates high voltage as a result.
3. The ignition capacitor is parallel-connected to the end of the magneto that is grounded through a contact breaker.

4. When the contact breaker is open, current flows through the condenser and charges it. The contact breaker is controlled by the cam.
5. The primary current flow is decreased as the condenser is now functioning as a charger, which lowers the overall magnetic field produced in the system. The condenser's voltage rises as a result.
6. An EMF created by the condenser's enhanced high voltage will ignite the appropriate spark plug through the distributor.
7. When the engine first starts, it rotates slowly, and as a result, the voltage produced by the magneto is likewise initially low. However, as the engine's speed increases, the voltage and current flow also increase. To avoid the engine starting slowly, we can utilize an external source, such as the battery, to jump start it.

The Magneto Ignition system has the following benefits

1. Since there is no battery or connecting wire, it is more reliable. Furthermore, with the coil ignition mechanism, the engine cannot be started if the battery is dead without a spare battery being present.
2. Medium to extremely high engine speeds are better for ignition, albeit the latter case has a tendency to produce excessive voltages unless the magneto is made specifically for these high speeds.
3. A relatively small amount of area can be taken up by current Magneto designs that make use of the more recent cobalt steel and nickel aluminum magnet metals.
4. Modern magnet alloys in more contemporary magnetos can produce ignition characteristics with very low starting speeds.
5. Now, as with coil ignition, it is possible to alter the automatic timing of the ignition.
6. By using adequate shunts on the Magneto, the strong spark at high engine speeds that previously caused the plug electrodes to burn away may now be avoided.

The Magneto Ignition technology has the following drawbacks

1. Not the best sparks for starting at low speeds.
2. Both its manufacture and component part replacement are expensive.
3. When compared to the coil ignition system, the half-speed engine driver is typically more complicated.
4. The spark timing adjustment with a regular magneto impacts the voltage or energy and has

an impact across the entire ignition timing range.

5. It is difficult to maintain.

Applications for the magneto ignition system include

1. In the engine, where other electricity-using devices like lawn mowers and chainsaws are not available.
2. Racing cars and piston engines for aircraft.
3. In the engine, where other electricity-using devices like lawn mowers and chainsaws are not available.
4. Racing cars and piston engines for aircraft.

CONCLUSION

At the conclusion of the compression stroke in spark-ignition engines, a mechanism is needed to ignite the compressed air-fuel mixture. This criterion is satisfied by the ignition mechanism. It is a component of the electrical system that delivers the electric current to the spark plug at the necessary voltage so that spark is generated at the appropriate time. It is made up of the appropriate wiring, a battery, a switch, a distributor ignition coil, and spark plugs. An ignition system is not necessary for a diesel engine using compression ignition. Because diesel is delivered into compressed air at a high temperature at the end of the compression stroke, the fuel-air mixture self-ignites.

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Centrifugal and Vacuum Advance Mechanism

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ABSTRACT: *The weights and springs that make up the centrifugal advance mechanism are housed inside the distributor. The weights push outward against the tension of the springs when the engine speed rises due to the centrifugal force produced by the revolving distributor shaft. This movement causes the distributor came to rotate, which ultimately modifies the ignition timing. Based on the engine-generated suction in the intake manifold, the vacuum advance mechanism functions. It normally consists of a vacuum capsule attached to the distributor and a diaphragm. The vacuum capsule transfers this motion to the distributor shaft when the diaphragm reacts to variations in the intake manifold vacuum.*

KEYWORDS: *Advance Mechanism, Centrifugal Advance, End Portion, Inner End, Internal Combustion*

INTRODUCTION

The ignition timing for full-load operation is controlled by the centrifugal advance mechanism. The adjustment mechanism is made to operate in a way that causes the spark to advance as intended. The flyweights, which are swung farther and farther outward as the speed rises, move the cam in the direction of shaft rotation thanks to the cam's movable mounting on the distributor shaft. As a result, the ignition point is shifted in the early or advance direction because the cam lobes come into touch with the breaker lever and rubbing block a little earlier. The weights are swung outward a larger or lesser amount from the center depending on the speed of the engine and, consequently, of the shaft. Then, a holding spring that perfectly balances the centrifugal force holds them in the stretched position, in a condition of equilibrium corresponding to the shifted time angle [1], [2].

We distinguish between the rolling contact type and the sliding contact type of centrifugal advance mechanism because the weights shift the cam either on a rolling contact or sliding contact basis. The size of the weights, the shape of the contact mechanisms rolling or sliding contact type, and the retaining springs, all of which can be of very different designs, determine the start of the timing adjustment in the range of low engine speeds and the continued adjustment based on the full load curve. The centrifugal force operated cam is equipped with a lower limit stop to mark the start of the adjustment and an upper limit stop to cap the largest full-load adjustment that is permitted. Depict a typical sliding contact type centrifugal advance mechanism. Advance Mechanism for Vacuum. Under part load operation, the vacuum advance mechanism changes the ignition point. The adjustment system is made to operate in a way that produces the desired part-load advance curve [3], [4].

The static vacuum that exists in the carburetor at any given time, which pressure relies on the position of the throttle valve at any given time, and which is at a maximum when this valve is approximately half open, serves as the adjustment control quantity in this mechanism. Thus, the vacuum maximum is explained. A vacuum unit's diaphragm is movable due to variations in gas pressure. The pressure difference between the current vacuum and ambient pressure at any particular time determines where this diaphragm is located. The tension that has been predetermined on a compression spring determines when adjustment will start. According to the part-load advance curve that will be used, the diaphragm area, spring force, and spring rigidity are all chosen and balanced with regard to one another. The moveable breaker plate is coupled to a vacuum advance arm that transmits the movement of the diaphragm, which under part-load conditions causes the breaker plate to shift farther in the opposite direction of the distributor shaft's rotation. The vacuum unit's base has limit stops on the vacuum advance arm that limit the range of adjustment [5], [6].

Centrifugal Advance Mechanism Benefits

Internal combustion engines frequently have the centrifugal advance mechanism, particularly in car engines. It is essential for enhancing engine performance. The centrifugal advance mechanism has the following advantages:

Increased Fuel Efficiency: Based on engine speed, the centrifugal advance system modifies the ignition timing. The device advances the ignition timing as engine speed rises, ensuring ideal spark timing for effective combustion. Because the fuel-air mixture is ignited at the ideal time to maximize power and minimize wasted energy, fuel economy is enhanced as a result [7], [8].

Increased Power Output: The centrifugal advance mechanism enables the combustion process to begin sooner in the engine cycle by advancing the ignition time at higher engine speeds. Because of this, the engine is able to produce more power during the power stroke, increasing overall power output. Better acceleration and increased performance are two benefits [9], [10].

Reduced Engine Knock: When the air-fuel mixture in the combustion chamber detonates early or unevenly, it can make an unwelcome knocking or pinging sound. By ensuring that the spark plug ignites at the proper moment during the compression stroke, the centrifugal advance mechanism aids in preventing engine knock. It guarantees that the combustion process proceeds smoothly and reduces the possibility of knock by modifying the ignition timing dependent on engine speed. The centrifugal advance system is built to give progressive timing advance as engine speed rises, ensuring optimal timing over the RPM range. This guarantees continuous ignition timing optimization over the whole RPM (revolutions per minute) range. It contributes to maintaining smooth engine running, responsiveness, and power delivery under various driving circumstances by reacting to changing engine speeds.

Reliability and Simplicity: Centrifugal advance mechanisms are comparatively straightforward mechanical components made up of weights and springs. They are resilient and less prone to failure because they don't require electrical or electronic components. Their ease of use adds to the engine's overall dependability.

Cost-Effectiveness: The centrifugal advance mechanism is less expensive to produce and maintain than more sophisticated electronic igniting systems. It is a practical option for many engines thanks to its simple design and mechanical functioning, especially in situations where cost is an important factor. Overall, the centrifugal advance system has a number of benefits, including higher power output, decreased engine knock, enhanced fuel efficiency, ideal timing across the RPM range, simplicity, dependability, and cost-effectiveness. It is a vital component in internal combustion engines because of these advantages, which helps to maximize their performance.

DISCUSSION

After the initial timing is set manually, there are two common techniques used in modern engines to advance and retard the ignition timing automatically based on engine speed and operating conditions. Most manufacturers refer to these techniques as

automated advance systems. The centrifugal advance mechanism is one. The Hoover advance system. A centrifugal advance device. It consists of a base plate, a cam, a spring, and two fly weights Figure 1. The distributor drive shaft transports the flyweights through a base plate that is connected to the drive shaft. On the base plate, the fly weights pivot, and are connected to the cam via springs. The springs, flywheel, and base plate connect the cam to the distributor drive shaft. Due to centrifugal force turning the plate and cam anticlockwise as engine speed increases, the weights displace. The intended advance is impacted by this movement. From no advance at low speed to full advance at high speed, the timing of the spark varies. The ignition timing is automatically advanced and retarded in accordance with the engine speed and operating conditions thanks to a diaphragm in this device.

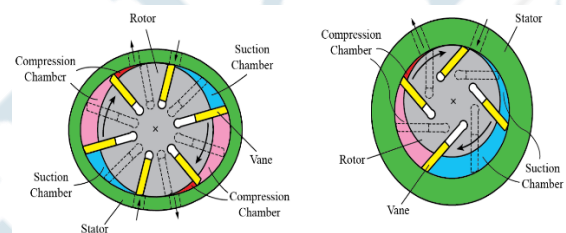


Figure 1: Sliding contact type centrifugal advance mechanism [MDPI].

The diaphragm divides into two chambers, one of which is open to the atmosphere and the other of which is connected to the induction manifold. The distributor and diaphragm are linked together via a linkage. The contact breaker is held fully retarded when the diaphragm is in its regular position. The induction manifold depression increases with increased engine speed, and atmospheric pressure pushes up the diaphragm. The diaphragm's movement causes the contact breaker to revolve in the opposite direction, advancing the ignition. As the vacuum is reduced, the diaphragm can return to its previous position and delay ignition. That the vacuum advance mechanism is used on the distributor, whereas the centrifugal advance mechanism primarily considers speed and ignores load conditions. Due to centrifugal force and vacuum created in the intake manifold, the ignition advance in this system is partially caused by both of these factors. For all driving situations, the two processes combined provide almost flawless spark timing.

A Centrifugal Advance Mechanism Comprising: A drive member and a driven member, the members being connected with one another to rotate about a common axis as they move toward and away from each other along the common axis; a resilient

element mounted to bias the drive and driven members toward each other; one of the drive and driven members having a ramp, the ramp having an inner end portion and an outer end portion, the inner end portion being closer to the common axis than the outer end portion, the ramp rising as it moves from the inner end portion toward the outer end portion and a ball in contact with the ramp and the other of the drive and driven members, the resilient element biasing the ball down the ramp, the ball traveling radially outward and up the ramp as the drive and driven members rotate about the common axis at increasing angular velocity thereby causing the drive and driven members to separate from each other against the bias of the resilient element and to rotate with respect to each other.

The centrifugal advance mechanism of claim wherein one of the drive and driven members has a second ramp, the ramp rising as it moves from the inner end portion towards the outer end portion, and further comprising a second ball in contact with the second ramp, the ramp having an inner end portion and an outer end portion, the inner end portion being closer to the axis of rotation than the outer end portion. The centrifugal advance mechanism of claim, further including a ramp coupled to a driving component. The centrifugal advance device of claim, where the ramp has a rail that directs the ball. The resilient element of the centrifugal advance mechanism of claim wherein a spring is used. The centrifugal advance mechanism of claim with a coil spring as the spring. The centrifugal advance mechanism of claim with a drive member that has teeth for a drive chain engagement. An internal combustion engine comprising a crankshaft, a camshaft and a centrifugal advance mechanism connected to one of the crankshafts and the camshaft, the centrifugal advance mechanism comprising a drive member and a driven member. The members being connected with one another to rotate about a common axis as they move toward and away from each other along the common axis a resilient element mounted to bias the drive and driven members toward each other one of the drive and driven members having a ramp, the ramp having an inner end portion and an outer end portion, the inner end portion being closer to the common axis than the outer end portion, the ramp rising as it moves from the inner end portion toward the outer end portion and a ball in contact with the ramp and the other of the drive and driven members, the resilient element biasing the ball down the ramp, the ball traveling radially outward and up the ramp as the drive and driven members rotate about the common axis at increasing angular velocity thereby

causing the drive and driven members to separate from each other against the bias of the resilient element and to rotate with respect to each other. The internal combustion engine of claim 8 where a drive chain is trained around a sprocket that is coupled to either the crankshaft or the camshaft and where the drive member has teeth that engage the chain.

The internal combustion engine of claim where in one of the drive and driven members has a second ramp, the ramp rising as it moves from the inner end portion towards the outer end portion, and further including a second bail in contact with the second ramp. The second ramp has an inner end portion and an outer end portion, the inner end portion being closer to the axis of rotation than the outer end portion. The ramp is coupled to the driving part in claim 's internal combustion engine. In accordance with claim of the internal combustion engine, the ramp includes a wall that directs the ball The resilient element in the internal combustion engine of claim is a spring, according to claim. The spring in the internal combustion engine of claim is a coil spring in clause. The drive component of the internal combustion engine described in claim has teeth that can engage a drive chain. A motor vehicle comprising an internal combustion engine, the internal combustion engine comprising a crankshaft, a camshaft and a centrifugal advance mechanism connected to one of the crankshaft and the camshaft, the centrifugal advance mechanism comprising a drive member and a driven member, the members being connected with one another to rotate about a common axis as they move toward and away from each other along the common axis a resilient element mounted to bias the drive and driven members toward each other one of the drive and driven members having a ramp.

The ramp having an inner end portion and an outer end portion, the inner end portion being closer to the common axis than the outer end portion, the ramp rising as it moves from the inner end portion toward the outer end portion; and a bail in contact with the ramp and the other of the drive and driven members, the resilient element biasing the ball down the ramp, the ball traveling radially outward and up the ramp as the drive and driven members rotate about the common axis at increasing angular velocity thereby causing the drive and driven members to separate from each other against the bias of the resilient element and to rotate with respect to each other. The motor vehicle of claim wherein one of the drive and driven members has a second ramp, the ramp rising as it moves from the inner end portion towards the outer end portion, and further including a second ball in contact with the second ramp, the ramp

having an inner end portion and an outer end portion, the inner end portion being closer to the axis of rotation than the outer end portion.

The ramp is attached to the driving part in accordance with claim for the automobile. The vehicle described in claim, where the ramp includes a wall that directs the bail. The car of claim when the resilient part is a spring. The car of claim where the drive member has teeth for securing a drive chain. The motor vehicle of claim, wherein the vehicle is a motorbike, is described in clause. The motor vehicle of claim, further including a V-twin engine. The invention relates to centrifugal advance mechanisms in general, and more specifically, to centrifugal advance mechanisms that change the angular relationship between two rotating members as a function of rotational speed. An internal combustion engine's ability to run depends on the precise timing of a number of actions, including the firing of one or more spark plugs and the opening and closing of a number of valves, including the intake and exhaust valves. The speed of the engine influences when these events should occur during its operational cycle. The performance of an engine can be impacted by timing, and in some prior art engines, devices have been given to change the angular relationship between parts rotating at different speeds.

A centrifugal advance mechanism is provided, which consists of a driven member and a drive member coupled to one another to rotate about a common axis as they move towards and away from one another along the common axis. To slant the members towards one another, a robust element is installed. The inner end portion of the ramp on one of the drive and driven members is closer to the common axis than the outer end portion, and the ramp rises as it advances from the inner end portion to the outer end portion. The ramp and one of the drive and driven parts are in touch with a ball. The ball is biased down the ramp by the robust element. The ball moves radially outward and up the ramp as the driving and driven components spin about the common axis with increasing angular velocity. The drive and driven members separate from one another and rotate with regard to one another when the ball is moved up the ramp, going against the resilient element's bias.

There is a crankshaft, a camshaft, and a centrifugal advance mechanism for an internal combustion engine. One of the crankshaft and camshafts is coupled to the centrifugal advance mechanism. The drive and driven members of the centrifugal advance mechanism are coupled to rotate around a common axis as they move towards and away from one

another along the common axis. One of the drive member and driven member has a ramp, and a resilient element is installed to bias the drive member and driven member towards one another. The ramp has two end portions, one on either side of the common axis, with the inner end section being closer to the common axis. Moving from the inner end piece to the outer end portion, the ramp rises. The resilient element pushes the ball down the ramp when it is in contact with the ramp and one of the drive and driven members. The drive and driven members separate from one another against the bias of the resilient element and rotate with regard to one another as the ball moves radially outward and up the ramp while rotating about the common axis at increasing angular velocity.

An internal combustion engine is featured in a motor vehicle. The engine has a centrifugal advance mechanism, a camshaft, and a crankshaft. One of the crankshaft and camshafts is coupled to the centrifugal advance mechanism. The drive member and the driven member of the centrifugal advance mechanism are coupled to rotate around a common axis as they move towards and away from one another along the common axis. One of the drive and driven members has a ramp, and a resilient element is installed to bias the drive and driven members towards one another. The ramp has two end portions, one on either side of the common axis, with the inner end section being closer to the common axis. Moving from the inner end piece to the outer end portion, the ramp rises. The ramp and one of the drive and driven parts are in contact with a bail, and the resilient element pushes the ball down the ramp. The drive and driven members separate from one another against the bias of the resilient element and rotate with regard to one another as the ball moves radially outward and up the ramp while rotating about the common axis at increasing angular velocity.

Applications

Numerous internal combustion engines, especially those with distributor-based ignition systems, use the centrifugal advance mechanism. Following are a few typical uses for the centrifugal advance mechanism:

Automotive Engines: Gasoline-powered cars, trucks, motorcycles, and other vehicles all use centrifugal advance mechanisms, which are commonly found in automotive engines. It is frequently included in the distributor, which controls ignition timing according to engine speed. It provides optimum combustion and engine performance by changing the timing advance as the

engine rpm rises. Lawnmowers, chainsaws, generators, and other power equipment all use tiny engines, which also contain the centrifugal advance mechanism. For varying engine speeds, these engines frequently use a straightforward distributor-based ignition system with a centrifugal advance mechanism. Boats, jet skis, and snowmobiles are just a few examples of recreational vehicles that use internal combustion engines and distributor-based ignition systems. By regulating the ignition timing in accordance with engine speed, the centrifugal advance system aids these engines in supplying the necessary power and performance for leisure activities.

Agricultural Machinery: Internal combustion engines with centrifugal advance systems are frequently used in agricultural machinery such as tractors, combines, and harvesters. These engines must deliver effective power production over a broad speed range, and the centrifugal advance system aids in adjusting the ignition timing for various operating circumstances.

Industrial Equipment: A centrifugal advance mechanism is a feature of internal combustion engines that are used in a number of industrial applications, including pumps, compressors, and power generators. The mechanism makes sure that the ignition timing is correct at various engine speeds, which contributes to the general functionality of these engines, which demand dependable and efficient operation.

Advance Mechanism for Vacuum

Under part load operation, the vacuum advance mechanism changes the ignition point. The adjustment system is made to operate in a way that produces the desired part-load advance curve. The static vacuum that exists in the carburetor at any given time, which pressure relies on the position of the throttle valve at any given time, and which is at a maximum when this valve is approximately half open, serves as the adjustment control quantity in this mechanism. Thus, the vacuum maximum is explained. A vacuum unit's diaphragm is movable due to variations in gas pressure. The pressure difference between the current vacuum and ambient pressure at any particular time determines where this diaphragm is located. The tension that has been predetermined on a compression spring determines when adjustment will start. According to the part-load advance curve that will be used, the diaphragm area, spring force, and spring rigidity are all chosen and balanced with regard to one another. The moveable breaker plate is coupled to a vacuum advance arm that transmits the movement of the

diaphragm, which under part-load conditions causes the breaker plate to shift farther in the opposite direction of the distributor shaft's rotation. The vacuum unit's base has limit stops on the vacuum advance arm that limit the range of adjustment.

CONCLUSION

A centrifugal advance mechanism is provided, which consists of a driven member and a drive member coupled to one another to rotate about a common axis as they move towards and away from one another along the common axis. To slant the members towards one another, a robust element is installed. The inner end portion of the ramp on one of the drive and driven members is closer to the common axis than the outer end portion, and the ramp rises as it advances from the inner end portion to the outer end portion. The ramp and one of the drive and driven parts are in touch with a ball. The ball is biased down the ramp by the robust element. The ball moves radially outward and up the ramp as the driving and driven components spin about the common axis with increasing angular velocity.

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A Brief Overview about Homogeneous Mixture

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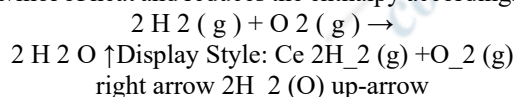
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ABSTRACT: A chemical reaction known as combustion occurs when certain fuel components, such as hydrogen and carbon, mix with oxygen to produce heat energy and raise the temperature of the gases. The presence of a combustible mixture and a way to start the process are prerequisites for combustion. The theory of combustion is a tremendously complicated subject that has long been the focus of extensive study. Despite this, there is little information on the phenomenon of combustion.

KEYWORDS: Combustion Chamber, Chemical Reaction, Flame Front, Flame Speed, Pressure Rise

INTRODUCTION

A fuel the reluctant and an oxidant, typically ambient oxygen, undergo a high-temperature exothermic redox chemical process that results in oxidized, frequently gaseous products and a mixture known as smoke [1], [2]. Since a flame only appears when components undergoing combustion vaporize, combustion does not always result in fire, but when it does, a flame is a distinctive sign of the reaction. The heat from a flame may be enough energy to make the reaction self-sustaining, even though the activation energy must be overcome to initiate combustion such as when using a lit match to start a fire [3], [4]. A convoluted series of simple radical reactions frequently occurs during combustion. Endothermic pyrolysis is the process that turns solid fuels like wood and coal into gaseous fuels that can be burned to generate the heat needed to create new solid fuels. A flame or a blazing incandescent light is frequently produced when combustion is hot enough. A straightforward illustration is provided by the reaction that occurs when hydrogen and oxygen combine to form water vapor, which is frequently utilized as a fuel for rocket engines. At constant temperature and pressure, this reaction produces 242 kJ/mol of heat and reduces the enthalpy accordingly:



Relatively high temperatures are required for catalyzed combustion in air. Complete combustion, where there is no residual fuel and, preferably, no residual oxidant, is stoichiometric with respect to the fuel. The chemical equilibrium of burning in air is predominantly on the side of the products from a thermodynamic perspective. Since the chemical equilibrium may not have been reached or may still

contain unburned substances like carbon monoxide, hydrogen, and even carbon, full combustion is almost impossible to achieve. As a result, the smoke that is created frequently contains unburned or only partially oxidized materials.

Since the combustion of nitrogen is thermodynamically favored at high temperatures but not at low temperatures, any combustion at high temperatures in atmospheric air, which is composed of 78 percent nitrogen, would also produce minor amounts of various nitrogen oxides, also known as NO_x. Burning rarely produces clean gas, thus catalytic converters or fuel gas cleaning may be mandated by law [5], [6]. Naturally occurring fires are started by lightning or volcanic materials. In the form of campfires and bonfires, combustion was the first regulated chemical reaction that mankind discovered. It is still the primary way that humanity produces energy. Carbon, hydrocarbons, or more complex mixes like wood that contain partially oxidized hydrocarbons are the most common types of fuel. Cooking, producing electricity, heating homes or businesses, and other purposes all make use of the thermal energy that results from the combustion of either fossil fuels like coal or oil, or renewable fuels like firewood. Additionally, the only reaction currently used to power rockets is combustion. The destruction of waste, both harmful and nonhazardous, is another purpose for combustion [7], [8].

DISCUSSION

A chemical reaction known as combustion occurs when certain fuel components, such as hydrogen and carbon, mix with oxygen to produce heat energy and raise the temperature of the gases. The presence of a combustible mixture and a way to start the process are prerequisites for combustion. The theory of

combustion is a tremendously complicated subject that has long been the focus of extensive study. Despite this, there is little information accessible about the phenomenon of combustion. Depending on the kind of engine, the process of combustion typically occurs in either a homogeneous or a heterogeneous fuel vapor-air combination [9], [10].

Homogeneous Mixture

A virtually homogenous combination of fuel and air is created in the carburetor of spark-ignition engines. Thus, a homogeneous mixture is created outside the engine cylinder, and at a specific moment near the conclusion of the compression stroke, combustion begins within the cylinder. With a specific velocity, the flame front spreads throughout a flammable mixture. Fuel and oxygen molecules are more or less evenly distributed in a homogenous gas mixture. When the mixture of fuel vapor and air ignites, a flame front develops and quickly spreads throughout it. Heat transmission and the diffusion of burning fuel molecules from the combustion zone to the surrounding layers of unburned mixture are what cause the flame to spread. The new mixture and the combustion byproducts are separated by a small area known as the flame front. The normal flame velocity is the speed at which the flame front advances towards the unburned mixture in a direction perpendicular to its surface.

The flame speed is typically in the range of 40 cm/s in a homogenous mixture with an equivalency ratio, the ratio of the actual fuel-air ratio to the stoichiometric fuel-air ratio close to 1.0. The maximum flame speed in a spark-ignition engine is achieved when is between 1.1 and 1.2, or when the mixture is only a little richer than stoichiometric. The flame speed rapidly decreases to a low value if the equivalency ratio is outside of this range. The flame extinguishes when the flame speed reaches a very low value, at which point the heat loss from the combustion zone equals the heat released during burning. For proper combustion, it is therefore much preferable to run the engine at an equivalency ratio of 1.1 to 1.2. In mixtures outside the aforementioned range, the flame speed can be increased by adding turbulence and appropriate air movement.

Heterogeneous Mixture

The velocity of mutual diffusion between fuel vapors and air determines the pace of combustion in a heterogeneous gas mixture, while the rate of chemical reaction plays a small role. The main factor affecting the combustion characteristics is the self-ignition or spontaneous ignition of the fuel-air combination at the high temperature produced by greater compression ratios. Since there are always

local zones where fluctuates between 1.0 and 1.2, which corresponds to the maximal rate of chemical reaction, combustion in heterogeneous mixtures can occur in an overall lean mixture. In this area, ignition begins, and the resulting flame aids in burning the fuel in the nearby areas where the mixture is leaner. Similar to this, where the mixture is rich, combustion takes place due to the high temperature created by combustion that began where is between 1.0 and 1.2. The next sections provide a thorough analysis of combustion in both spark-ignition and compression-ignition engines.

Spark-Ignition Engine Combustion

In a normal spark-ignition engine, as was already said, the fuel and air are uniformly mixed in the intake system before being introduced through the intake valve into the cylinder, where they combine with residual gases and are then compressed. Under typical operating circumstances, an electric discharge at the spark plug starts combustion towards the end of the compression stroke. Following ignition, a turbulent flame forms and spreads through this premixed charge of fuel and air as well as the leftover gas in the clearance volume until it reaches the walls of the combustion chamber. The SI engine's combustion can be broadly classified into two categories: normal combustion and aberrant combustion.

Si Engines' Stages of Combustion

Depicts a typical theoretical pressure-crank angle diagram for the compression, combustion, and expansion phases of an ideal four-stroke spark-ignition engine. The diagram shows that in an ideal engine, the entire pressure rise during combustion occurs at constant volume, or at T DC. This does not occur in a real engine, though. Below is a detailed description of the combustion process in a real SI engine. Depicts a typical theoretical pressure-crank angle diagram for the compression, combustion, and expansion, phases of an ideal four-stroke spark-ignition engine. The diagram shows that in an ideal engine, the entire pressure rise during combustion occurs at constant volume, or at T DC. This does not occur in a real engine, though. Below is a detailed description of the combustion process in a real SI engine. The combustion process in a SI engine, according to Sir Ricardo, who is regarded as the founder of engine development, has three stages: depicts the pressure variation brought on by combustion in a real engine. In this diagram, A represents the spark passing point, B represents the starting point of the pressure rise, and C represents the achievement of peak pressure. As a result, AB

stands for the first stage, BC for the second stage, and CD for the third stage.

The first stage (A–B) is known as the ignition lag or preparation phase, during which a self-propagating flame nucleus grows and develops. This is a chemical process that is influenced by temperature, pressure, fuel type, and the amount of exhaust residual gas. Additionally, it depends on how the temperature and response rate are related. The physical second stage (B–C) is concerned with how the flame spreads across the combustion chamber. The indicator diagram's first measurable rise in pressure, or the point where the line of combustion diverges from the compression line (point B), marks the beginning of the second stage. The departure from the motoring curve demonstrates this. The flame spreads nearly at a steady speed during the second stage. Because just a small portion of the burning mixture makes contact with the cylinder wall during this time, there is little heat transmission to the wall.

The amount of turbulence present and the reaction rate, which is influenced by the makeup of the mixture, both have a significant impact on the rate of heat release. Since the piston is close to the top dead center at this point and the combustion chamber capacity is essentially constant, the rate of pressure rise and the rate of heat release are proportionate. The moment the maximum pressure on the indicator diagram is reached (point C), is typically used as the third stage's beginning point. In this phase, the flame velocity reduces. Low flame velocity and reduced flame front surface cause the rate of combustion to decrease. There can be no pressure rise at this stage of combustion since the expansion stroke begins before it, with the piston moving away from the top dead center.

Propagation of Flame Front

The flame front's pace of propagation inside the cylinder is very important for effective combustion. The response rate and the transposition rate are the two key variables that affect how quickly the flame front moves across the combustion chamber. The process of pure chemical combination in which the flame consumes the unburned charge yields the reaction rate. The pressure difference between the burning gases and the unburned gases in the combustion chamber and the physical movement of the flame front in relation to the cylinder wall both contribute to the transposition rate. The rate of flame propagation. Area I, (A–B), has a low transposition rate and little turbulence, which causes the flame front to advance rather slowly. Due to the relatively tiny initial charge burnt mass, there is virtually little

flame front transposition. The slow advance of the flame is mostly caused by the low reaction rate.

Additionally, the lack of turbulence lowers the reaction rate and, consequently, the flame speed since the spark plug must be placed in a gas layer that is quiescent and close to the cylinder wall. The flame front advances faster and at a constant rate (BC) as it moves out of the quiescent zone and into more turbulent regions (area II), where it consumes a larger amount of mixture. The amount of unburned charge is significantly less at the conclusion of flame travel, and as a result, the transposition rate again declines, slowing the flame speed. Since the flame is entering a region (area III) with comparatively moderate turbulence (C–D), the reaction rate is likewise decreased once more. An internal combustion engine has a combustion chamber where the fuel/air mixture is burnt. The name has also been applied to an addition to the firebox used in steam engines to provide a more thorough combustion process. In an internal combustion engine, the pressure produced by the combustion of the air and fuel applies direct force to a specific area of the engine for a piston engine, the force is applied to the top of the piston, converting the gas pressure into mechanical energy often in the form of a rotating output shaft. An external combustion engine, in contrast, has the combustion occur in a different area of the engine from where the gas pressure is transformed into mechanical energy.

Spark-Ignited Motors

The combustion chamber is often found in the cylinder head of spark ignition engines, such as petrol engines. The bottom of the combustion chamber is frequently positioned such that it roughly aligns with the top of the engine block in engines.

Side-Valve Motor

Above the piston and valve of a flathead engine is the combustion chamber. Between the piston and the valves is the combustion chamber of an OHC engine. The top of the piston, when it is close to top dead center, serves as the bottom of the combustion chamber in modern engines with overhead valves or overhead camshaft. The spark plug, exhaust valves, and sidewalls of the combustion chamber's roof are located above this. With no side protrusions i.e., the entire combustion chamber is positioned immediately above the piston, this creates a comparatively small combustion chamber. Common combustion chamber forms include those that resemble one or more half-spheres, such as the kidney, hemi, pent-roof, or wedge shapes. The combustion chamber of the earlier flathead engine has an elongated shape that lies above the piston and

the valves, which are situated next to the piston. This shape is referred to as the bathtub combustion chamber. The intake valve is positioned above the combustion chamber while the exhaust valve is positioned below it in IOE engines, which combine elements of overhead valve and flathead engines.

To achieve efficient combustion and maximize power output, consideration must be given to the geometry of the combustion chamber, intake ports, and exhaust ports. The "swirl" pattern a rotating component to the gas flow and turbulence that are commonly achieved by cylinder heads improve mixing and boost gas flow rate. The quantity of swirl is also influenced by the piston top's shape. Squish, in which the fuel mixture is squished under high pressure by the rising piston, is another design element to encourage turbulence for good fuel/air mixing. Since the flame front the leading edge of the burning gases originates at the spark plug and moves downward towards the piston from there, the location of the spark plug is also a significant consideration. A well-designed engine should avoid having any small cracks where stagnant end gas could become trapped, limiting engine power and perhaps causing engine banging. Most engines only use one spark plug per cylinder, however some like the Alfa Romeo Twin Spark engine from 1986 to 2009 use two. One of the key components of gas turbine plants, air jet engines and rocket engines is the combustion chamber, which uses the calorific power of the burned fuel H_u to heat the original components from a starting temperature T_0 to a predetermined T_g temperature.

The equation for an air jet engine's heat delivery to one kilogram me of air in a typical combustion chamber at constant pressure, accounting for combustion efficiency and heat losses through the walls. where L_0 is the working medium to fuel flow rate and depends on the oxidizing medium, such as air, where C_{po} and C_{ps} are the specific heat capacities of the initial working medium and the combustion products, respectively. Theoretically, L_0 of oxidizing medium is required to completely burn 1 kilogram me of fuel. The surplus coefficient is the multiplier used to account for excess air when calculating the stoichiometric air requirement. $L_0 = 0.115C + 0.345H + 0.043O$, where C, H, and O are the mass fractions of carbon, hydrogen, and oxygen in the fuel, is the formula needed to burn hydrocarbon (petroleum) fuel in air. In the case of aviation kerosene (84–86% C, 14–16% H), $L_0 = 14.9$. L_0 is 17.2 and 34.5 for CH_4 and H_2 , respectively.

The amount of heat in Joules released as a result of the full combustion of 1 kilogramme of fuel in air at $t_0 = 15^\circ C$ and $p = 0.1$ MPa during cooling of

combustion products to $15^\circ C$ is known as calorific power, or the lowest heat of fuel combustion. This ignores the heat of condensation and the amount of water vapour present. In general, it is estimated by: For instance, $H_u = 49,500$ and $116,700$ kJ/kg for CH_4 and H_2 , respectively, and $42,900$ to $43,100$ kJ/kg for aircraft paraffin. Water, hydrocarbons C_xH_y , CO_2 and CO , NO and NO_2 , and other gases are produced during the combustion of hydrocarbon fuels. From an environmental perspective, the combustion chamber is impacted by their composition. A decrease in combustion efficiency, η , increases the amount of CO , C_xH_y , and causes soot and smoke to be produced. As combustion temperature rises and the amount of time combustion products spend in the combustion zone lengthens, nitrogen oxides (NO_x) are ejected from the atmosphere at a higher rate. State regulation therefore governs the permissible levels of NO_x , CO , C_xH_y , and smoke for the majority of engine types. The combustion regime is represented by the oxidizer (air) excess coefficient, which is $= G_a/G_f$. Stoichiometric conditions are reached when $= 1$, rich conditions are reached when > 1 , and lean conditions are reached when < 1 . The heat consumption of the surplus fuel and oxidizer causes the temperature of the combustion products, T_g , to be lower than the level closest to stoichiometric whether there is an excess or deficiency of oxidizer. The steady-state combustion in the chamber comes to an end with a major change in. Both of these are referred to as rich and poor flame-outs. Power plant combustion chambers should have the following characteristics high combustion efficiency, low pressure losses of the working medium flow across the chamber ($=$ pout/pin in gas-turbine engines, is 0.94–0.96), high reliability, and longer service life (in gas-turbine engines, up to 10,000 hours). The absence of overheating, carbon deposits, etc. can guarantee these.

The value of preheating T_g/T_0 and the variation of η and coefficients with the air flow rate (or $M_{comb.ch}$) are believed to be the features of the combustion chamber. Decreases as $M_{comb.ch}$ and T_g/T_0 expand. When plotted against $M_{comb.ch}$, combustion efficiency improves with increasing T_g/T_0 and reaches a flat maximum. High uniformity of the fields of circumferential gas temperatures at the combustion chamber outlet for a nozzle device's reliability and of the temperature versus radius profile for blade reliability, with temperature diminishing towards the upper and lower ends of the blade, are particularly significant in gas-turbine engines. The growth of oxidizer and fuel flows in the combustion and mixing zones results in the fields.

The range of flammability for homogenous hydrocarbon fuel-air mixes is 0.5 to 1.7.

Flame front propagation speeds range from 0.5 to 2.0 m/s for kerosene and 210 m/s for hydrogen, which are not very fast. Therefore, a combustion stabilizer with a reverse current zone can be designed to ensure reliable mixture inflammation in the combustion zone for all operating regimes of the combustion chamber. This will ensure stable combustion at mean flow velocities much higher than the velocity of the flame front propagation. The structure of such a flow in the combustion zone behind the combustion stabilizer. In the mixing zone, adding air to combustion products lowers average temperature values and increases values. In the combustion chamber of an air jet engine, for instance, the characteristic values of for a poor flame-out typically range from 20 to 50. In rocket engines, the system of vortices around the oxidizer and fuel jets typically has an impact on the stabilization of the flame front.

Propagation of the Flame Front

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The slow advance of the flame is mostly caused by the low reaction rate. Additionally, the lack of turbulence lowers the reaction rate and, consequently, the flame speed since the spark plug must be placed in a gas layer that is quiescent and close to the cylinder wall. The flame front advances faster and at a constant rate (BC) as it moves out of the quiescent zone and into more turbulent regions, where it consumes a larger amount of mixture, as illustrated. The amount of unburned charge is significantly less at the conclusion of flame travel, and as a result, the transposition rate again declines, slowing the flame speed. Since the flame is entering a with comparatively moderate turbulence (C–D), the reaction rate is likewise decreased once more.

Factors Impacting Flame Rate

Since the flame velocity affects the rate of pressure rise in the cylinder and is connected to several types of anomalous combustion that happen in spark-ignition engines, the study of factors that impact the flame velocity is crucial. The fuel-air ratio and turbulence are the two most significant parameters that, to varying degrees, influence flame speed. Below, several influences on flame speed are examined in further detail. In mixtures with little turbulence, the flame speed is relatively modest, and it rises as turbulence increases. This is mostly because the burning and unburned particles are physically mixed together more at the flame front, speeding up reaction by increasing the rate of contact. The fuel-air mixture is admitted into the incoming mixture during the suction stroke through relatively tiny sections of the intake pipe, valve openings, etc., creating turbulence in the process.

The rate of response and flame speed appear to be higher in turbulence that is made up of numerous little swirls as opposed to fewer, larger swirls. The turbulence during the compression stroke is increased by an effective combustion chamber design that takes into account the geometry of the cylinder head and piston crown. Turbulence typically makes it easier for heat to transfer to the cylinder wall. Intimate mixing of the fuel and oxygen speeds up the chemical process as well, reducing the need for a spark advance. This facilitates combustion of lean mixtures as well. Turbulence-induced increases in flame speed shorten the period of combustion, which lessens the likelihood of aberrant combustion. However, extreme turbulence may put out the flame, causing the engine to run rough and loudly. The fuel-air ratio significantly affects how quickly a flame burns. The influence of mixture strength on the rate of burning as shown by the time required for complete combustion in a given engine where the highest flame velocities lowest time for complete combustion are obtained with somewhat richer mixtures (point A). The flame speed is reduced when the mixture is made richer or leaner. In the case of lean mixtures, less thermal energy is emitted, resulting in a lower flame temperature.

Incomplete combustion is caused by extremely rich mixes, which again results in the release of less heat energy. As input temperature and pressure rise, flame speed rises as well. A better homogeneous air-vapor combination may be formed with a higher initial pressure and temperature, which aids in accelerating the flame. A general rise in the charge's density makes this possible. A higher compression ratio raises the working mixture's pressure and

temperature, which shortens the early phase of combustion preparation and necessitates less ignition advance. The compressed mixture's high pressures and temperatures hasten the second stage of combustion as well. As a result of the clearance volume being reduced due to increased compression ratio, the cylinder gases' density during burning is increased. As a result, the peak pressure and temperature rise, and the overall combustion time shortens. Consequently, engines with higher compression ratios have faster flames. As the engine power rises, the cycle pressure rises as well. The cylinder fills to a higher density with the increased throttle opening. This causes the flame to burn more quickly. Throttling reduces production by increasing the working mixture's dilution while decreasing the beginning and final compression pressures. It becomes unstable and challenging for the self-propagating flame nucleus to evolve smoothly. The main drawbacks of SI engines are their poor combustion at low loads and the requirement for mixture enrichment (1.2 to 1.3), which wastes fuel and releases unburned hydrocarbons and incomplete combustion byproducts like carbon monoxide and other noxious gases into the atmosphere. As the engine speed increases, the turbulence inside the cylinder intensifies, increasing the flame speed practically linearly. If the engine speed is doubled, the time for the flame to go through the combustion region would be cut in half. The same number of crank degrees would be required for flame propagation at twice the engine speed and, thus, at half the initial time. At all speeds, the crank angle needed for flame propagation during the entire combustion phase will be almost constant. The pace of flame propagation is not greatly influenced by the size of the engine. Given that the flame must travel a greater distance in large engines, the time needed for full combustion is longer. This necessitates a longer crank angle during combustion. One of the reasons huge engines are built to run at low speeds is because of this.

CONCLUSION

One of the key components of air jet and rocket engines are combustion chambers. A chemical reaction known as combustion occurs when certain fuel components, such as hydrogen and carbon, mix with oxygen to produce heat energy and raise the temperature of the gases. The presence of a combustible mixture and a way to start the process are prerequisites for combustion. The theory of combustion is a tremendously complicated subject that has long been the focus of extensive study. Despite this, there is little information accessible

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